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EXPERIMENTAL VERIFICATION AND REVISION OF THE VENTING RATE MODE--ETC(U)

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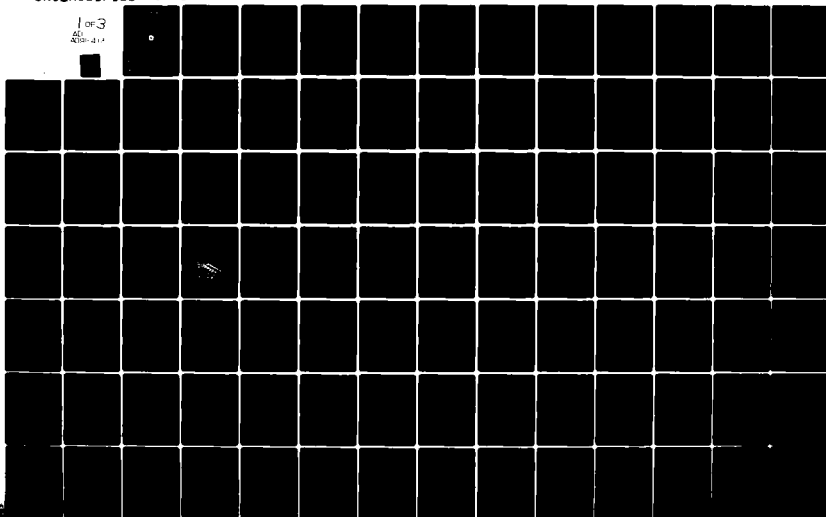
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EXPERIMENTAL VERIFICATION AND REVISION  
OF THE VENTING RATE MODEL OF THE  
HAZARD ASSESSMENT COMPUTER SYSTEM  
AND THE VULNERABILITY MODEL

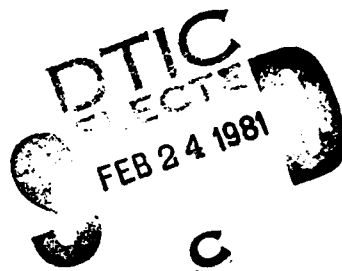
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16. Abstract A model developed previously as part of the Hazard Assessment Computer System to predict the discharge rate of a chemical cargo from a punctured ship tank has been revised and verified. The computerized model can now treat: (1) a puncture above, below, or intersecting the waterline; (2) vacuum relief valve actuations; (3) simultaneous liquid and gas discharge; (4) volatile or nonvolatile cargo; and (5) air or water ingestion into the tank, to permit additional cargo to be discharged. The model is based on an analysis of such thermodynamic and fluid flow phenomena as two-phase and choked flow and realistic evaporation rates within the tank. The Chemical Properties File was examined for its applicability to the revised model, and some desirable improvements were indicated. Experimental verification of the model consisted of a series of extensively instrumented tests of small tanks to investigate: (1) discharge coefficients of irregularly shaped punctures; (2) air and water ingestion; and (3) discharge of volatile and boiling liquids. The good comparison of the model to the tests adequately verified the model.		
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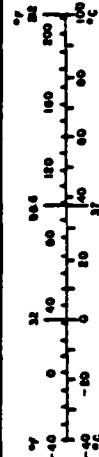
# METRIC CONVERSION FACTORS

## Approximate Conversions to Metric Measures

Symbol	When You Know	Multiply by	To Find	Symbol
<b>LENGTH</b>				
in	inches	2.5	centimeters	cm
ft	feet	30	centimeters	cm
y	yards	0.9	meters	m
mi	miles	1.6	kilometers	km
<b>AREA</b>				
sq in	square inches	6.5	square centimeters	cm <sup>2</sup>
sq ft	square feet	0.09	square meters	m <sup>2</sup>
sq yd	square yards	0.8	square meters	m <sup>2</sup>
sq mi	square miles	2.6	square kilometers	km <sup>2</sup>
ac	acres	0.4	hectares	ha
<b>MASS (weight)</b>				
oz	ounces	28	grams	g
lb	pounds	0.45	kilograms	kg
sh	short tons (2000 lb)	0.9	tons	t
<b>VOLUME</b>				
cc	centimeters	0	milliliters	ml
fl oz	fluid ounces	30	milliliters	ml
cu in	cubic inches	0.24	liters	l
qt	quarts	0.97	liters	l
gal	gallons	3.8	liters	l
cu ft	cubic feet	0.03	cubic meters	m <sup>3</sup>
cu yd	cubic yards	0.76	cubic meters	m <sup>3</sup>
<b>TEMPERATURE (Celsius)</b>				
°F	Fahrenheit temperature	5/9 (after subtracting 32)	Celsius temperature	°C

## Approximate Conversions from Metric Measures

Symbol	When You Know	Multiply by	To Find	Symbol
<b>LENGTH</b>				
cm	centimeters	0.4	inches	in
m	meters	3.3	yards	y
km	kilometers	0.6	miles	mi
<b>AREA</b>				
cm <sup>2</sup>	square centimeters	0.16	square inches	sq in
m <sup>2</sup>	square meters	1.2	square yards	sq yd
ha	hectares (10,000 m <sup>2</sup> )	0.4	square miles	sq mi
km <sup>2</sup>	square kilometers	2.5	acres	ac
<b>MASS (weight)</b>				
g	grams	0.035	ounces	oz
kg	kilograms	2.2	pounds	lb
t	tons (1000 kg)	1.1	short tons	sh
<b>VOLUME</b>				
ml	milliliters	0.03	fluid ounces	fl oz
l	liters	2.1	quarts	qt
l	liters	1.06	gallons	gal
m <sup>3</sup>	cubic meters	0.26	cubic feet	cu ft
m <sup>3</sup>	cubic meters	1.3	cubic yards	cu yd
<b>TEMPERATURE (Celsius)</b>				
°C	Celsius temperature	9/5 (plus add 32)	Fahrenheit temperature	°F



\* 1 in = 2.54 precisely. For other exact conversions and more detailed tables, see NBS Spec. Publ. 280, Units of Weight and Measure, Price \$7.50, SO Catalog No. C12.10-280.

## SUMMARY

This final report covers all four tasks of a project to revise and verify experimentally the Venting Rate Model of the Hazards Assessment Computer System. The report documents (1) all experimental procedures and representative test data, and (2) the analysis, development, and verification of the final form of the model.

Background. An analytical/computer model had been previously developed for the U. S. Coast Guard for use in predicting the discharge of a liquid or gas cargo from a punctured ship tank. A later independent review found that the model contained a number of errors and inappropriate assumptions. The review generally concluded that the model should be revised and verified experimentally. The present program was therefore sponsored by the Coast Guard to accomplish the recommended revision and verification.

### Tasks 1 and 2 - Review and Reformulation of the Model.

During the first two tasks of this program, the Venting Rate Model and other relevant literature were reviewed, and the model was then completely reformulated. The new model is applicable to nearly every practical combination of puncture location, cargo properties, intermittent outflows, and relief valve actuations. In summary, the new model incorporates:

- o puncture locations above or below the waterline;
- o vacuum relief valve actuations;
- o possibility of air in the cargo vapor space;
- o changes of phase during the discharge of a volatile cargo;
- o choked flow discharges of gases and volatile liquids;
- o simultaneous liquid and gas discharges;
- o air or water ingestion through the puncture; and
- o effects of realistic evaporation rates of volatile liquids within the tank.

Temperature stratification of a volatile liquid, one of the phenomena suggested by the review as perhaps being important enough to be included in the model, was investigated experimentally and found to be negligible. Thus, stratification is not included in the new model.

The model has been programmed for computerized solution. Program listings and flow charts are included in this report.

The Chemical Properties File was also reviewed for its applicability to the revised Venting Rate Model. Data correlations for cargo properties of nonvolatile cargos were found to be satisfactory. However, several serious limitations were discovered for volatile cargos; namely: (1) the compressibility and the specific heat of the saturated vapor should be included; (2) the vapor pressure vs temperature correlations are moderately inaccurate; and (3) a correlation giving the latent heat of vaporation as a function of temperature should be included. More accurate and complete correlations for six representative cargos were also developed.

Task 3 - Experimental Program. The third task of the program was an extensive series of model-scale experiments. The overall objectives of these tests were:

- o obtain data to evaluate the assumptions used in developing the analytical models;
- o gain insight into air and water ingestion; and
- o acquire discharge rate data for a variety of test conditions to validate the overall Venting Rate Model.

The first test series determined the discharge coefficients for a variety of realistically shaped punctures over a range of Reynolds numbers. These tests showed that a discharge coefficient of 0.65 is adequate for nearly all puncture shapes; the exceptions are vertical or horizontal slot-like punctures with ragged edges pointing outwards, for which the recommended discharge coefficient is 0.825.



The second test series was designed to study the ingestion of air or water (for submerged punctures) through the puncture. It was found that air or water ingestion can result in the discharge of more cargo than would be predicted by the previous Venting Rate Model when the ambient pressure exceeds the tank pressure at the puncture location. The tests were used to obtain physical insight about air and water ingestion, from which relatively simple but realistic analytical models were developed.

The third test series studied the discharge of volatile cargos. The specific objectives of these tests were: (1) evaluate the importance of liquid stratification; (2) gain insight and obtain quantitative data about evaporation within the tank; (3) evaluate the effects of phase changes of the discharge (two-phase outflow); and (4) obtain discharge data (e.g., time durations, temperatures, tank pressure, mass rate of outflow) which could be used for the overall verification of the revised analytical/computer model.

As indicated above, these tests showed that liquid stratification was negligible. The effects of evaporation rate, however, were found to be significant when the vacuum relief valve is jammed shut or when the cargo is transported at super-atmospheric pressure. Evaporation within the tank tends to keep the partial pressure of the vapor in the tank at some value intermediate between the saturation pressure of the liquid and the pressure that corresponds to the expansion of a gas whose mass does not increase as the vapor space volume increases during the discharge. Since the tank pressure is an important factor in predicting the discharge rate, a new model of evaporation rates appropriate for closed tanks was developed. The tests also showed that the discharge rate of liquids was limited by changes of phase to smaller values than would be predicted by the single-phase flow relations used in the previous Venting Rate Model. Two-phase flow models were therefore developed.

Task 4 - Model Verification. In this task, the Venting Rate Model was compared to the test data and revised as indicated by the comparisons. This was, in fact, a continuing process during the entire project.

Each submodel of the overall model was first compared to representative test data relevant to that model and then corrected when necessary. In addition, the tests were used to determine the best values for any empirical constants appearing in the model. It should be noted that no submodel required more than one such empirical constant and that the values of the constants were all physically reasonable.

The entire model was then compared to tests for many different kinds of cargo discharge:

- o nonvolatile liquids, with and without operable vacuum relief valves;
- o air ingestion during the discharge of volatile and nonvolatile liquids;
- o water ingestion during the discharge of nonvolatile liquids heavier and lighter than water; and
- o volatile liquids transported at slightly super-atmospheric pressures.

The comparisons of the model to these tests were very good.

It is concluded that the revised model is now satisfactory for use in the Hazards Assessment Computer System.

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## PRINCIPAL SYMBOLS

### English Symbols

$A_p$	area of puncture
$A_{Lo}, A_{Vo}$	puncture areas occupied by liquid and gas discharges
$A_{in}, A_{out}$	puncture areas occupied by air ingestion inflow and cargo liquid outflow
$A_T$	tank cross-sectional area
$C_D$	discharge coefficient
$C_{pA}$	specific heat of air (constant pressure)
$C_{pL}$	specific heat of cargo liquid (constant pressure)
$\bar{C}_{pV}$	specific heat of cargo vapor (along saturation line)
$g$	acceleration of gravity
$h_L$	specific enthalpy of liquid
$K$	empirical constant in water ingestion models
$M_A, M_L, M_V$	masses of air, liquid, and vapor in tank
$P_V$	partial pressure of vapor in tank
$P_T$	tank vapor-space pressure
$P_\infty$	ambient pressure (includes hydrostatic head of water when the puncture is submerged)
$R$	gas constant of cargo vapor
$R_m$	gas constant of air/vapor mixture
$t$	time
$T$	cargo temperature
$T_{air}$	air temperature
$v_L, v_V$	specific volume of liquid and vapor
$V_T$	tank volume

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# PRINCIPAL SYMBOLS (Cont'd)

$W_A$	mass flow rate of air into tank
$W_e$	mass rate of evaporation at liquid/vapor interface
$W_{Ao}, W_{Lo}, W_{Vo}$	mass flow rates of air, liquid, and vapor discharges
$W_w$	mass flow rate of water into tank
$x_c$	quality of two-phase choked flow
$z$	compressibility factor of vapor
$z_L$	elevation of cargo liquid surface above tank bottom
$z_{Lh}, z_{Gh}$	elevation of liquid and gas discharge streams above tank bottom
$z_T$	elevation of tank top above bottom
$z_w, z_{wt}$	elevation of outside water surface and water surface in tank

## Greek Symbols

$\beta$	two-phase flow parameter, see eq. (II.7b)
$\gamma_m$	ratio of specific heats for vapor or vapor/air mixture
$\Delta P_{valve}$	vacuum set-point opening of relief valve
$\lambda$	latent heat of evaporation
$\rho_L, \rho_w$	density of cargo liquid and water

## Other Subscripts

$c$	related to choked flow
$Eq.$	related to equilibrium of outside pressure and tank pressure at puncture elevation
$i$	initial value
$s$	related to constant entropy process in which the pressure is reduced to ambient



## I. INTRODUCTION

As part of the Hazards Assessment Computer System (HACS) of the Chemical Hazards Response Information System, a model was previously developed to predict the rate of liquid or gaseous cargo discharged to the environment by an accidental rupture of a ship tank [1]. An independent review of this model later revealed several errors in its basic equations and a lack of realism in some of its underlying assumptions [2]. The Coast Guard has sponsored the present program to correct these deficiencies and to validate the corrected model experimentally.

Errors in the basic equations and incorrect assumptions made in developing the previous model included:

- o the thermodynamic energy balance used to predict tank pressure and temperature for volatile cargos was incorrect;
- o incompressible flow equations were used to predict the venting of a volatile liquid, even though the liquid partially vaporizes during the venting;
- o the discharge was assumed to cease when the pressure differential across the puncture was zero, even if the liquid cargo level were still above the puncture; that is, ingestion of air or water through the puncture was ignored although this would allow the remaining liquid cargo above the puncture to be discharged; and
- o incorrect values were recommended for the discharge coefficients of the puncture.

Other stated or implied assumptions that may be unrealistic or inappropriate included:

- o no provision was made for vacuum relief valves;
- o all punctures were above the waterline;

- o simultaneous venting of liquid and gas did not occur;
- o air was not present in the vapor space;
- o the specific heat of superheated vapor was used to compute enthalpy changes of saturated vapor;
- o evaporation at the liquid-gas interface in the tank was computed by assuming the vapor remained saturated; and
- o temperature stratification in the liquid resulting from evaporation was negligible.

The consequences of these errors and assumptions could not always be determined solely by examining the model or its predictions. For example, potential errors in the evaporation model could not be evaluated because a more realistic model was not available. Further, the overall accuracy of the venting rate model was unknown.

The present program was designed to correct these errors and to validate the models. It had the following scope.

1. Review the venting rate model and any other relevant literature, with the aim of uncovering errors, inconsistencies, and unrealistic assumptions.
2. Correct and reformulate the venting rate model and reprogram it for computerized solution.
3. Review the Chemical Properties File of HACS for applicability to the corrected venting rate model.
4. Conduct a sensitivity analysis of the computed results to the input variables.
5. Design and conduct a model-scale test program to obtain venting rate data for tank conditions and cargo conditions analogous to full scale.
6. Revise the venting rate model as indicated by the correlation of the test data and the model predictions.

These efforts were arranged into four tasks. Task 1 covered the first effort given above; Task 2 covered the next three; Task 3 covered the fifth effort; and Task 4 covered the last effort. To present a more readable narrative, this final report is not arranged strictly by tasks. The complete analytical model development is presented first (Tasks 1, 2, and 4). Then the test program (Task 3) is described. Comparisons of the tests and the model predictions are given next, followed by the conclusions and appendices, which incorporate the sensitivity analysis and the review of the Chemical Properties File.

## II. TASK II. REFORMULATION OF MODEL

The venting rate model was reformulated in detail in the Interim Report [3]. A summary of that development is presented in this section, with a complete derivation presented in Appendix A. More complete details of the air and water ingestion submodels and the evaporation rate model will be given, however, since they were not fully developed at the time the Interim Report was issued.

### II.1 Reformulation of Tank Energy Balance

The total mass of cargo discharged can sometimes be easily estimated (for example, the quantity of liquid above the puncture). But the venting rates and the time required for the venting can only be determined by coupling energy and mass balances which describe the time-varying pressure and temperature of the cargo within the tank cargo to models of the discharge rate through the puncture. As in AMSHAH [1], two limiting conditions are assumed in formulating the energy balances:

- (1) all processes occurring in the tank are adiabatic,  
or
- (2) all processes occurring in the tank are isothermal.

These assumptions eliminate the need to consider energy transfers with the outside and thus greatly simplify the models. Furthermore, predictions made using these assumptions will bound the actual tank conditions. In addition, isothermal and adiabatic conditions are practically identical for a nonvolatile cargo, since the temperature changes little during the venting, and so for this case the assumption of which process actually occurs is not important.

The critical review [2] of AMSHAH's model concluded that temperature stratification as a result of heat and mass transfer between the liquid and vapor phases of the cargo within the tank might need to be accounted for in the energy balances. Experimental results to be described later show, however, that

the temperature stratification is small and has only a negligible effect on the tank energy balance, even for very rapid venting of volatile cargos. In agreement with the original models, then, a "well stirred reactor" approach is assumed.

Considering the control volume shown in Figure II.1 and assuming for the moment that the processes within the tank are adiabatic, the correct form of the energy balance for the cargo in the tank is [3]:

$$\begin{aligned} (M_L C_{PL} + M_V \bar{C}_{PV} + M_A C_{PA}) \frac{dT}{dt} = v_T \left( \frac{dP_T}{dt} \right) \\ + \lambda \left( \frac{dM_L}{dt} + W_{Lo} \right) + W_A C_{PA} (T_{air} - T) \end{aligned} \quad (II.1)$$

The negligible changes in gravitational potential energy and kinetic energy have been ignored in this equation [1,2,3]. The bar over  $\bar{C}_{PV}$  is meant to emphasize that the enthalpy change of saturated vapor ( $dh_V = \bar{C}_{PV} dT$ ) is the relevant quantity for the vapor, for which  $C_p$  of superheated vapor is a poor approximation. For isothermal processes, equation (II.1) reduces simply to  $T = \text{constant} = \text{initial temperature}$ .

The mass balances which must be used with equation (II.1) are:

$$\frac{dM_L}{dt} = -W_{Lo} - W_e \quad (II.2a)$$

$$\frac{dM_V}{dt} = -W_{Vo} + W_e \quad (II.2b)$$

$$\frac{dM_A}{dt} = -W_{Ao} + W_A \quad (II.2c)$$

Since  $v_T = v_L M_L + v_V M_V$ , where the specific volume of the vapor is based on its partial pressure in the air/vapor mixture in the vapor space, and since  $dv_T/dt = 0$ , equations (II.2a) and (II.2b) can be combined to give a relation for the evaporation rate:

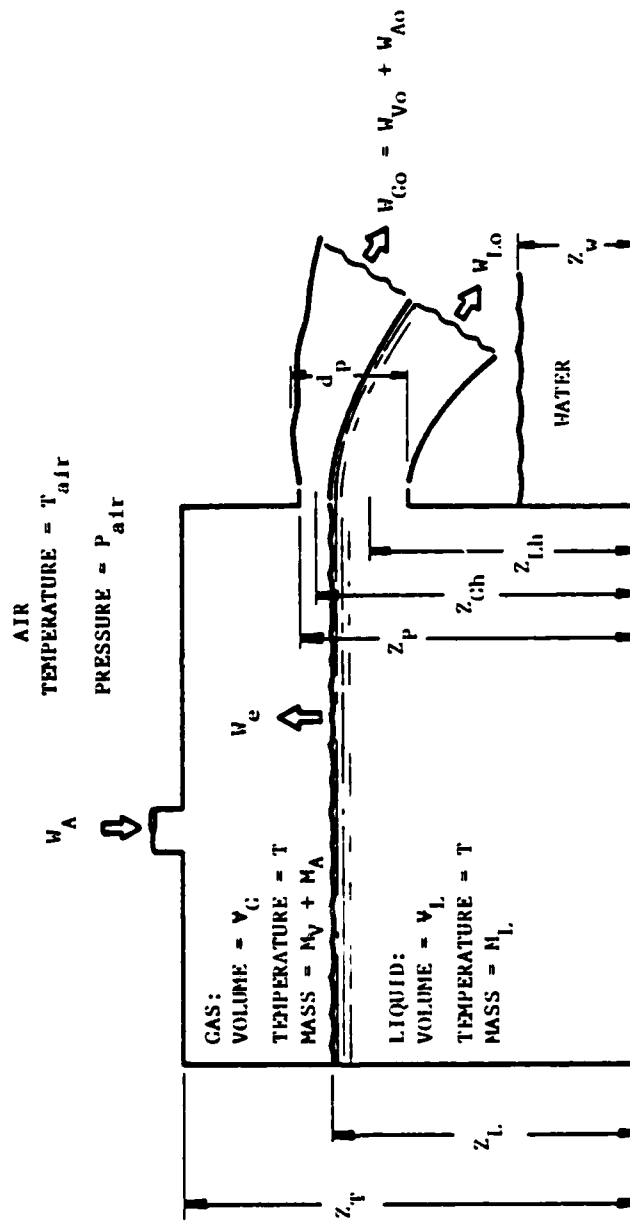


FIGURE 11.1. NOMENCLATURE FOR TANK ENERGY BALANCE

$$W_e = \left[ v_L W_{Lo} + v_V W_{Vo} - M_L \frac{dv_L}{dt} - M_V \frac{dv_V}{dt} \right] / (v_V - v_L) \quad (\text{II.3})$$

Equations (II.1) - (II.3) constitute the reformulated tank energy and mass balances that are appropriate for volatile cargos.

For a nonvolatile cargo (vapor pressure at normal temperature is much smaller than atmospheric pressure), the previous mass and energy balances include many zero or near-zero terms, and a numerical solution using these equations would be unduly complicated. Only the mass balances (equations (II.2) with  $W_e = 0$ ) and a relation that gives the change in the vapor space pressure when the vapor space volume changes are required. If the vacuum relief valve is jammed shut, the pressure relation follows from the adiabatic expansion relation:

$$P_T = P_{Ti} \left( \frac{v_T - M_{Li} v_{Li}}{v_T - M_L v_L} \right)^{\gamma_m} \quad (\text{II.4})$$

where the subscript i indicates the initial conditions at the time the puncture is created. The exponent  $\gamma_m$  is equal to one for isothermal conditions, and equal to the ratio of specific heats of the vapor/air mixture in the vapor space for adiabatic conditions. If the relief valve is operable, equation (II.4) is still valid as long as the tank pressure remains above the set-point vacuum of the valve. Once the set-point is exceeded, the tank pressure relation is

$$P_T = P_{atm} - \Delta P_{valve} \quad (\text{II.5})$$

since it is assumed that the valve always remains open thereafter.

## II.2 Flow Rate Models

The next part of the venting rate model relates the discharge rates  $W_{Lo}$ ,  $W_{Vo}$ , and  $W_{Ao}$  to the tank conditions.

When its vapor pressure is higher than atmospheric pressure, a venting liquid may partially vaporize and a venting vapor

may partially condense. These phenomena must be included in the flow models; the fact that they were not included in the models described in [1] limits their predictive accuracy, as will be discussed further in Section III.4. Although there are many correlations available to compute two-phase flow [4], they are all mathematically very complicated when a change of phase occurs within the flow stream. For this reason, a thermodynamic-equilibrium type of flow is assumed, but an empirically determined factor is applied to the effective density to increase the reliability of the predictions.

In general, a thermodynamic-equilibrium flow is given by

$$W_o = C_D A_o \rho_{out} [2 (h_{in} - h_{out}) + 2g (z_{in} - z_{out})]^{1/2} \quad (II.6)$$

where the subscript "in" refers to a location within the tank where the fluid velocity is small and "out" refers to the exit plane of the puncture. The discharge coefficient  $C_D$  is meant to account for irreversible flow and other nonideal effects. Equation (II.6) can be specialized to include volatile and nonvolatile liquids and gases. For a volatile liquid, the appropriate form is derived by assuming an isentropic process through the puncture; the result, as shown in Appendix A, is:

$$W_{Lo} = C_D A_{Lo} \rho_{Lo} \left\{ 2 [C_{PL} (T - T_s) - C_{PL} T_s \ln \frac{T}{T_s} + (P_T - P_V)/\rho_L + g (z_L - z_{Lh})] \right\}^{1/2} \quad (II.7a)$$

where  $\rho_{Lo}$  is the exit-plane density:

$$\rho_{Lo} = \left\{ v_{Ls} + \left[ s T_s (v_{Vs} - v_{Ls}) C_{PL} \ln \frac{T}{T_s} \right] / \lambda_s \right\}^{-1} \quad (II.7b)$$

Here, the subscript s indicates the saturated state at which the pressure is equal to the pressure  $P_o$  outside the tank (atmospheric, plus the static head of water if the puncture is submerged). The temperature  $T_s$  and the other properties ( $v_{Ls}$ ,



$v_{Vs}$ , etc.) can be computed from the Chemical Properties File data. Average values over the temperature range  $T$  to  $T_s$  should be used for  $C_{pL}$  and  $\rho_L$ . The term  $(P_T - P_V)$  in equation (II.7a) allows for the possibility that the vapor space pressure differs from the saturation pressure  $P_V$  corresponding to the liquid temperature.

The factor  $\beta$  in equation (II.7b) represents an empirical way to incorporate the experimental observation [25] that the actual fraction of the liquid vaporized is somewhat less than would be predicted by thermodynamic equilibrium relations. According to the tests described later,  $\beta = 0.12$  gives a good correlation to the data. It is recognized that equations (II.7a) and (II.7b) are just an approximation to a complex two-phase flow [4], but the comparison of the model to the tests indicates that the approximation is satisfactory for this application.

The flow relation for a volatile vapor is similarly:

$$w_{Vo} = C_D A_{Vo} \rho_{Vo} \left\{ 2 \left[ \bar{C}_{pV} (T - T_s) + ZRT_s \ln \frac{P_T}{P_\infty} - \bar{C}_{pV} T_s \ln \frac{T}{T_s} \right] \right\}^{1/2} \quad (II.8a)$$

where

$$\rho_{Vo} = \left\{ v_{Vs} - \beta \left[ ZRT_s \ln \frac{P_T}{P_\infty} - T_s \bar{C}_{pV} \ln \frac{T}{T_s} \right] \frac{(v_{Vs} - v_{Ls})}{\lambda_s} \right\}^{-1} \quad (II.8b)$$

When a mixture of vapor and air is venting, the air tends to suppress the fairly small phase change effects of the vapor, so equation (II.6) in this case reduces to [5]:

$$w_{Vo} = \frac{C_D A_{Vo} P_T}{\sqrt{R_m T}} \left( \frac{v_A}{v_V + v_A} \right) \left\{ \frac{2\gamma_m}{\gamma_m - 1} \left[ \left( \frac{P_T}{P_\infty} \right)^{\frac{2}{\gamma_m}} - \left( \frac{P_T}{P_\infty} \right)^{\frac{\gamma_m + 1}{\gamma_m}} \right] \right\}^{1/2} \quad (II.9a)$$

and

$$W_{Ao} = \left( \frac{v_v}{v_a} \right) W_{Vo} \quad (\text{II.9b})$$

The specific volumes of the vapor and air in the tank are based on their respective partial pressures.

For nonvolatile cargos, the flow equations for vapor or a vapor/air mixture are the same as (II.9a) and (II.9b). The flow equation for discharge of a nonvolatile liquid can be obtained by standard Bernoulli-type relations; it is:

$$W_{Lo} = C_D A_{Lo} \rho_L \left\{ 2 \left[ (P_T - P_\infty) / \rho_L + g (Z_L - Z_{Lh}) \right] \right\}^{1/2} \quad (\text{II.10})$$

In all these models, as well as in the water and air ingestion models to be described subsequently, both vapor and liquid outflows can occur simultaneously. This is handled analytically by partitioning the total puncture area into a liquid outflow area  $A_{Lo}$  and a vapor outflow area  $A_{Go}$ .

### II.3 Evaporation Rate Model

The discharge rates of liquid and gas depend on the vapor space pressure,  $P_T$ . Since evaporation of liquid into the vapor space during the discharge tends to keep  $P_T$  from decreasing as the vapor space volume increases, it is important to model the evaporation realistically.

Given the cargo temperature, the equilibrium pressure of the vapor (i.e., the liquid saturation pressure) can be computed from the Chemical Properties Files. The pressure and temperature fix the vapor density, so equation (II.3) can be used to determine the evaporation rate required to maintain the pressure at its equilibrium value. AMSHAH [1] used exactly this procedure. However, it is clear that to create the required evaporation rate there must be a driving force; the vapor space pressure must be less than the liquid saturation pressure [6,7]. The tests conducted during the present program confirmed this conclusion. Thus,  $v_v$  cannot be related only to the cargo

temperature, and equation (II.3) must be augmented by an evaporation rate model.

Evaporation rates from a liquid into the atmosphere are commonly related to the difference between the saturation pressure of the liquid and the actual partial pressure of the vapor in the air above the liquid [8]:

$$W_e = \alpha A_T (P_{V,sat} - P_V) / \sqrt{2\pi RT} \quad (II.11)$$

where the coefficient  $\alpha$  must be determined experimentally. For the present tests, where evaporation in a closed container was obtained, equation (II.11) could not be made to correlate the data very well; further, the best values of  $\alpha$  appeared to vary widely from test to test, as a function of venting rates and chemical properties. By examining several kinds of correlations, it was found that a better data fit for these tests was given by:

$$W_e = W_{e,sat} - 1.9 (W_{e,sat})^{5/4} / (A_T \rho_L \sqrt{RT})^{1/4} \quad (II.12)$$

where  $W_{e,sat}$  is the evaporation rate required to maintain the vapor partial pressure at its equilibrium value. The constant in equation (II.12) varied from about 1.6 to 2.2 from test to test, with the best average being 1.9. The rationale for equation (II.12) is given in Appendix A.

Although equation (II.12) implies that  $P_{V,sat} - P_V$  is non-zero, it cannot be put in a form that reveals this dependency explicitly. However, the model does show that larger evaporation rates give larger differences between  $W_{e,sat} - W_e$ , and therefore larger pressure differences,  $P_{V,sat} - P_V$ .

In the computerized solution procedure, the evaporation rate model is used as follows. At any instant,  $W_{e,sat}$  is calculated from equation (II.3) by assuming that the vapor is in equilibrium with the liquid. Equation (II.12) is then used to compute the corrected  $W_e$ . Using this value of  $W_e$ , equation

(II.3) is solved for  $v_v$ , and finally the actual vapor pressure is computed from the vapor equation of state:  $P_v = ZRT/v_v$ . Several iterations are required for convergence.

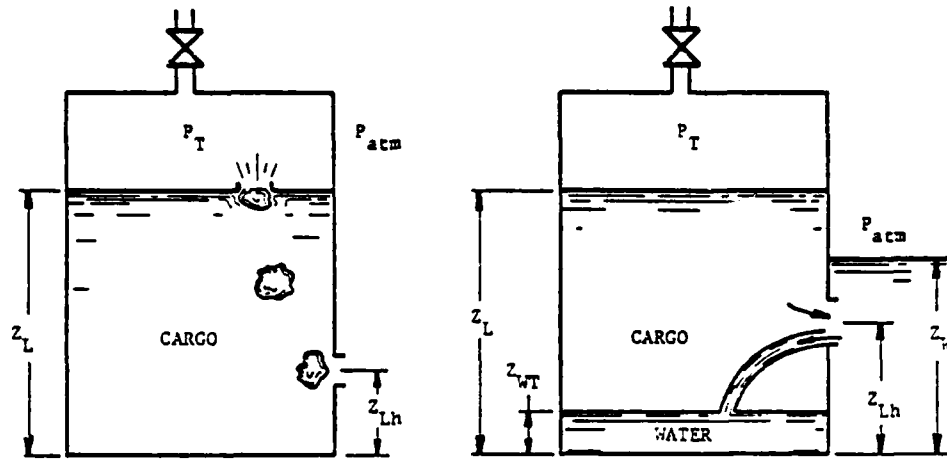
When the pressure at the puncture, e.g.,  $P_T + \rho_L g(Z_L - Z_{Lh})$ , is substantially larger than  $P_\infty$ , the changes in  $P_T$  during the venting have a relatively small effect on the discharge. Therefore, it is apparent that the effects of the evaporation rate on the discharge are likely to be small for large tanks.

#### II.4 Air and Water Ingestion Models

There are many cases in which air or water ingested into the tank through the puncture can cause additional cargo to be discharged. For example, when liquid is venting and the vacuum relief valve is jammed shut, a vacuum may eventually be created in the vapor space that is sufficient to lower the tank pressure at the puncture to the ambient pressure. Even though the net pressure differential across the puncture is now zero, the discharge will not cease if the liquid level is still above the puncture, although this was assumed to be the case in [1]. Instead, water or air (depending on whether the puncture is submerged) will be ingested to relieve the vacuum. For submerged punctures, water can be ingested even when the relief valve is operable.

##### II.4.1 Air Ingestion Model

Air can only be ingested when (1) liquid is being vented, (2) the puncture is not submerged, and (3) the static pressure,  $P_T + \rho_L g(Z_L - Z_{Lh})$ , at the puncture is nearly equal to atmospheric pressure. When all these conditions are met, typically when the relief valve is jammed shut, tests have shown that the outflow becomes intermittent. Air bubbles are ingested to relieve the vacuum in the vapor space, thereby halting the outflow temporarily; more liquid is then discharged, until  $P_T + \rho_L g(Z_L - Z_{Lh})$  is again decreased to near atmospheric pressure; more air bubbles are ingested, and so on, until the liquid level falls below the top of the puncture; see Figure II.2a.



a) AIR INGESTION MODEL

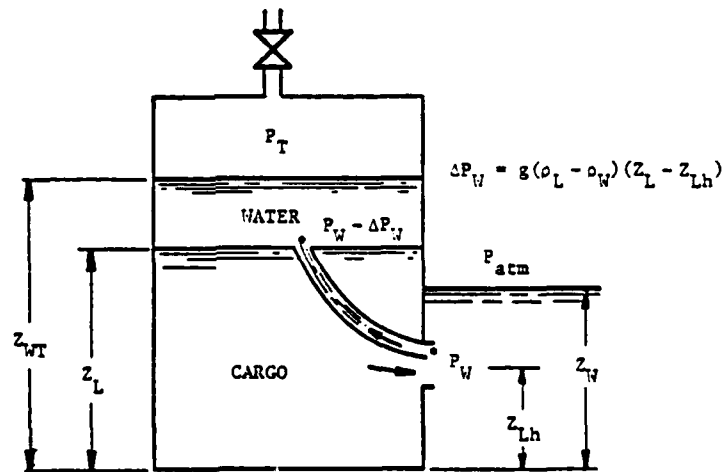
b) WATER INGESTION MODEL,  
 $\rho_L < \rho_W$ c) WATER INGESTION MODEL,  
 $\rho_L > \rho_W$ 

FIGURE II.2. AIR AND WATER INGESTION MODELS

The available literature on this kind of intermittent flow is sparse [9]. A relatively simple but plausible model has therefore been developed for the venting rate model. The model contains one empirical parameter (the bubble size factor) that is selected to give agreement with test results. In summary, the model is described as follows:

- o Air ingestion begins when the combination of tank liquid level and vapor space pressure has decreased nearly to atmospheric pressure:
- $$P_T - P_{atm} + g \rho_L (Z_L - Z_{Lh}) \leq 0.001 P_{atm} \quad (II.13)$$
- o When (II.13) is satisfied, an air bubble is ingested whose volume is a specified multiple (the bubble size factor) of a sphere having a cross-sectional area equal to the puncture area.
  - o The vapor space pressure is increased by an amount  $\Delta P_T$  corresponding to the addition of this bubble to the vapor space; for a volatile cargo, the  $\Delta P_T$  calculation is accomplished by integrating equation (II.1) over an infinitesimal time step,  $\Delta t$ , with  $W_A \Delta t$  set equal to the ingested bubble mass; for a nonvolatile cargo, the change in the specific volume of the vapor space gas mixture is computed, and  $\Delta P_T$  is then calculated by ideal gas laws.
  - o Since the left hand side of equation (II.13) is now larger than the right hand side, additional liquid is discharged, in accordance with the appropriate energy balances and flow relations; the outflow continues until equation (II.13) is again satisfied.
  - o Another air bubble is ingested, and the process is repeated.

The right hand side of equation (II.13) is set equal to  $0.001 P_{atm}$ , rather than to zero, to avoid numerical problems in the computerized model, by ensuring that the outflow has not ceased completely before the bubbles are ingested. Experiments have indicated that the value of the right hand side is between zero and about  $0.002 P_{atm}$ , so equation (II.13) is reasonably accurate. As the vapor space volume increases, the increase in the pressure  $\Delta P_T$  due to the ingestion of a single bubble becomes smaller, and the net  $\Delta P$  driving the liquid outflow therefore becomes smaller; further, the hydrostatic head  $Z_L - Z_{Lh}$  is smaller than it was originally. To prevent an excessive number of computational iterations, more than one bubble can be admitted when necessary. The criterion used in the computerized model is that the value of  $P_T + \rho_L g (Z_L - Z_{Lh})$  after each bubble (or bubbles) has been admitted should be reasonably close to the value obtained after the very first bubble is ingested. This procedure also tends to bring the vapor space pressure smoothly to atmospheric pressure just as the puncture is uncovered.

By comparison of the venting rate model to the air ingestion tests, the best value of the bubble volume factor mentioned above has been found to be 5.6. This gives an effective bubble diameter of about 1.8 times the puncture diameter, which agrees qualitatively with visual observations.

For very large punctures, and especially for long vertical punctures, the outflow may not completely cease during the air ingestion; that is, the bubbles may be ingested only over the top part of the puncture. This kind of ingestion was not observed in the smaller punctures used in the present tests, but if it does occur, the model described herein will provide a first approximation to the actual outflow rate.

#### II.4.2 Water Ingestion Models

When the puncture is below the water line, the tests have shown that water will be ingested whenever the net pressure

difference across the puncture approaches zero. Once started, water ingestion continued until the puncture was under water on both sides. In contrast to air ingestion where the outflow became intermittent, here the outflow remained steady but shared the puncture area with the water inflow. In other words, there was a water stream entering the tank and simultaneously a cargo liquid stream leaving the tank.

The models developed here assume the cargo and the ingested water do not mix or react chemically. This is a conservative assumption for many cargos, since the venting rate will be predicted to be larger than will actually occur. In any event, a model that included mixing or chemical reactions in the tank would be excessively complicated. Since the cargo and the liquid do not mix, the ingested water will displace a lighter cargo upward, thereby exposing more cargo to the puncture and increasing the static pressure at the puncture; see Figure II.2b. When the water is lighter than the cargo, the ingested water will float on top and thereby increase the static pressure at the puncture; see Figure II.2c. In both cases, the ingested water allows additional cargo to be discharged.

The driving force for water ingestion is the buoyancy (negative or positive) of the water relative to the cargo, which forces the water into the tank when the cargo outflow momentum is low (i.e., when the net positive pressure difference across the puncture is near zero). The tests showed that the water entering the tank under these conditions soon broke up into a stream of discrete droplets. Since an analysis of water ingestion could not be found in the literature, several new but approximate models were developed. The models contain one empirical constant that can be chosen to give better correlation to the test data. In summary, they are described as:



- o Tank conditions for which the pressure differential across the puncture is zero are determined by an equation similar to equation (II.13) of the air ingestion model, but with the right hand side replaced by the hydrostatic head  $g\rho_W (Z_W - Z_{Lh})$ .
- o Ingestion begins just prior to this condition of zero net  $\Delta P$ , such that a small positive pressure differential, the magnitude of which is determined with the aid of the empirical constant mentioned earlier, will continue to force out cargo.
- o The water inflow remains in a stream tube; this assumption allows the calculation of a  $\Delta P$  acting on the water in the direction to force water into the tank.
- o The puncture area is shared between the cargo outflow and the water inflow such that the inflow and outflow rates maintain the same pressure difference driving the cargo outflow as at the start of ingestion.

When the water is lighter than the cargo and the cargo static pressure at the puncture is equal to the outside pressure (atmospheric plus water hydrostatic head), the net pressure difference from the "inlet" to the "outlet" of the water streamtube mentioned above (Figure II.2c) can be calculated by considering the various static heads; the result is:

$$\Delta P_W = g (\rho_L - \rho_W) (Z_L - Z_{Lh}) \quad (II.14)$$

which is just the buoyancy-induced pressure on a vertical water column of equal length. When the water is heavier than the cargo, the pressure differential is similarly:

$$\Delta P_W = g (\rho_W - \rho_L) (Z_{Lh} - Z_{WT}) \quad (II.15)$$

In both cases, the water inflow rates are calculated by:

$$W_W = C_D A_{in} \rho_W \sqrt{2 \Delta P_W / \rho_W} \quad (II.16)$$

where  $A_{in}$  is the part of the puncture occupied by the water stream. There is little or no change of phase (evaporation) of the liquid outflow during water ingestion since the tank pressure is nearly in equilibrium with the outside pressure; see equations (II.7b) and (II.8b). Thus, for simplicity, the cargo venting rate is calculated using an incompressible flow relation. The water inflow rate is supposed to be just large enough to maintain the original net  $\Delta P$  at the time when ingestion started, so the cargo outflow rate is:

$$W_{Lo} = C_D A_{out} \rho_L \left\{ 2 \left[ (P_{Ti} - P_{T,Eq.}) / \rho_L + g (Z_{Li} - Z_{L,Eq.}) \right] / K \right\}^{1/2} \quad (II.17)$$

where  $P_{T,Eq.}$  and  $Z_{L,Eq.}$  are the values of  $P_T$  and  $Z_L$  which first satisfy the water ingestion form of equation (II.13).  $P_{Ti}$  and  $Z_{Li}$  are values at some slightly earlier time; in the computerized model, these correspond to the values at the time step preceding the start of water ingestion. The actual  $\Delta P$  across the puncture may be only a fraction of the value  $(P_{Ti} - P_{T,Eq.}) / \rho_L + g (Z_{Li} - Z_{L,Eq.})$  selected by the computer scheme. Thus, an empirical constant  $K$  is inserted in equation (II.17); the best value obtained from the test comparisons is  $K = 83$ . It will be shown later that there is reason to believe that  $K = 83$  is a universal value, and not just a value specific to a given test and computer prediction.

When the ingested water is lighter than the cargo, it floats on the liquid surface and suppresses any further evaporation. When it is heavier and thus sinks, evaporation is possible, but it is neglected in the models because the energy balances needed to predict temperature changes become much more complicated. In effect, the cargo is treated as an incompressible liquid and a non-condensable gas after water is ingested.

As mentioned earlier, the inflow rate of water is adjusted so as to maintain the pressure differential driving the cargo outflow. When the water is heavier than the cargo (Figure

II.2b), this merely requires that the two volumetric flow rates be equal so that  $Z_L$  remains constant; in other words

$$W_w = \rho_w W_{Lo} / \rho_L \quad (\text{II.18})$$

Equations (II.16), (II.17), and (II.18) are all made compatible by allocating the puncture area between the two streams as:

$$A_{in} + A_{out} = A_p \quad (\text{II.19})$$

These equations are sufficient to determine all the flow rates.

When the water is lighter than the cargo (Figure II.2c), the relation between  $W_w$  and  $W_{Lo}$  depends on whether the relief valve is operable. When it is, the vapor space pressure remains constant. Thus, to maintain a constant  $\Delta P$  across the puncture, the weight of the water ingested must equal the weight of the cargo discharged, for otherwise the hydrostatic head above the puncture would change. The flow relation, consequently, is

$$W_w = W_{Lo} \quad (\text{II.20a})$$

On the other hand, when the relief valve is jammed, the conditions specified by equation (II.20a) would compress the vapor space gas since a larger volume of water would be admitted than the volume of cargo discharged. Thus, the increase in  $P_T$  and the hydrostatic pressure changes must be balanced in order to hold constant the  $\Delta P$  driving the water outflow. The appropriate relation is derived by expressing the  $\Delta P$  at the puncture in terms of the vapor space compression and the changes in water and cargo levels; the result, as shown in Appendix A, is:

$$W_w = W_{Lo} \left\{ \frac{g (Z_T - Z_{WT}) + \gamma_m P_T / \rho_L}{g (Z_T - Z_{WT}) + \gamma_m P_T / \rho_W} \right\} \quad (\text{II.20b})$$

Equations (II.13) - (II.20) constitute the models of water ingestion for those cases where some cargo liquid is discharged

prior to the start of ingestion. There are also situations where water can be ingested immediately upon puncturing the tank. This would occur whenever

$$P_{Ti} - P_{atm} + g\rho_L (Z_{Li} - Z_{Lh}) < g\rho_W (Z_W - Z_{Lh}) \quad (II.21)$$

For these conditions, cargo would never be discharged unless water ingestion permits it.

When  $\rho_W < \rho_L$ , it is assumed that water will enter the tank and rise to the top without any cargo being discharged, until the tank pressure at the puncture is equal to the outside pressure:

$$\begin{aligned} P_T - P_{atm} + g\rho_W (Z_{WT,Eq.} - Z_{Li}) + g\rho_L (Z_{Li} - Z_{Lh}) \\ = g\rho_W (Z_W - Z_{Lh}) \end{aligned} \quad (II.22)$$

Of course, if the required level of water  $Z_{WT,Eq.}$  is greater than the elevation of the top of the tank, the tank will fill up with water before equation (II.22) is satisfied; this case is modeled separately. If  $Z_{WT,Eq.} < Z_T$ , water will continue to flow into the tank, but now cargo will be discharged since there is a net positive  $\Delta P$  across the puncture. At this point, the venting model is the same as for the similar water ingestion models described earlier, although the cargo outflow rate equation can be simplified, as shown in Appendix A, to give:

$$W_{Lo} = C_D A_{out} \rho_L \sqrt{0.2g (Z_{WT,Eq.} - Z_{Lh})/K} \quad (II.23)$$

$Z_{WT,Eq.}$  can exceed  $Z_T$  when the relief valve is operable, and  $P_T$  does not increase as the water enters. When this occurs, it is assumed that the water completely fills up the vapor space. The pressure thereafter in the vanishingly small volume between the top of the tank and the water is assumed to remain constant at the initial value maintained by the relief valve. Thus, as further water enters in an attempt to satisfy

equation (II.22), cargo will be forced out at the same volumetric flow rate. For this case, there is a net  $\Delta P$  across the puncture forcing water into the tank, so

$$W_W = C_D A_{in} \rho_W \left\{ 2 \left[ (P_{atm} - P_{Ti} / \rho_W + g (Z_W - Z_T) - g (\rho_L - \rho_W) (Z_L - Z_{Lh})) \right] \right\}^{1/2} \quad (II.24a)$$

and

$$W_{Lo} = \rho_L W_W / \rho_W \quad (II.24b)$$

There are no other relations available to determine the way the puncture area is shared, so it is assumed that  $A_{in} = A_{out} = A_p/2$ .

When the relief valve is jammed shut, the vapor space pressure will increase as the water enters, and the condition that  $Z_{WT,Eq.} \geq Z_T$  is impossible. For this case, equation (II.22) must be solved simultaneously with the relation between vapor space volume and pressure:

$$P_T = P_{Ti} \left( \frac{Z_T - Z_{Li}}{Z_T - Z_{WT,Eq.}} \right)^{\gamma_m} \quad (II.25)$$

where the subscript "i" denotes the initial conditions.

When  $\rho_W > \rho_L$ , the water initially ingested into the tank will lift the cargo until there is a hydrostatic balance at the puncture. Since the original depth of cargo is  $Z_{Li}$ , the equilibrium relation is

$$P_T - P_{atm} + g \rho_L (Z_{Li} + Z_{WT,Eq.} - Z_{Lh}) = g \rho_W (Z_W - Z_{Lh}) \quad (II.26)$$

where  $Z_{Li} + Z_{WT,Eq.}$  is the new elevation of the cargo surface. Assuming for the moment that (1)  $Z_{Li} + Z_{WT,Eq.} < Z_T$  (i.e., the new liquid level is below the top of the tank), and (2) the water in the tank does not cover the vent, further water will

be ingested, but now cargo will be discharged at the same volumetric rate. The model for this kind of venting is the same as the analogous models described earlier, except again it must be assumed that  $A_{in} = A_{out} = A_p/2$ . Ingestion will continue until the puncture is covered by water on both sides. If the water initially covers the vent, cargo is never discharged. If  $Z_{Li} + Z_{WT,Eq.} > Z_T$ , which can occur only when the relief valve is operable, it is assumed that the cargo is lifted entirely to the top of the tank and the vapor space pressure remains at its initial value. Then, as further water is ingested, cargo is discharged at the same volumetric flow rate. The water inflow rate is then given by

$$W_w = C_D A_{in} \rho_w \left\{ 2 \left[ (P_{atm} - P_{Ti}) / \rho_w + g (Z_w - Z_{Lh}) - g (\rho_L / \rho_w) (Z_T - Z_{Lh}) \right] \right\}^{1/2} \quad (II.27)$$

which is based on the net  $\Delta P$  from inside to outside at the center of the puncture. When the relief valve is jammed shut, an expression analogous to equation (II.25) is used to compute  $Z_{WT,Eq.}$ .

This completes all the water ingestion models.

## II.5 Choked Flow Models

Some cargos (e.g., butane) have vapor pressures at normal temperatures that are substantially higher than atmospheric and are carried in pressurized tanks. If one of these tanks is punctured, the pressure differential across the opening may be so large that the outflow velocity is limited by the speed of sound [5]; in other words, the flow is "choked" to some lower value than that predicted by the models described in II.2, all of which assume that the pressure in the outflow stream at the puncture exit is the outside ambient pressure. For these cases, choked flow models are developed.

For venting of a gas, the limiting exit pressure for choked flow is [5]:

$$P_c = P_T \left( \frac{2}{\gamma_m + 1} \right)^{\frac{\gamma_m}{\gamma_m - 1}} \quad (\text{II.28})$$

If the calculated  $P_c$  is greater than ambient (atmospheric, plus water hydrostatic head if the puncture is submerged), the outflow will be choked, and  $P_c$  must be used in equation (II.9a) instead of  $P_\infty$  to compute  $W_{VO}$ . As the vapor continues to vent,  $P_T$  will gradually decrease until eventually  $P_c = P_\infty$ ; at this point, the outflow is no longer choked, and equation (II.9a) can be used directly. Note that equation (II.28) considers the effects of two-phase flow for vapor as being negligible.

Ordinarily, flow of a liquid cannot be choked because of the very high speed of sound for liquids. For a volatile liquid, however, the change in phase of the outflow can easily lower the speed of sound to a fairly small value. The exit conditions for choked flow of a volatile liquid can be determined by maximizing the mass flow per unit area:

$$\frac{\dot{m}_{Lo}}{C_D A_O} = [2 (h_L - h_{Lc} - x_c \lambda_c)]^{1/2} / [v_{Lc} + \beta x_c (v_{Vc} - v_{Lc})] \quad (\text{II.29})$$

where  $x_c$  represents the relative mass fraction of vapor in the outflow. (The static head above the puncture,  $\rho g_L (Z_L - Z_{Lh})$ , is neglected in this equation, since it should be small compared to the tank pressure  $P_V$  whenever choked flow is likely.) The maximum of equation (II.29) corresponds to

$$\frac{h_L - h_{Lc} - x_c \lambda_c}{v_{Lc} + \beta x_c (v_{Vc} - v_{Lc})} = -\frac{1}{2} \left\{ \frac{\frac{d}{dT_c} (h_{Lc} + x_c \lambda_c)}{\frac{d}{dT_c} [v_{Lc} + \beta x_c (v_{Vc} - v_{Lc})]} \right\} \quad (\text{II.30})$$

Equation (II.30) can be solved numerically using the known cargo pressure and temperature to determine the choked flow exit temperature  $T_c$  and corresponding saturation pressure,  $P_c$ . If this pressure is greater than  $P_\infty$ , the flow is choked, and

these values of  $T_c$  and  $P_c$  must be used in equations (II.7) to compute  $W_{LO}$ . As the liquid vents,  $P_T$  may eventually decrease to the point where the exit pressure  $P_c$  required to choke the flow is smaller than  $P_\infty$ ; after this point, equations (II.7) can be used directly. Depending on the saturation pressure of the liquid, however, it is entirely possible for the liquid flow to remain choked throughout.

The choked flow model given by equations (II.29) and (II.30), although similar to that described in the Interim Report [3], assumes equilibrium flow rather than attempting to incorporate nonequilibrium effects [10,11] as suggested in [3]. The assumption of equilibrium flow is in general agreement with the assumptions used for nonchoked flows.

In the computer code, equation (II.30) is solved by expanding the left hand side in powers of  $\Delta T$ , where  $T_c = T - \Delta T$ . The derivatives on the right side are evaluated numerically by assuming two values for  $T_c$ . As shown in Appendix A, the temperature difference  $\Delta T$  is then found from:

$$\Delta T = T \left\{ -\frac{\beta v_V}{\lambda} [D] + \sqrt{\left(\frac{\beta v_V}{\lambda} [D]\right)^2 + \frac{4v_L [D]}{C_{PL} T} \left[1 - \frac{\beta v_V E}{\lambda} [D]\right]} \right\} \\ \div \left\{ 2 \left[1 - \frac{\beta v_V A}{\lambda} [D]\right] \right\} \quad (II.31a)$$

where  $[D]$  is the computed approximate value of the derivatives:

$$[D] = \frac{d}{dT_c} \left[ C_{PL} (T - T_c) - x_c \lambda_c \right] / \frac{d}{dT_c} \left[ v_{Lc} (1 - \beta x_c) + \beta x_c v_{Vc} \right] \quad (II.31b)$$

The parameter  $E$  is related to the saturation pressure correlation (see equation (II.33) of Section II.6) and is:

$$E = [B_0 T / (C_0 + T^*)^2] \ln 10 - 1.5 \quad (II.31c)$$



( $T^*$  is in  $^{\circ}\text{C}$ , to conform to the Chemical Properties File usage; all the other temperatures are  $^{\circ}\text{K}$ .) One or two iterations of equations (II.31) have been shown to be sufficient to determine  $T_c$ .

## II.6 Review of Chemical Properties File

The previous technical review [2] indicated that the thermodynamic data in the Chemical Properties File were in error in some cases and the temperature range covered by the correlating equations was not sufficient in other cases. These conclusions have been confirmed by the present review. Six representative chemicals, all of which might be transported in a state where phase changes can occur so that thermodynamic properties are important, were reviewed in detail to document any errors; these chemicals are: anhydrous ammonia, butane, chlorine, LNG, methyl alcohol, and propane.

Based on this review of the representative chemicals, the following conclusions have been drawn about the adequacy of the present Chemical Properties File for use in the venting rate model.

1. Correlating equations for liquid density and specific heat are slightly inaccurate, but probably acceptable.
2. Correlating equations for saturated vapor properties are probably not acceptable; in particular, the specific heat for saturated vapor (not superheated vapor) and the compressibility factor should be included.
3. Correlating equations for saturation pressure as a function of temperature are moderately inaccurate and can lead to inaccurate predictions of tank pressure during venting.
4. The single value of latent heat given for each chemical differs by as much as 20% from the true value over the range of applicable temperatures

needed in the venting rate model energy balance and flow equations.

Recommended correlations for the six reviewed chemicals, based on the indicated data sources, are given in Appendix B. The recommended correlations for latent heat are derived from the Clausius-Clapeyron equation,  $\lambda = (T^* + 273.17)(v_V - v_L)(dP_V/dT)$ , using the new correlations given for vapor compressibility, liquid density, vapor compressibility factor, and vapor pressure. Thus, a separate correlation for  $\lambda$  is not needed, since it can be computed from the included correlations by the relation

$$\lambda = ZRB_O \ln 10 [(T^* + 273.2)/(T^* + C_O)]^2 - 5.501 \times 10^{-4} B_O P_V (T^* + 273.2)/(T^* + C_O) \rho_L \quad (\text{II.32})$$

Here,  $B_O$  and  $C_O$  are constants in the vapor pressure correlation:

$$\log_{10} P_V = A_O - B_O/(T^* + C_O) \quad (\text{II.33})$$

(In both equations (II.32) and (II.33), the units of temperature  $T^*$  are °C.)

Nonvolatile cargos are usually transported at near atmospheric pressures. Venting of gas is therefore unlikely to be important, and there will be little or no change of phase during liquid venting. Under these restrictions, the liquid density is the only cargo property that is important. The present Chemical Property File data, consequently, are sufficiently accurate for these cases.

### III. EXPERIMENTAL PROGRAM

There were three primary objectives of the test program:

1. Obtain data to evaluate the assumptions of the analytical models
2. Gain insight into air and water ingestion
3. Obtain discharge rate data for a variety of test conditions to validate the overall reliability of the venting rate model.

#### III.1 Scaling Considerations

All the tests were conducted using tanks that were much smaller than full scale. It was important, then, that the simulated cargos and the initial tank conditions be selected so as to fairly represent full scale venting behavior. The liquid/vapor properties and the test conditions needed to achieve the desired simulation were determined from a similitude analysis [12] in which the important cargo and tank parameters were arranged in various nondimensional groups. This analysis showed clearly that an exact simulation would be difficult to obtain. For example, the static head of liquid in the tank,  $g\rho_L Z_L$ , was reduced considerably in the tests in comparison to full scale. To obtain similarity between the pressure differentials driving the discharges would have required a reduction of both the tank vapor-space pressure and the atmospheric pressure in direct proportion to  $g\rho_L Z_L$ . Although it would have been feasible to use a simulated cargo whose vapor pressure was less than full scale in accordance with this scaling criterion, conducting the entire test in a vacuum chamber or other apparatus to reduce the atmospheric pressure would have been extremely difficult and expensive. For this reason, tests achieving exact similarity to some given prototype were not attempted. The correct overall phenomena were, however, represented fairly well, as will be described in the next section. Lack of exact similarity does not

seriously limit the validation of the analytical model because the model can be used to predict results for the conditions used in the tests and the predictions compared with these test results.

### III.2 Summary of Tests

#### III.2.1 Test Series No. 1: Discharge Coefficients

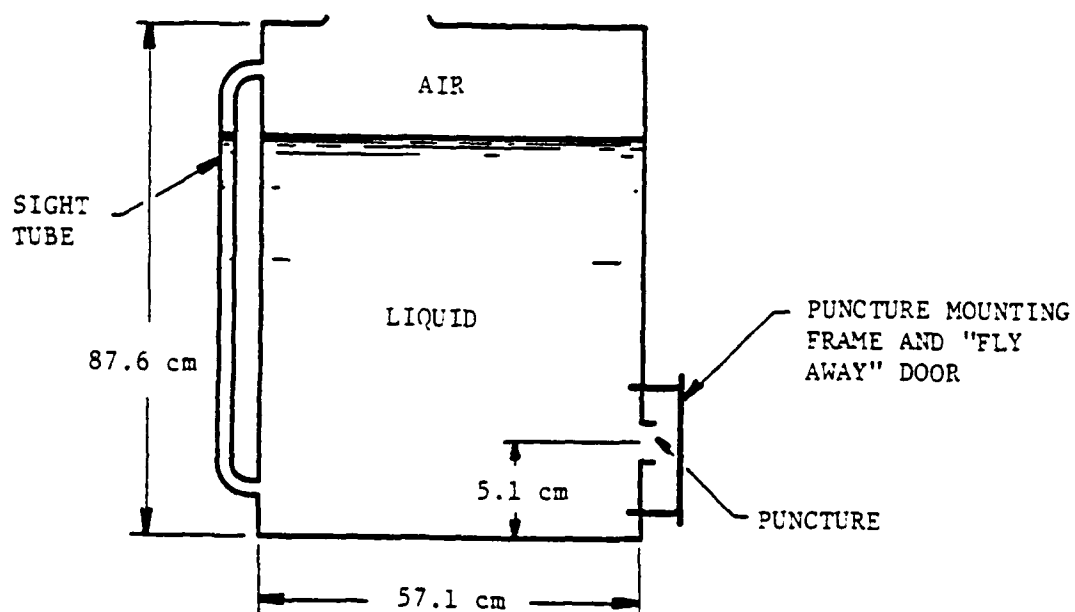
The original venting rate model [1] incorporated a fixed value of 0.8 for the discharge coefficient of the puncture. Since the predicted venting rate is directly proportional to the discharge coefficient, the first test series was designed to evaluate the implicit assumption that there is a negligible variation of  $C_D$  with puncture geometry, cargo properties, and discharge velocity.

The tests were conducted with an open cylindrical steel tank (Figure III.1a) having a diameter of 57.2 cm (22.5 in.), height of 87.6 cm (34.5 in.), and total capacity of 224.8 liters (59.4 gallons). A circular hole approximately 61 cm in diameter was machined in the wall of the tank near the bottom, around which a "picture frame" fixture was mounted flush to the outer surface. Simulated punctures were cut in sheet metal of about the same thickness as the tank wall (0.16 cm or 1/16 in.) and attached to the fixture, as indicated in Figure III.1a. In this way, three different puncture shapes could be readily tested using the same setup:

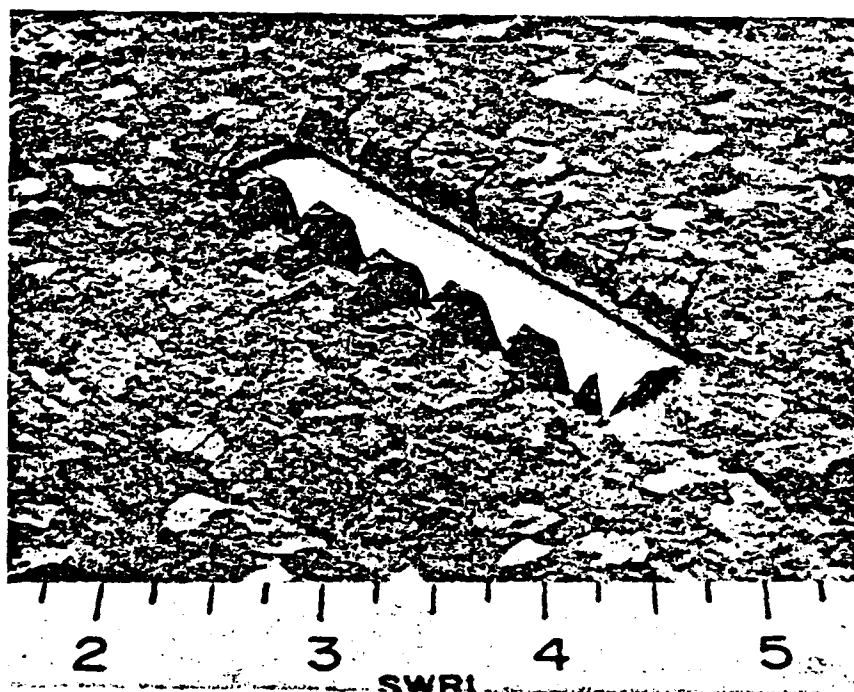
- o circular
- o horizontal rectangular (width/height = 5)
- o vertical rectangular (height/width = 5)

Each puncture shape was tested with four different edge conditions:

- o smooth
- o jagged petals, all pushed inward uniformly
- o jagged petals, all pushed outward uniformly
- o jagged petals, pushed inward randomly



a) CYLINDRICAL STEEL TANK



b) TYPICAL SIMULATED PUNCTURE WITH JAGGED EDGE

FIGURE III.1. TEST APPARATUS FOR DETERMINATION OF DISCHARGE COEFFICIENTS

The smooth-edge punctures served as a baseline with which to compare the more realistic jagged punctures. A photograph of a typical jagged edge puncture is shown in Figure III.1b. Except for some preliminary tests, the open area of all the punctures was  $11.4 \text{ cm}^2$ , corresponding to circular holes of diameter 3.8 cm (1.5 in.) and to rectangular holes 7.55 cm by 1.51 cm (2.97 in. by 0.59 in.). To vary the Reynolds number of the discharge over a wide range, three different nonvolatile liquids were used: water, aqueous glycerine solution, and hydraulic oil. The densities and viscosities of these liquids are:

<u>LIQUID</u>	<u>DENSITY (<math>\text{g/cm}^3</math>)</u>	<u>VISCOSITY (cp)</u>
Water	0.998	0.98
40% Water - 60% Glycerine	1.157	8.63
Hydraulic Oil	0.878	64.90

For a given test, the tank was filled to a preselected level, generally 64.8 cm (35.5 in.) above the centerline of the puncture. The discharge was started by opening a quick-release valve. Liquid levels in the tank were measured visually during the test by a graduated sight tube attached to the tank, and flow times were measured with a stop watch. The tests were terminated when the liquid level was still above the puncture (generally 14.6 cm, or 5.75 in.). Each test was repeated at least three times. Test-to-test correlation was excellent, with the total discharge time never varying by more than 0.5 second.

As is the usual practice [13], the discharge coefficient was calculated by using Torricelli's equation to derive a relation between  $C_D$  and the test measurements:

$$C_D = \left( \frac{A_T}{A_P} \right) \left[ \sqrt{2H_i/g} - \sqrt{2H_f/g} \right] / \Delta t \quad (\text{III.1})$$

where  $H_i$  and  $H_f$  are the initial and final liquid depths in the

tank, and  $\Delta t$  is the total discharge time. A representative Reynolds number for the discharge is given by  $\rho_L \sqrt{g(H_i + H_f)A_0} / \mu$ , which is based on the discharge velocity corresponding to the average liquid depth.

Summaries of the test results are given in Table III.1. The discharge coefficient varied much more with geometry than with Reynolds number, at least for the Reynolds numbers that are of most interest. In particular, the horizontal or rectangular punctures with outward-pushed petals give  $C_D$ 's that are 25% larger than for the other puncture geometries. The test values of  $C_D$  for the smooth-edge circular puncture agree reasonably well with published data [13] which show that  $C_D$  should decrease from about 0.68 to about 0.61 as the Reynolds number increases from about 1000 to over 100,000 or more; this favorable comparison validates the experimental procedure.

To summarize, these data show that a discharge coefficient of 0.65 should be adequate in practice for most punctures. The single exception to this recommendation would be a rectangular puncture having ragged edges pointing predominately outward;  $C_D$  for such a puncture averages about 0.825 for large Reynolds numbers. Note that the value of  $C_D = 0.8$  used in the AMSHAH model [1] roughly corresponds to the singular exception of the tests.

The wall thicknesses of the tanks used in Test Series 2 and 3 were comparable to the hole diameters; for these holes,  $C_D$  was found to be about 0.73. It is believed that these kinds of punctures are not relevant to actual ship tank punctures, and so the data are not included in Table III.1.

### III.2.2 Test Series No. 2 : Air and Water Ingestion Studies

The second series of tests was designed both to obtain insight about liquid discharge during air or water ingestion and to obtain quantitative data. In order to make visual observations, the tank was constructed of transparent acrylic

TABLE III.1. DISCHARGE COEFFICIENTS AS A FUNCTION OF  
PUNCTURE GEOMETRY AND REYNOLDS NUMBER

Representative Reynolds Numbers: Water = 96,000; Glycerine Solution = 12,600; Hydraulic Oil = 1275

Puncture Geometry	Edge Condition	Orientation	Discharge Coefficient		
			Water	Glycerine Solution	Hydraulic Oil
Circular	Smooth	--	0.650	0.663	0.648
Circular	Uniform Jagged Inward	--	0.622	0.663	0.608
Circular	Random Jagged Inward	--	0.577	0.577	0.571
Circular *	Uniform Jagged Outward	--	0.619	0.620	0.627
Rectangular *	Smooth	Horizontal	0.625	0.645	0.652
Rectangular *	Smooth	Vertical	0.624	0.643	0.652
Rectangular *	Uniform Jagged Inward	Horizontal	0.629	0.629	0.611
Rectangular *	Uniform Jagged Inward	Vertical	0.626	0.608	0.609
Rectangular *	Random Jagged Inward	Horizontal	0.609	0.605	0.586
Rectangular *	Random Jagged Inward	Vertical	0.613	0.627	0.584
Rectangular *	Uniform Jagged Outward	Horizontal	0.826	0.798	0.739
Rectangular *	Uniform Jagged Outward	Vertical	0.828	0.802	0.741

\* Aspect Ratio = 5:1



plastic. The tank, as sketched in Figure III.2, was a closed cylinder 27.6 cm (10.87 in.) in diameter and 61.0 cm (24 in.) high, with a total capacity of 36.5 liters (9.65 gallons). Two simulated circular punctures were drilled into the tank wall 4.4 cm (1.75 in.) above the bottom. The smaller puncture was 2.2 cm (0.875 in.) in diameter, and the larger was 3.2 cm (1.25 in.). Discharge coefficients for these punctures were determined to be 0.72 - 0.73, values higher than might be expected from Table III.1, but which are the result of the much larger wall thickness (about 1.2 cm) of this tank.

Air Ingestion Studies. For these tests, water was used as a convenient simulation of a nonvolatile cargo. Several preliminary tests were conducted to determine what vapor space vacuum pressures during the discharge could be held nearly constant by manually adjusting the simulated relief valve (Figure III.2).

The initial set of data-tests was conducted to determine the functional dependency of the starting point of air ingestion with liquid level and vapor space pressure. Within the accuracy of the measurements of liquid level and pressure, air ingestion was observed to begin when the tank pressure at the puncture (vapor space pressure plus hydrostatic) was within 0.25 KPa (1 in. of water) or less of atmospheric pressure; that is:

$$P_T - P_{atm} + g\rho_L (Z_L - Z_{Lh}) < 0.0025 P_{atm} \quad (III.2)$$

Since the vapor space pressure  $P_T$  could not be held exactly constant during a test, with pressure variations of  $\pm 1$  cm water not being uncommon, the value of the left hand side of equation (III.2) at the instant air ingestion started could not be calculated with exact certainty, but it was in the range of 0 to 0.0025 times  $P_{atm}$ .

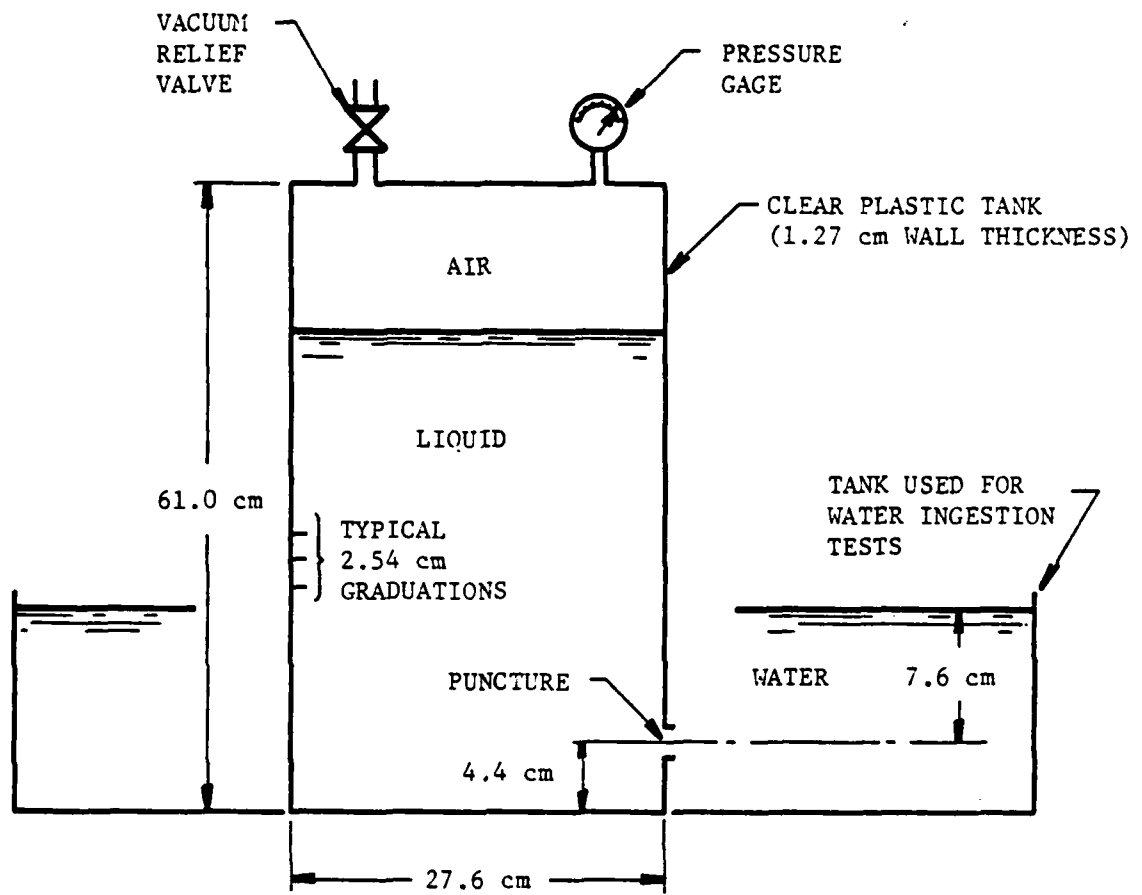


FIGURE III.2. TEST APPARATUS FOR AIR AND WATER INGESTION STUDIES

The next set of tests continued the venting past the point of air ingestion until the liquid level dropped to within 2.5 cm (1 in.) of the puncture centerline. These tests confirmed the assumptions used in developing the model of air-ingestion/intermittent outflow; that is: (1) the flow does not cease when  $P_T - P_{atm} + \rho_L (Z_L - Z_{Lh})$  first equals zero, if  $Z_L > Z_{Lh}$ ; (2) air is ingested in the form of discrete bubbles, comparable in average diameter to about two or three times the puncture diameter; (3) during the time the bubble blocks the puncture, the outflow ceases; and (4) the vapor space pressure increases each time a bubble is ingested. Table III.2 summarizes the results of these tests. Again, each test was conducted at least three times, with excellent repeatability of the results (less than 1 second variation in discharge time).

Rise times of the bubbles were measured in several tests. The bubbles appeared to attain their terminal velocity very quickly; the average velocity was computed to be 50.2 cm/sec  $\pm$  4.5 cm/sec.

Water Ingestion Studies. For these tests, the tank was installed in a shallow but larger diameter sheet-metal tank, which was then filled with water until the puncture was submerged to a preselected depth. Three different liquids were tested: carbon tetrachloride, which is heavier than water; isopropyl alcohol, which is lighter than water; and aqueous glycerine, which is heavier than water. Carbon tetrachloride and isopropyl alcohol are both virtually immiscible with water, but water dissolves readily in aqueous glycerine. For these tests, the simulated safety relief valve was closed completely to emphasize the water ingestion aspects of the discharges.

The vapor space pressure during these tests dropped almost immediately to a vacuum about numerically equal to the hydrostatic head of the liquid, and water was ingested into the tank almost immediately. After passing through the puncture,

TABLE III.2. RESULTS OF AIR INGESTION STUDIES\*

Puncture Diameter, cm	Vapor-Space Vacuum, cm of H <sub>2</sub> O	Initial Water Height, cm	Air-Ingestion Water-Height, cm	Final Water Height, cm	Time to Air Ingestion, sec	Time to Final Height, sec
2.2	15.2	26.7	17.8	2.5	21.6	88.1
2.2	10.2	16.2	12.4	2.5	11.5	56.7
3.2	25.4	33.5	28.4	2.5	9.2	63.3
3.2	20.3	27.9	24.8	2.5	6.5	49.5

\* Test liquid was water; all heights measured above puncture centerline.

the water stream broke up into droplets which rose to the surface for the carbon tetrachloride tests, or sank to the tank bottom for the isopropyl alcohol tests, or mixed with the tank liquid for the aqueous glycerine tests. The water ingestion continued thereafter until, for the carbon tetrachloride and isopropyl alcohol tests, the level of water in the tank covered the puncture. For the aqueous glycerine tests, the density of the tank liquid continuously decreased as a result of mixing with the ingested water, and the outflow decreased steadily; however, even after 45 minutes, some outflow was still observed. (The aqueous glycerine tests were conducted to study the basic phenomena of ingestion more completely; however, the venting rate model does not include any provisions for mixing of the ingested water with the cargo.) Table III.3 summarizes the results of the tests using carbon tetrachloride and isopropyl alcohol. For all these tests, the tank was submerged in water to a depth 7.6 cm (3 in.) above the centerline of the puncture, and only the smaller puncture (2.2 cm) was used.

The tank vacuum pressure during the carbon tetrachloride test slowly decreased as water was ingested, with the net effect being that the total pressure (vapor space + liquid head) at the puncture remained fairly constant. At the instant that water was first ingested, for example, the total pressure at the puncture was 11.4 cm of water, or just slightly greater than the hydrostatic head of water outside the tank (7.6 cm of water); as the carbon tetrachloride reached the top of the puncture, a point where the water level in the tank was 27.6 cm above the puncture, the total pressure at the puncture was 8.5 cm of water. It can be concluded that the mass flow rates of the water and the cargo adjusted themselves to maintain this constancy of pressure, and the analytical model of water ingestion for a cargo heavier than water is based upon this conclusion.

TABLE III.3. WATER INGESTION STUDIES

III.3a Carbon Tetrachloride,  $\rho = 1.59 \text{ g/cm}^3$ 

Liquid Height, cm	Tank Pressure cm of H <sub>2</sub> O Vacuum	Time, sec
27.9	33.0	~ 0
22.9	30.5	110
17.8	27.9	245
12.7	24.9	380
7.6	22.1	510
2.5	20.3	635
top of puncture	19.1	665

At end of test, water height in tank = 27.6 cm

III.3b Isopropyl Alcohol,  $\rho = 0.783 \text{ g/cm}^3$ 

Liquid Height, cm	Tank Pressure cm of H <sub>2</sub> O Vacuum	Time, sec
27.9	12.7	~ 0
27.3	12.7	613

At end of test, water height in tank = 5.6 cm, or level with puncture top.

For the isopropyl alcohol tests, the vapor space pressure remained nearly constant, as did the level of the isopropyl alcohol in the tank. The ingested water, which sank to the bottom of the tank, appeared to displace the tank liquid such that the inflow and outflow volumetric flow rates always remained equal, and neither the tank pressure nor the cargo liquid height varied greatly. In effect, all the cargo below the top of the puncture was discharged and replaced by water. (The slight drop of 0.6 cm in the level of the isopropyl alcohol may be due to its solubility in water, which is small but nonzero.) The analytical model for water ingestion into a tank containing a cargo lighter than water is based upon this conclusion about the flow rates.

### III.2.3 Test Series No. 3: Discharge of Volatile Cargos

The final series of tests were designed to evaluate such thermodynamic phenomena for volatile cargos as liquid stratification, finite evaporation rates, and two-phase flow, and to obtain quantitative data for verification of the venting rate model.

A closed rectangular tank 30.5 cm by 30.5 cm (12 in. by 12 in.) and 61.0 cm (24 in.) high, made of 0.63 cm (0.25 in.) thick aluminum plate, with a total capacity of 56.6 liters (15 gallons), was used for these tests; see Figure III.3. The tank exterior was insulated by fiberglass 5 cm (2 in.) thick to approximate adiabatic test conditions. Two 600-watt heating tapes were attached to the wall to bring the simulated cargo to the desired temperature before a test. A small exterior pump continuously circulated the liquid in the tank during the pretest period to prevent any initial stratification; the spatial variation of liquid temperature was always kept below 0.27°C (0.5°F) prior to a test by this procedure.

Two simulated punctures, 0.32 cm (0.125 in.) and 2.2 cm (0.875 in.) diameter, were drilled into one wall near the vertical centerline and 7.6 cm (3 in.) above the bottom; most of

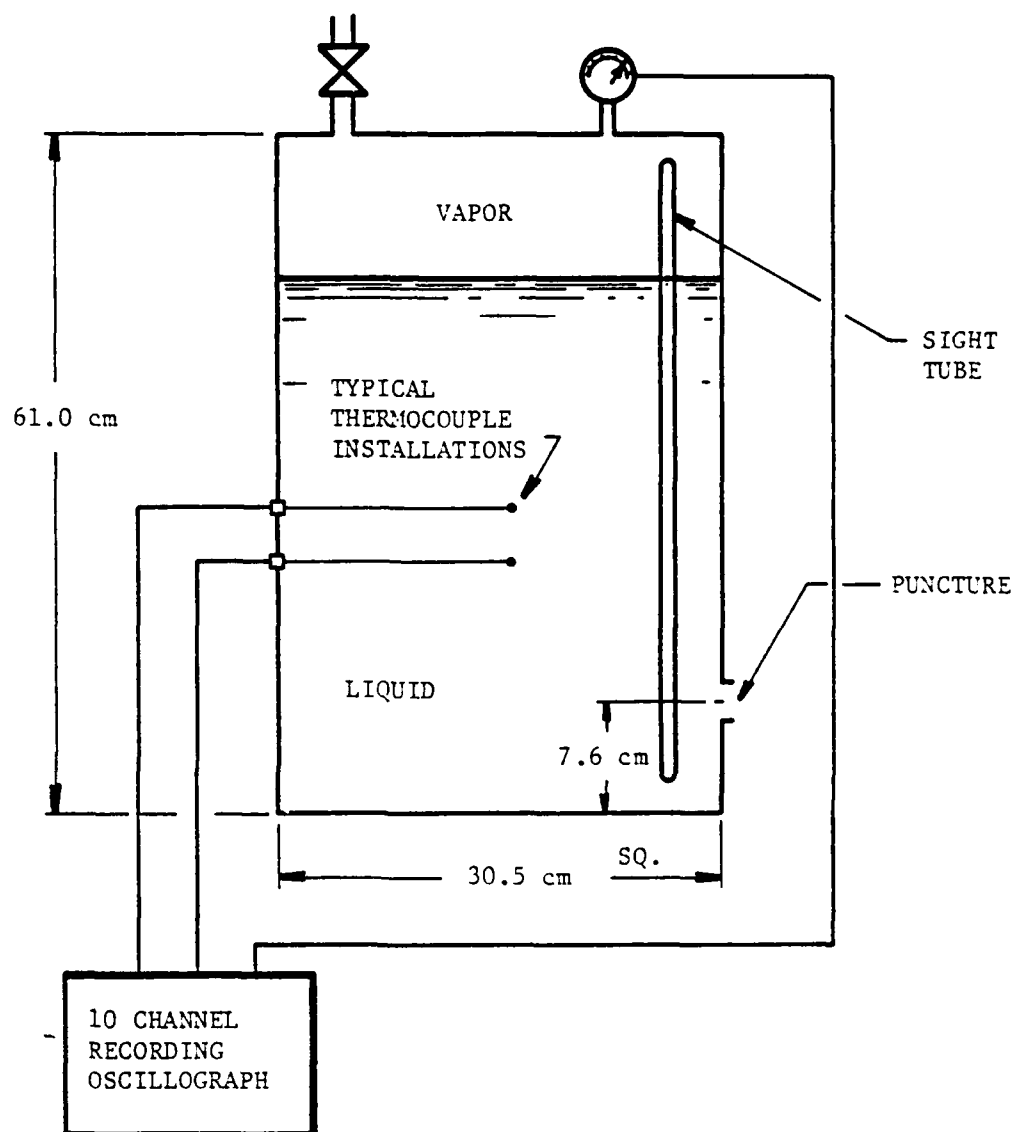


FIGURE III.3. INSULATED AND HEATED TEST TANK FOR DISCHARGE TESTS OF VOLATILE LIQUIDS



the testing was done with the larger puncture. A valve mounted on the tank top was used to vent vapor during the preliminary heating, and for some tests, as a simulated relief valve. Tank liquid levels were measured visually from a graduated sight tube, marked along each 2.54 cm (1.0 in.) of its length; in some tests, these measurements were verified, with excellent correlation, by thermocouple measurements as explained below.

The tank was instrumented with 20 Type J bayonet thermocouples inserted in one wall through drilled holes and extending to the vertical centerline of the tank; the vertical spacing of the holes was 2.54 cm. The vapor space pressure was measured with a diaphragm type of transducer (Validyne DP7). The thermocouple and pressure measurements were continuously recorded on a light beam, strip chart oscillograph (Honeywell Model 1858 Visicorder). A stop watch was used to measure discharge times; these measurements were verified by the timing marks on the strip chart.

Isopentane and methylene chloride (also called dichloromethane), both of which have boiling points slightly above room temperature at atmospheric pressure, were used in these tests as simulated volatile cargos. The properties of these liquids are shown in Table III.4. (Thermodynamic data on methylene chloride are incomplete and somewhat contradictory; thus, these tabulations are not recommended for the Chemical Properties File without further verification.)

Test Preparations. Preliminary testing revealed that obtaining reproducible data for venting of volatile liquids would be difficult, primarily because of the difficulty in controlling the initial air content of the vapor space. Any small amount of air mixed with the vapor in the vapor space would result in an initial tank pressure that was larger than would be expected on the basis of saturation conditions for a pure vapor. This, in turn, would cause a rapid decrease in the tank pressure during venting. That is, the partial pressure of

TABLE III.4. SATURATED THERMODYNAMIC PROPERTIES  
OF ISOPENTANE AND METHYLENE CHLORIDE

$P_v$  = saturation pressure (KPa);  $C_{PL}$ ,  $\bar{C}_{PV}$  = specific heats of saturated liquid and vapor (cal/g-°C);  $\lambda$  = latent heat of evaporation (cal/g);  
 $\rho_L$  = liquid density (g/cm<sup>3</sup>);  $Z$  = vapor compressibility; and  $T$  = temperature (°C).

a. ISOPENTANE (molecular weight = 72.147) [14]

$$\log_{10} P_v = 5.9666 - 1045.87/(T + 236.18)$$

$$\rho_L = 0.6405 - 1.00255 \times 10^{-3}(T)$$

$$Z = 0.984 - 1.126 \times 10^{-4}(T) - 2.182 \times 10^{-6}(T^2)$$

$$C_{PL} = 0.5218 + 1.0584 \times 10^{-3}(T)$$

$$\bar{C}_{PV} = 0.360 + 2.12 \times 10^{-4}(T)$$

$$\lambda = [65.2257 - 7.4638 \times 10^{-2}(T) - 1.864 \times 10^{-4}(T^2)] \frac{(T + 273.2)^2}{(T + 236.18)^2} - 0.57530 P_v (T + 273.2)/\rho_L (T + 236.18)^2$$

b. METHYLENE CHLORIDE (molecular weight = 84.93) [15]

$$\log_{10} P_v = 5.4971 - 822.48/(T + 194.92)$$

$$\rho_L = 1.35 - 1.75 \times 10^{-3}(T)$$

$$Z = 0.9995 - 1.588 \times 10^{-3}(T) + 8.505 \times 10^{-6}(T^2)$$

$$C_{PL} = 0.285 + 3.50 \times 10^{-4}(T)$$

$$\bar{C}_{PV} = 0.129 - 1.651 \times 10^{-3}(T)$$

$$\lambda = [44.26 - 7.032 \times 10^{-2}(T) + 3.766 \times 10^{-4}(T^2)] \frac{(T + 273.2)^2}{(T + 194.92)^2} - 0.45242 P_v (T + 273.2)/\rho_L (T + 194.92)^2$$

the constant mass of air decreased in accordance with the increase in vapor space volume as the liquid discharged. The partial pressure of the vapor, on the other hand, remained more nearly constant since its mass was increased by evaporation from the liquid surface at a rate that tended to compensate for the increase in vapor space volume. Depending upon the quantity of air initially contained in the vapor space, the drop in the vapor space pressure would lead to air ingestion through the puncture fairly quickly for some tests, while for other tests, with seemingly identical conditions, air ingestion would occur only near the end of the discharge, if at all. After this was realized, great care was taken to eliminate all the air from the vapor space, by the following procedure.

The tank was filled with liquid from the main reservoir until liquid flowed out the valve in the top of the tank. The valve was then closed, the filling halted, and the liquid heated until a small positive pressure existed. The valve was then opened to vent a liquid/vapor mixture, all the while continuing to heat the liquid to maintain a positive pressure in the tank and thus prevent air from entering through the valve. The boil-off of liquid was continued until the desired liquid level in the tank was obtained. The valve was then closed. The heating and the circulating of the liquid throughout the tank interior by the pump were continued, however, until the desired temperature and pressure conditions were obtained. Liquid (or vapor) discharge was then started by manually removing a rubber cork from the simulated puncture. This procedure usually led to reproducible tests, although some variation was still encountered occasionally, apparently because of air that became dissolved in the liquid isopentane and methylene chloride in their reservoirs. Altogether, nearly 40 tests were needed to obtain reliable venting data for each of the desired sets of initial conditions.

Liquid Stratification. Several of the tests were designed to evaluate the assumption that venting of the liquid would cause only a negligible degree of stratification.

Figures III.4 and III.5 show typical results of the thermocouple measurements for fairly rapid discharges of isopentane and methylene chloride through the 2.2-cm puncture. Some stratification is evident in the liquid near the liquid-vapor interface as a consequence of the heat transfer needed to provide the latent heat of vaporization, and there is a  $1^{\circ}$ - $2^{\circ}$ C temperature drop from the liquid to vapor. However, this small amount of stratification has a negligible effect on the energy content of the cargo, and the assumption that the temperature is uniform is acceptable for the purposes of formulating the energy balances, as will be shown later by the test and model comparisons. For slower discharge rates through the 0.32-cm puncture, stratification was even less noticeable.

The thermocouple strip-chart records, which clearly displayed the  $1^{\circ}$ - $2^{\circ}$ C temperature decrease at the liquid/vapor interface, provided a convenient check on the sight tube measurements of liquid level. The comparison of these two independent determinations was excellent.

Evaporation Rate. Figures III.6 and III.7 give typical results of vapor space pressure and liquid level for liquid discharges, for the same two tests given in Figures III.4 and III.5. The measured vapor space pressure decreased rapidly upon opening the puncture, and thereafter decreased more slowly. Also shown is the calculated saturation pressure of the liquid, based on the average liquid temperature estimated from the thermocouple records and on the liquid surface temperature; it can be seen that these two pressures differ very little, a further confirmation of the negligible effect of stratification. However, the actual vapor space pressure is significantly less than the saturation pressure, as it should be for the reasons discussed in Section II.3. A drop in vapor space pressure was observed even for the much slower venting through the 0.32-cm puncture, but the magnitude of the drop was much smaller, as shown for a typical case in Figure III.8.

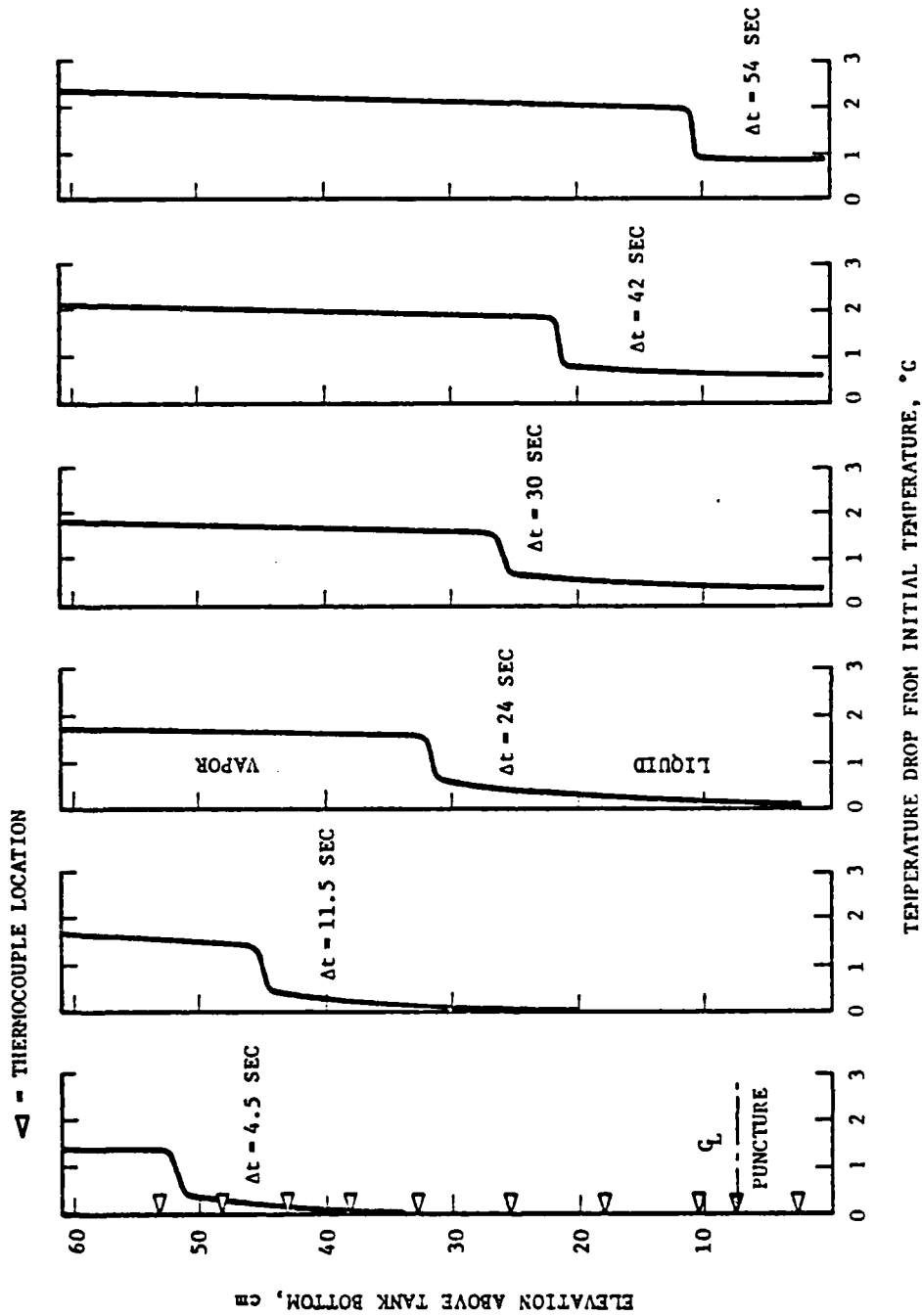


FIGURE III.4. TEMPERATURE STRATIFICATION FOR ISOPENTANE TEST NO. 6.  
 $z_{L1} = 56.2 \text{ cm}$ ;  $T_1 = 29.2^\circ\text{C}$ ;  $P_{T1} - P_{\text{atm}} = 84 \text{ cm H}_2\text{O}$

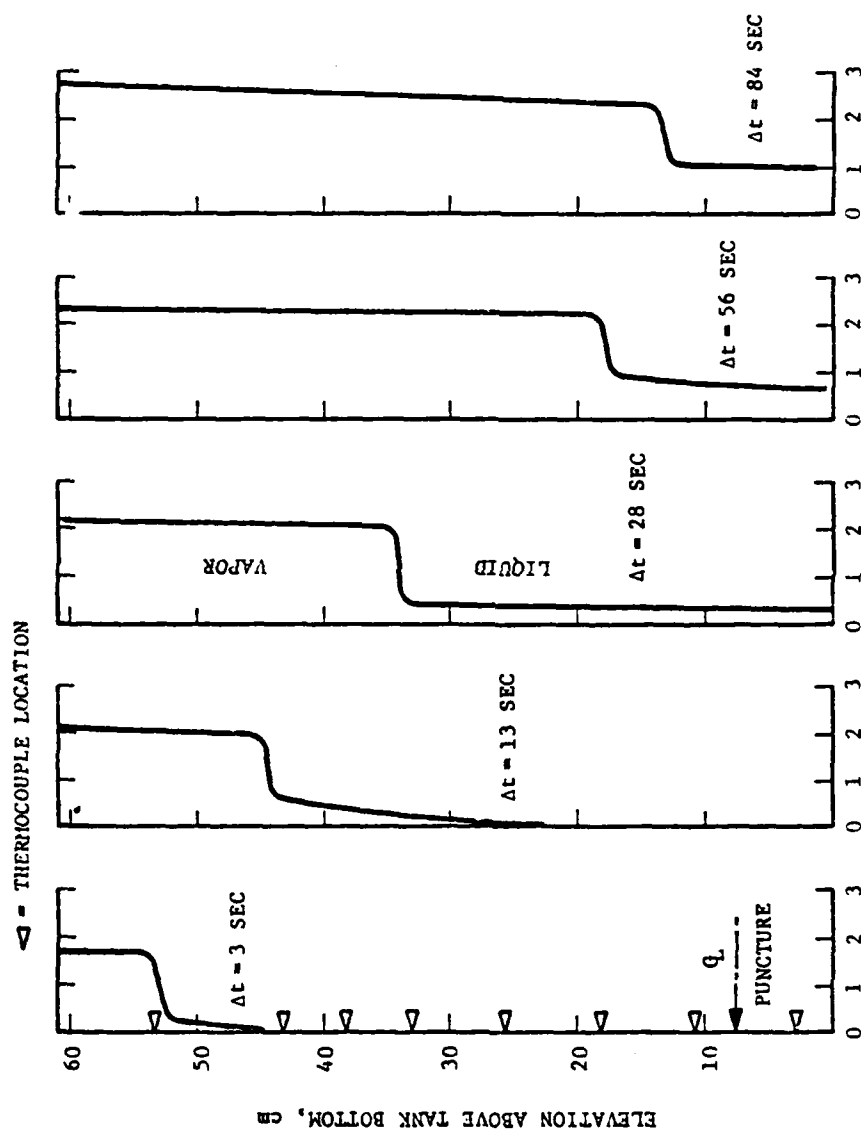


FIGURE III.5. TEMPERATURE STRATIFICATION FOR METHYLENE CHLORIDE TEST NO. 18.  
 $z_{Li} = 54.6$  cm;  $T_i = 41.4^\circ\text{C}$ ;  $P_{T1} - P_{atm} = 61$  cm  $\text{H}_2\text{O}$ .

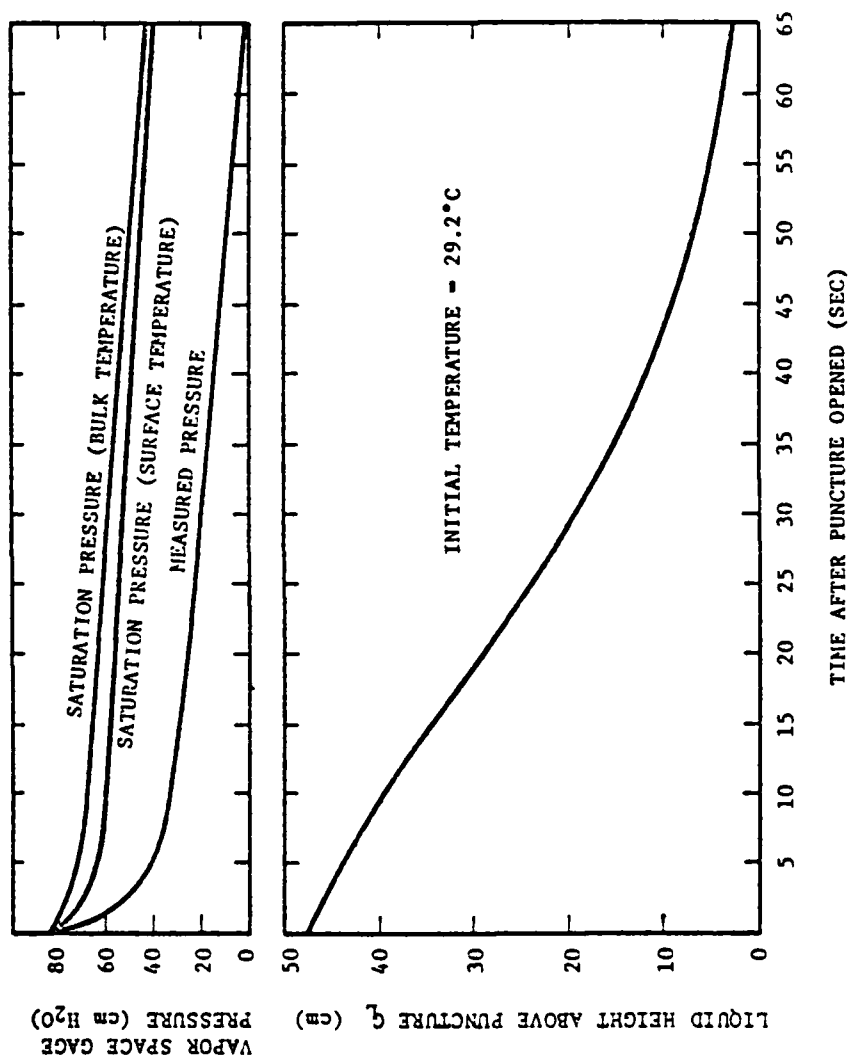


FIGURE III.6. DISCHARGE OF ISOPENTANE; PUNCTURE DIAMETER = 2.2 cm.  
TEST NO. 6.

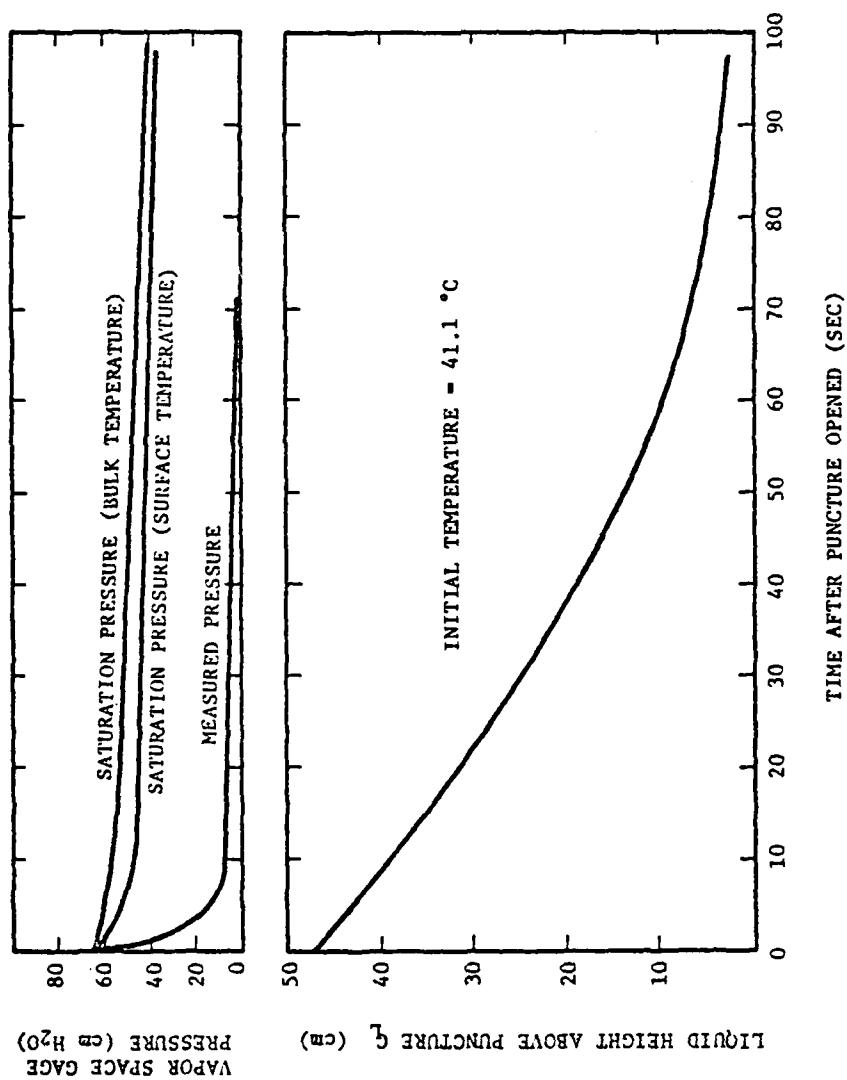


FIGURE III.7. DISCHARGE OF METHYLENE CHLORIDE; PUNCTURE DIAMETER = 2.2 cm.  
TEST NO. 18.



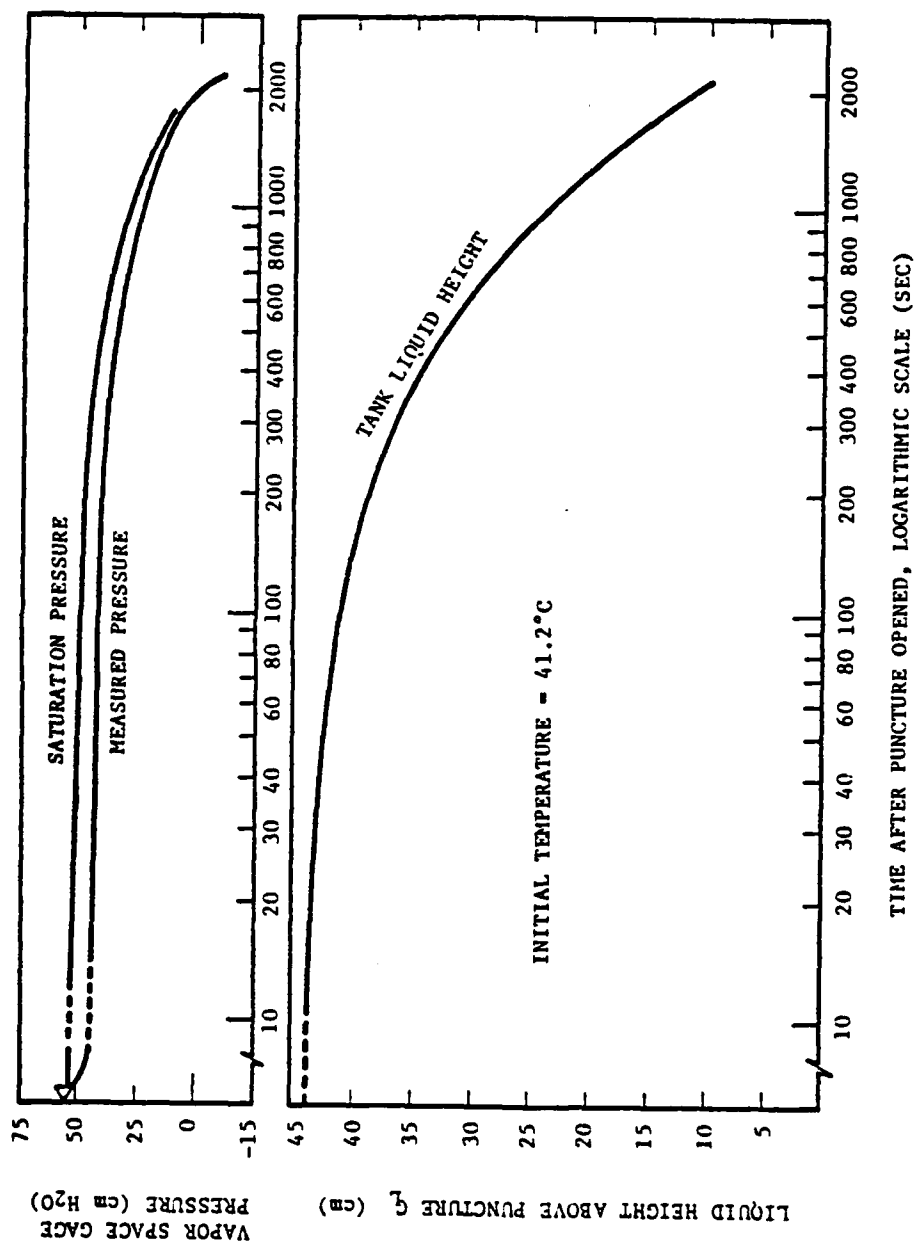


FIGURE III.8. DISCHARGE OF METHYLENE CHLORIDE;  
PUNCTURE DIAMETER = 0.32 cm. TEST NO. 22

The increase in the mass of vapor contained in the vapor space for these tests was computed by using the measured vapor space pressure and temperature records in conjunction with the measured liquid levels (i.e., vapor space volumes). The evaporation rate model described earlier in Section II.3 was also used to predict the evaporated mass, since  $W_{e,sat}$  could be computed from the temperature measurements of the liquid, using the known saturation pressure vs temperature relation for the tested liquid to compute the needed saturation pressure. Comparisons of the model predictions (integrated with respect to time to give vapor mass change rather than evaporation rate) to the tests are shown in Figure III.9. As can be seen, the comparison is very close. These kinds of test and model results were used, in fact, to deduce the best value of the empirical constant (1.9) in equation (II.12). Predictions of the model given by equation (II.11) are also shown; these comparisons are much poorer.

It ought to be mentioned that there is some uncertainty in the accuracy of the adiabatic behavior of the liquid late in each test. The large thermal inertia of the test tank probably moderated the decrease of the liquid temperature somewhat, so that both the true "adiabatic" liquid temperature and, therefore, the saturation pressure should actually be slightly less near the end of the tests than are shown in Figures III.4 through III.8.

Two-Phase Flow. The tank liquid level data as a function of time were used both to evaluate the implicit assumption made in AMSHAH [1] that two-phase flow effects were negligible, and to determine the best correlating value of the empirical factor  $\beta$  in the present two-phase flow model.

For the test results shown in Figure III.8, the average discharge rate calculated from the test data is 23.2 g/sec for the 130 seconds required for the liquid level to drop 2.54 cm from an average level of 34.3 cm above the puncture centerline. Since the tank gage pressure at this time was

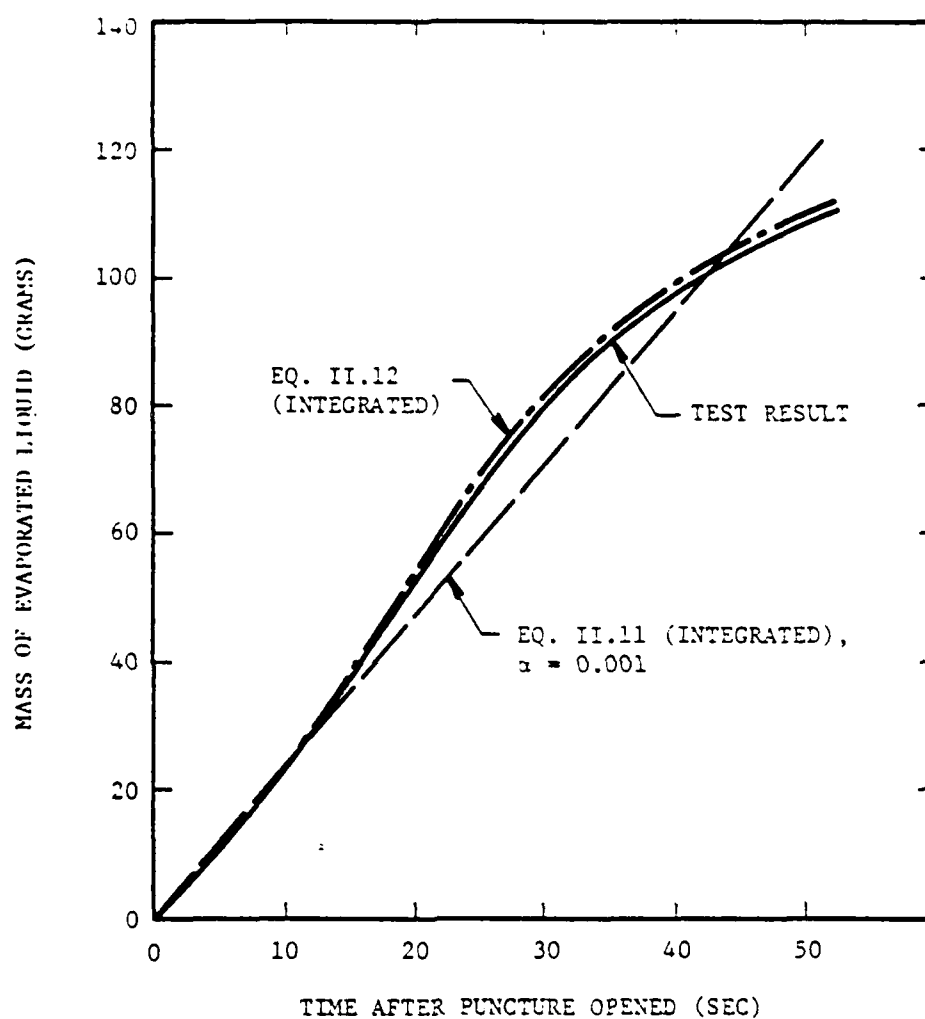


FIGURE III.9. COMPARISON OF PREDICTED AND MEASURED EVAPORATION.  
ISOPENTANE TEST NO. 6.

36.8 cm of water, the flow model used in [1] predicts that the discharge rate should be

$$W_{Lo} = 1.273 \frac{g}{cm^3} * 0.07917 cm^2 * C_D * \left\{ 2 * 980.5 \frac{cm}{sec^2} * \left[ \frac{36.8}{1.273} + 34.3 \right] cm \right\}^{1/2} \quad (III.3)$$

or 35.6  $C_D$  g/sec. The discharge coefficient required by this model to agree with the measured  $W_{Lo}$  is 0.65, but the actual  $C_D$  (as determined by prior testing) is 0.73. The model therefore overpredicts the test results by 12%. Likewise, for the test data shown in Figure III.4, the single-phase flow model predicts a discharge rate of 708 g/sec (with  $C_D = 0.73$ ) at the instant when the tank pressure is 30.5 cm  $H_2O$  and the liquid level is 35.5 cm, although the discharge rate calculated from the test data is about 600 g/sec; thus, the model also overpredicts this test by about 16%.

All the tests were analyzed at representative points to determine the best correlating value of  $\beta$  for the two-phase flow model (equations (II.7a) and (II.7b)). The best average value of  $\beta$  was found to be 0.12, and there was little test-to-test variation. For example, the discharge rates predicted by the model with  $\beta = 0.12$  for the two tests described above are 23.7 g/sec and 636 g/sec, respectively, both in excellent agreement with the tests.\*

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\* For the methylene chloride test shown in Figure III.3, the test measurements at the indicated time were  $P_T = 101.60$  KPa,  $P_{atm} = 98.0$  KPa,  $T = 41.5^\circ C$ , and  $P_T - P_V = -7.5$  cm  $H_2O = -0.75$  KPa. The corresponding calculated liquid properties are  $\rho_L = 1.278$  g/cm<sup>3</sup>,  $T_s = 39.68^\circ C$ ,  $\lambda_s = 74.6$  cal/g,  $v_{vs} = 300.0$  cm<sup>3</sup>/g, and  $C_{pL}$  (average) = 0.299 cal/g-°C. From eq. (II.7b) with  $\beta = 0.12$ ,  $v_{Lo} = 1.036$  cm<sup>3</sup>/g (or  $\rho_{Lo} = 0.965$  g/cm<sup>3</sup>). The discharge velocity (the radical in eq. (II.7a) is 425 cm/sec. Thus,  $W_{Lo} = 0.07917 * 0.73 * 425/1.036 = 23.7$ . (Footnote continued on next page)

Vapor Discharge. Several tests were conducted in which only vapor was discharged, using the valve in the top of the tank as a simulated puncture. There was an initial blowdown of the tank pressure to very near atmospheric pressure, followed by a much slower discharge as the liquid remaining in the tank evaporated. These tests were not analyzed in detail for two reasons: test durations were so long that the tank processes could not validly be considered as adiabatic, and the test pressure instrumentation, which was designed primarily for liquid discharges, could not resolve the small pressure difference between the tank vapor space and the atmosphere which would be needed to make discharge rate comparisons with the model. It was noted, however, that the temperature of the vapor space was consistently much lower than the liquid as a result of the initial blowdown; a typical temperature difference was 17°C.

Simultaneous Liquid and Vapor Discharge. No unusual or unexpected phenomena were encountered during these tests. The vapor space pressure temperature again was nearly 17°C less than the liquid. These tests were not analyzed in detail for the same reasons given for the vapor discharge tests.

Air Ingestion. In a few tests, air was trapped in the vapor space deliberately during the tank filling process. In this way, a partial vacuum could be caused almost immediately after the start of the liquid discharge, and air was ingested through the puncture during most of the test. The air ingestion began at a tank pressure and liquid level that was in agreement with equation (III.2). No unusual or unexpected phenomena were observed, so the tests were not analyzed further.

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\* For the isopentane test shown in Figure III.4,  $P_T = 100.73$  KPa,  $P_{atm} = 97.74$  KPa,  $T = 28.9^\circ\text{C}$ ,  $P_T - P_v = -42.0$  cm  $\text{H}_2\text{O} = -4.11$  KPa,  $c_L = 0.611$  g/cm<sup>3</sup>;  $T_s = 26.83^\circ\text{C}$ ,  $\lambda_s = 84.7$  cal/g,  $v_{vs} = 346$  cm<sup>3</sup>/g, and  $C_{pL}$  (average) = 0.551 cal/g-°C. The discharge specific volume = 2.181 cm<sup>3</sup>/g and the discharge velocity = 490 cm/sec. Thus,  $W_{Lo} = 3.879 \times 0.73 \times 490/2.181 = 636$  g/sec.

#### IV. VALIDATION OF MODEL

By using results of typical tests, all the empirical constants in the venting rate model have been determined by the methods discussed in Sections II and III. The predictive reliability of the overall model and its computerized solution are demonstrated in this section by comparison to results of all the test series.

##### IV.1 Discharge of Nonvolatile Liquids into Air; Operable Relief Valve

The discharge of a nonvolatile liquid into air was studied in Test Series No. 1 as a means of determining the variation of discharge coefficients with the shape of the puncture. With  $C_D$  supplied as input to the computerized model, the total time required to discharge the cargo initially above the puncture centerline was computed. The predicted venting times were virtually identical to the test results. For example, the predicted time was 66.5 seconds to discharge enough liquid to drop the level from 64.8 cm to 14.6 cm above the puncture centerline for the tank shown previously in Figure II.1. The average discharge time for a set of three tests was also 66.5 seconds.

##### IV.2 Discharge of Nonvolatile Liquids: Air Ingestion

Experimental results for liquid discharges in which air was ingested (the vacuum relief valve was jammed) were presented in Table II.2. All these data were analyzed in conjunction with predictions of the venting rate model to determine the best correlating value of the ingested air bubble volume factor. This value turned out to be 5.6; the implied average bubble diameter is therefore about 1.8 times the average open dimension of the puncture, which is in good agreement with the visual observations of bubble size.

Comparisons of a typical test of Test Series No. 2 (where the simulated cargo was water) and the predictions of the model are given below. In the test, the simulated relief valve was manually adjusted to maintain a pressure in the vapor space vacuum of 15.25 cm of water, so the relief set-point used in the prediction was also input as this value (i.e., 1.494 KPa).

INITIAL HEIGHT ABOVE PUNCTURE, cm		TIME TO INGESTION START, sec		HEIGHT AT INGESTION START, cm		TOTAL DISCHARGE TIME, sec	
Test	Model	Test	Model	Test	Model	Test	Model
26.7	26.7	21.6	22.3	17.8	21.3	88.1	86.0

As can be seen, the model-to-test comparison is excellent, both with respect to when ingestion begins and to the total time required to discharge the liquid.

#### IV.3 Discharge of Nonvolatile Liquids: Water Ingestion

Experimental results for the discharge of liquids when water was ingested were given in Table III.3. These tests were analyzed in conjunction with the venting rate model to determine the best correlating value of the empirical parameter  $K$  of the model (see, for example, equations (II.17)); the best overall correlation for all the tests was found to be given for  $K = 83$ .

When the simulated cargo was heavier than water, the model predicted the tests excellently, as shown in the table in the next page. In the tests, the relief valve was kept closed to create a vacuum pressure such that water was ingested almost immediately upon starting the discharge. In the model, nearly the same vacuum was predicted, with water being ingested after 0.8 second. For both the test and the model, the external water level was 12.7 cm above the puncture centerline. As can be seen, the model and the test are well correlated, including the trend and magnitude of the vapor space vacuum as a function of time.

INITIAL HEIGHT ABOVE PUNCTURE, cm		INITIAL VACUUM, cm H <sub>2</sub> O		FINAL VACUUM, cm H <sub>2</sub> O		TOTAL DISCHARGE TIME, sec	
Test	Model	Test	Model	Test	Model	Test	Model
27.9	27.9	33.0	28.9	17.8	13.1	665	641

By referring to equations (II.17), it can be seen that the model prediction of the discharge rate depends on  $P_{Ti}$  and  $Z_{Li}$ , the values of tank pressure and liquid height at the time increment just before water ingestion starts. For a given set of initial conditions, the numerical incrementing scheme will always arrive at these same values of  $P_{Ti}$  and  $Z_{Li}$ , and will predict the same values of  $P_{T,Eq.}$  and  $Z_{L,Eq.}$ . For slightly different initial conditions,  $Z_{Li}$  and  $P_{Ti}$  would also be slightly different. On a percentage basis, however, the changes in  $P_{Ti} - P_{T,Eq.}$  and  $Z_{Li} - Z_{L,Eq.}$  would be substantial. Thus, the discharge rate prediction might change significantly for, say, a small change in the initial liquid level, although the actual discharge rate would be expected to change only slightly. Consequently,  $K$  would also need to be changed to bring the predictions into agreement, which is an undesirable result. To determine whether the predictions are in fact sensitive to small changes in the initial conditions, the initial liquid level in the computations was decreased by 0.1 cm, to 27.8 cm above the puncture. The corresponding change in  $Z_{Li} - Z_{L,Eq.}$  was fifty percent. However, the predicted total discharge time changed by only 3 seconds, to 638 seconds, and the changes in the computed initial and final vacuum pressures were also negligible. The reason for the lack of sensitivity is that the slight increase of the initial vapor space volume resulted in a slight increase of  $P_{Ti}$ , which increased  $P_{Ti} - P_{T,Eq.}$  just enough to cancel out the fifty percent decrease of  $Z_{Li} - Z_{L,Eq.}$ . It is concluded that there is very little sensitivity to slight changes in the initial conditions. The corollary to this conclusion is that the value  $K = 93$  is general and not specific



to the particular values of  $Z_{Li}$  and  $P_{Ti}$  that result from a given set of initial conditions or the numerical solution technique.

Correlation of the model to the test results for cargos lighter than water was not as good. For example, for the isopropyl alcohol tests described in Section III, the measured time required for the ingested water level to reach the top of the puncture was 613 seconds, at which time the discharge ceased. The computed results were that 194 seconds are required to ingest enough water to cover the puncture; moreover, this prediction was not very sensitive at all to the value of  $K$  used in the range of 83 to 1000, so the value  $K = 83$  selected from the tests with cargos heavier than water is also recommended for cargos lighter than water. The primary reason for the discrepancy in the model predictions is believed to be as follows: the conditions for water ingestion were met at the end of the first mass discharge incrementation cycle of the model. This meant that  $P_{Ti}$  (the pressure at the iteration preceding the start of ingestion) was selected as the initial vapor space pressure; that is,  $P_{Ti}$  was equal to atmospheric pressure. This value of tank pressure is the pressure that is subsequently maintained constant by the water ingestion model, rather than the 12.7 cm of water vacuum observed in the tests. The result is that the predicted discharge is too large. For a ship tank, the static head of the cargo above the puncture would be much larger than was used in these tests. Thus, the model would predict that a definite volume of liquid would be discharged before a vacuum large enough to cause water ingestion would be created. Therefore, the difficulty encountered in these predictions of the discharge time for the small tanks used in the present tests is probably not a serious one.

#### IV.4 Discharge of Volatile Liquids

Experimental results for the discharge of a volatile liquid were discussed in Section III. By analyzing the test

results, an evaporation rate model, as described in Section II.3, was formulated and the empirical constants of the model were determined. The empirical factor,  $\beta$ , in the model of two-phase flow was determined similarly. The entire venting rate model for volatile cargos is now compared to the tests.

Figure IV.1 shows a comparison between the predictions of liquid height, evaporated mass, and vapor space pressure, and representative results of a typical isopentane test taken from the strip-chart recordings. The predicted total time to discharge the cargo is about 63 seconds, compared to about 67 in the test; equally good agreement is apparent for the other parameters of interest. In particular, the vapor space pressure decreased with time in the same way as did the test; this indicates that the evaporation rate model includes the appropriate physical phenomena, even though the total mass evaporated into the vapor space is slightly overpredicted. Note that  $P_T - P_{atm} \approx 0$  over most of the test and that  $P_{sat}$  always remains noticeably larger than  $P_{atm}$ . Consequently, basing the predictions on  $P_{sat}$  (as AMSHAH does) would overpredict the discharge rate, especially when the hydrostatic head was small, and might cause the onset of air or water ingestion, when they occur, to be missed entirely.

The discharge rate near the start of the test is slightly overpredicted because of the slight overprediction of the evaporation rate, and this causes the prediction of tank liquid height to be less than the measured height throughout the rest of the discharge time. Note, however, that the discharge rate (i.e., the slope of the liquid height curve) is well predicted except for this initial discrepancy. Air ingestion was predicted to occur during the last 12 seconds of the discharge. The time at which air was first ingested in the test was difficult to determine since the tank was not transparent, but oscillations in the sight tube liquid level and in the pressure record became noticeable during the last 6 to 8 seconds, which indicate that air was being ingested for at least this period.

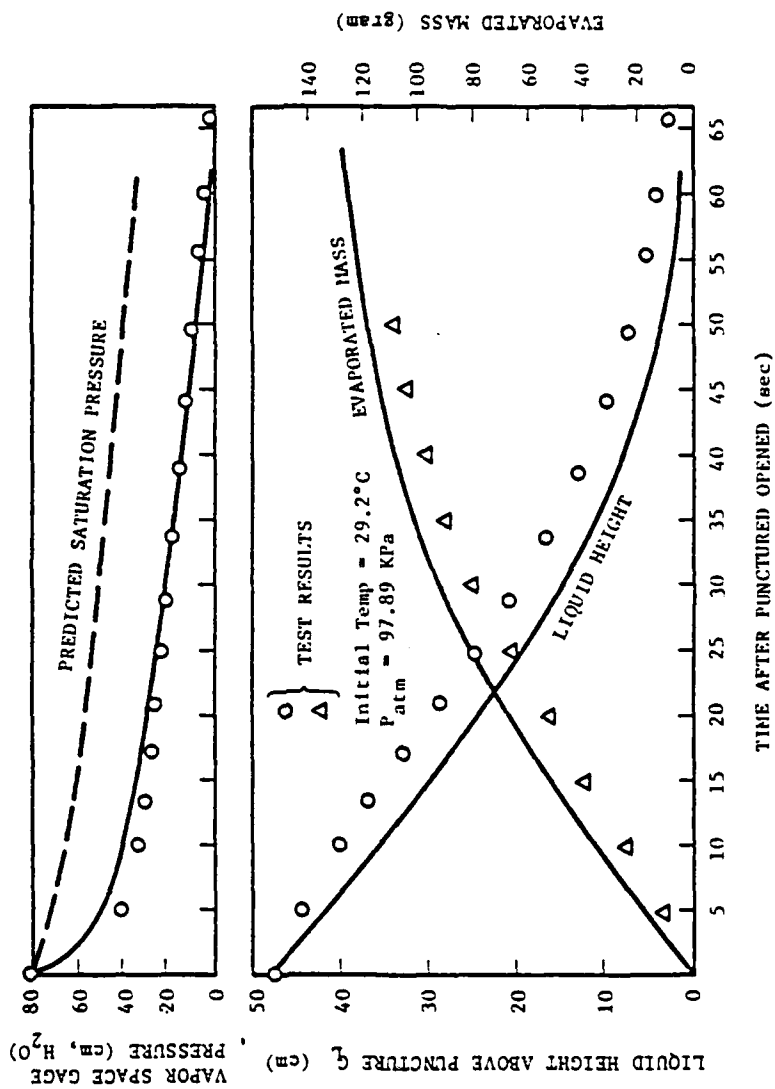


FIGURE IV.1.1. COMPARISON OF VENTING RATE MODEL AND TEST RESULTS FOR VENTING OF LIQUID ISOPENTANE; TEST NO. 6

Thus, the model and test are also in good agreement concerning air ingestion.

Figure IV.2 shows similar comparisons for a methylene chloride test. The predicted discharge time of 93 seconds compares well to the measured result of nearly 100 seconds. Predictions of the other phenomena also compare well, except that the magnitude of the initial rapid decrease of the vapor space pressure was underpredicted. This is perhaps the result in the test of the existence of a small quantity of air in the vapor space.

The initial temperature of the liquid for the predictions shown in both Figures IV.1 and IV.2 had to be adjusted slightly (less than  $0.5^{\circ}\text{C}$ ) in comparison to the test measurement in order to bring the initial vapor space pressure into agreement with the tests. The experimental measurements of temperature, vapor space gage pressure, and local atmospheric pressure (all of which enter into the comparison of predicted and measured gage pressures) could have been in error by enough to explain this adjustment of initial temperature, but it is also possible that the industrial quality liquids used in the tests may not have had exactly the same vapor pressure versus temperature relation that was used in the model predictions.

#### IV.5 Other Validation Checks

Data from other tests, such as venting of vapor of volatile cargoes, were not reliable enough to serve as checks on the validity of the model. The computer model has been exercised, however, for all the various kinds of possible discharges of liquid and gas. These results have been verified against the analytical model by (1) single-point hand calculations and (2) reasonableness of predictions. As far as could be determined by these limited checks, the computerized model accurately reproduces these kinds of discharges.

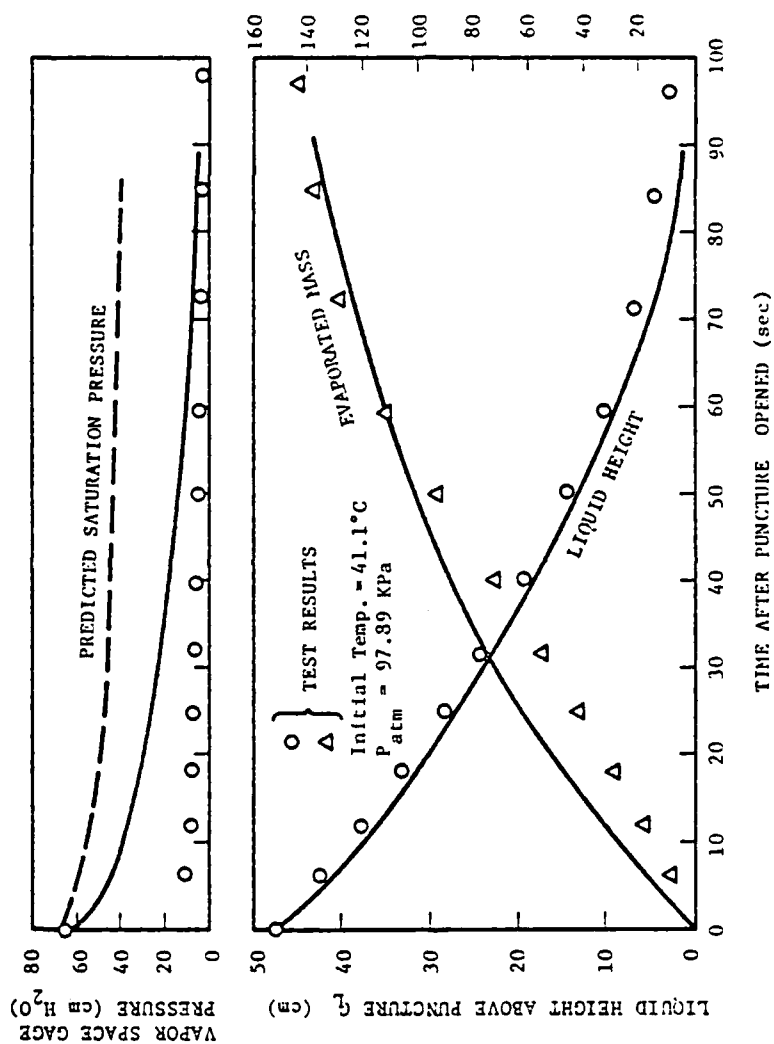


FIGURE IV.2. COMPARISON OF VENTING RATE MODEL AND TEST RESULTS FOR VENTING OF LIQUID METHYLENE CHLORIDE; TEST NO. 18

## V. CONCLUSIONS

The Venting Rate Model previously developed by others for the Hazards Assessment Computer System has been completely reformulated. All errors in the basic equations have been corrected, and the limitations imposed by inappropriate assumptions have been removed. The revised model now covers nearly every practical combination of puncture location, cargo properties, intermittent outflows, and relief valve actuations. In summary, the revised model incorporates the following new features:

- o punctures above or below the waterline;
- o relief valve actuations;
- o the possibility of air in the vapor space;
- o simultaneous liquid and gas discharges;
- o phase changes during the discharge of volatile cargos;
- o choked flow of gases and volatile liquids;
- o air or water ingestion; and
- o realistic effects of evaporation within the tank.

Four of the submodels of the Venting Rate Model contain empirical constants whose "best" values have been determined by comparison to experimental data. Three of the constants are related to observable physical phenomena--the air bubble size factor of 5.6 in the air ingestion model, the constant 1.9 in the tank cargo evaporation model, and the constant  $\beta = 0.12$  in the two-phase flow model. The fourth constant,  $K = 83$ , is used in the computerized solution of the water ingestion model. The relative importance of these constants in the predictions depends upon the type of cargo discharge encountered. In some cases the predictions are completely independent of the values of the constants. For example, when the relief valve is operable, air is admitted as needed to keep the vapor space pressure near atmospheric. Thus, air will not be ingested through

the puncture until the liquid discharge is nearly finished, and as a result the value of the bubble size factor is unimportant in the model predictions. Further, the vapor space pressure is controlled by the relief valve and not by evaporation into the vapor space, so the numerical constant in the evaporation model is likewise practically irrelevant. On the other hand, water can still be ingested, but only if the puncture is far enough below the waterline to overcome the hydrostatic head of the cargo. The onset of water ingestion will be considerably delayed, however, compared to a case when the relief valve is jammed shut. The effects of two-phase flow, and thus the value of  $\beta$ , are only important for discharges of cargos whose vapor pressure is greater than atmospheric and are carried in pressurized tanks. Otherwise the liquid discharge is predicted to be nearly incompressible, in agreement with observations. In any event, the two-phase flow model and the tank cargo evaporation model affect only the predicted time required for the discharge and not the total quantity of liquid cargo discharged. A faster discharge rate and a more conservative estimate of the hazard can be obtained by setting either or both of the empirical constants ( $\beta$  and 1.9) in these models to zero. The air and water ingestion models do influence the computation of the total liquid discharged, but the value of their empirical constants (the bubble size factor and  $K = 83$ ) only change the discharge rate. A conservative estimate of the hazard can be obtained by decreasing the value of  $K$  (now equal to 83) in the water ingestion model or by increasing the value of the bubble size factor (now equal to 5.6) in the air ingestion model; either of these changes will increase the predicted discharge rate.

Predictions of the computerized Venting Rate Model have been compared to scale-model test results for many kinds of discharges. Generally good correlation of the model to the tests was found. In one sense, these good correlations do not necessarily represent a true verification of the model because

part of the data used in making the comparisons was used to determine the empirical constants of the various submodels. Usually, model verification requires a comparison of the model to an independent set of data taken under different experimental conditions from those which are used to determine the empirical constants. Because of the constraints in the scope of the project, it was not possible to conduct a separate set of experiments for the purpose of model verification. Therefore, an alternative technique was used.

Each submodel containing an empirical constant was first fitted to the relevant test data to determine the constant. For example, in determining  $\beta$  of the two-phase flow model, test measurements of liquid temperature, liquid level, vapor space pressure, etc., were used to make predictions of mass discharge rates, using the flow model alone; these were then compared to the measured discharge rates alone. Note that actual test data were used to determine  $\beta$  rather than predictions of liquid temperature, vapor space pressure, and the other required submodel inputs made from exercising the entire Venting Rate Model. Since only the flow submodel was used to find  $\beta$ , it was considered an appropriate test of the entire Venting Rate Model to input the initial conditions of the same experiments to compare its output to the actual test results. The good correlation of the entire model to test data thus does show that the energy balances have been formulated correctly, that the interaction of the energy balance with tank cargo evaporation has been modeled realistically, that the interaction of the air ingestion model with the vapor space pressure model has also been incorporated realistically, and so on. It can be concluded, then, that the Venting Rate Model has been verified to a significant degree. However, we would recommend that further experimentation and subsequent model validation be carried out.

The Chemical Properties File has been reviewed and found to be satisfactory for nonvolatile cargos. For volatile cargos, however, several additional correlations are required:



the compressibility and specific heat of saturated vapor, and the latent heat of evaporation. It also appears that the accuracy of the saturation pressure correlations should be improved.

## REFERENCES

1. P. P. K. Raj and A. S. Kalelkar, "Assessment Models in Support of the Hazard Assessment Handbook," Report CG-D-65-74, Arthur D. Little, Inc., Cambridge, Mass., 1974. (NTIS AD 776-617)
2. N. A. Eisenberg, C. J. Lynch, R. M. Kumar, R. J. Bredina, and M. N. Durrani, "A Critical Technical Review of Six Hazard Assessment Models," Report CG-D-122-76, Enviro Control, Inc., Rockville, Md., 1975. (NTIS AD A035-599)
3. F. T. Dodge and R. E. White, "A Study to Experimentally Verify and Revise the Venting Rate Model of the Hazard Assessment Computer System and the Vulnerability Model. Interim Report: Reformulation of the Venting Rate Model," Contract DOT-CG-73623-A, SwRI Project 02-5295, Southwest Research Institute, San Antonio, Texas, Dec. 1976.
4. J. D. Parker, J. H. Boggs, and E. F. Blick, Introduction to Fluid Mechanics and Heat Transfer, Addison-Wesley Publishing Co., Reading, Mass., 1970.
5. A. H. Shapiro, The Dynamics and Thermodynamics of Compressible Fluid Flow, Vol. I, The Ronald Press Co., New York, 1953.
6. R. B. Bird, W. E. Stewart, and E. N. Lightfoot, Transport Phenomena, John Wiley, 1960.
7. D. C. Gray, "Solvent Evaporation Rates," American Industrial Hygiene Association Journal, 35, pp. 695-710, Nov. 1974.
8. J. P. Hirth and G. M. Pound, Condensation and Evaporation: Nucleation and Growth Kinetics, MacMillan Co., 1963.
9. G. B. Wallis, C. J. Crowley, and Y. Hagi, "Conditions for a Pipe to Run Full When Discharging Liquid into a Space Filled with Gas," Trans. ASME, J. Fluids Engineering, vol. 99, pp. 405-413, June 1977.
10. R. E. Henry and H. K. Fauske, "The Two-Phase Critical Flow of One-Component Mixtures in Nozzles, Orifices, and Short Tubes," Trans. ASME, J. Heat Transfer, 93, pp. 179-187, May 1971.
11. R. F. Tangren, C. H. Dodge, and H. S. Siefert, "Compressibility Effects in Two-Phase Flow," J. Applied Physics, 20, pp. 637-645, July 1949.

12. W. E. Baker, P. S. Westine, and F. T. Dodge, Similarity Methods in Engineering Dynamics. Theory and Practice of Scale Modeling, Hayden Books, Washington, D. C., 1973.
13. Anon., "Flow of Fluids Through Valves, Fittings, and Pipe," Tech. Paper No. 410, The Crane Co., Chicago, Ill., 1965.
14. Anon., Refrigerating Data Book, 5th ed., American Society of Heating, Refrigerating, and Air Conditioning Engineers, New York, 1942.
15. K. E. Starling, Fluid Thermodynamic Properties for Light Petroleum Systems, Gulf Publishing Co., 1973.
16. Handbook of Chemistry and Physics, 48th edition, The Chemical Rubber Co., 1967-68.
17. G. J. van Wylen and R. E. Sonntag, Fundamentals of Classical Thermodynamics, Table A.2, Thermodynamic Properties of Ammonia, John Wiley and Sons, 1973.
18. Handbook of Tables for Applied Engineering Science, 2nd edition, The Chemical Rubber Co., 1973.
19. T. Boublik, V. Fried, and E. Hala, The Vapour Pressure of Pure Substances, Elsevier Scientific Publishing Co., 1973.
20. F. Din, Thermodynamic Functions of Gases, Vol. 2, Butterworths Publishing Co., 1962.
21. Chlorine: Its Manufacture, Properties, and Use, ed. J. S. Sconce, Chapter 3., American Chemical Society Monograph Series, Reinhold Publishing Co., 1962.
22. Anon., "Methane," Cryogenic Engineering News, pp. 100-101, Jan. 1960.
23. T. Lee, S. Wyatt, S. Desai, and K. Chao, "Benedict-Webb-Rubin Equation of State for Methane at Cryogenic Conditions," Advances in Cryogenic Engineering, 14, pp. 49-54, Plenum Press, 1969.
24. Methanol Technology and Economics, ed. G. A. Danner. "A Review of Volumetric, Thermodynamic, and Other Physical Properties for Methanol," by P. T. Eubank, AIChE Symposium Series 93, 66, 1970.
25. J. F. Bailey, "Metastable Flow of Saturated Water," Trans. ASME, 73, pp. 1109-1116, Nov. 1951.

## APPENDIX A. DERIVATION OF EQUATIONS

A.1 Tank Energy Balance

The tank contents are assumed to be "well stirred," and the temperature of the gas and the liquid are assumed to be equal. Considering a control volume around the liquid shown in Figure II.1, the general well-stirred reactor energy equation is:

$$\dot{Q}_{cv} + \left[ \dot{m}_{in} \left( h_{in} + \frac{1}{2} V_{in}^2 + g z_{in} \right) \right] = \dot{W}_{cv} + \frac{dE_{cv}}{dt} + \left[ \dot{m}_{out} \left( h_{out} + \frac{1}{2} V_{out}^2 + g z_{out} \right) \right] \quad (A.1)$$

The work term  $\dot{W}_{cv}$  includes the  $P_T dV$  contribution of the moving boundaries. Equation (A.1) is specialized for adiabatic processes,  $\dot{Q}_{cv} = 0$ . The internal energy of the control volume is:

$$E_{cv} = M_L \left( u_L + \frac{1}{2} V_L^2 + g z_{Lcm} \right) \quad (A.2)$$

where  $u_L$  is the liquid internal energy,  $V_L$  is the average liquid velocity in the tank and  $z_{Lcm}$  is the vertical location of the liquid center of mass.

Stagnation enthalpies are defined as follows:

$$H_L = h_L + \frac{1}{2} V_L^2 = h_{Lo} + \frac{1}{2} V_{Lo}^2 \quad (A.3a)$$

$$H_V = h_V + \frac{1}{2} V_G^2 = h_{Vo} + \frac{1}{2} V_{Go}^2 \quad (A.3b)$$

$$H_A = h_{Ao} + \frac{1}{2} V_{Go}^2 \quad (A.3c)$$

$$H_{\infty} = h_A + \frac{1}{2} V_A^2 \quad (A.3d)$$

where the first two relations follow from the fact that the liquid (or the vapor) in the tank and in the discharge stream both have the same reservoir conditions. Since  $u_L = h_L - P_T v_L$ , equation (A.1) can be re-written as:

$$\begin{aligned} \frac{d}{dt} (M_L H_L + g Z_{Lcm}) - \frac{d}{dt} (P_T v_L) &= -A_T P_T \frac{dz_L}{dt} \\ &- W_e \left( H_V + \frac{1}{2} v_e^2 - \frac{1}{2} v_G^2 + g Z_L \right) - W_{Lo} (H_L + g Z_{Lh}) \end{aligned} \quad (A.4)$$

An energy equation for a control volume around the vapor space can be written similarly:

$$\begin{aligned} \frac{d}{dt} [M_V H_V + M_A H_A + (M_A + M_V) g Z_{Gcm}] - \frac{d}{dt} (P_T v_V) &= \\ A_T P_T \frac{dz_L}{dt} + W_e \left( H_V + \frac{1}{2} v_e^2 - \frac{1}{2} v_G^2 + g Z_L \right) + W_A (H_{\infty} + g Z_m) \\ &- W_{Vo} (H_V + g Z_{Gh}) - W_{Ao} (H_A + g Z_{Gh}) \end{aligned} \quad (A.5)$$

Here, the assumption has been made that the sum of the air and the vapor partial pressures is equal to the tank pressure, so that  $P_A M_A v_A + P_V M_V v_V$  can be set equal to  $P_A V_G + P_V V_G = (P_A + P_V) V_G = P_T V_G$ .

Equations (A.4) and (A.5) can be combined by using conservation of mass (e.g., equations (II.2)) to eliminate the  $dM_V/dt$  and  $dM_A/dt$  terms, and by using the definition of the latent heat of vaporization:

$$H_V = H_L + \lambda + (v_G^2 - v_L^2)/2 \quad (A.6)$$

The result is:

$$\begin{aligned}
M_L \frac{dH_L}{dt} + M_V \frac{dH_V}{dt} + M_A \frac{dH_A}{dt} &= \dot{V}_T \frac{dP_T}{dt} + W_A (H_{\infty} - H_A) \\
&+ \left[ \lambda + \frac{1}{2} V_e^2 + (V_G^2 - V_L^2)/2 \right] \left( \frac{dM_L}{dt} + W_{Lo} \right) \\
&+ g \left\{ W_A Z_T - W_{Lo} Z_{Lh} - (W_{Vo} + W_{Ao}) Z_{Gh} \right. \\
&\left. - \frac{d}{dt} [M_L Z_{Lcm} + (M_A + M_V) Z_{Gcm}] \right\} \quad (A.7)
\end{aligned}$$

According to the numerical examples given in [2], both the kinetic and potential energy terms are negligible during the discharge of a volatile substance. Thus, since  $dH/dt = dh/dt = C_p dT/dt$  when kinetic energy is negligible, equation (A.7) can be finally written as

$$\begin{aligned}
(M_L C_{PL} + M_V \bar{C}_{PV} + M_A C_{PA}) \frac{dT}{dt} &= \dot{V}_T \left( \frac{dP_T}{dt} \right) + \lambda \left( \frac{dM_L}{dt} + W_{Lo} \right) \\
&+ W_A C_{PA} (T_{air} - T) \quad (A.8)
\end{aligned}$$

which is equation (II.1) of the report.

## A.2 Flow Equations for Volatile Substances

Considering an isentropic discharge of liquid from a saturated, or equilibrium, initial state, to an exit pressure  $P_{\infty}$  corresponding to a saturation temperature  $T_s$ , where the fluid is a liquid-vapor mixture, the thermodynamic quality of the exit mixture is:

$$x = \frac{s_{out} - s_{L_{\infty}}}{s_{V_s} - s_{L_s}} = \frac{s_L - s_{L_{\infty}}}{s_{V_s} - s_{L_s}} \quad (A.9)$$

where  $s_L$  is the entropy of the saturated liquid in the tank. ( $s_{out} = s_{in} = s_L$  for an isentropic process). But

$$s_L - s_{L_s} = \int_{T_{\infty}}^T \frac{C_{PL} dT}{T} = C_{PL} \ln \frac{T}{T_s} \quad (A.10a)$$

and

$$s_{Vs} - s_{Ls} = \lambda_s / T_s \quad (A.10b)$$

Thus, the enthalpy change is:

$$h_{in} - h_{out} = h_{in} - [h_{Ls} + x\lambda_s] = C_{PL}(T - T_s) - C_{PL}T_s \ln \frac{T}{T_s} \quad (A.11)$$

If the liquid is compressed, i.e., if the tank pressure is greater than the saturation pressure, there is an additional enthalpy change,  $v_L (P_T - P_V)$ , when the pressure decreases in the exit stream to the saturation value. Altogether, then, the exit velocity,  $\sqrt{2\Delta h + g\Delta Z}$ , is

$$V_{Lo} = \left\{ 2 \left[ C_{PL} (T - T_s) - C_{PL}T_s \ln \frac{T}{T_s} + (P_T - P_V)/\rho_L + g(Z_L - Z_{Lh}) \right] \right\}^{1/2} \quad (A.12)$$

The mass flow expression, equation (II.7a) follows directly.

The density of the exit mixture, for an isentropic discharge, would be similarly:

$$\begin{aligned} \rho_{Lo} &= \frac{1}{v_{Lo}} = \left\{ v_{Ls} + x (v_{Vs} - v_{Ls}) \right\}^{-1} \\ &= \left\{ v_{Ls} + T_s C_{PL} (v_{Vs} - v_{Ls}) \ln \frac{T}{T_s} / \lambda_s \right\}^{-1} \end{aligned} \quad (A.13)$$

However, previous tests (e.g., [25]) have shown that the evaporation from the liquid occurs mainly at the surface of the stream and is a function of the length-to-diameter ratio of the orifice. The net effect is that the exit mixture density has an average value that is larger than that predicted by equation (A.13). For that reason, the empirical factor  $s$  has been inserted in equation (II.7b). Presumably, a factor would also be needed in the velocity relation, equation (A.12), but the results of [25] indicate otherwise.

The results for a vapor discharge can be derived similarly, but for a vapor the entropy change  $s_v - s_{vs}$  is

$$s_v - s_{vs} = Z R \ln \frac{T}{T_s} - \bar{C}_{pv} \ln \frac{T}{T_s} \quad (\text{A.14})$$

Equation (II.8a) then follows.

### A.3 Evaporation Rate Correlation

By observing the trends exhibited by the tests, it seemed that the larger the evaporation rate was, the larger the discrepancy was between the measured tank pressure and the liquid saturation pressure. Consequently, a model of the form:

$$\Delta W_e = (W_e)_{\text{sat}} - (W_e)_{\text{test}} = (W_e)_{\text{sat}}^n \quad (\text{A.15})$$

was assumed.  $(W_e)_{\text{sat}}$  could be calculated for any test from the measured liquid temperatures and vapor space volumes;  $(W_e)_{\text{test}}$  could also be computed from the measured vapor space pressures, temperatures, and volumes. The following table shows some typical numerical results (in g/sec):

Test No.	$(W_e)_{\text{sat}}$	$(W_e)_{\text{test}}$	$\Delta W_e$	$\frac{\Delta W_e}{(W_e)_{\text{sat}}}$	$\frac{\Delta W_e}{(W_e)_{\text{sat}}^2}$	$\frac{\Delta W_e}{(W_e)_{\text{sat}}^{5/4}}$
6	2.43	2.31	0.12	0.050	0.020	0.039
20	3.27	3.17	0.10	0.031	0.009	0.023
22	0.079	0.078	0.0011	0.014	0.176	0.026

The value  $n = 1.25$  best correlates the wide range of evaporation rates shown in the table. To make equation (A.15) dimensionally correct, the right hand side must be divided by  $\{A_T \rho_L \sqrt{RT}\}^{1/4}$ . The proportionality constant was found from linear regression to be 1.9.



#### A.4 Derivation of Water Ingestion Model Equations

Derivation of equation (II.20b). When the relief valve is jammed shut, the vapor space gas is compressed by the ingested water floating on the cargo; see Figure II.2c. Thus:

$$P_T = P_{Ti} \left( \frac{Z_T - Z_{Li}}{Z_T - Z_{WT}} \right)^{\gamma_m} \quad (A.16)$$

where  $\gamma_m = 1$  for isothermal compression, and  $\gamma_m =$  ratio of gas specific heats for adiabatic compression.

The pressure differential at the puncture is:

$$\Delta P = P_T + g \rho_W (Z_{WT} - Z_L) + g \rho_L (Z_L - Z_{Lh}) - P_a \quad (A.17)$$

Since it is required that  $d(\Delta P)/dt = 0$ , equations (A.16) and (A.17) can be combined to give:

$$\left( \frac{\gamma_m P_T}{Z_T - Z_{WT}} \right) \frac{dZ_{WT}}{dt} + g \rho_W \left( \frac{dZ_{WT}}{dt} - \frac{dZ_L}{dt} \right) + g \rho_L \left( \frac{dZ_L}{dt} \right) = 0 \quad (A.18)$$

But  $dZ_{WT}/dt = (W_w/\rho_w - W_{Lo}/\rho_L)/A_T$  and  $dZ_L/dt = -W_{Lo}/\rho_L A_T$ . Equation (II.20) is derived by substituting these relations into equation (A.18) and solving for  $W_w$ .

Derivation of equation (II.23). When  $Z_{WT,Eq.} < Z_T$ , the cargo outflow is assumed to begin when the water level in the tank is slightly greater than  $Z_{WT,Eq.}$ ; in particular, it begins at

$$Z_{WT} = 1.1 (Z_{WT,Eq.} - Z_{Lh}) + Z_{Lh} \quad (A.19)$$

This is analogous to the use of  $P_{Ti} - P_{T,Eq.}$  and  $Z_{Li} - Z_{L,Eq.}$  in equation (II.17), but here the  $\Delta P$  at the puncture is just  $Z_{WT} - Z_{Lh} - (Z_{WT,Eq.} - Z_{Lh}) = 0.1 (Z_{WT,Eq.} - Z_{Lh})$ . Equation (II.23) follows directly.

### A.5 Derivation of Choked Flow Equations (II.31)

As shown previously in equations (A.9) and (A.10), the thermodynamic quality of the discharge mixture for an isentropic expansion to temperature  $T_c$  is:

$$x = C_{PL} T_c \ln \frac{T}{T_c} / \lambda_c \quad (\text{A.20})$$

Making the substitution  $T_c = T (1 - \frac{\Delta T}{T})$ , and expanding in powers of  $\Delta T/T$  gives:

$$x \lambda_c = C_{PL} \Delta T \left[ 1 - \frac{1}{2} \left( \frac{\Delta T}{T} \right) + \dots \right] \quad (\text{A.21})$$

Likewise

$$h_L - h_{Lc} = C_{PL} \Delta T \quad (\text{A.22})$$

The vapor density  $v_{Vc}$  can be written as:

$$v_{Vc} = \frac{ZRT_c}{P_c} = \frac{ZRT}{P} \frac{\left(1 - \frac{\Delta T}{T}\right)}{\left(1 - \frac{\Delta P}{P}\right)} \approx v_V \left(1 - \frac{\Delta T}{T}\right) \left(1 + \frac{\Delta P}{P} + \dots\right) \quad (\text{A.23})$$

From the Chemical Properties File, the pressure  $P$  is given by

$$\frac{\ln P}{\ln 10} = A - \frac{B}{(C + T^*)} \quad (\text{A.24})$$

so

$$\frac{\Delta P}{P} \approx \ln 10 \left[ \frac{B \Delta T}{(C + T^*)^2} \right] \quad (\text{A.25})$$

Therefore, the product  $x_c v_{Vc}$  can be expressed as

$$\begin{aligned}
 x_c v_{Vc} &= \frac{C_{PL} \Delta T}{\lambda_c} \left[ 1 - \frac{1}{2} \left( \frac{\Delta T}{T} \right) \right] v_V \left\{ 1 + \frac{\Delta T}{T} \left[ \frac{BT \ln 10}{(C + T^*)^2} - 1 \right] \right\} \\
 &= \frac{C_{PL} \Delta T v_V}{\lambda_c} \left\{ 1 + \frac{\Delta T}{T} \left[ \frac{BT \ln 10}{(C + T^*)^2} - \frac{3}{2} \right] \right\} \quad (A.26)
 \end{aligned}$$

Since  $3x_c$  is generally small, the term  $v_{Lc} - 3x_c v_{Lc}$  in the denominator of equation (II.30) is approximated as  $v_{Lc} = v_L$ . Likewise,  $\lambda_c = \lambda$ , so altogether equation (II.30) can be written as:

$$\frac{\frac{1}{2} C_{PL} \left( \frac{\Delta T}{T} \right)^2}{v_L + \frac{3C_{PL} v_V T}{\lambda} \left( \frac{\Delta T}{T} \right) \left[ 1 + \frac{\Delta T}{T} E \right]} = - \frac{1}{2} [D] \quad (A.27)$$

where  $E$  and  $[D]$  are given by equations (II.31b) and (II.31c). Equation (A.27) is quadratic in  $(\Delta T/T)$  and can be solved to give equation (II.31a).

## APPENDIX B. CHEMICAL PROPERTIES FILE

Units are generally in the cgs system: pressure in multiples of  $10^5$  dyne/cm<sup>2</sup> =  $10^3$  N/m<sup>2</sup> = 1 KPa; specific heat in cal/gr-°C; latent heat in cal/gr; temperature in °C; and density in gr/cm<sup>3</sup>. The temperature range of validity for the correlations generally correspond to vapor pressures from about 0.1 to 2 to 4 atmospheres. Maximum errors of the correlations with respect to the cited data sources are also given. Correlations from the present CPF are given in parentheses, as well as their maximum error over the same temperature range.

In most of the data sources, English units are used. The new correlations were initially developed therefore in English units and then converted to cgs. For the same reason, the English units correlations given in the CPF were reviewed and converted to cgs.

1. ANHYDROUS AMMONIA [16,17,18], molecular weight = 17.698

Range of validity: -40°C to 38°C (71.8 kPa to 1461 kPa)

- $\log_{10} P_V = 6.61724 - \frac{976.98}{T + 245.19}$  ; max. error = -0.73%  
 (=  $7.1465 - \frac{1232.78}{T + 273.2}$  ; max. error = 4.05%)
- $\rho_L = 0.6371 - 1.413 \times 10^{-3} T$  ; max. error = -0.29%  
 (=  $0.6538 - 1.299 \times 10^{-2} T - 1.36 \times 10^{-5} T^2$  ; max. error = 2.43%)
- $Z = 0.932 - 1.647 \times 10^{-3} T - 1.215 \times 10^{-5} T^2$  ; max. error = 0  
 (= 1.0 ; max. error = 17.4%)
- $C_{PL} = 1.104 + 1.20 \times 10^{-3} T$  ; max. error = 0.73%  
 (1.100 +  $1.000 \times 10^{-3} T$  ; max. error = 1.56%,

- $\overline{C_{PV}} = 0.257 - 4.02 \times 10^{-3}T - 1.504 \times 10^{-5}T^2$ ; max. error = -2.61%  
 $(C_{PV} = 0.533 + 2.616 \times 10^{-3}T + 3.756 \times 10^{-7}T^2 + 1.644 \times 10^{-10}$ ; max. error = 449.9%)
- $\lambda = [244.37 - 0.432T - 1.1023 \times 10^{-3}T^2] \left( \frac{T + 273.2}{T + 245.2} \right)^2 - \frac{0.53741 P_V (T + 273.2)}{(T + 245.2)^2 (0.6371 - 1.413 \times 10^{-3}T)}$ ; max. error = -0.31%  
 (= 327.0 ; max. error = 18.82%)

2. BUTANE [15,16,18,19,20], molecular weight = 58.121

Range of validity: -55°C to 43°C (6.9 kPa to 415 kPa)

- $\log_{10} P_V = 5.9496 - \frac{943.46}{T + 239.7}$ ; max. error = 0.49%  
 (= 5.9545 -  $\frac{946.11}{T + 240}$ ; max. error = 0.76%)
- $\rho_L = 0.5952 - 9.267 \times 10^{-4}T$ ; max. error = 0.14%  
 (= 0.6009 -  $9.000 \times 10^{-4}T$ ; max. error = 1.37%)
- $z = 0.95746 - 1.1405 \times 10^{-3}T - 1.0404 \times 10^{-5}T^2$ ; max. error = -0.38%  
 (= 1.0 ; max. error = 12.55%)
- $C_{PL} = 0.5540 + 2.7378 \times 10^{-4}T + 1.5116 \times 10^{-5}T^2$ ; max. error = 1.64%  
 (= 0.5800 +  $1.300 \times 10^{-3}T$ ; max. error = 9.02%)
- $\overline{C_{PV}} = 0.335$ ; max. error = 3.72%  
 $(C_{PV} = 0.3811 + 1.150 \times 10^{-3}T - 6.384 \times 10^{-7}T^2 + 1.445 \times 10^{-10}T^3$ ; max. error = 28.38%)

$$\bullet \quad \lambda = [71.069 - 8.4656 \times 10^{-2}T - 7.7226 \times 10^{-4}T^2] \left( \frac{T+273.2}{T+239.7} \right)^2 - \frac{0.51897 P_V (T+273.2)}{(T+239.7)^2 (0.5952 - 9.267 \times 10^{-4}T)} ; \text{max. error} = 1.57\% \\ (= 92.0 ; \text{max. error} = 13.58\%)$$

3. CHLORINE [21], molecular weight = 70.91

Range of validity:  $-51^\circ$  to  $38^\circ$  (45 kPa to 1067 kPa)

$$\bullet \quad \log_{10} P_V = 6.0540 - \frac{860.19}{T+246.52} ; \text{max. error} = 0.16\% \\ (= 6.5425 - \frac{1086.11}{T+273.17} ; \text{max. error} = 5.02\%)$$

$$\bullet \quad \rho_L = 1.4676 - 2.8738 \times 10^{-3}T ; \text{max. error} = 0.49\% \\ (= 1.4599 - 2.599 \times 10^{-3}T ; \text{max. error} = 0.93\%)$$

$$\bullet \quad Z = 0.948 - 1.3064 \times 10^{-3}T - 9.383 \times 10^{-6}T^2 ; \text{max. error} = 0.10\% \\ (= 1.0 ; \text{max. error} = 11.5\%)$$

$$\bullet \quad C_{PL} = 0.2334 + 1.35 \times 10^{-4}T ; \text{max. error} = -2.05\% \\ (= 0.1200 + 5.00 \times 10^{-4}T ; \text{max. error} = -58.19\%)$$

$$\bullet \quad \overline{C_{PV}} = 0.0832 - 5.6571 \times 10^{-4}T - 4.3393 \times 10^{-6}T^2 ; \text{max. error} = 0.61\% \\ (C_{PV} = 0.1130 + 5.8933 \times 10^{-5}T - 9.166 \times 10^{-8}T^2 ; \text{max. error} = 89.73\%)$$

$$\bullet \quad \lambda = [52.585 - 7.2465 \times 10^{-2}T - 5.2047 \times 10^{-4}T^2] \left( \frac{T+273.2}{T+246.52} \right)^2 - \frac{0.47317 P_V (T+273.2)}{(T+246.52)^2 (1.4676 - 2.8738 \times 10^{-3}T)} ; \text{max. error} = -0.55\% \\ (= 68.72 ; \text{max. error} = 19.28\%)$$

4. LNG\* [15,13,19,22,23], molecular weight = 16.03

Range of validity:  $-177^{\circ}$  to  $-153^{\circ}$  (22.3 kPa to 202.5 kPa)

- $\log_{10} P_V = 5.7365 - \frac{389.89}{T + 266.0}$  ; max. error = -1.14%  
(= same)
- $\rho_L = 0.1833 - 1.665 \times 10^{-3} T$  ; max. error = 0%  
(=  $0.1880 - 1.399 \times 10^{-3} T$  ; max. error = 1.98%)
- $Z = -2.5724 - 3.97015 \times 10^{-2} T - 1.10412 \times 10^{-4} T^2$  ; max. error = 0.45%  
(= 1.0 ; max. error = 4.21%)
- $C_{PL} = 1.1995 + 2.2639 \times 10^{-3} T$  ; max. error = -0.57%  
(=  $1.1961 + 2.1996 \times 10^{-3} T$  ; max. error = 1.33%)
- $\overline{C}_{PV} = 2.2639 + 1.0904 \times 10^{-2} T$  ; max. error = -4.48%  
( $C_{PV} = 0.5166 + 6.6838 \times 10^{-4} T + 5.311 \times 10^{-7} T^2$  ; max. error = 12.1%)
- $\lambda = - [236.63 + 4.4238 T + 1.23028 \times 10^{-2} T^2] \left( \frac{T + 273.2}{T + 266.0} \right)^2$   

$$- \frac{0.21447 P_V (T + 273.2)}{(T + 266.0)^2 (0.1833 - 1.665 \times 10^{-3} T)}$$
 ; max. error = 1.51%  
(= 121.8 ; max. error = -4.65%)

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\* Assumed to be methane.

5. METHYL ALCOHOL [2,16,18,19,24], molecular weight = 32.043

Range of validity:  $-10^{\circ}$  to  $80^{\circ}$  (2 kPa to 179 kPa)

- $\log_{10} P_V = 7.20585 - \frac{1582.27}{T+239.73}$  ; max. error = -5.45% at  $-10^{\circ}$   
 ; max. error = 0.32%,  $T > 15^{\circ}$   
 (=  $7.3265 - \frac{2002.22}{T+273.17}$  ; max. error = 2.86%)
- $\rho_L = 0.8099 - 9.1979 \times 10^{-4} T$  ; max. error = 0.10%  
 (same)
- $z = 0.959 + 5.000 \times 10^{-4} T - 1.0833 \times 10^{-5} T^2$  ; max. error =  
 = 0.13%  
 (= 1.0 ; max. error = 7.53%)
- $C_{PL} = 0.5660 + 1.622 \times 10^{-3} T$  ; max. error = -0.43%  
 (=  $0.5260 + 3.200 \times 10^{-3} T$  ; max. error = 9.26%)
- $\overline{C_{PV}} = 0.3018 - 1.601 \times 10^{-3} T - 1.905 \times 10^{-6} T^2$  ; max. error  
 = 3.00%  
 ( $C_{PV} = 0.2922 + 6.886 \times 10^{-4} T + 2.497 \times 10^{-7} T^2$  ; max.  
 error = 100.72%)
- $\lambda = [216.56 + 1.129 \times 10^{-1} T - 2.446 \times 10^{-3} T^2] \left( \frac{T+273.2}{T+239.7} \right)^2$   
 $- \frac{0.87037 P_V (T+273.2)}{(T+239.7)^2 (0.8099 - 9.1979 \times 10^{-4} T)}$  ; max. error = -1.96%  
 (= 262.8 ; max. error = -6.58%)



6. PROPANE [15,16,19,20,25], molecular weight = 58.12

Range of validity:  $-84^{\circ}$  to  $44^{\circ}$  (1.5 kPa to 1520 kPa)

- $\log_{10} P_V = 5.9683 - \frac{818.56}{T+248.68}$  ; max. error = -1.67% at  $44^{\circ}$   
 (=  $5.9545 - \frac{813.33}{T+248.0}$  ; max. error = -1.20%,  $T < 27^{\circ}$ )
- $\rho_L = 0.5301 - 1.425 \times 10^{-3}T - 2.788 \times 10^{-6}T^2$  ; max. error = 0.46%  
 (=  $0.5350 - 1.100 \times 10^{-3}T$  ; max. error = 5.65%)
- $Z = 0.8895 - 2.499 \times 10^{-3}T - 1.486 \times 10^{-5}T^2$  ; max. error = 0.85%  
 (= 1.0 ; max. error = 33.3%)
- $C_{PL} = 0.5731 + 1.6913 \times 10^{-3}T + 1.705 \times 10^{-5}T^2$  ; max. error = -6.46%  
 (=  $0.6599 + 2.50 \times 10^{-3}T$  ; max. error = -15.41%)
- $\overline{C}_{PV} = 0.2373 - 1.6045 \times 10^{-3}T - 1.4184 \times 10^{-5}T^2$  ;  
 max. error = -21.28% at  $-70^{\circ}$   
 max. error = 2.52%,  $T > -70^{\circ}$   
 ( $C_{PV} = 0.3700 + 1.158 \times 10^{-3}T - 4.083 \times 10^{-7}T^2$  ; max. error = 154.32%)
- $\lambda = [75.512 - 2.121 \times 10^{-1}T - 1.2615 \times 10^{-3}T^2] \left( \frac{T+273.2}{T+248.7} \right)^2$   
 $- \frac{0.45026 P_V (T+273.2)}{(T+248.7)^2 (0.5301 - 1.10 \times 10^{-3}T)}$  ; max. error = 2.55%  
 (= 101.78 ; max. error = 46.72%)

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EXPERIMENTAL VERIFICATION AND REVISION OF THE VENTING RATE MODE--ETC(U)

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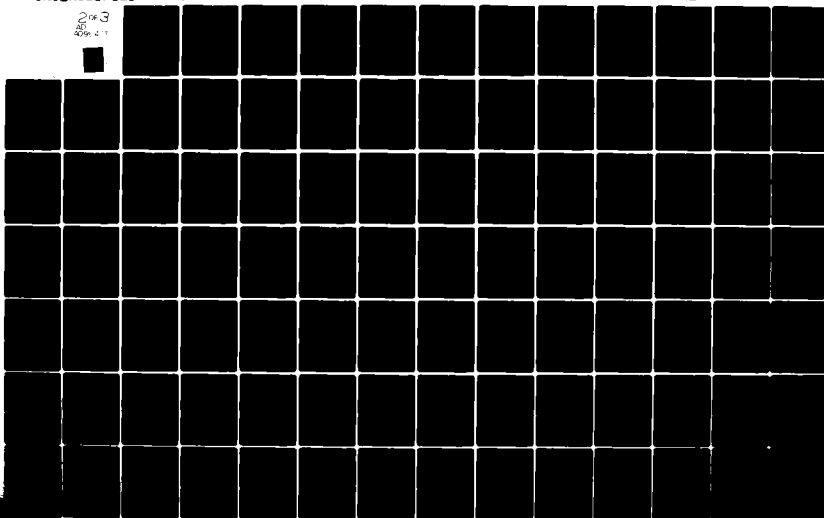
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## APPENDIX C. COMPUTER PROGRAM

The analytical model presented in Section II has been programmed for numerical solution in the computer code TVENT, which is outlined in this appendix. The basic technique used in the code is to assume that a small percentage of the cargo mass is discharged and then compute the time increment for the discharge to occur, using the various energy and mass balances and flow relations. After the model has been "solved" to a prescribed degree of accuracy for this incremental discharge, an additional discharge is assumed, and the solution process repeated. This continues until tank conditions are computed that make further discharges impossible.

TVENT is arranged into a "MAIN" code and a number of subroutines that generally correspond to the various submodels comprising the entire venting rate model. Flow charts giving the logic for the MAIN program and the subroutines are given in Figures C.1 through C.12 and Tables C.2, C.3, and C.4. Table C.1 gives the required inputs for the program in its present form.

To describe the code briefly, the MAIN program: (1) organizes the input data; (2) determines whether the cargo is volatile; (3) determines whether liquid or vapor or both are discharged; (4) selects the increment of mass to be discharged; (5) determines whether the calculations for any one iteration have converged; (6) determines when the discharge is completed; and (7) calls the various subroutines that perform the detailed computations. Before each calculation cycle, MAIN also calls the subroutine INGEST, which determines whether the tank conditions are appropriate for air or water ingestion. A test for choked flow conditions (CHOKTEST) is called before the first cycle and each cycle thereafter so long as choked outflow can occur.

Subroutines are used to make most of the calculations. ENERGY solves the energy and mass balances for a volatile cargo and calls FLOW1 to compute discharge flow rates. NONENER and FLOW2 are used similarly for nonvolatile cargos. INGEST performs all the logic operations needed to determine whether air or water ingestion can occur at any instant. AIRIN computes the change in tank conditions when air bubbles are ingested by calling ENERGY2 (for volatile cargos) or NONENER2 (for nonvolatile cargos) to determine the vapor space pressure rise; a return to ENERGY or NONENER is then made to compute the subsequent liquid discharge. WATIN computes the cargo discharge and water inflow when water is ingested, and calls FLOW2 to compute the outflow. NOOUT is used to compute the discharge when water can be ingested at the instant the puncture is opened, as well as for certain other water ingestion cases similar to this. Other self-explanatory subroutines compute cargo properties and miscellaneous data.

A computer listing of each routine is also given in this appendix. Table C.5 explains the correspondence between the symbols used in the analysis and those used in the computer code. Table C.6 gives the formats to be used to enter the input data in TVENT.

Other notes on the use of the computer code include:

1. The normal outputs are the total mass of cargo liquid, cargo vapor, and air released, and the total discharge time.
2. The code assumes that only liquid or only vapor can be discharged through circular or horizontal slot-like punctures; when the liquid surface intersects this type of puncture, only liquid is discharged until the level drops below the puncture. Simultaneous liquid and vapor discharges are allowed for vertical slot-like punctures when the liquid level intersects the puncture.

3. Similarly, circular or horizontal slot-like punctures must be either completely open to the air or completely submerged. Only for a vertical slot-like puncture is the water level allowed to intersect the puncture.
4. The code assumes the cross-sectional area of the tank is constant; for spherical or other tank shapes, an effective tank height should be input such that  $V_T/Z_T$  gives a representative or average area.
5. For nonvolatile substances, only the liquid density correlation must be input accurately. The other correlations, if they are not known for that cargo, can be input as any convenient values (say, one). Care should be taken, however, to make sure that the fictitious vapor pressure correlation predicts positive pressures that are significantly less than atmospheric.

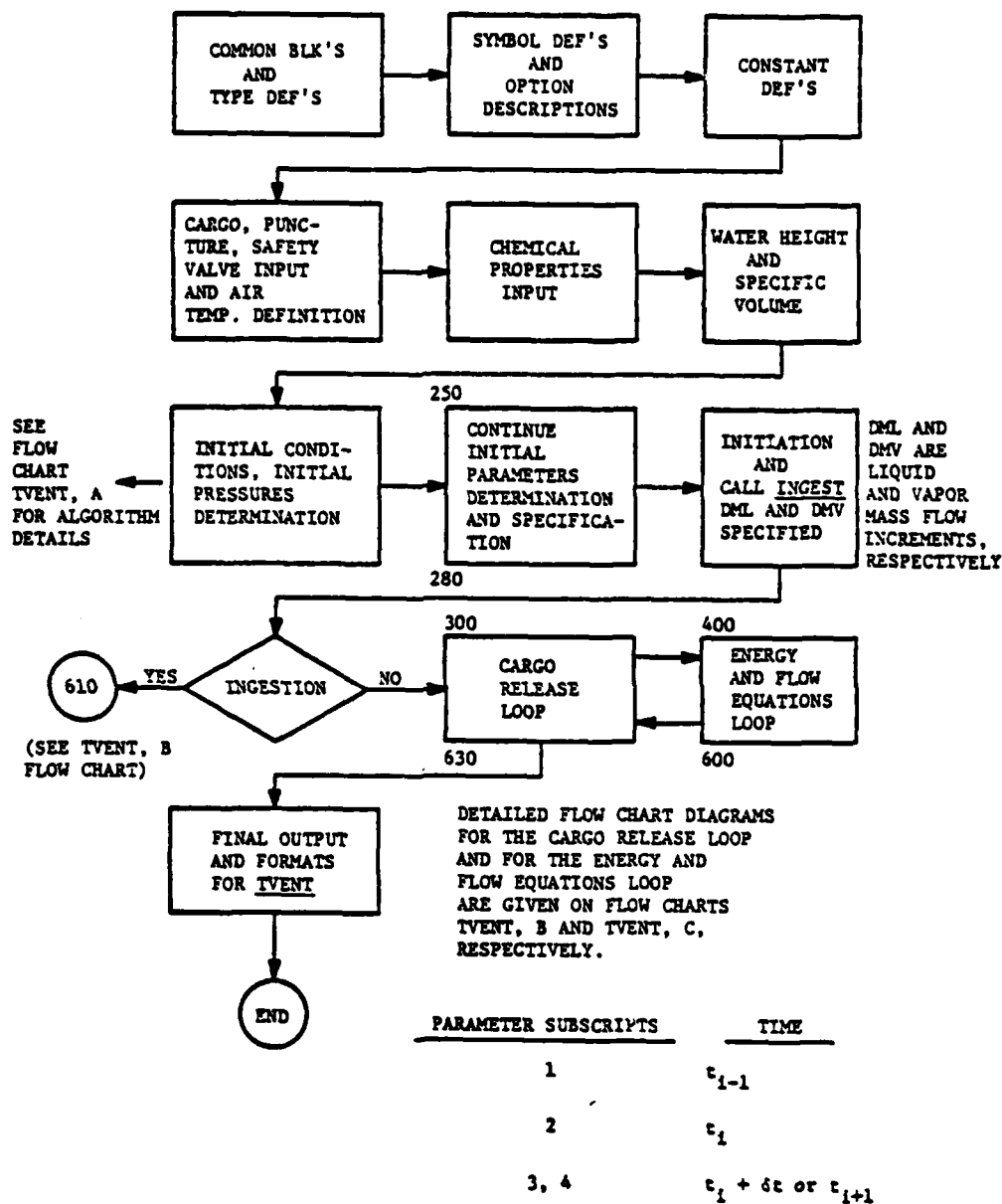


FIGURE C-1. PROGRAM TVENT (MAIN PROGRAM) FLOW CHART

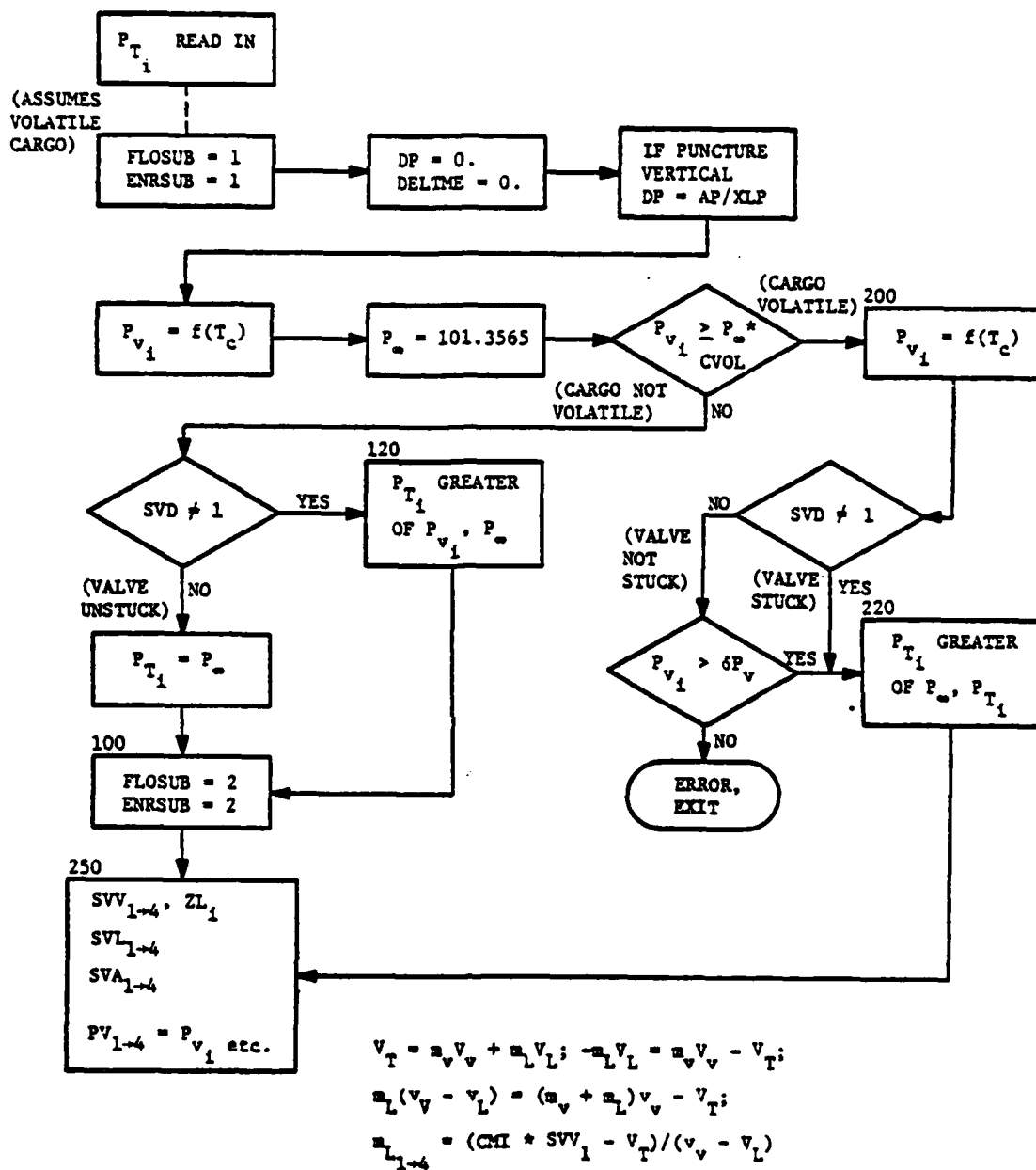


FIGURE C-1. PROGRAM TVENT (MAIN PROGRAM) FLOW CHART (Cont'd)

(TVENT, A - Initial Conditions, Initial Pressures  
Determinations Algorithm Flow Chart)

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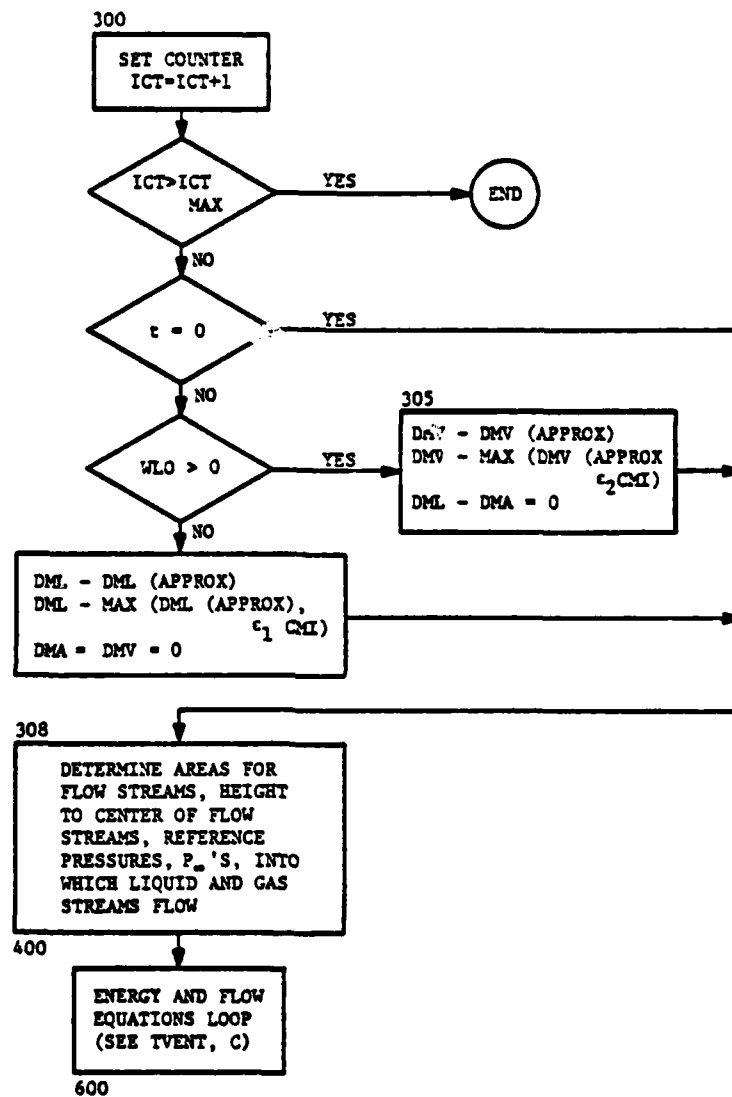


FIGURE C-1. PROGRAM TVENT (MAIN PROGRAM) FLOW CHART (Cont'd)  
(TVENT, B - Cargo Release Loop Flow Chart)



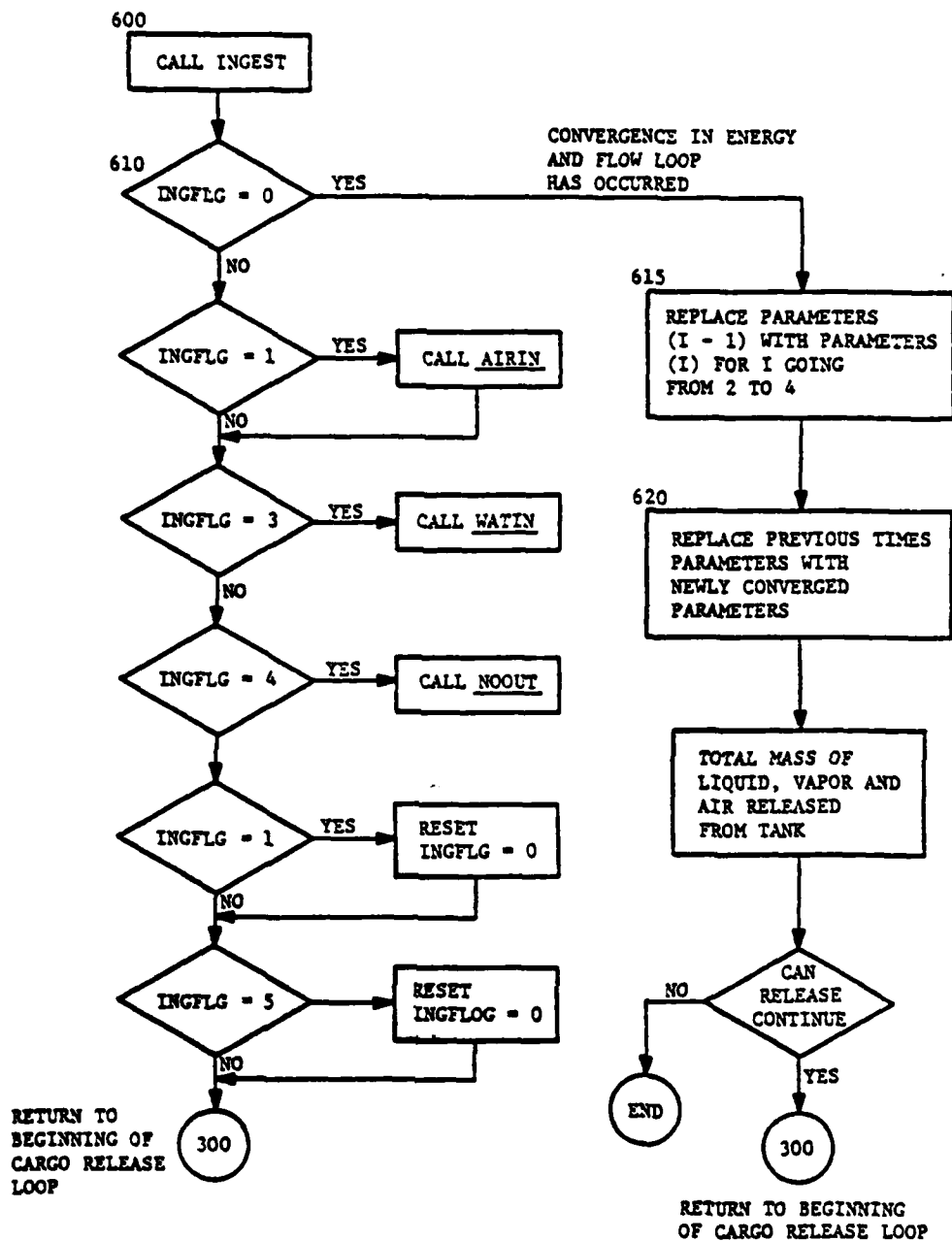


FIGURE C-1. PROGRAM TVENT (MAIN PROGRAM) FLOW CHART (Cont'd)  
(TVENT, B - Cargo Release Loop Flow Chart)

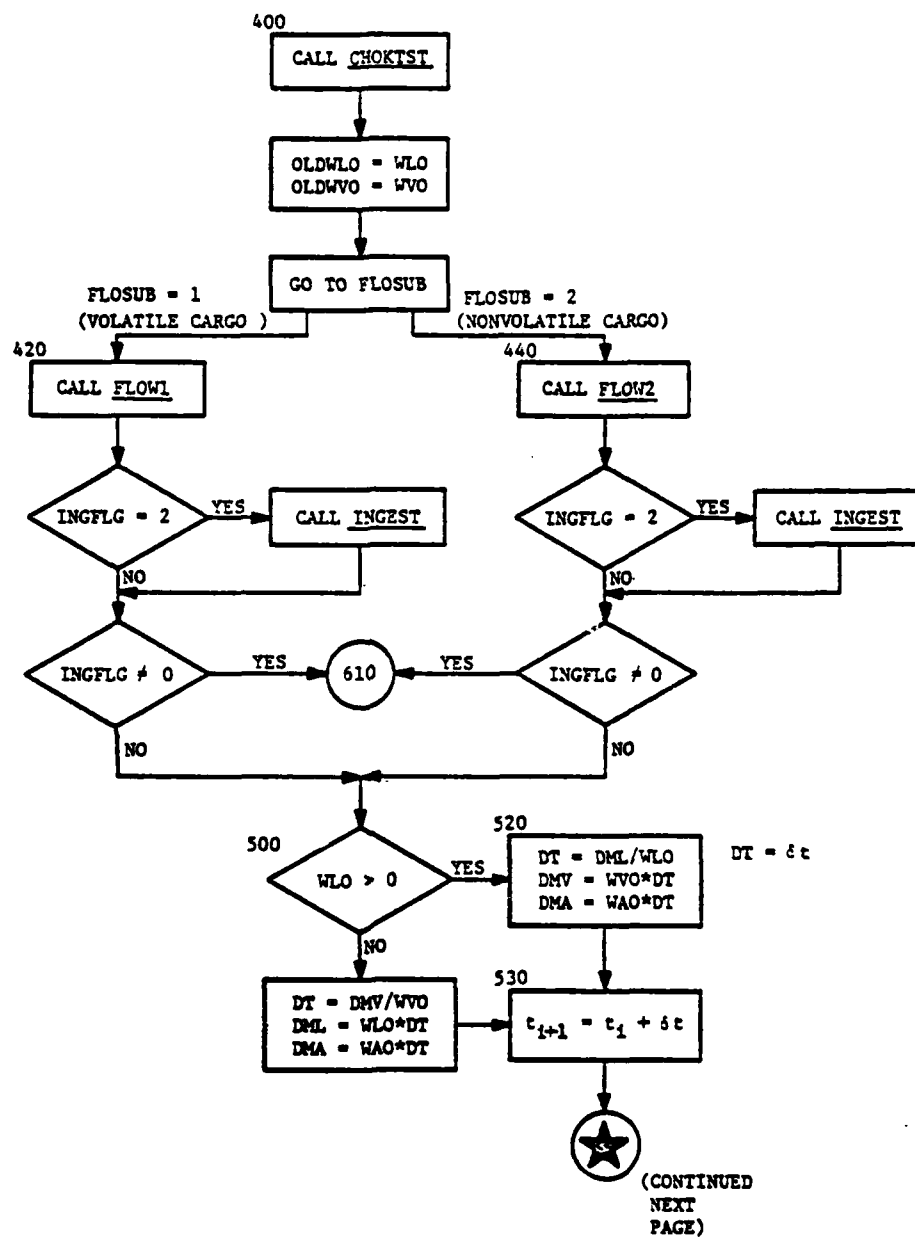


FIGURE C-1. PROGRAM TVENT (MAIN PROGRAM) FLOW CHART (Cont'd)  
(TVENT, C - Energy and Flow Equations Loop Flow Chart)

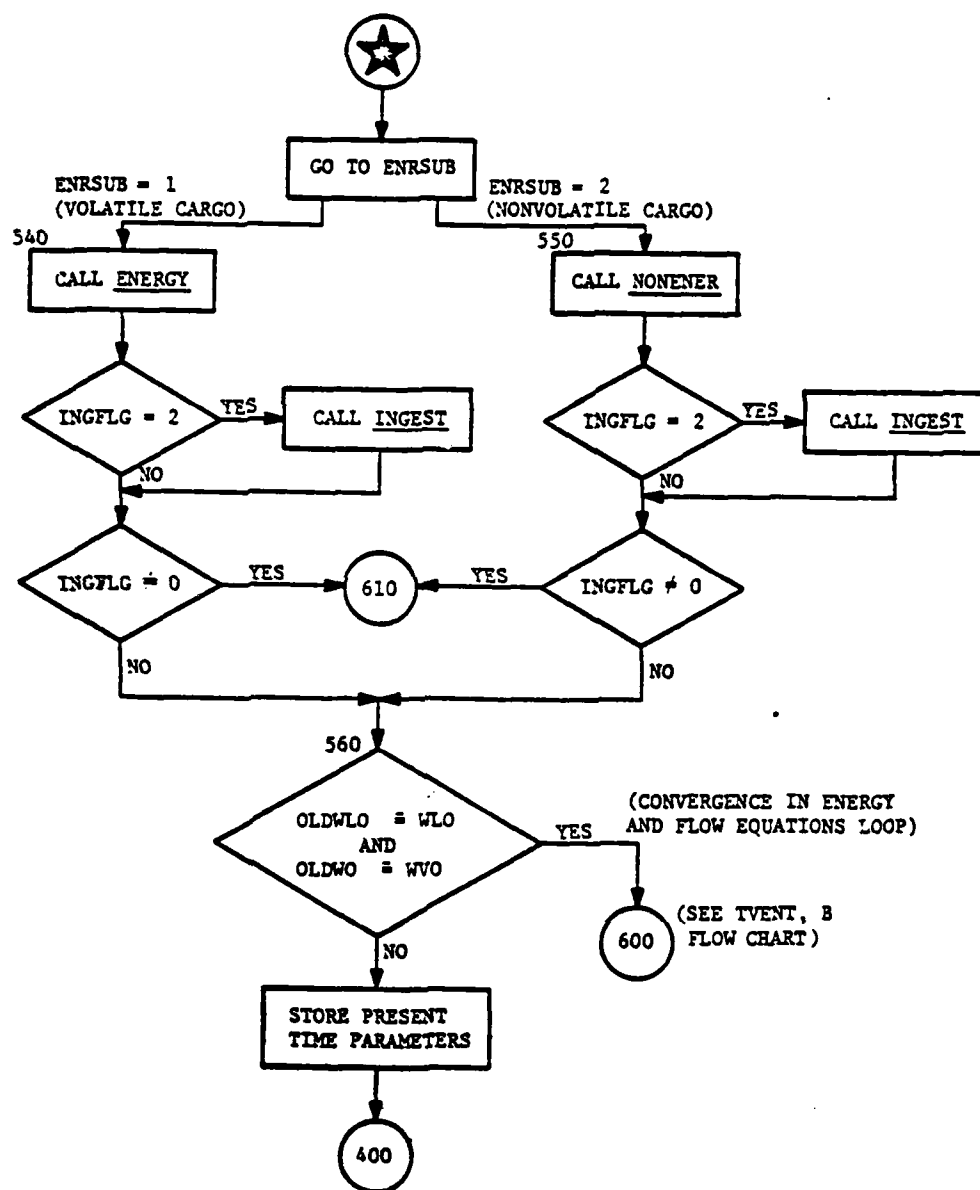


FIGURE C-1. PROGRAM TVENT (MAIN PROGRAM) FLOW CHART (Concl'd)  
(TVENT, C - Energy and Flow Equations Loop Flow Chart)

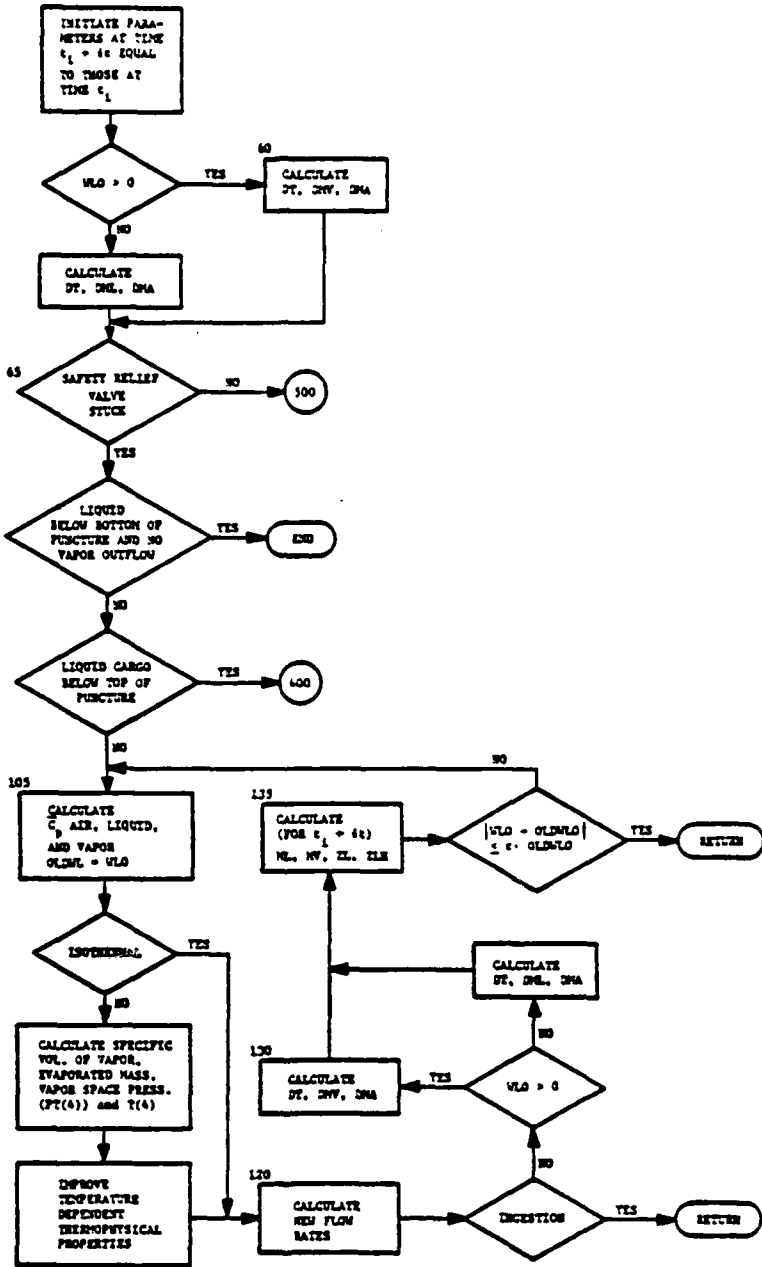


FIGURE C-2. ENERGY SUBROUTINE FLOW CHART

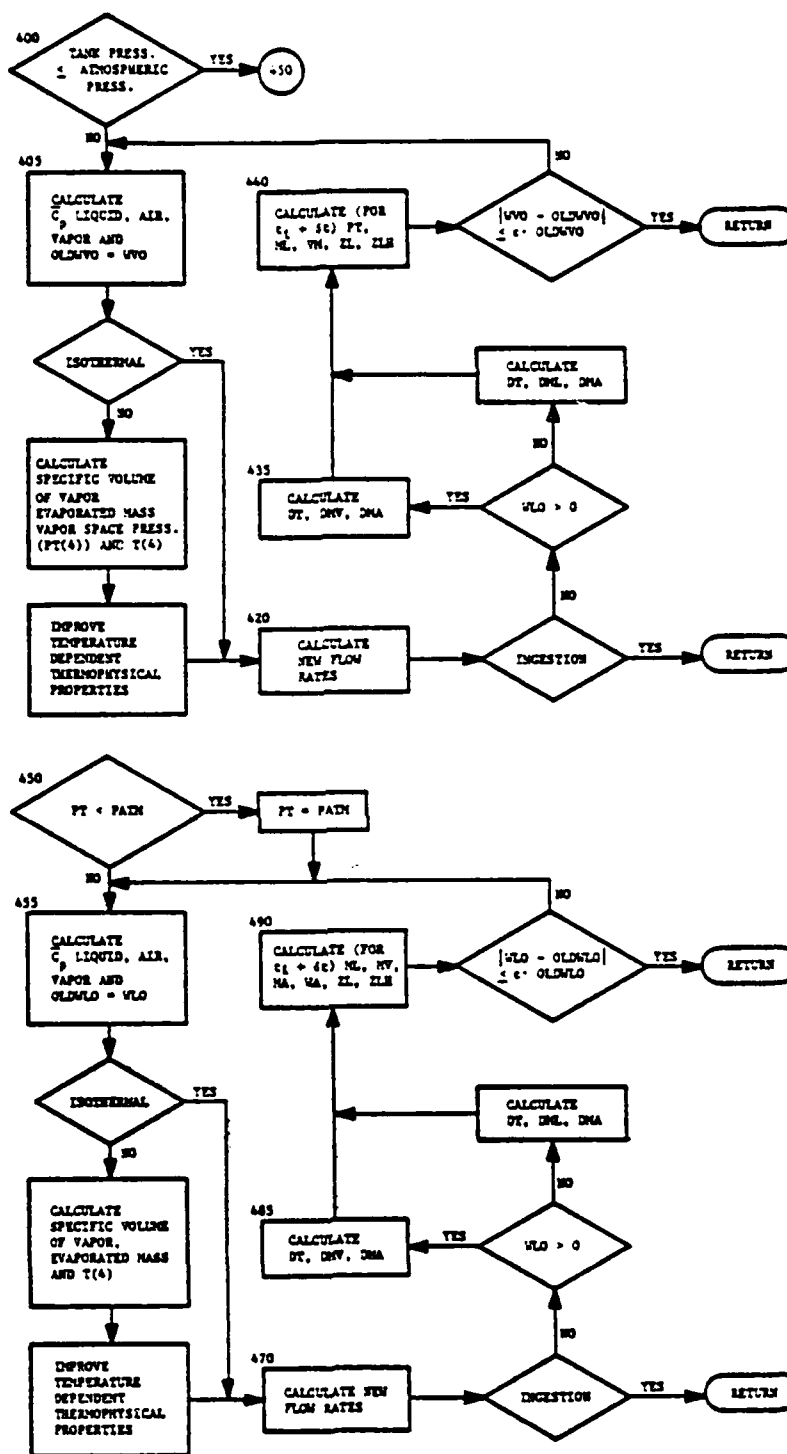


FIGURE C-2. ENERGY SUBROUTINE FLOW CHART (Cont'd)

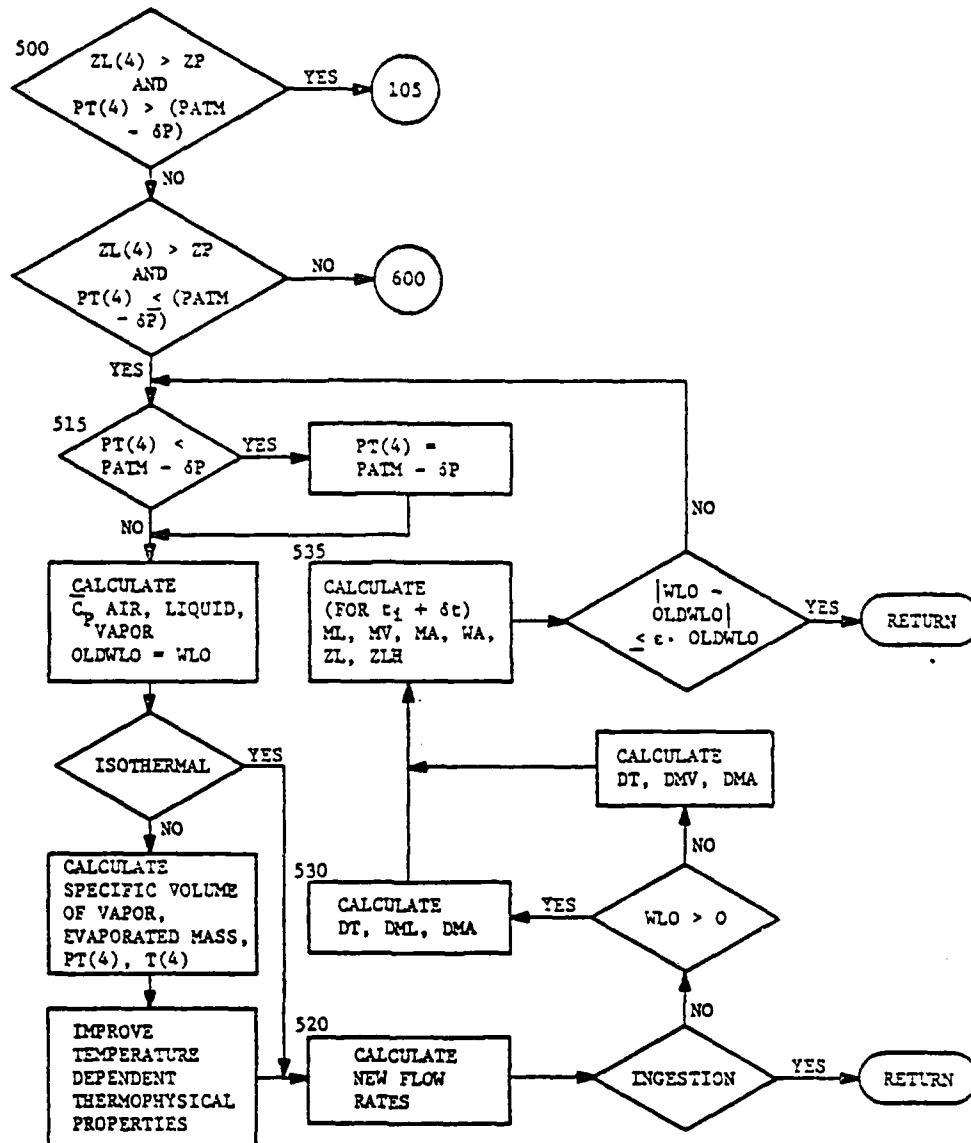


FIGURE C-2. ENERGY SUBROUTINE FLOW CHART (Cont'd)

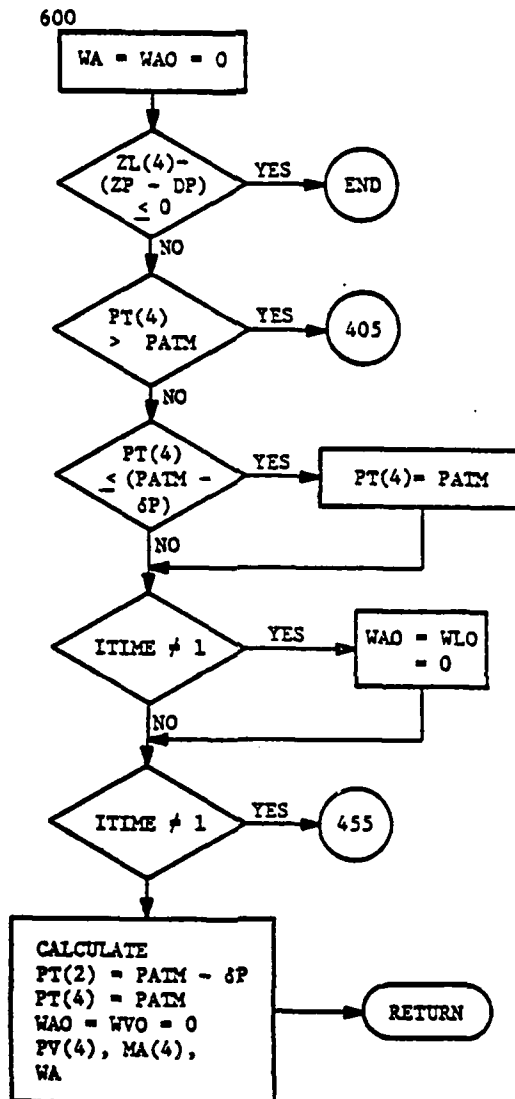


FIGURE C-2. ENERGY SUBROUTINE FLOW CHART (Concl'd)

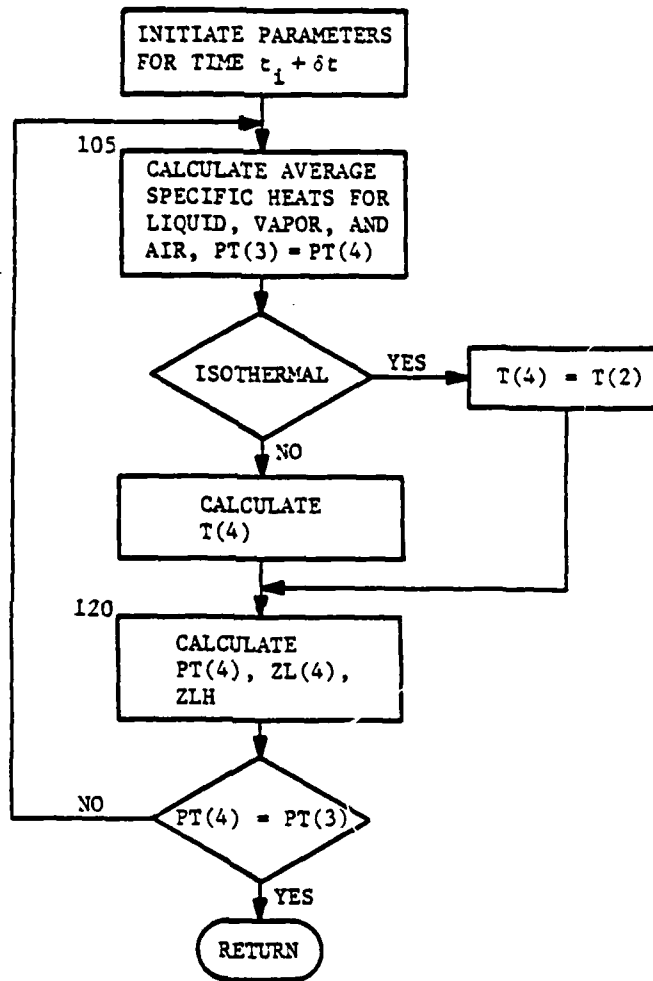
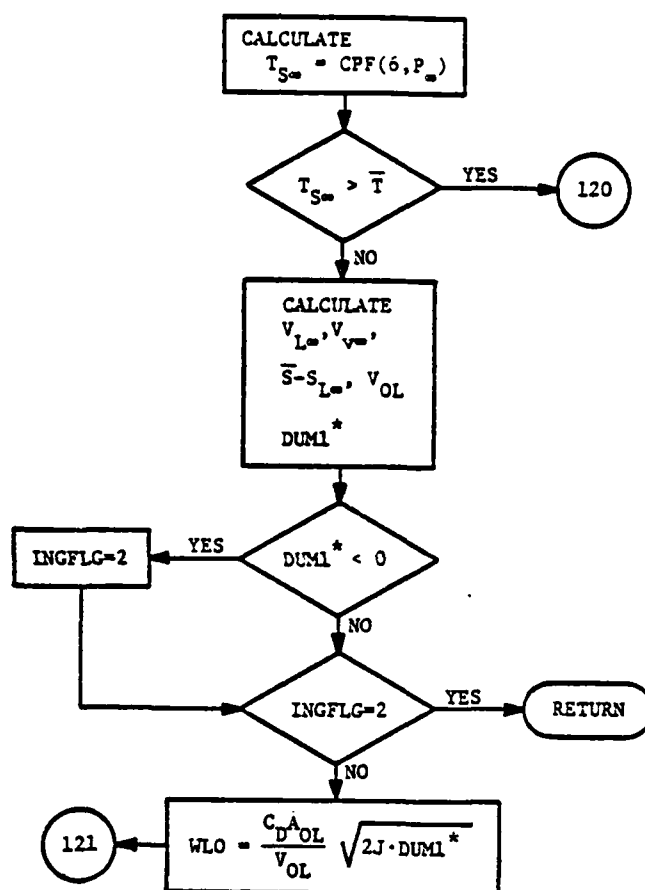


FIGURE C-3. SUBROUTINE ENERGY2 (KOUNT) FLOW CHART





$$*DUM1 = \bar{c}_{pL}(\bar{T} - T_{S_{\infty}}) - T_{S_{\infty}}(\bar{s} - s_{L_{\infty}}) + 10^4(\bar{P}_T - \bar{P}_v)\bar{v}_L/J + g(\bar{z}_L - \bar{z}_{Lh})/J$$

$$**DUM1 = 2(10^4(\bar{P}_T - P_{\infty})\bar{v}_L + g(\bar{z}_L - \bar{z}_{Lh}))$$

FIGURE C-4. SUBROUTINE FLOW1 FLOW CHART

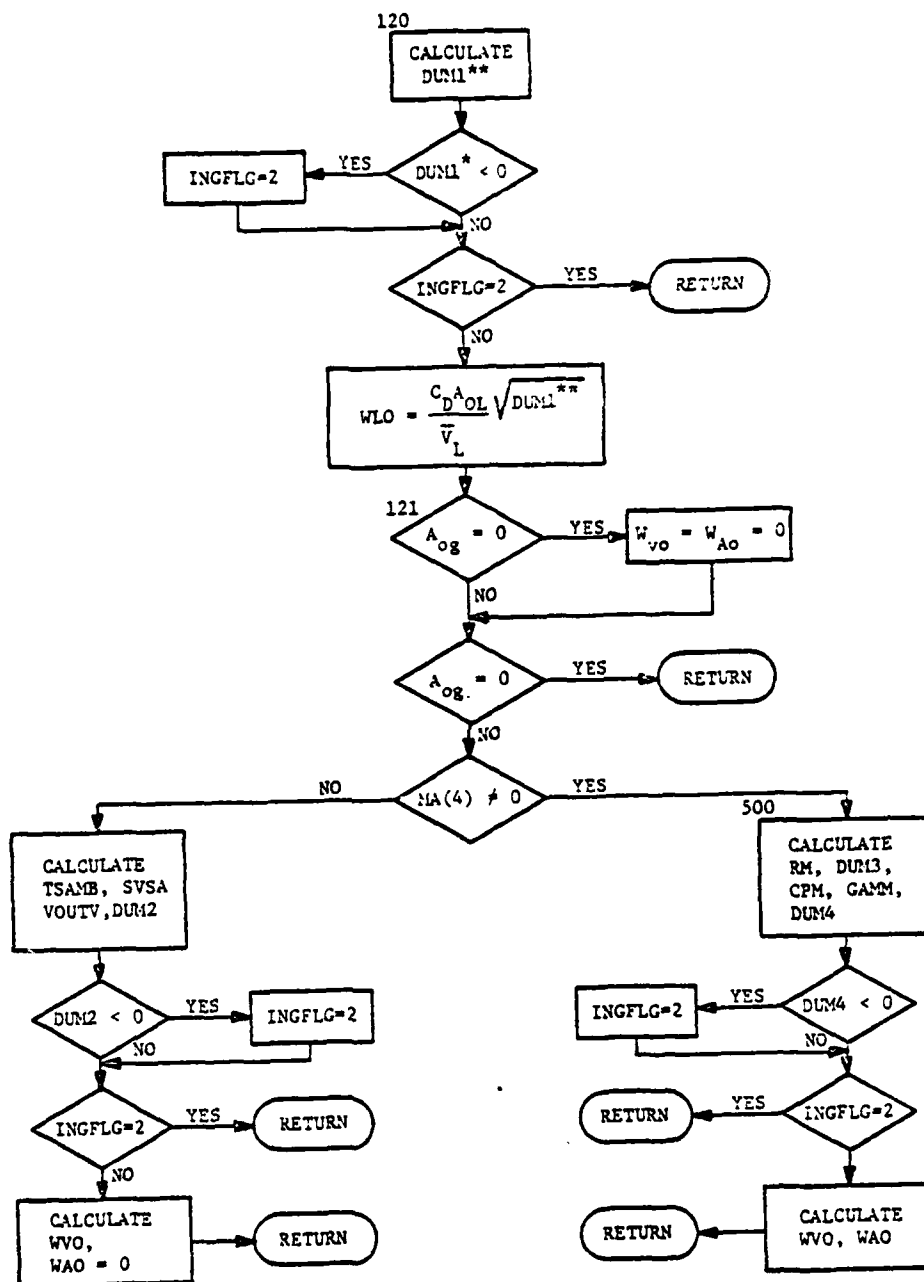


FIGURE C-4. SUBROUTINE FLOW1 FLOW CHART (Concl'd)

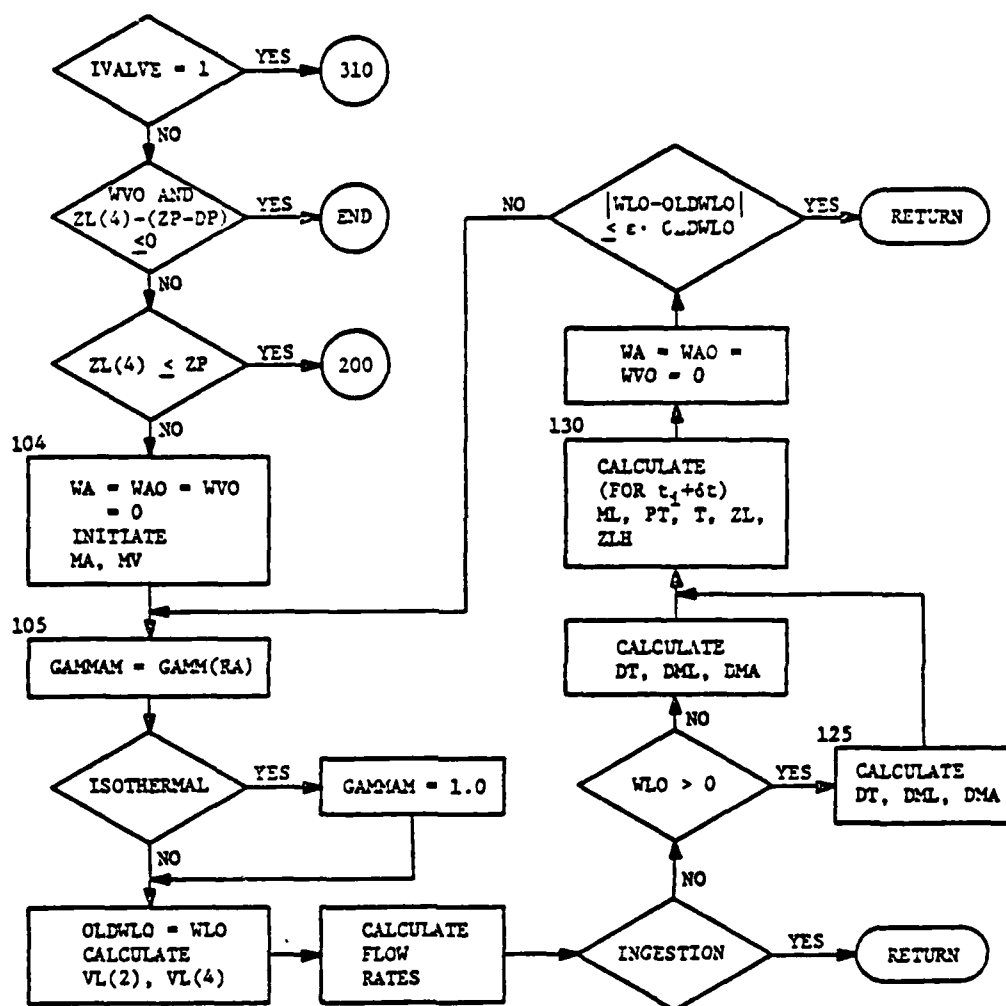


FIGURE C-5. SUBROUTINE\_NONENER FLOW CHART

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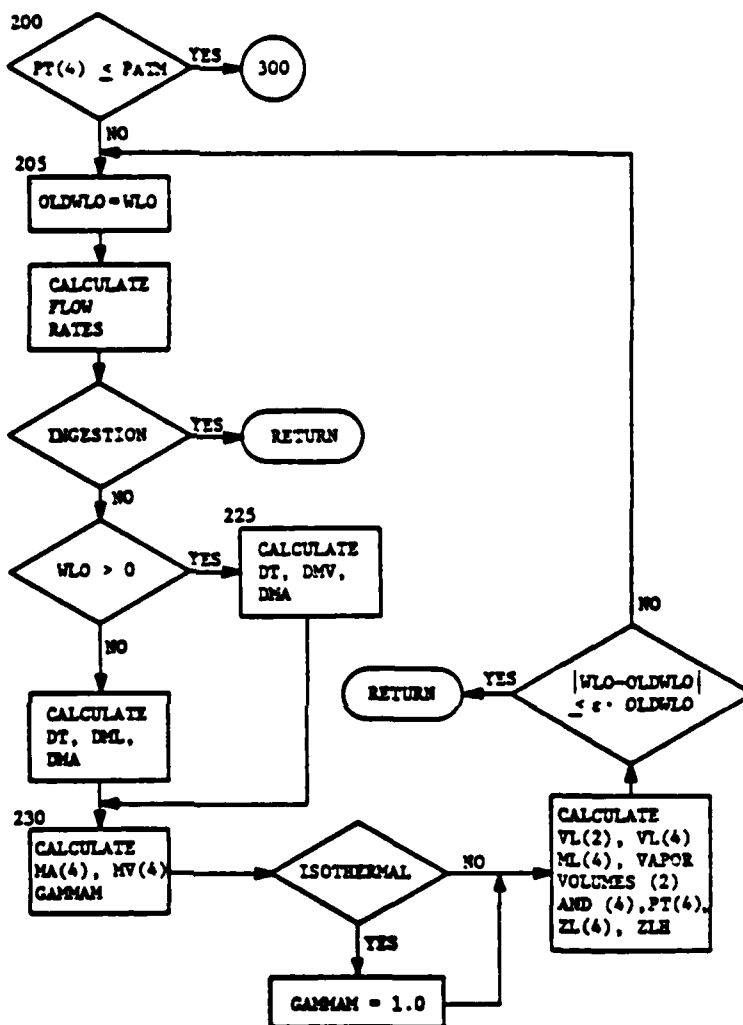


FIGURE C-5. SUBROUTINE NONENER FLOW CHART (Cont'd)



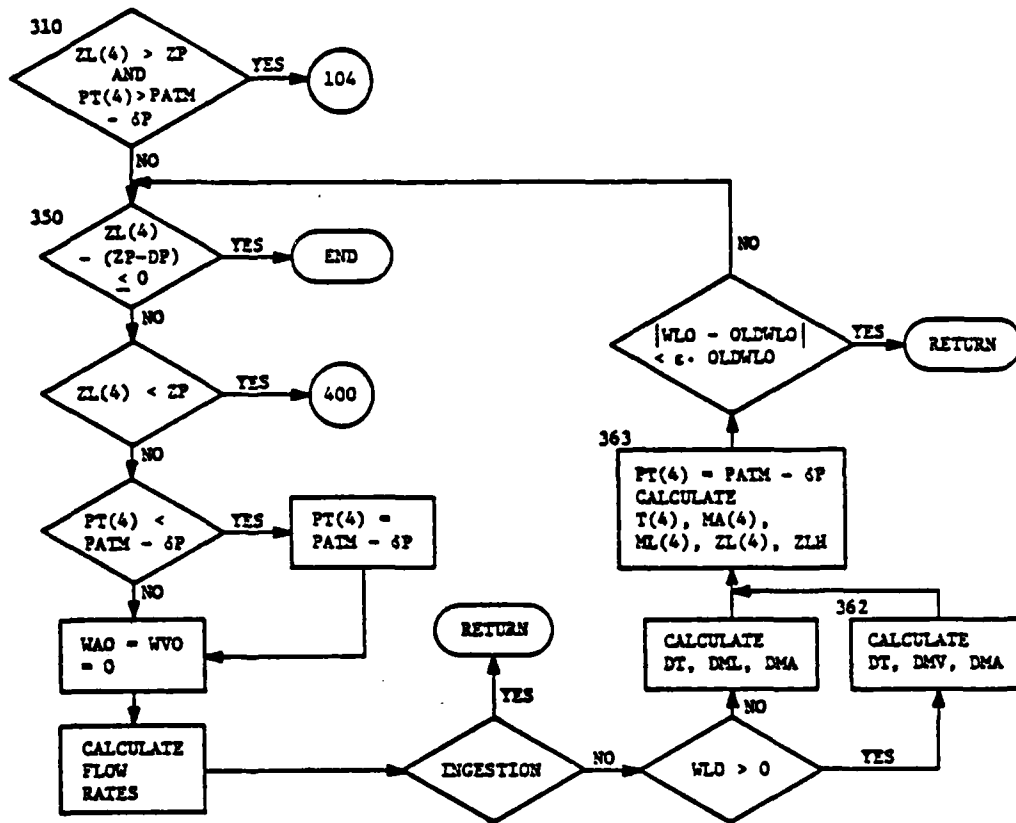


FIGURE C-5. SUBROUTINE NONENER FLOW CHART (Cont'd)

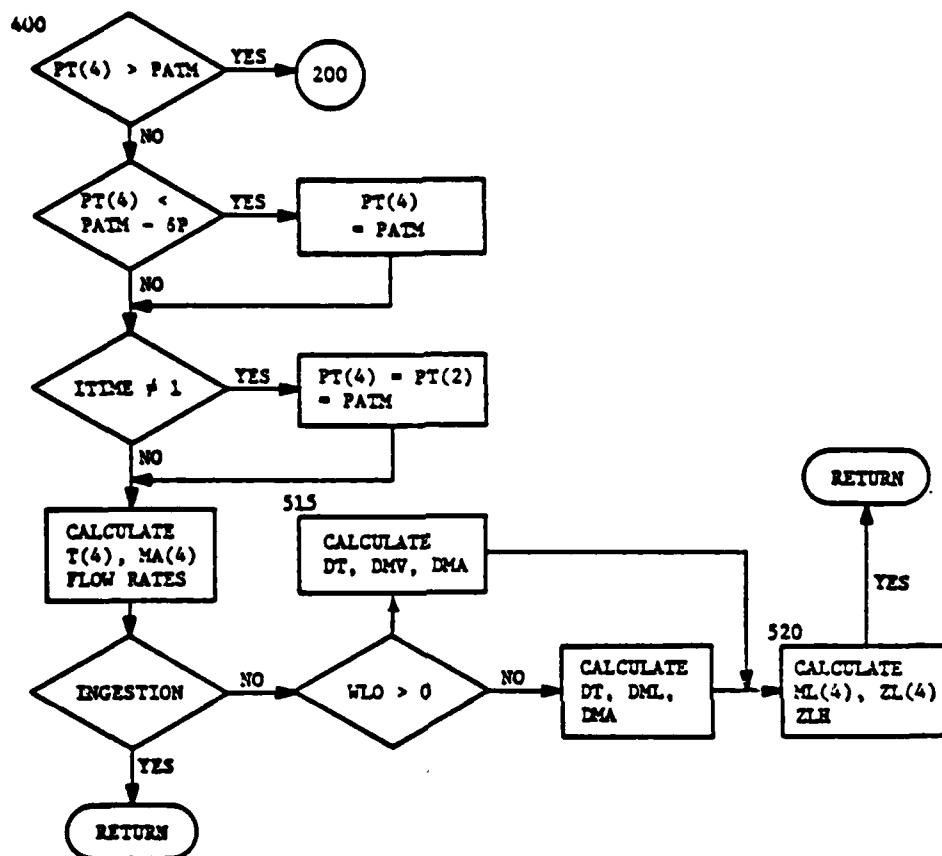


FIGURE C-5. SUBROUTINE NONENER FLOW CHART (Concl'd)

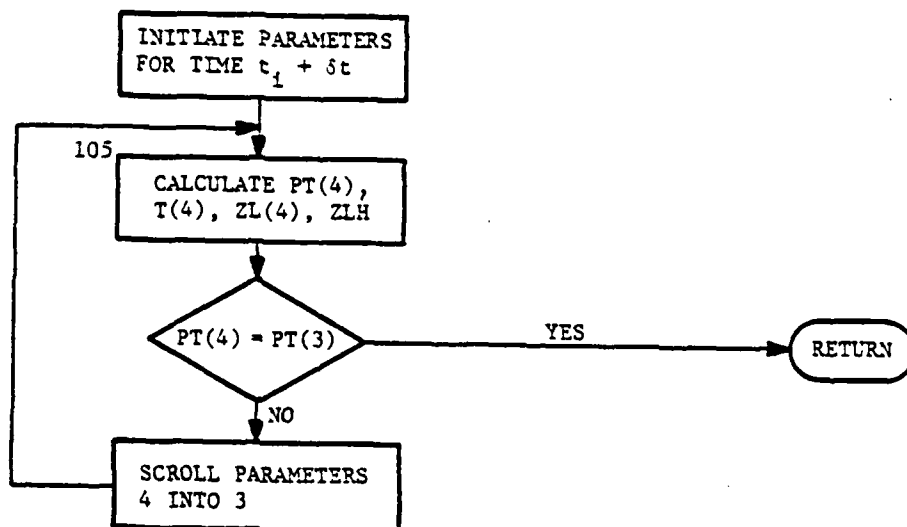
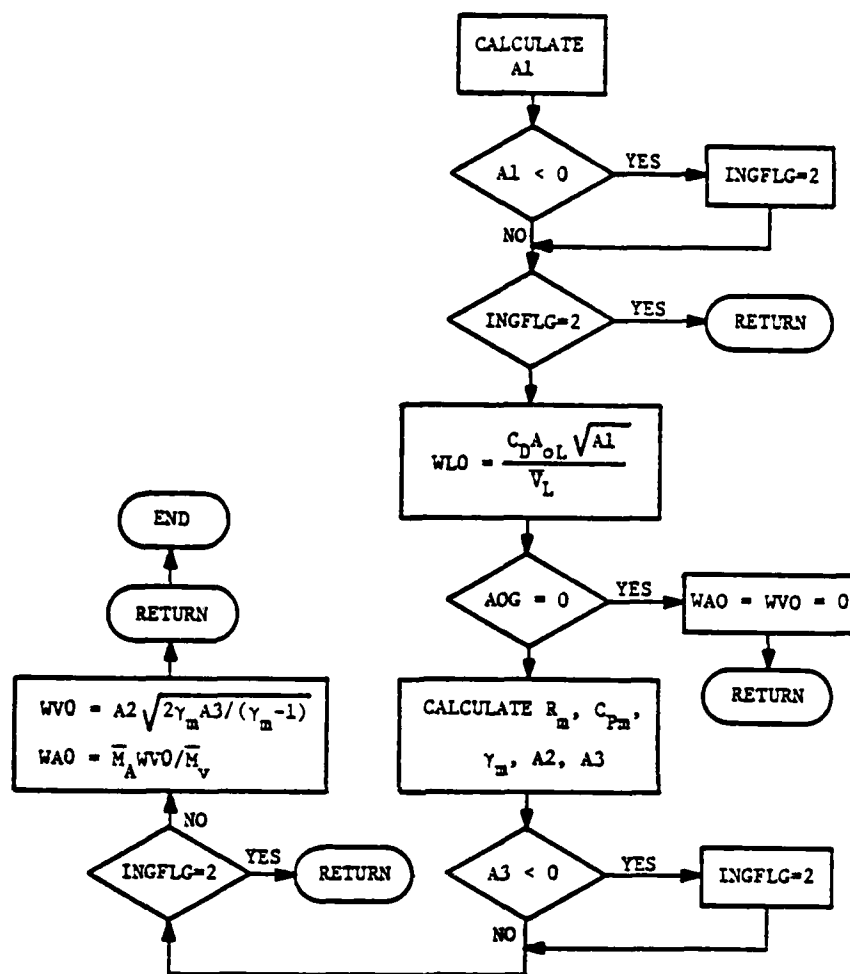


FIGURE C-6. SUBROUTINE NONENG2 (KOUNT)  
FLOW CHART





$$A1 = 2(10^4 (\bar{P}_T - \bar{P}_m) \bar{V}_L + g (\bar{Z}_L - \bar{Z}_{LH}))$$

$$A2 = 10^2 \bar{M}_V C_{D_{OG}} \bar{P}_T / ((\bar{M}_V + \bar{M}_A) \sqrt{R_m \bar{T}})$$

$$A3 = (\bar{P}_m g / \bar{P}_T)^{2/\gamma_m} - (\bar{P}_m g / \bar{P}_T)^{(\gamma_m + 1)/\gamma_m}$$

FIGURE C-7. SUBROUTINE FLOW2 FLOW CHART

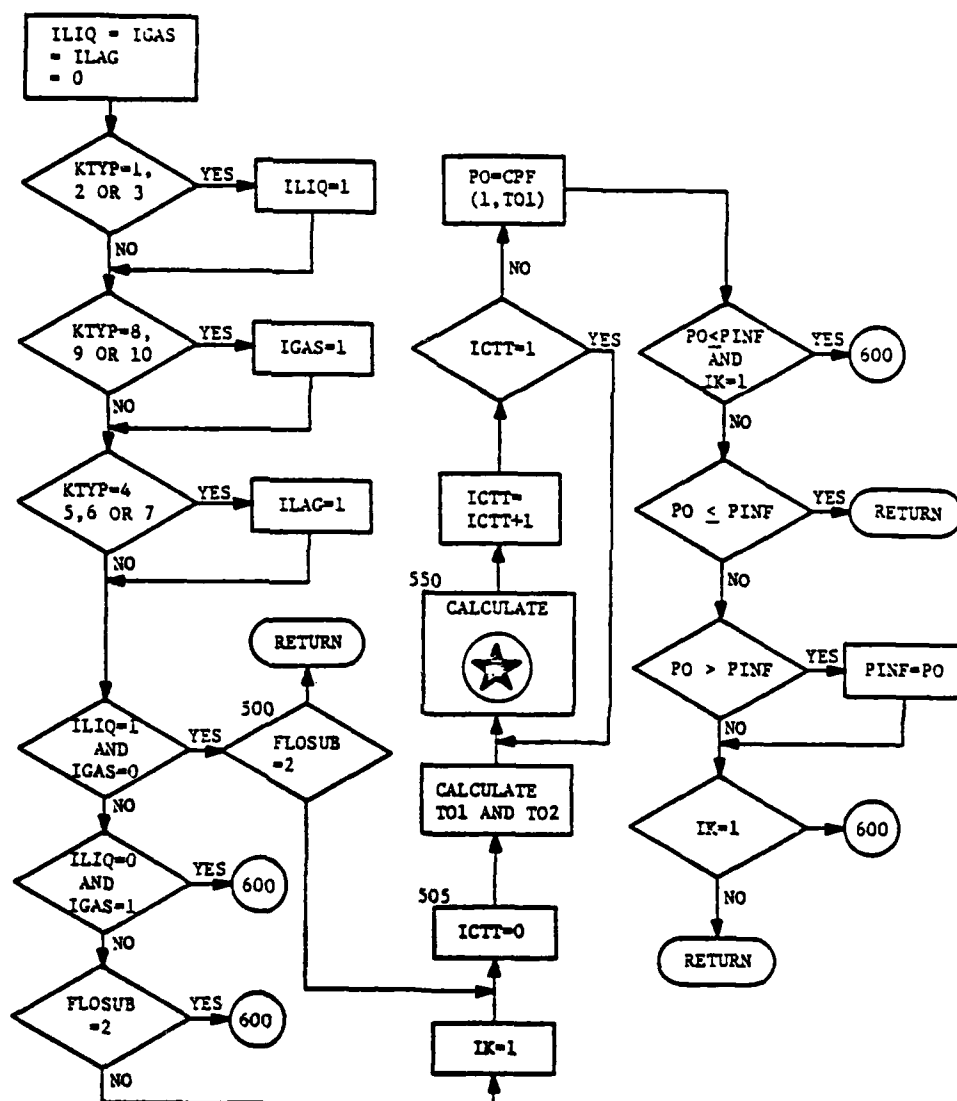


FIGURE C-8. SUBROUTINE CHOKTST (KTYP) FLOW CHART

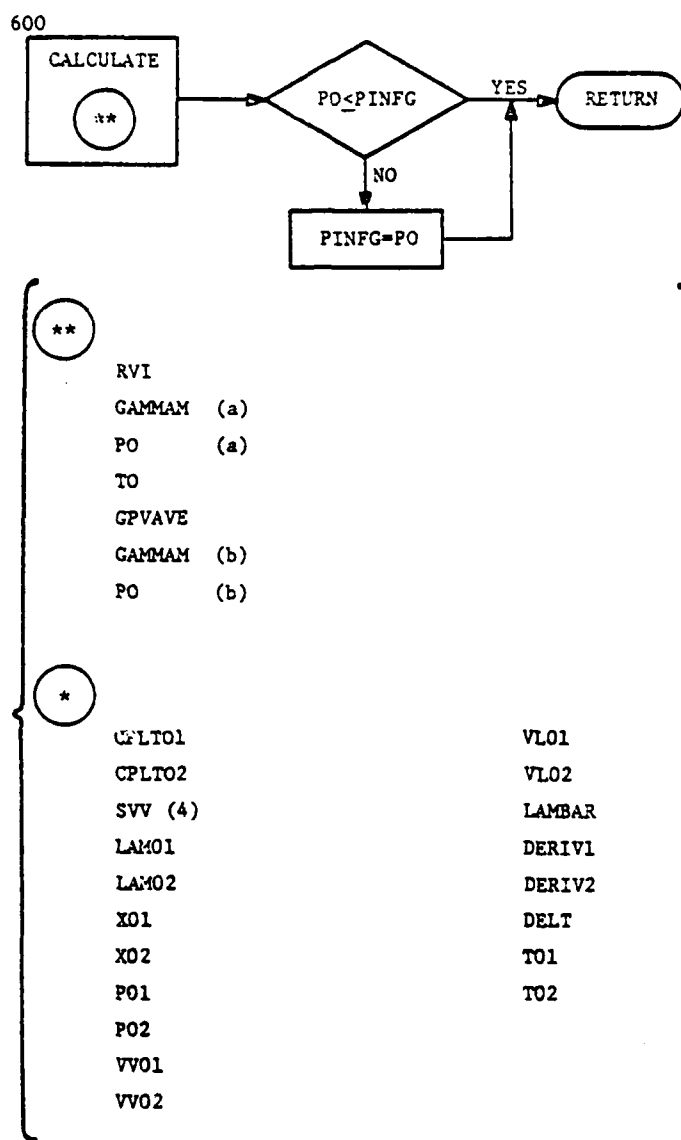


FIGURE C-8. SUBROUTINE CHOKTST (KTYP) FLOW CHART  
(Concl'd)

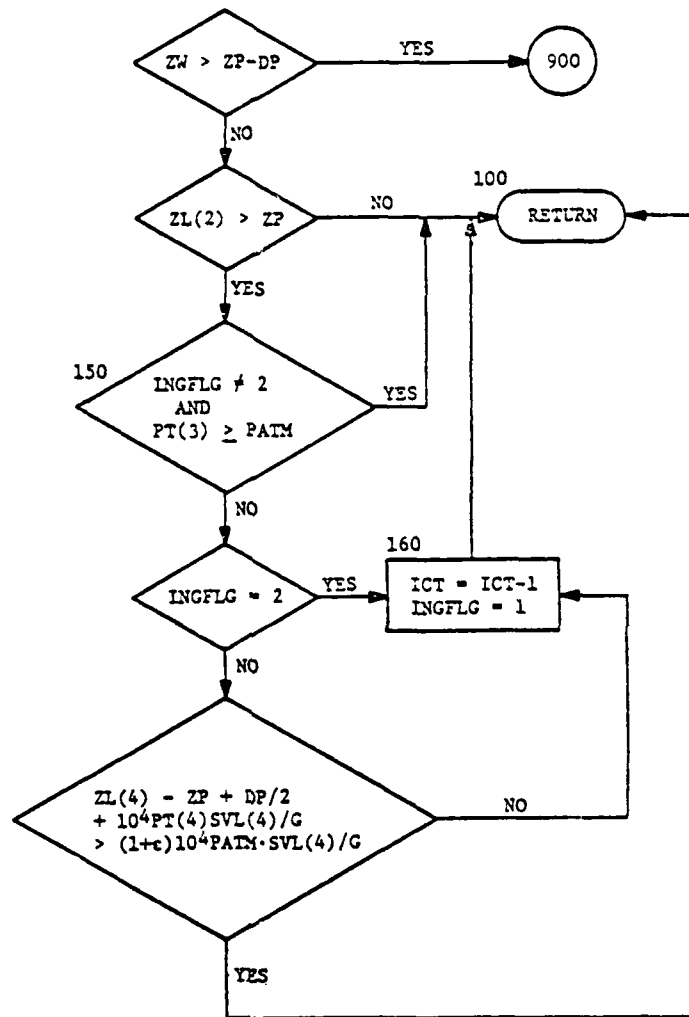


FIGURE C-9. SUBROUTINE INGEST FLOW CHART

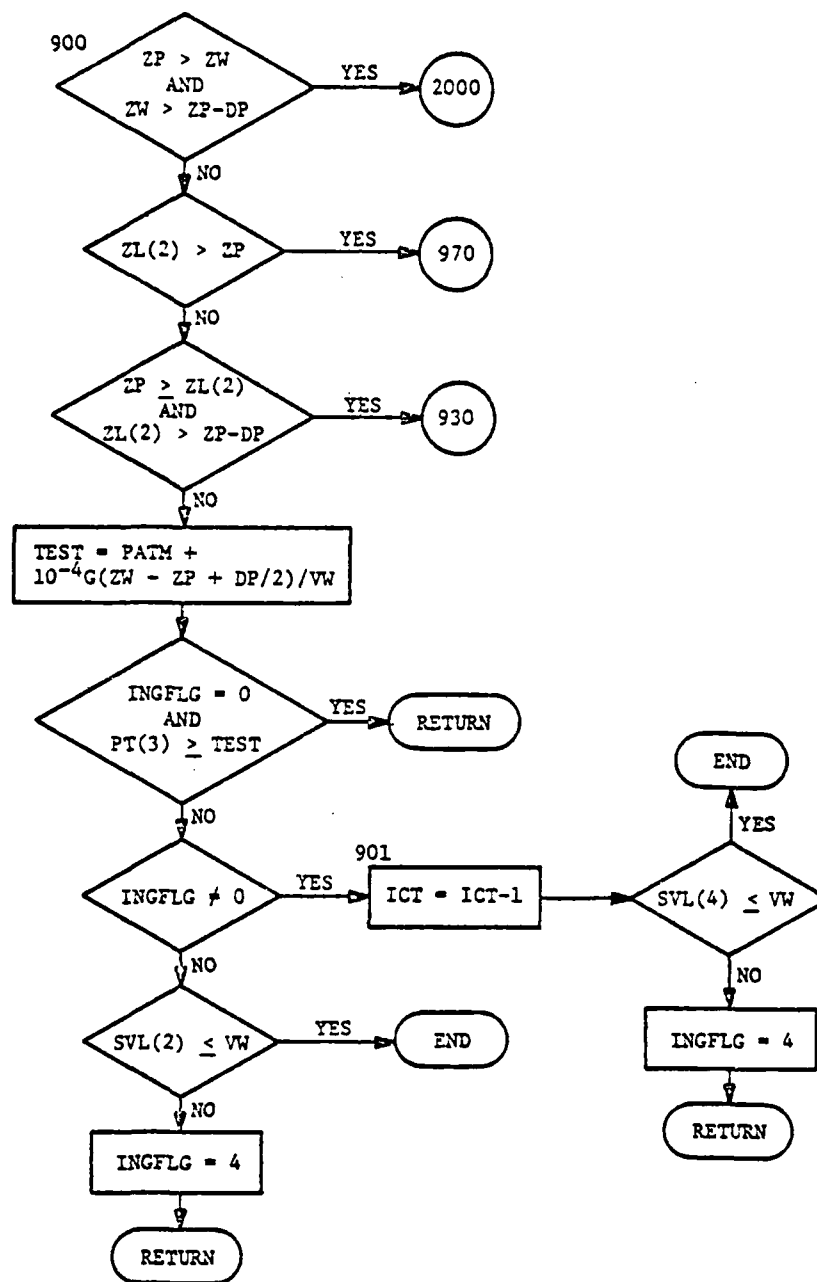


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)

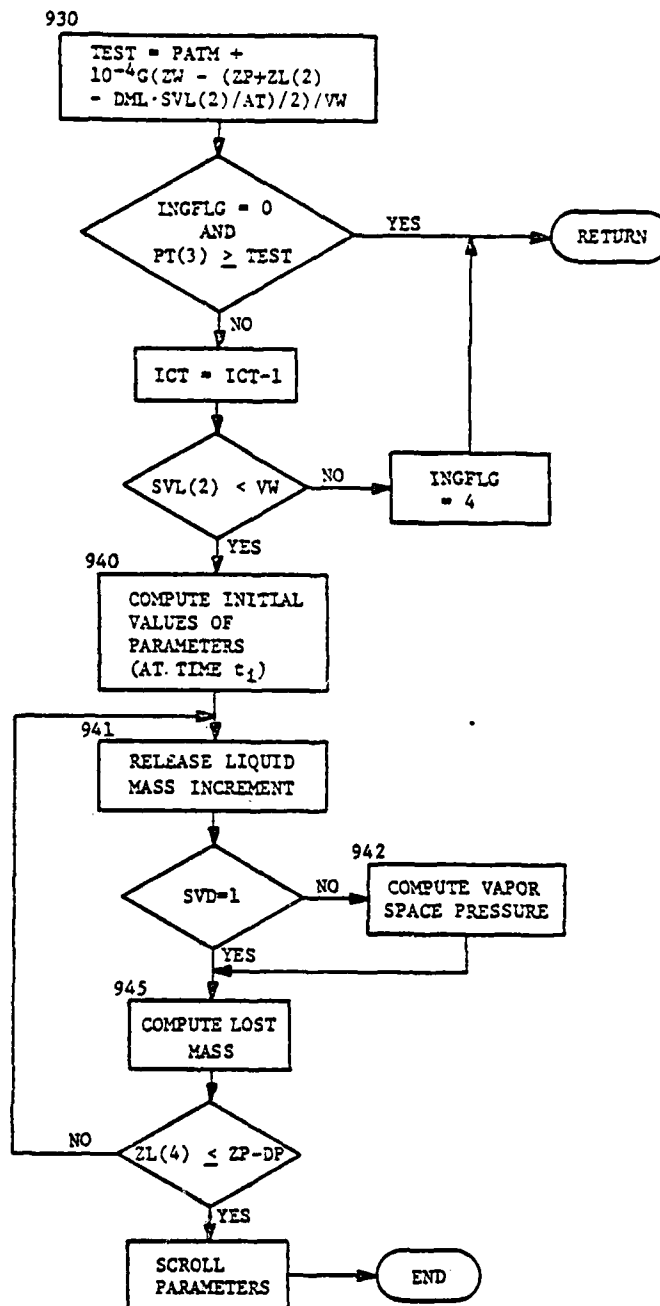


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)

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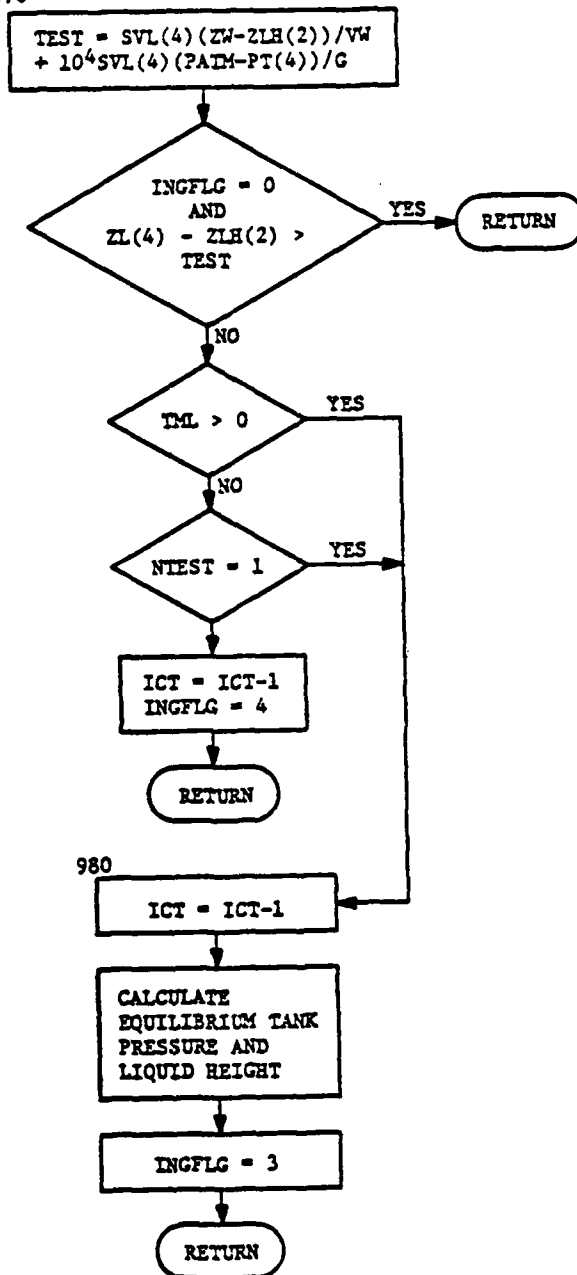


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)

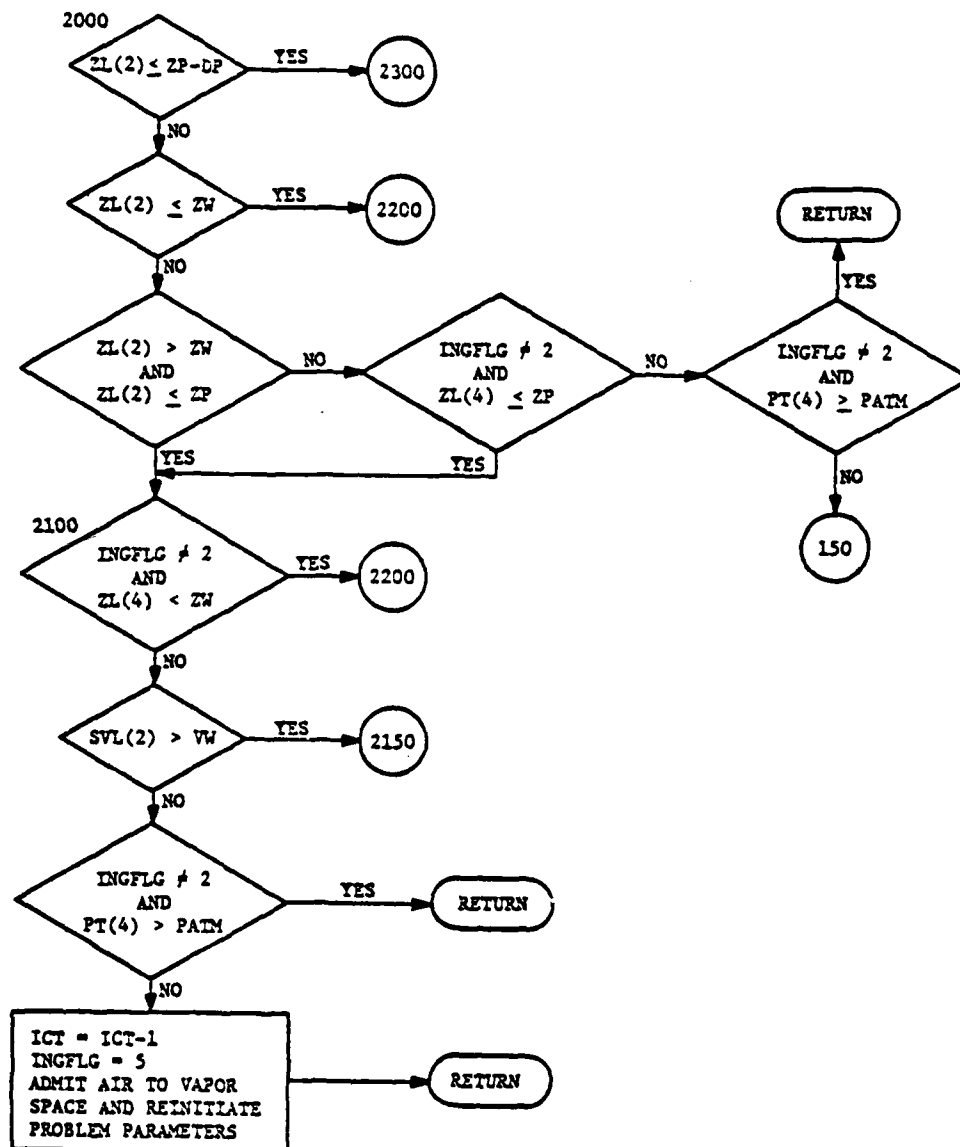


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)



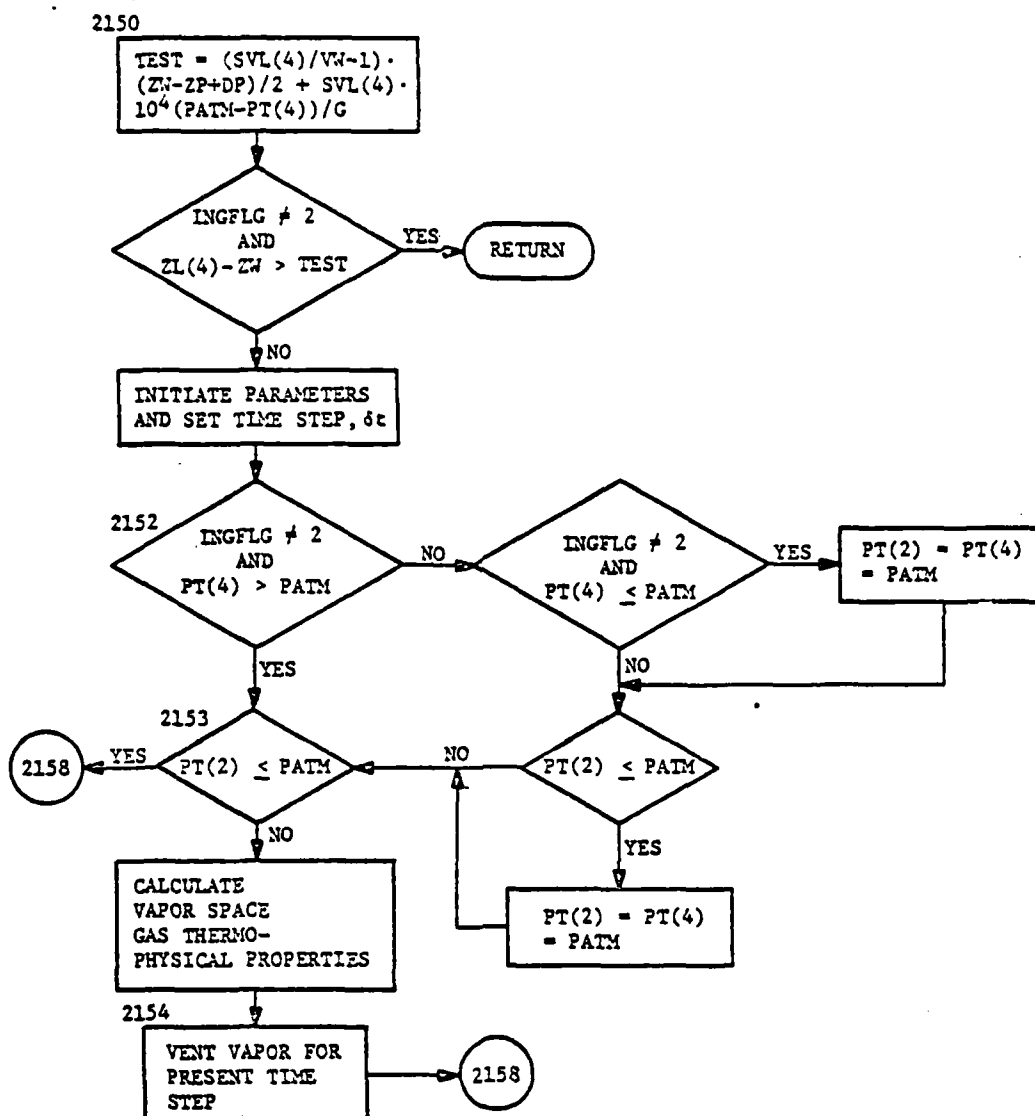


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)

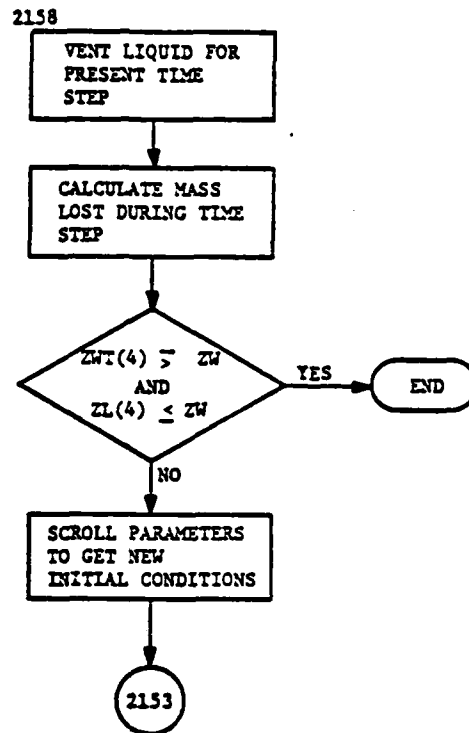


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)

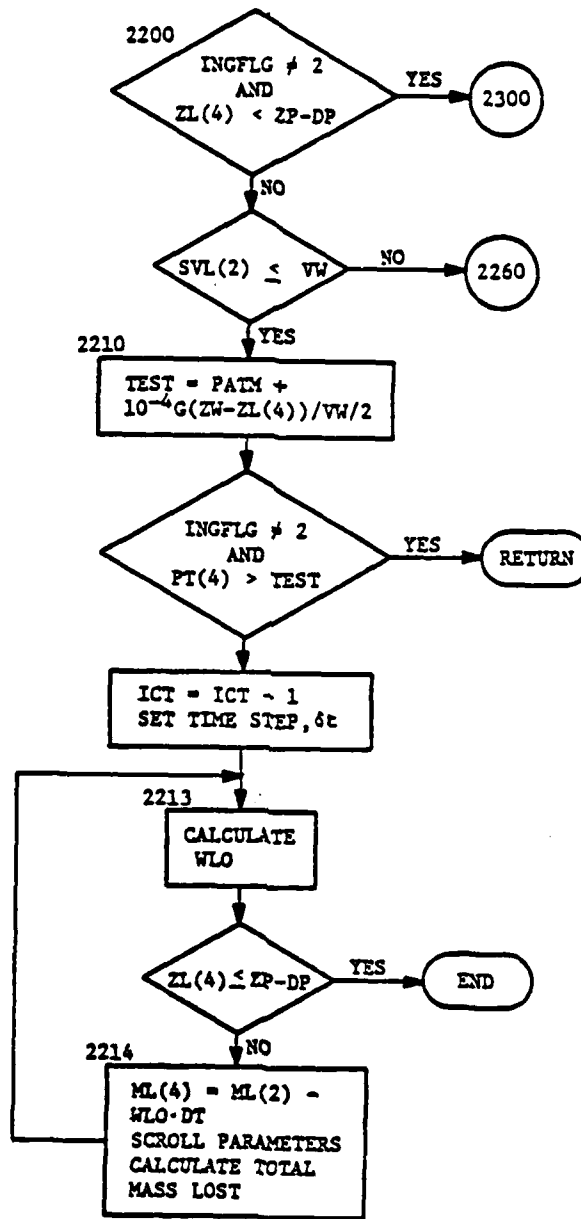


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)

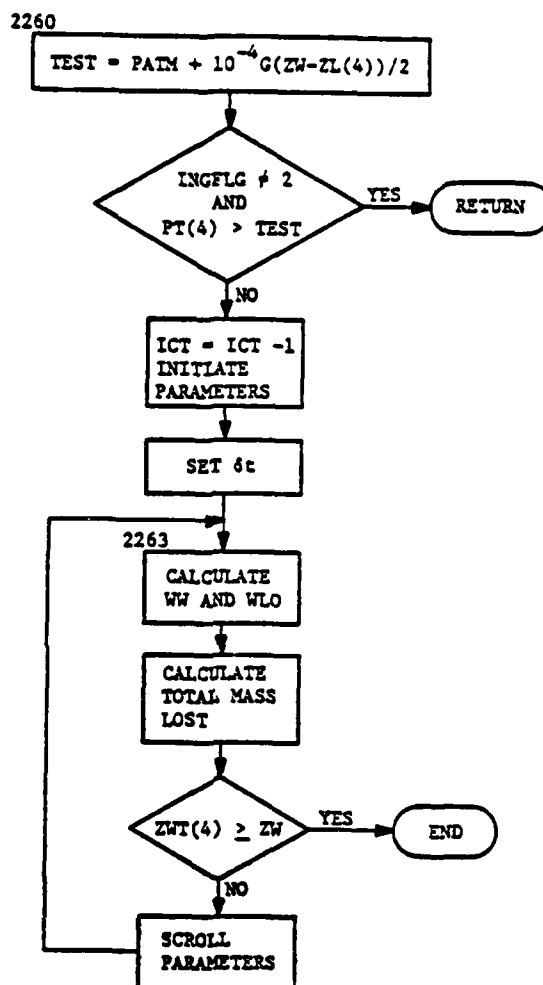


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Cont'd)

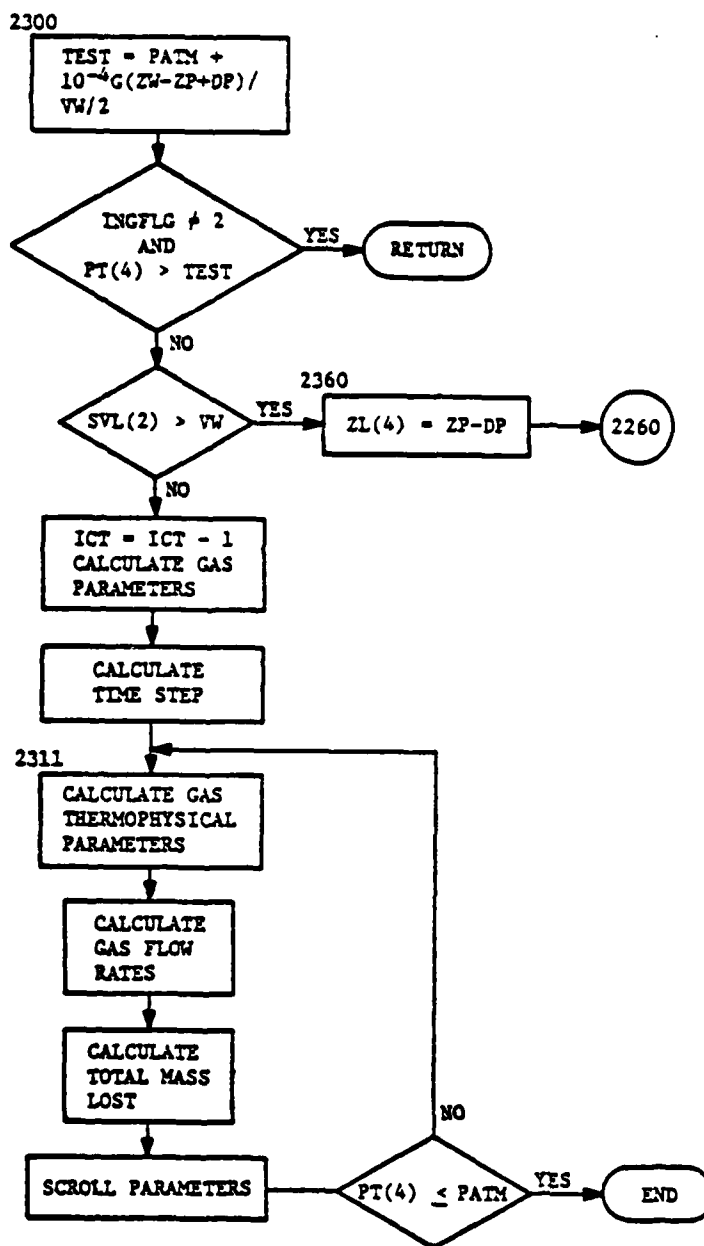


FIGURE C-9. SUBROUTINE INGEST FLOW CHART (Concl'd)

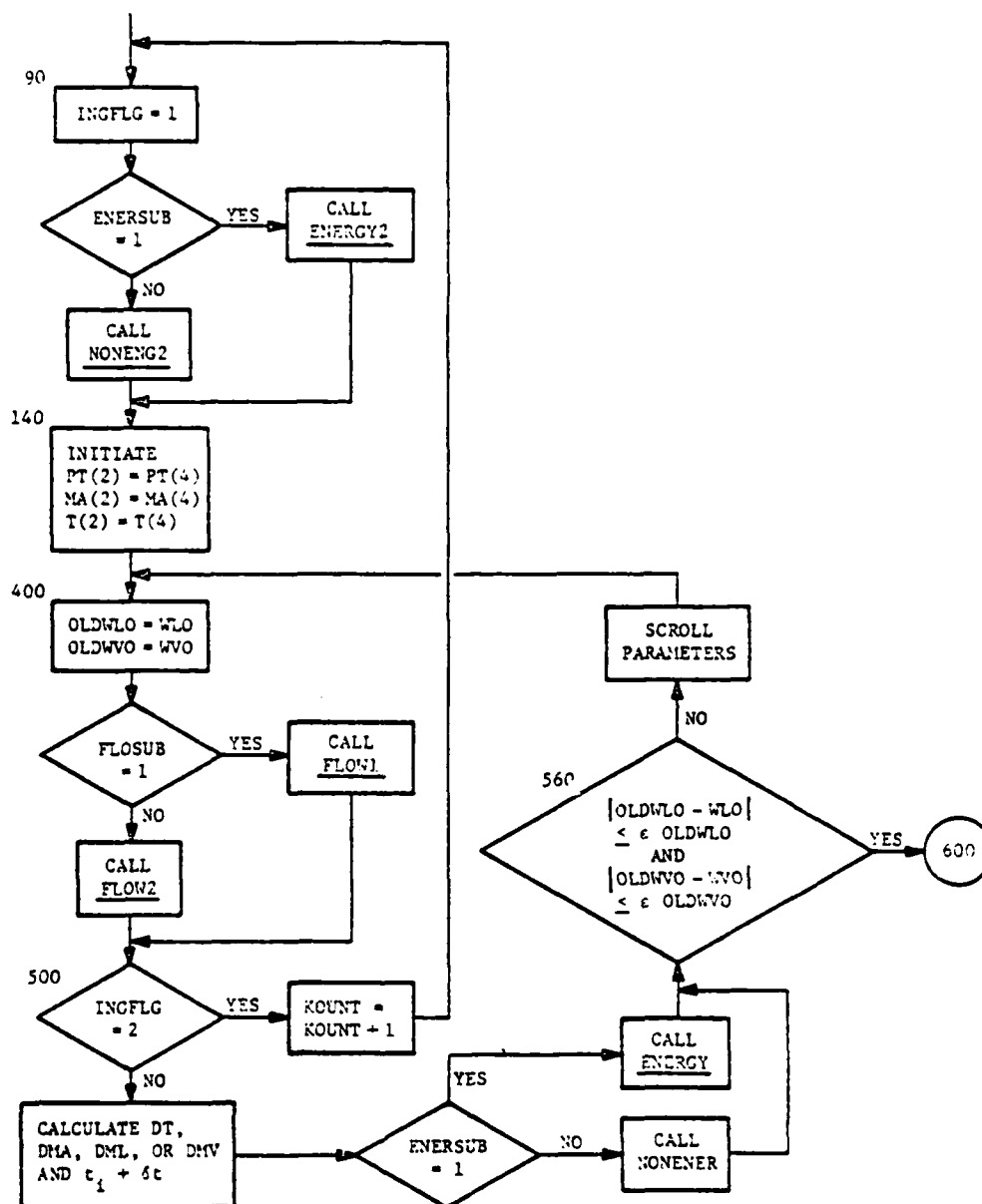


FIGURE C-10. SUBROUTINE AIRIN (ITIME) FLOW CHART

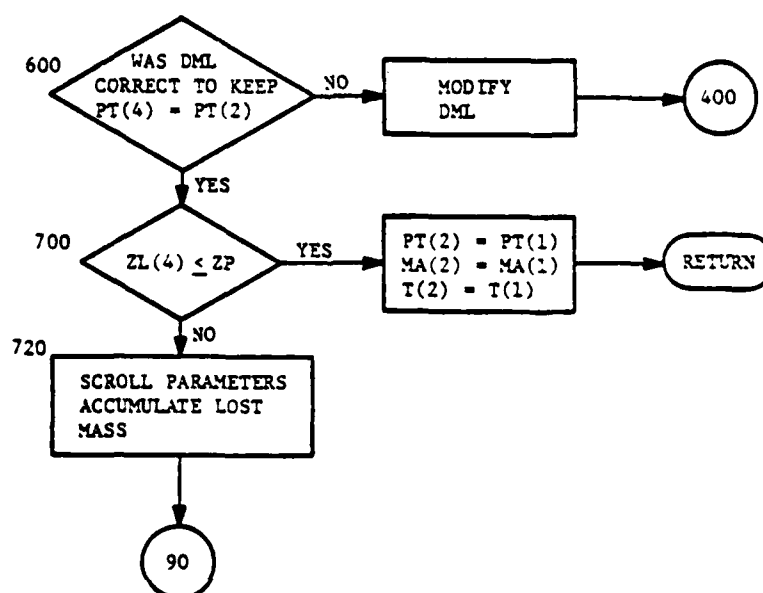


FIGURE C-10. SUBROUTINE AIRIN (ITIME) FLOW CHART  
(Concl'd)

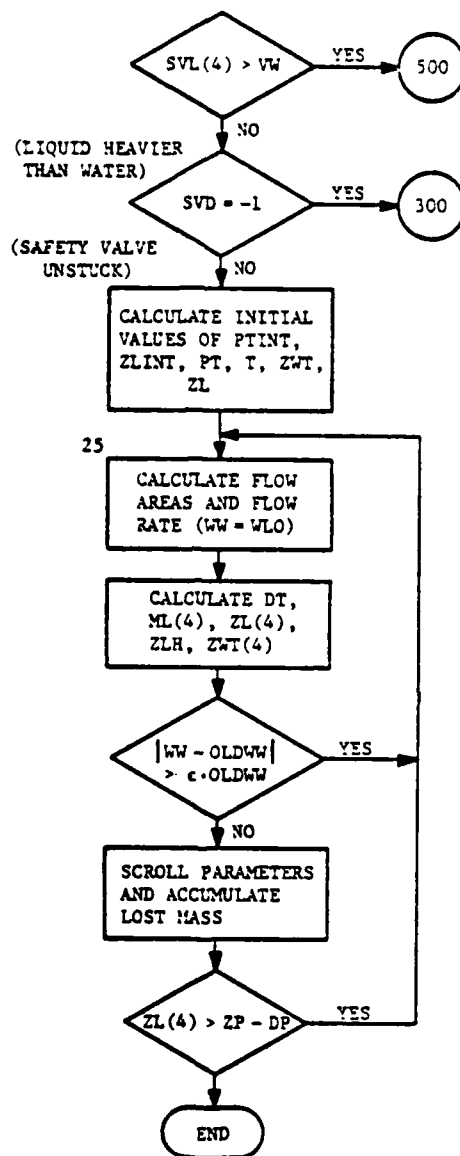


FIGURE C-11. SUBROUTINE WATIN FLOW CHART



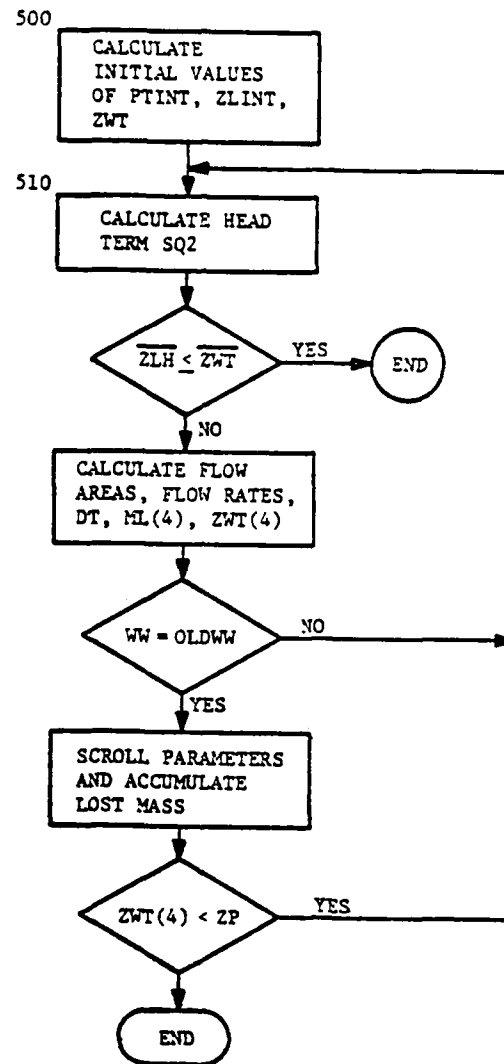


FIGURE C-11. SUBROUTINE WATIN FLOW CHART  
(Cont'd)

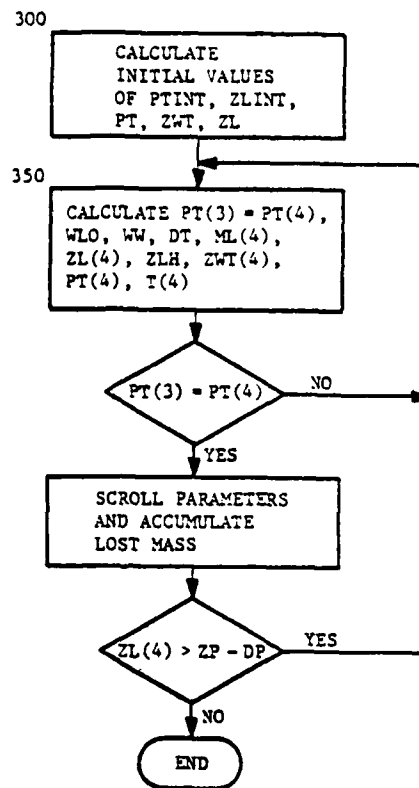


FIGURE C-11. SUBROUTINE WATIN FLOW CHART  
(Concl'd)

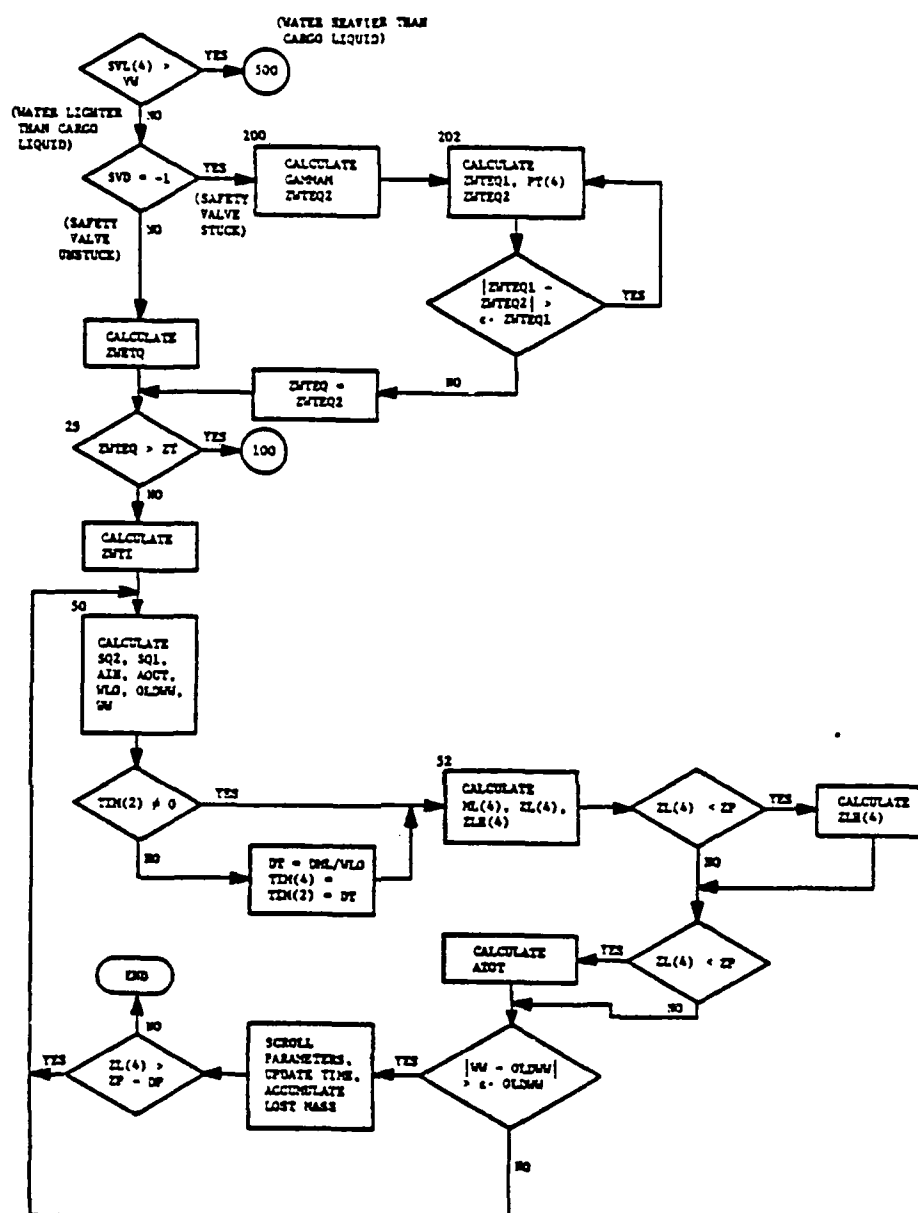


FIGURE C-12. SUBROUTINE NOOUT FLOW CHART

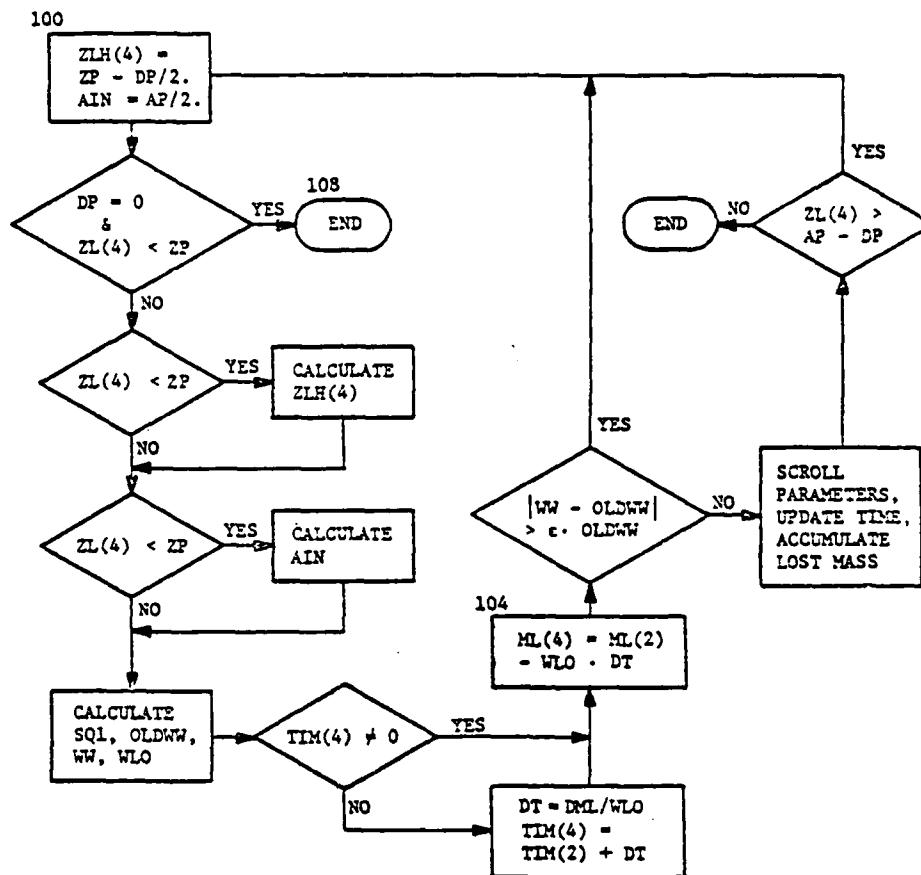


FIGURE C-12. SUBROUTINE NOOUT FLOW CHART  
(Cont'd)

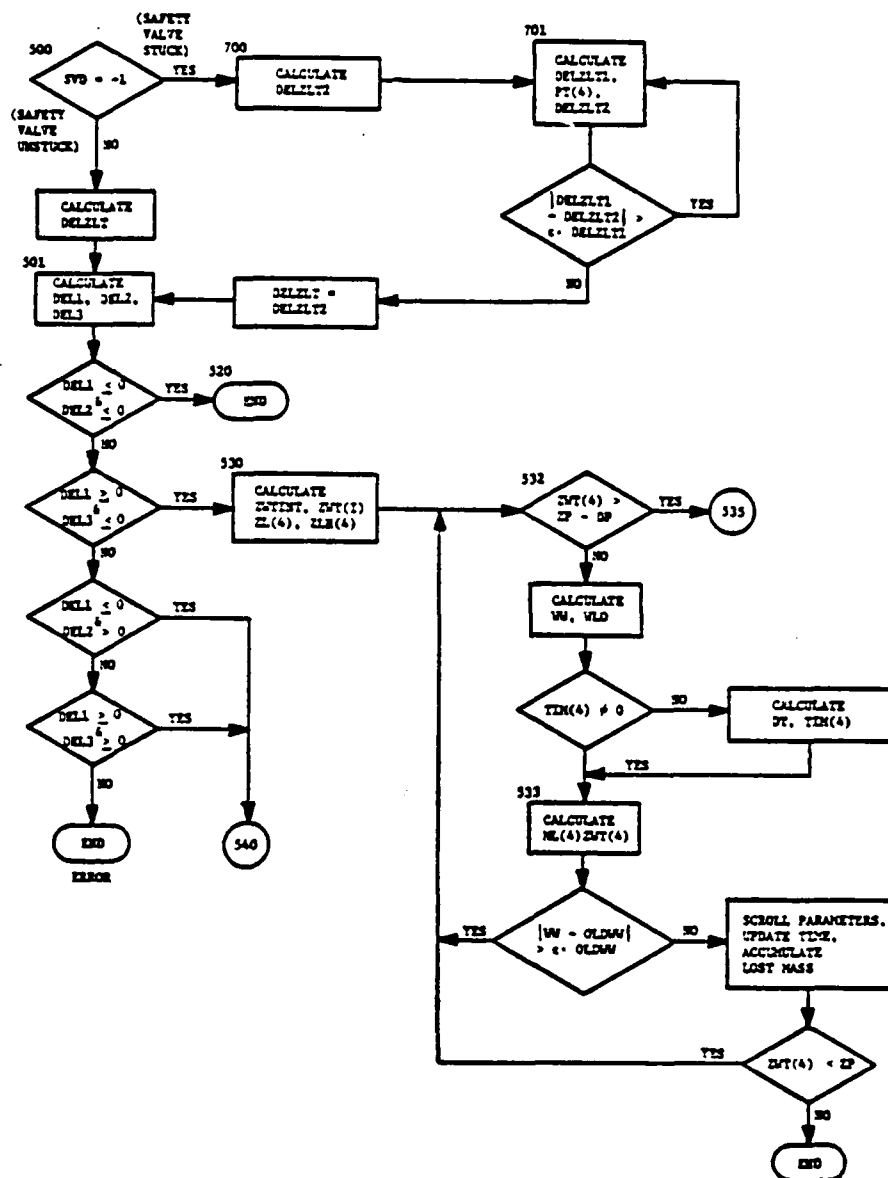


FIGURE C-12. SUBROUTINE NOOUT FLOW CHART  
(Cont'd)

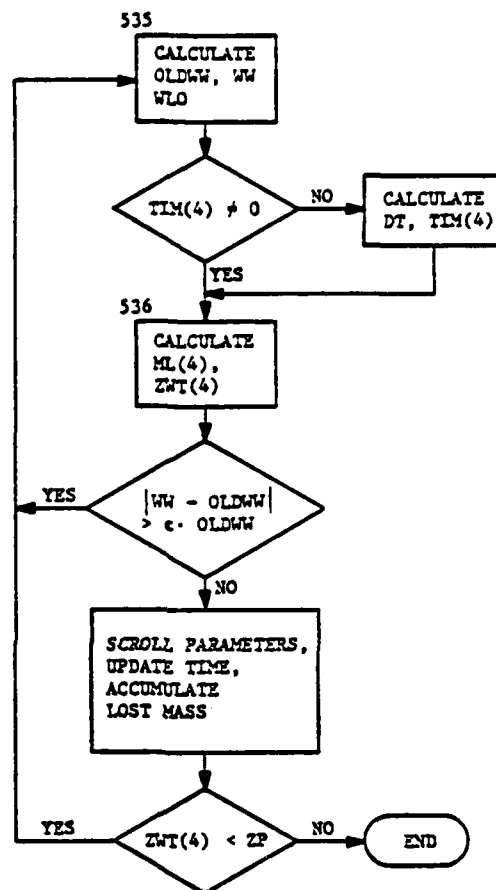


FIGURE C-12. SUBROUTINE NOOUT FLOW CHART  
(Cont'd)

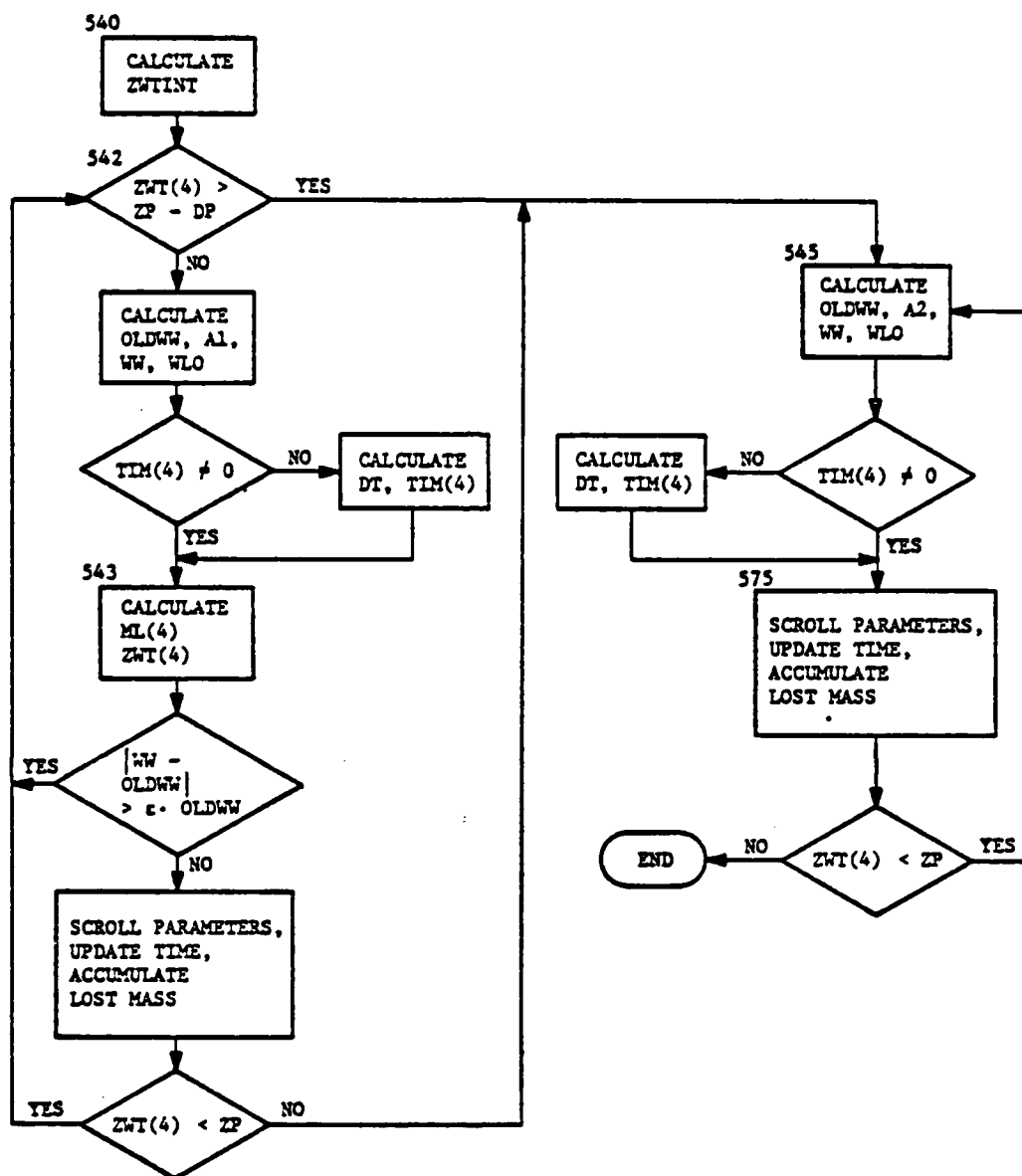


FIGURE C-12. SUBROUTINE NOOUT FLOW CHART  
(Concl'd)

TABLE C.1. VENTING RATE MODEL INPUT

1. Cargo identification code (to be used eventually with the HACS Chemical Properties File)
2. Cargo molecular weight
3. Total initial mass of cargo, grams
4. Cargo initial temperature, °C
5. Volatility constant; used in determining whether the energy balance for volatile or nonvolatile cargos will be used, by comparing saturation pressure at the cargo initial temperature to (volatility constant) times (atmospheric pressure); a value of 0.8 has been found acceptable.
6. Air temperature, °C
7. Air pressure, KPa (Standard Atmosphere = 101.4 KPa)
8. Tank volume, cm<sup>3</sup>
9. Tank height, cm
10. Estimate of tank pressure, KPa; is compared to atmospheric pressure and initial estimate of tank pressure is set to larger of the two.
11. Puncture description: round, horizontal, or vertical slot
12. Height of puncture above tank bottom, cm
13. Puncture area, cm<sup>2</sup>
14. Puncture width, if horizontal or vertical slot, cm
15. Puncture discharge coefficient
16. Safety valve condition: stuck or unstuck
17. Safety valve vacuum differential setting, if unstuck, KPa
18. Height of outside water above tank bottom, cm
- 19-24. Constants in thermodynamic property correlations for vapor pressure (KPa), liquid density (g/cm<sup>3</sup>), vapor compressibility (-), liquid specific heat (cal/g-°C), vapor specific heat (cal/g-°C), and latent heat of evaporation (cal/g).
25. Descriptor for adiabatic or isothermal conditions.



TABLE C-2. MISCELLANEOUS SUBPROGRAMS

## SUBROUTINE AIRMASS:

Uses ideal gas equation of state to determine mass of air in the cargo tank vapor space.

## SUBROUTINE CORRECT:

Updates latent heat, specific volume of vapor, specific volume of liquid, liquid specific heat, and vapor specific heat with the most recent value of temperature at time  $t_{i+1}$ .

## FUNCTION CPLHF (T):

Computes the latent heat of vaporization of the cargo liquid with temperature, T.

## FUNCTION GAMM (RA):

Computes the average value of the ratio of specific heats of the vapor space gases (cargo vapor and, when present, air). The average is taken over times  $t_i$  and  $t_{i+1}$ .

## SUBROUTINE TERMIN (END CODE):

A terminal printout of time, mass, pressure, height, and temperature. TERMIN is entered from another subprogram that is identified by the name given by the Hollerith Field designated by the parameter end code.

## SUBROUTINE ZLHCALC:

Computes the flow center height, ZLH.

TABLE C-3 . CPF (IFLAG,T) FUNCTION SUBPROGRAM

IFLAG - Integer to describe which thermophysical property is to be determined

T - Independent parameter

<u>IFLAG</u>	<u>T</u>	<u>CPF</u>
1	Cargo Temperature	Cargo Material Vapor Pressure, $P = f(T)$
2	Cargo Temperature	Cargo Material Liquid Density
3	Cargo Temperature	Cargo Vapor Compressibility Factor, $Z$
4	Cargo Temperature	Cargo Material Liquid Specific Heat
5	Cargo Temperature	Cargo Material Saturated Vapor Specific Heat at Constant Pressure
6	Cargo Material Vapor Pressure	Cargo Temperature, $T = f^{-1}(P)$

TABLE C-4. SUBROUTINE INGEST FLAG PARAMETERS

One key parameter that determines the computational process is called INGFLG. Values that INGFLG can have and the effect on computation progression are given below:

<u>INGFLG</u>	<u>RESULT</u>
0	Uninterrupted venting - normal mode. Adequate pressure within the tank for liquid cargo and/or gas (cargo vapor and possibly air) to vent through the tank puncture.
2	Insufficient pressure within the tank to vent to the outside media (air and/or water). Initiated in subroutine FLOW1 or FLOW2. Used as a transfer flag to route computation process to subroutine INGEST. INGFLG will be changed to another appropriate value in INGEST.
1	Initiated in subroutine INGEST to route computation process to subroutine AIRIN. If INGFLG is set to 1 in INGEST, return from INGEST to the main program, TVENT, is followed by a call to subroutine AIRIN. Return to TVENT from AIRIN causes INGFLG to be reset to 0.
3	Initiated in subroutine INGEST. Causes return to the main program, TVENT, and an immediate call to subroutine WATIN.
4	Initiated in subroutine INGEST. Causes return to the main program, TVENT, and an immediate call to subroutine NOOUT.
5	Initiated in subroutine INGEST. Causes air to be admitted to the vapor space in the tank instantaneously. Return to the main program, TVENT, follows, whereupon in TVENT, INGFLG is reset to 0 and the cargo release loop is reentered.

TABLE C.5. LIST OF VARIABLES IN COMPUTER PROGRAM

<u>Variable</u>	<u>Units</u>	<u>Common Label</u>	<u>Description</u>	<u>Routines Where Referenced</u>
AIRK	—	ENG	Air bubble volume coefficient	TVENT, AIRIN
AOG	cm <sup>2</sup>	GEN	Area of gas flow	TVENT, FLOW1, FLOW2, INGEST
AOL	cm <sup>2</sup>	GEN	Area of liquid flow	TVENT, FLOW1, FLOW2
AP	cm <sup>2</sup>	ENG	Area of puncture	TVENT, AIRIN, NOOUT, WATIN
AT	cm <sup>2</sup>	GEN	Tank cross-sectional area	TVENT, AIRMASS, ENERGY, ENERGY2, INGEST, NONENER, NONENG2, NOOUT, WATIN
C1, C2	—	CP	Constants used in calculating specific heat of cargo	TVENT, CPF
CDORF (CD)	—	GEN	Discharge coefficient	TVENT, FLOW1, FLOW2, INGEST, NOOUT, WATIN
CMI	g	GEN	Initial cargo mass	TVENT
CMT (MOLE)	$\frac{g}{g-mole}$	GEN	Cargo molecular weight	TVENT, AIRMASS, CHOKTST, CORRECT, ENERGY, ENERGY2, FLOW1, FLOW2, GAMM, INGEST, NONENER, NONENG2
CPA	$\frac{cal}{g-^{\circ}C}$	GEN	Specific heat of air	TVENT, ENERGY, ENERGY2, FLOW1, FLOW2, GAMM, INGEST
CPL	$\frac{cal}{g-^{\circ}C}$	GEN	Specific heat of cargo liquid	TVENT, AIRIN, CORRECT, ENERGY, ENERGY2, GAMM, INGEST, WATIN
CPVBAR	$\frac{cal}{g-^{\circ}C}$	GEN	Specific heat of cargo vapor	TVENT, AIRIN, CORRECT, ENERGY, ENERGY2, GAMM, INGEST, WATIN
D1, D2	—	CP	Constants used in calculating liquid density	TVENT, CPF
DELTIME	s	ENG	Evaporated mass per time step	TVENT, ENERGY, ENERGY2, NOOUT
DMA	g	ENG	Mass of air released per time step	TVENT, AIRIN, ENERGY, NONENER
DML	g	ENG	Mass of liquid released per time step	TVENT, AIRIN, ENERGY, INGEST, NONENER, NOOUT, WATIN

TABLE C.5. LIST OF VARIABLES IN COMPUTER PROGRAM (Cont'd)

<u>Variable</u>	<u>Units</u>	<u>Common Label</u>	<u>Description</u>	<u>Routines Where Referenced</u>
DMV	g	ENG	Mass of vapor released per time step	TVENT, AIRIN, ENERGY, INGEST, NONENER
DP	cm	ENG	Vertical extent of vertical slots	TVENT, ENERGY, INGEST, NONENER, NOOUT, WATIN, ZLRALC
DPV (DELPVAL)	KPa	GEN	Safety valve relief vacuum setting	TVENT, ENERGY, INGEST, NONENER
ENRSUB	—	ENG	Selects energy subroutine	TVENT, AIRIN
FLOSUB	—	ENG	Selects flow subroutine	TVENT, AIRIN, CRCKTST
G	$\frac{\text{cm}}{\text{sec}^2}$	GEN	Gravitational acceleration	TVENT, AIRIN, FLOW1, FLOW2, INGEST, NOOUT, WATIN
H1 + H7	—	CP	Constants for latent heat of cargo	TVENT, CPLYH
HTD (IHEATTR)	—	GEN	Heat transfer descriptor	TVENT, ENERGY, ENERGY2, INGEST, NONENER, NONENG2, WATIN
ICT	—	ENG	Time step counter	TVENT, INGEST
INGFLG	—	ENG	Subroutine flag used in INGEST	TVENT, AIRIN, ENERGY, FLOW1, FLOW2, INGEST, NONENER
J	$\frac{\text{g-cm}^2}{\text{cal-sec}^2}$	GEN	Mechanical equivalent of heat	TVENT, AIRMASS, FLOW1
LAMBDA	$\frac{\text{cal}}{\text{g}}$	GEN	Latent heat of cargo	TVENT, CORRECT, ENERGY, ENER2
MA	g	GEN	Air mass in tank	TVENT, AIRIN, AIRMASS, ENERGY, ENERGY2, FLOW1, FLOW2, GAMM, INGEST, NONENER, NONENG2, NOOUT, TERMIN, WATIN
ML	g	GEN	Liquid mass in tank	TVENT, AIRIN, AIRMASS, CORRECT, ENERGY, ENERGY2, INGEST, NONENER, NONENG2, NOOUT, TERMIN, WATIN
MV	g	GEN	Vapor mass in tank	TVENT, AIRIN, AIRMASS, CORRECT, ENERGY, ENERGY2, FLOW1, FLOW2, GAMM, INGEST, NONENER, NONENG2, NOOUT, TERMIN, WATIN

TABLE C.5. LIST OF VARIABLES IN COMPUTER PROGRAM (Cont'd)

Variable	Units	Common Label	Description	Routines Where Referenced
NTEST	—	ENG	Loop counter through main program	TVENT, ENGEST
P1-P3	—	CP	Constants for calculating vapor pressure	TVENT, CHOKTST, CFF
PA	KPa	GEN	Air pressure in the tank	TVENT, AIRIN, ENERGY, WATIN
PATM	KPa	GEN	Atmospheric pressure	TVENT, ENERGY, ENGEST, NONENER
PENT	KPa	GEN	Ambient pressure external to puncture	TVENT, CHOKTST, FLOW1, FLOW2, ENGEST, NOOUT
PENFG	KPa	GEN	Ambient pressure external to puncture for gas flow	TVENT, CHOKTST, FLOW1, FLOW2
PT	KPa	GEN	Tank pressure	TVENT, AIRIN, AIRMASS, CHOKTST, CORRECT, ENERGY, ENERGY2, FLOW1, FLOW2, ENGEST, NONENER, NONENG2, NOOUT, TERMIN, WATIN
PTEQI	KPa	WAT	Tank pressure at equilibrium	TVENT, ENGEST, WATIN
PV	KPa	GEN	Vapor pressure in the tank	TVENT, AIRIN, AIRMASS, CHOKTST, CORRECT, ENERGY, ENERGY2, ENGEST, NONENER, NONENG2, NOOUT, TERMIN, WATIN
RG	$\frac{\text{KPa-cm}^3}{\text{g-mole-}^\circ\text{K}}$	GEN	Universal gas constant, mechanical units	TVENT, AIRMASS, CHOKTST, CORRECT, ENERGY, ENERGY2, FLOW1, FLOW2, GAMM, ENGEST, NONENER, NONENG2, NOOUT, WATIN
RG1	$\frac{\text{cal}}{\text{g-mole-}^\circ\text{K}}$	GEN	Universal gas constant, thermal units	TVENT, FLOW1, FLOW2, GAMM, ENGEST
SVA	$\frac{\text{cm}^3}{\text{g}}$	GEN	Specific volume of air	TVENT, ENERGY2, NONENG2
SVD (IVALTE)	—	GEN	Safety valve descriptor	TVENT, ENERGY, ENGEST, NONENER, NOOUT, WATIN
SVL (VL)	$\frac{\text{cm}^3}{\text{g}}$	GEN	Specific volume of liquid	TVENT, AIRIN, CORRECT, ENERGY, ENERGY2, ENGEST, NONENER, NONENG2, NOOUT, WATIN

TABLE C.5. LIST OF VARIABLES IN COMPUTER PROGRAM (Cont'd)

<u>Variable</u>	<u>Units</u>	<u>Common Label</u>	<u>Description</u>	<u>Routines Where Referenced</u>
SVV (VV)	$\frac{\text{cm}^3}{\text{g}}$	GEN	Specific volume of vapor	TVENT, AIRIN, CHOKIST, CORRECT, ENERGY, ENERGY2, INGEST, WATIN
T	°C	GEN	Cargo temperature	TVENT, AIRIN, AIRMASS, CHOKIST, CORRECT, CPF, CPLEF, ENERGY, ENERGY2, FLOW1, FLOW2, GAMM, INGEST, NONENER, NONENG2, NOOUT, TERMIN, WATIN
TC	°C	GEN	Initial cargo temperature	TVENT, NONENER, NONENG2
TIME (TIM)	sec	GEN	Time	TVENT, AIRIN, ENERGY, INGEST, NOOUT, TERMIN, WATIN
TINF (TATM)	°C	GEN	Air temperature	TVENT, ENERGY, ENERGY2
TMA	g	ENG	Total mass of air released from tank	TVENT, AIRIN, INGEST, NOOUT, TERMIN, WATIN
TML	g	ENG	Total mass of liquid released from tank	TVENT, AIRIN, INGEST, NOOUT, TERMIN, WATIN
TMV	g	ENG	Total mass of vapor released from tank	TVENT, AIRIN, INGEST, NOOUT, TERMIN, WATIN
V1-V3	—	CP	Constants used in calculating specific heat of vapor	TVENT, CPF
VOLB	$\text{cm}^3$	ENG	Volume of air bubble	TVENT, AIRIN, ENERGY2, NONENG
VT (VOLT)	$\text{cm}^3$	GEN	Tank volume	TVENT, AIRMASS, ENERGY, ENERGY2, INGEST, NONENER, NONENG2, NOOUT, WATIN
VW	$\frac{\text{cm}^3}{\text{g}}$	GEN	Specific volume of water	TVENT, INGEST, NOOUT, WATIN
WAO	$\frac{\text{g}}{\text{sec}}$	GEN	Discharge rate of air released	TVENT, AIRIN, ENERGY, ENERGY2, FLOW1, FLOW2, INGEST, NONENER, NONENG2
WLO	$\frac{\text{g}}{\text{sec}}$	GEN	Discharge rate of liquid released	TVENT, AIRIN, ENERGY, ENERGY2, FLOW1, FLOW2, INGEST, NONENER, NONENG2, NOOUT, WATIN
WVO	$\frac{\text{g}}{\text{sec}}$	GEN	Discharge rate of vapor released	TVENT, AIRIN, ENERGY, ENERGY2, FLOW1, FLOW2, INGEST, NONENER, NONENG2

TABLE C.5. LIST OF VARIABLES IN COMPUTER PROGRAM (Concl'd)

<u>Variable</u>	<u>Units</u>	<u>Common Label</u>	<u>Description</u>	<u>Routines Where Referenced</u>
SLP	cm	ENG	Slot width	TVENT, INGEST, NOOUT
Z1 - Z3	—	CP	Constants used for calculating vapor compressibility	TVENT, CPF
ZL	cm	GEN	Height of liquid in- side tank	TVENT, AIRIN, AIRMASS, ENERGY, ENERGY2, FLOW1, FLOW2, INGEST, NONENER, NONENG2, NOOUT, TERMIN, WATIN, ZLECALC
ZLEQI	cm	WAT	Equilibrium height of liquid in the tank	TVENT, INGEST, WATIN
ZLC	cm	GEN	Height of center of flow through puncture	TVENT, AIRIN, ENERGY, ENERGY2, FLOW1, FLOW2, INGEST, NONENER, NONENG2, NOOUT, WATIN, ZLECALC
ZP	cm	GEN	Puncture height	TVENT, AIRIN, ENERGY, INGEST, NONENER, NOOUT, WATIN, ZLECALC
ZT	cm	GEN	Tank height	TVENT, INGEST, NOOUT
ZW	cm	GEN	Height of water out- side tank	TVENT, INGEST, NOOUT, ZLECALC
ZWT	cm	ENG	Height of water in- side tank	TVENT, INGEST, NOOUT, WATIN



TABLE C.6. FORMAT FOR INPUT DATA TO TVENT PROGRAM

Data may be input to the TVENT program through a series of 13 input card sets. Each series corresponds to one data case. Multiple cases may be tested by making only one run of TVENT. To run more than one case, replace the INPUT CARD THIRTEEN (end-of-file card) with an end-of-record card (a 7-8-9 punch in column one). Then repeat the series of 13 input card sets for each case to be tested. For the last case use the INPUT CARD SET THIRTEEN instead of the end-of-record card.

INPUT CARD SET ONE (READ (12,5000) CID, CMWT, CMI, TC, CVOL)

Input the cargo parameters using the standard FORTRAN format statement: FORMAT (A3, 7X, 6E10.0).

<u>Variable</u>	<u>Description</u>	<u>Units</u>
1. CID	Cargo identification	--
2. CMWT	Cargo molecular weight	gram/gram-mole
3. CMI	Initial mass of cargo	gram
4. TC	Initial cargo temperature	°C
5. CVOL	Constant used in calculating volatility of cargo	--

INPUT CARD SET TWO (READ (12,5005) ZT, VT, PTI, HTD)

Input the tank parameters using the input format: FORMAT (3E10.0,I3).

<u>Variable</u>	<u>Description</u>	<u>Units</u>
1. ZT	Tank height	cm
2. VT	Tank volume	cm <sup>3</sup>
3. PTI	Initial tank pressure	KPa
4. HTD	Heat transfer descriptor: 1 = Insulated walls (ADIABATIC process) -1 = Isothermal walls (ISOTHERMAL process)	--

INPUT CARD SET THREE (READ 12,5020) PDTSC, ZP, AP, XLP, CDORF)

Input the puncture parameters using the input format: FORMAT (1I10, 4E10.0).

TABLE C.6. FORMAT FOR INPUT DATA TO TVENT PROGRAM  
(Cont'd)

<u>Variable</u>	<u>Description</u>	<u>Units</u>
1. PDTSC	Puncture descriptor: 1 = Round hole 2 = Horizontal slot 3 = Vertical slot	--
2. ZP	Puncture height	cm
3. AP	Puncture area	cm <sup>2</sup>
4. XLP	Width of vertical slots (Note: Enter zero for round holes or horizontal slots)	cm
5. CDORF	Discharge coefficient	--

INPUT CARD SET FOUR (READ 12,5030) SVD, DPV)

Input the safety valve parameters using the input format: FORMAT (1I10, 1E10.0).

<u>Variable</u>	<u>Description</u>	<u>Units</u>
1. SVD	Safety valve descriptor: 1 = Unstuck -1 = Stuck	--
2. DPV	Safety valve relief setting	KPa

INPUT CARD SET FIVE (READ (12,5010) TINF, PATM)

Input the air properties using the input format: FORMAT (8E10.0).

<u>Variable</u>	<u>Description</u>	<u>Units</u>
1. TINF	Air temperature	°C
2. PATM	Atmospheric pressure	KPa

Input card sets 6 through 11 contain the chemical property file constants used in FUNCTION CPF and FUNCTION CPLHF. The variable T denotes the temperature in °C in the following formulas.

TABLE C.6. FORMAT FOR INPUT DATA TO TVENT PROGRAM  
(Concl'd)

INPUT CARD SET SIX (READ (12,5010) P1, P2, P3)

Input the vapor pressure constants using the input format: FORMAT (8E10.0).

INPUT CARD SET SEVEN (READ (12,5010) D1, D2)

Input the liquid density constants using the input format: FORMAT (8E10.0).

INPUT CARD SET EIGHT (READ (12,5010) Z1, Z2, Z3)

Input the vapor compressibility constants using the input format: FORMAT (8E10.0).

INPUT CARD SET NINE (READ (12,5010) C1, C2)

Input the specific heat of cargo constants using the input format: FORMAT (8E10.0).

INPUT CARD SET TEN (READ (12,5010) V1, V2, V3)

Input the specific heat of vapor constants using the input format: FORMAT (8E10.0).

INPUT CARD SET ELEVEN (READ (12,5010) H1, H2, H3, H4, H5, H6, H7)

Input the latent heat of cargo constants using the input format: FORMAT (8E10.0).

INPUT CARD SET TWELVE (READ (12,5010) ZW, VW)

Input the water characteristics using the input format: FORMAT (8E10 (8E10.0).

<u>Variable</u>	<u>Description</u>	<u>Units</u>
1. ZW	Height of water outside of tank	cm
2. VW	Specific volume of water	cm <sup>3</sup> /gram

INPUT CARD SET THIRTEEN (End-of-file card)

This card ends the data used by TVENT. Input a 6-7-8-9 punch in card column 1 for the end-of-file.

```

1      PROGRAM IVENT (INPUT, OUTPUT, TAPE9=INPUT, TAPE12=OUTPUT)
2
3      C      CHEMICAL PROPERTIES COEFFICIENTS
4      COMMON /CP/ M1,M2,M3,M4,M5,M6,M7
5      COMMON /GEN/ PT(C),T(C),TL(C),ZLN(C),MA(C),MV(C),ML(C),PINF,
6      VM,G,J,EDONT,AOL,AOC,RG),RG,TIME(C),CRUT,
7      VT,ZM,IP,DPV,SVD,MFD,CPL,SVL(C),SVC(C),MO,MVO,
8      MAD,PATM,AT,PV(C),LAMBDA(C),CPL(C),CPDAR(C),TC,SVA,
9      PINIG,ZI,CNI,PA(C)
10     COMMON/ING/ DP,TML,MV,TMA,DNL,MV,DNA,FLOSUB,ENMSUB,AP,INGYLE,
11     ICI,XLP,DELM,ZUT(C),VOLU,MTEST,ACHANGE
12     COMMON/VAL/ ZLOI,PICOI
13     COMMON /UNAME/ NUOVO
14     INTEGER MFD,PBTSC,PE,SVD,FLOSUB,ENMSUB
15     REAL ML,MV,M3,J,LAMBDA
16     *** DEFINITION OF INPUT DATA ***
17
18     C      SYMBOL      DESCRIPTION      UNITS
19     C      AT          TANK SURFACE AREA      CM**2
20     C      CCONF       DISCHARGE COEFFICIENT  CM**2
21     C      CIO         CAVGO ID              --
22     C      CNI         INITIAL CARGO MASS     CM
23     C      CMU         CARGO MOLECULAR WT     CM/GM-MOLE
24     C      CVOL        CONST USED TO CALC. VOLATILE --
25     C      C1-C2       CONST FOR SP HEAT OF CARGO --
26     C      DPV         SAFETY VALVE RELIEF SETTING KPA
27     C      D1-D2       CONST FOR LIQUID DENSITY --
28     C      MFD         HEAT TRANSFER DESCRIPTOR --
29     C      M1          1 = ADIABATIC          --
30     C      M2          -1 = ISOINTERNAL        --
31     C      M3          CONST FOR LATENT HEAT LIQUID --
32     C      PBTSC       PUNCTURE DESCRIPTION   --
33     C      P1          1 = ROUND              --
34     C      P2          2 = HORIZONTAL          --
35     C      P3          3 = VERTICAL            --
36     C      PVI         INIT TANK PRESSURE     KPA
37     C      P1-P3       CONST FOR VAPOR PRESSURE --
38     C      SVD         SAFETY VALVE DESCRIPTOR --
39     C      ZI          -1 = STUCK              --
40     C      TC          INITIAL CARGO TEMPERATURE DEG C
41     C      TINF        AMBI TEMPERATURE      DEG C
42     C      VT          TANK VOLUME            CM**3
43     C      VM          SPECIFIC VOLUME OF WATER CM**3/GM
44     C      V1-V3       CONST FOR SP HEAT OF VAPOR --
45     C      XLP         SLOPE WIDTH           CM
46     C      ZP          PUNCTURE HEIGHT       CM
47     C      ZT          TANK HEIGHT           CM
48     C      ZM          HEIGHT OF WATER OUTSIDE CM
49     C      Z1-Z3       CONST FOR VAPOR COMPRESSIB. --
50     C      ZI          1 = ENERGY
51
52     C      *** PROGRAM PARAMETERS ***
53     C      SYMBOL      DESCRIPTION      UNITS
54     C      AOL         AREA OF GAUG FLOW     CM**2
55     C      ENMSUB      SELECTS ENERGY SUBROUTINE CM**2
56     C      1          1 = ENERGY

```



```

115      WRITE(MW,7001) CRI,TC,CVOL
      C    TANK PARAMETERS
      READ(MR,5005)ZT,VI,PTI,MTD
      AT=VI/ZT
      MTASC=10*INSULATIO
      IF (CIB.EQ.-1) MTASC=10*ISOTHERMAL
      WRITE(MW,7010)ZT,AT,VI,PTI,MTASC
      C
      C    PUNCTURE PARAMETERS
      READ(MR,5020)PDISC,IP,AP,XLP,CDOFF
      PDASC=10*HQUOMB
      IF (PDISC.EQ.-1) PDASC=10*H. SLOF
      IF (PDISC.EQ.-1) PDASC=10*H. SLOF
      WRITE(MW,7020)PDASC,IP,AP,XLP,CDOFF
      C
      C    SAFETY VALVE PARAMETERS
      READ(MR,5030)SVD,DPV
      SVASC=10*HSTUCK
      IF (SVD.LQ.-1) SVASC=10*HUNSTUCK
      WRITE(MW,7030)SVASC,DPV
      C
      C    AIR PROPERTIES
      READ(MR,5040)TINF,PATM
      PINT,PINF=PATM
      WRITE(MW,7040)TINF,PATM
      C
      C    CHEMICAL PROPERTY FILE
      C    VAPOR PRESSURE
      READ(MR,5010)P1,P2,P3
      WRITE(MW,7050)P1,P2,P3
      C    LIQUID DENSITY
      READ(MR,5010)D1,D2
      WRITE(MW,7060)D1,D2
      C    VAPOR COMPRESSIBILITY
      READ(MR,5010)Z1,Z2,Z3
      WRITE(MW,7070)Z1,Z2,Z3
      C    SPECIFIC HEAT OF CAPCO
      READ(MR,5010)CT,C2
      WRITE(MW,7080)CT,C2
      C    SPECIFIC HEAT OF VAPOR
      READ(MR,5010)H1,H2,H3
      WRITE(MW,7090)H1,H2,H3
      C    LATENT HEAT OF LIQUID
      READ(MR,5010)N1,N2,N3,N4,N5,N6,N7
      WRITE(MW,7100)N1,N2,N3,N4,N5,N6,N7
      C
      C    WATER CHARACTERISTICS:
      WRITE(MW,7110)
      READ(MR,5010)ZM,ZM
      WRITE(MW,7110)ZM,ZM
      WRITE(MW,7250)
      C
      C    ... CALCULATE INITIAL CONDITIONS ...
      INTCG=0
      TIME=1
      FLOSH=1
      LMSUB=1

```

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PROGRAM IVENT 76/74 OPT=1

```

175      DP=0.0
          DELTIME=0.0
          IF(PRTSC.EQ.1) DP=AP/NLP
          PVI=CPFI(TC)
          C
          C
          C      IS CARGO VOLATILE?
          C      IF (PVI -GE- PINF*CVOL) GO TO 201
          C
          C      NON VOLATILE CARGO
          C      RELIEF VALVE STUCK?
          C      IF (SVB-NE.1) GO TO 120
          C
          C      VALVE NOT STUCK
          C      PFI=PINF
          C      FLOSH=2
          C      EMRSUP=2
          C      GO TO 250
          C
          C      VALVE STUCK
          C      120 PFI=AMAX(PINF,PFI)
          C      GO TO 100
          C
          C      CARGO VOLATILE
          C      200 PVI=CPFI(TC)
          C
          C      RELIEF VALVE STUCK?
          C      IF (SVB-NE.1) GO TO 220
          C
          C      VALVE NOT STUCK
          C      IF (PVI-LT-0PV) GO TO 220
          C      INPUT ERROR
          C      WRITE(NW,9000)PVI,DPV
          C      GO TO 1000
          C
          C      RELIEF VENT STUCK
          C      220 PFI = AMAX(PVI,PINF)
          C
          C      COMPUTE INITIAL LIQUID LEVEL AND MASSES
          C      SVL(1)=SVL(2)+SVL(3)+SVL(4)+CPFI(1,TC)*RG*(TC+273.2)/(C*MT*PVI)
          C      PV(1)=PV(2)+PV(3)+PV(4)+PVI
          C      SVL(1)=SVL(1)+SVL(3)+SVL(4)+1.0/CPFI(2,TC)
          C      SVA=RG*(TIME+273.2)/(28.96*PINF)
          C      ML(1)=ML(2)+ML(3)+ML(4)-(C*MT*SVL(1)-VT)/(SVL(1)+SVL(1))
          C      MV(1)=MV(2)+MV(3)+MV(4)-C*MT-ML(4)
          C      ZLI=ML(1)+SVL(1)/AF
          C      IF (ZLI-LT-.75) GO TO 270
          C      WRITE(NW,9010)ZLI,ZLI,ZI
          C      ZLI=ZI
          C      270 ZL(1)=ZL(2)+ZL(3)+ZL(4)+ZLI
          C      TVL=ZL(1)+AT
          C      TVV=MV(1)+SVV(1)
          C      PA(1)=PA(2)+PA(3)+MA(4)+28.96*(PFI-PVI)*TVV/(RG*(TC+273.2))
          C      IVA=MA(1)+SVA
          C      PT(1)=PT(2)+PT(3)+PT(4)+PTI
          C      PA(1)=PA(2)+PA(3)+PA(4)-C*MT(4)+RG/28.96*(TC+273.2)/(PFI-ML(4)+
          C      SVL(4))
          C      TIME(1)=TIME(2)+TIME(3)+TIME(4)+0.

```





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FIN 4.6433F

76/74 OPT-1

PROGRAM TWEI

```

C 7 INTERSECTING PUNCTURE AND INTERSECTING PUNCTURE
C 8 BELOW WATER LEVEL BELOW PUNCTURE
C 9 BELOW PUNCTURE ABOVE PUNCTURE
C 10 BELOW PUNCTURE INTERSECTING PUNCTURE
C *****
C PUNCTURE NOT SUBMERGED BY WATER
C IF (ZM-ST-ZP-OP) GO TO 310
C CARGO BELOW PUNCTURE
C IF (ZL(2)-G[-ZP-OP]) GO TO 310
C AOG=AP
C AOL=O.
C PINT=PINTG=PINT-PAIN
C KTOP=B
C GO TO 400
C CARGO ABOVE PUNCTURE
C 310 CONTINUE
C IF (ZL(2)-LT-ZP) GO TO 320
C AOG=O.
C AOL=AP
C PINT=PINTG=PINT-PAIN
C ZLN(2)=ZP-O.5*OP
C ZLN(4)=ZLN(1)-ZLN(2)
C KTOP=1
C GO TO 400
C CARGO INTERSECTS PUNCTURE
C 320 CONTINUE
C AOL=AP*(ZL(2)-ZP*OP)/OP
C AOG=AP-AOL
C ZLN(2)=O.5*(ZL(2)+ZP-OP)
C ZLN(4)=ZLN(3)-ZLN(2)
C PINT=PINTG=PINT-PAIN
C KTOP=4
C GO TO 400
C ***
C PUNCTURE SUBMERGED BY WATER
C 330 CONTINUE
C IF (ZM-LT-ZP) GO TO 360
C CARGO BELOW PUNCTURE
C IF (ZL(2)-GT-ZP-OP) GO TO 340
C AOG=AP
C AOL=O.
C PINT=PINTG=PINT-PAIN*1.E-4*(ZM-ZP*OP*O.5)/VW
C KTOP=9
C GO TO 400
C CARGO ABOVE PUNCTURE
C 340 CONTINUE
C IF (ZL(2)-LT-ZP) GO TO 3.0
C AOG=O.
C AOL=AP
C ZLN(2)=ZP-OP*O.5
C ZLN(4)=ZLN(3)-ZLN(2)
C PINT=PINTG=PINT-PAIN*1.0E-4*(ZM-ZLN(2))/VW
C KTOP=2

```

```

365      GO TO 400
      CARGO INTERSECTS PUNCTURE
350  A0L=AP*(ZL(2)-ZP*DP)/DP
      A0G=AP-A0L
      ZLN(2)=ZL(2)+ZP*DP*0.5
      ZLN(4)=ZLN(1)+ZLN(2)
      PINTF=PINTF-PINTF-PATRN1.DE-4*G*(1/M-0.5*(ZP+ZL(2)))/VM
      KTFP=5
      GO TO 400
      C ***
      C PUNCTURE INTERSECTS WATER LEVEL
      C CARGO FLOW PUNCTURE
360  CONTINUE
      IF (ZL(2)-G1-ZP*DP) GO TO 570
      A0G=AP
      A0L=0.
      PINTF=PINTF-PATRN
      A1=ZP-ZM
      A2=ZM-ZP*DP
      PINTF=(1./(A1+A2))*((A1+PATRN2*(PATRN1.E-4*G*(ZM-ZP*DP)/2.)/VM)
      KTFP=10
      GO TO 400
      C CARGO ABOVE PUNCTURE
370  CONTINUE
      IF (ZL(2)-A1-ZP) GO TO 380
      A0L=AP
      A0G=0.
      ZLN(2)=ZM-.5*DP
      ZLN(4)=ZLN(1)+ZLN(2)
      ZLN(2)=(ZM-ZP-.P)*0.5
      A1=ZP-ZM
      A2=ZM-ZP*DP
      PINTF=PINTF-PINTF=(1./(A1+A2))*((A1+PATRN2*(PATRN1.E-4*
      G*(ZM-ZLN(2)/VM))
      KTFP=3
      GO TO 400
      C CARGO INTERSECTS PUNCTURE, BUT IS ABOVE WATER
380  CONTINUE
      IF (ZL(2)-A1-ZM) GO TO 390
      A0L=AP*(ZL(2)-ZP*DP)/DP
      A0G=AP-A0L
      ZLN(2)=.5*(ZL(2)+ZP*DP)
      ZLN(4)=ZLN(1)+ZLN(2)
      A2=ZL(2)-ZM
      A1=ZM-ZP*DP
      ZLN(2)=.5*(ZM+ZP*DP)
      PINTF=PINTF
      PINTF=PINTF*(1./(A1+A2))*((A2+PATRN1.E-4*G*(ZM-ZLN(2)/VM)
      KTFP=6
      GO TO 400
      C CARGO INTERSECTS PUNCTURE, BUT IS BELOW WATER
390  A0L=AP*(ZL(2)-ZP*DP)/DP
      A0G=AP-A0L
      ZLN(4)=ZLN(1)+ZLN(2)-(ZL(2)+ZP*DP)*.5
      A1=ZP-ZM
      A2=ZM-ZL(2)
      A3=ZL(2)-ZP*DP

```

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PROGRAM TVCNI 76/74 OPT=1

```

400      ZLN3=0.5*(ZL(2)+ZP-DP)
        PINT=PINT+PINT*(1-E-4)*G*(ZU-ZLN3)/VU
        PING=(1./(A1/A2))*(A1+PINT/A2*(PINT+1.E-4)*G*(ZU-ZL(2))+.5/VU))
        KTOP=7
        C ***
405      PV(3) = CPF(1,I(3))
        I1=0
        400 CONTINUE
        PINT=PINT+1
        CALL CHOKTST(KTOP,K5)
        OLDULO=ULO
        OLDVVO=VVO
410      IF(NDVVO.EQ.1.AND-PT(4).LT.PING) GO TO 430
        IF(NDVVO.EQ.1) GO TO 500
        GO TO (420,410),FLOSUM
415      420 CALL FLOW1
        IF(PING.EQ.2) CALL INGEST
        IF(PING.NE.0) GO TO 610
        GO TO 500
420      440 CALL FLOW2
        IF(PING.EQ.2) CALL INGEST
        IF(PING.NE.0) GO TO 610
        500 CONTINUE
        IF (ULO.GT.1.) GO TO 520
        DT=DRV/VVO
        DMA=ULO*DT
        DMA=VVO*DT
        GO TO 530
425      520 DT=DRL/ULO
        DMV=VVO*DT
        DMA=VVO*DT
        530 CONTINUE
        TIME(4)=TIME(3)+TIME(2)*DT
        GO TO (540,550),ENRSUB
430      540 CALL ENERGYTIME
        IF(CHANGE.EQ.2) GO TO 500
        IF(PING.EQ.2) CALL INGEST
        IF(PING.NE.0) GO TO 610
        GO TO 560
440      550 CALL NONINERTTIME
        IF(PING.EQ.2) CALL INGEST
        IF(PING.NE.0) GO TO 610
        560 CONTINUE
        IF (ADS(OLDULO-ULO).LE.-(.001*OLDULO).AND-
          1 ABS(OLDVVO-VVO).LE.-(.001*OLDVVO)) GO TO 600
        C
        C CONVERGENCE NOT ACHIEVED, SCROLL PLY 4 INTO PLY 3
        ML(3)=ML(4)
        MV(3)=MV(4)
        MA(3)=MA(4)
        ZL(3)=ZL(4)
        SVL(3)=SVL(4)
        SVV(3)=SVV(4)
        ZLN(3)=ZLN(4)
        PT(3)=PT(4)
        TIME(3)=TIME(4)
        PA(3)=PA(4)
445      C
450      C
455      C

```

```

460      T(3)=T(4)
          PV(3) = PV(4)
          GO TO 490
461      CONTINUE
          IF(MOUVO.EQ.1) MUO=MVO*((PT(4)-PINF6)/(PT(2)-PINF6))**.5
          NTEST=1
          CALL INGLEST
          IF(INGFLG.EQ.0) GO TO 615
          IF(INGFLG.EQ.1) CALL AJREM(TIME)
          IF(INGFLG.EQ.3) CALL WATM
          IF(INGFLG.EQ.4) CALL MOOUT
          IF(INGFLG.EQ.4) GO TO 615
          IF(INGFLG.EQ.4) GO TO 615
          IF(INGFLG.EQ.2) GO TO 500
          IF(INGFLG.EQ.1) INGTG=0
          WRITE(MU,2200) TIME(2),ML(2),MV(2),MA(2),INL,INV,TRA,I(2),PT(2),
1          ZL(2)
          IF(INGFLG.EQ.0) GO TO 630
          IF(INGFLG.EQ.5) INGTG=0
          GO TO 500
475      CONVERGENCE ACHIEVED ,SCROLL ALL PLOTS
          CONTINUE
          IF(EL(4)-GT.0.0)GO TO 614
          IZERO=IZERO+1
          IF(IZERO.GT.1) GO TO 616
          INL=INL*ML(2)
          IF(HTD.EQ.-1) ENMSUM=2
          IF(HTD.EQ.-1) FLOSHR=2
          ZL(2)=0.0
          ML(2)=0.0
          ZL(4)=ZL(3)-ZL(2)
          ZLH(4)=ZLH(3)-ZLH(2)-0.0
          ML(4)=ML(3)-ML(2)
          PT(4)=PT(3)-PT(2)
          T(4)=T(3)-T(2)
          MA(4)=MA(3)-MA(2)
          MV(4)=MV(3)-MV(2)
          PV(4)=PV(3)-PV(2)
          TIME(4)=TIME(3)-TIME(2)
          ZLH(4)=ZLH(3)-ZLH(2)
          PA(4)=PA(3)-PA(2)
          SVL(4)=SVL(3)-SVL(2)
          SVV(4)=SVV(3)-SVV(2)
          CPL(4)=CPL(3)-CPL(2)
          CPVHAR(4)=CPVHAR(3)-CPVHAR(2)
          DMV=MV(2)*.01
          WRITE(12,2201)
          FORMAT(1M ,2HTESTAB1)
          WRITE(MU,2200) TIME(2),ML(2),MV(2),MA(2),INL,INV,TRA,I(2),PT(2),
1          ZL(2)
          GO TO 500
480      DO 620 I=2,4
          ML(I-1)=ML(I)
          MV(I-1)=MV(I)
          MA(I-1)=MA(I)
          PT(I-1)=PT(I)
          T(I-1)=T(I)

```

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FIN 4.60431E

PROGRAM TVENE 76/74 OPT=1

```

515 ZL(1)=ZL(1)
ZL(1)=ZL(1)
PAC(1)=PAC(1)
TIME(1)=TIME(1)
PV(1)=PV(1)
SVL(1)=SVL(1)
SVL(1)=SVL(1)
CPL(1)=CPL(1)
CPVBAR(1)=CPVBAR(1)
620 CONTINUE
PT(2)=PT(2)-P(1)
T(2)=T(2)-T(1)
ML(2)=ML(2)-P(1)
MA(2)=MA(2)-P(1)
MV(2)=MV(2)-P(1)
PV(2)=PV(2)-P(1)
TIME(2)=TIME(2)
ZL(2)=ZL(2)-ZL(1)
ZL(2)=ZL(2)-ZL(1)
PAC(2)=PAC(2)-PAC(1)
SVL(2)=SVL(2)-SVL(1)
SVL(2)=SVL(2)-SVL(1)
CPL(2)=CPL(2)-CPL(1)
CPVBAR(2)=CPVBAR(2)-CPVBAR(1)
TML=TML+TML
TML=TML+TML
TML=TML+TML
TML=TML+TML
525 TIME=TIME+1
WRITE(10,720) TIME(2),ML(2),MV(2),MA(2),TML,TMV,TMA,PT(2),
ZL(2)
1 IF(CABS(PT(2))-PINF)AND(10.0,ML(2)-PINF)AND(1.0E-06/SVL(2))
1 -GE.0.001)PINF) GO TO 300
530 WRITE(10,720)
WRITE(10,720) TIME(2),TML,TMV,TMA
C
C HEAD FORMATS
540 5000 FORMAT (A3,7A,6E10.0)
5005 FORMAT(3E10.0,13)
5010 FORMAT (3E10.0)
5020 FORMAT (11E10,6E10.0)
5030 FORMAT(11D,7E10.0)
555 C
C WRITE FORMATS--HEADINGS
6000 FORMAT(1H1,111,50X,21H TANK VENTING PROGRAM ,1,
1 1H0,60X,11H INPUT DATA,11)
6100 FORMAT(1H1,111,50X,21H TANK VENTING PROGRAM ,1,
1 1H0,60X,7H OUTPUT ,11)
C
C WRITE FORMATS
7000 FORMAT(1H0,50X,10H CARGO PARAMETERS,11,1H0,
1 27X,10H CARGO 10,24X,A3,5X,15H MOLECULAR WT.,7X,E15.5)
7001 FORMAT(1H0,27X,12H INITIAL MASS,3X,E15.5,5X,12H INITIAL TEMP,8X,
1 F15.5,1,28X,15H VOLATILE CONST,7X,E15.5)
7010 FORMAT(1H0,50X,21H TANK PARAMETERS,11,1H0,
1 27X,6H HEIGHT,14X,E15.5,5X,12H SURFACE AREA,8X,E15.5,1,
2 27X,6H VOLUME,14X,E15.5,5X,12H SURFACE,12X,E15.5,1,
3 27X,6H WALLS)

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7020 FORMAT(1H0,5X,17PUNCTURE PARAMETERS,/,1H0,
1 2X,1INDEX,14X,110,5X,/,
2 2X,6HSLIGHT,14X,115.5,5X,4HARCA,10X,115.5,/,
3 2X,10HSLQ,14X,10X,115.5,5X,14HDISCHARGE COEF,6X,115.5)
7030 FORMAT(1H0,5X,21HSAFETY VALVE PARAMETERS,/,1H0,
1 2X,9HCONDITION,14X,110,5X,14HRELIEF SETTING,6X,115.5)
7040 FORMAT(1H0,5X,14HHAIR PROPERTIES,/,1H0,
1 2X,15HAIR TEMPERATURE,5X,115.5,5X,12HAIR PRESSURE,6X,115.5)
7050 FORMAT(1H0,5X,24HCHEMICAL PROPERTIES FILE,/,1H0,
1 2X,11HVAPOR PRESSURE,10X,115.5)
7060 FORMAT(2X,14HLIQUID DENSITY,10X,115.5)
7070 FORMAT(2X,21HVAPOR COMPRESSIBILITY,5X,115.5)
7080 FORMAT(2X,21HSPECIFIC HEAT OF LIQUID,1X,115.5)
7090 FORMAT(2X,21HSPECIFIC HEAT OF VAPOR,2X,115.5)
7100 FORMAT(2X,21HENTHALPY HEAT OF LIQUID,3X,115.5,/,
1 5X,115.5)
7105 FORMAT(1H0,5X,10HWATER PARAMETERS,/,/)
7110 FORMAT(1H0,27X,21HHEIGHT OF WATER OUTSIDE TANK,615.5,5X,
1 17HSPEC VOL OF WATER,3X,115.5)
C WRITE FORMATS--ERRORS
7200 FORMAT(1H0,10L12.5)
7250 FORMAT(1H0,1X,115.5)
7260 FORMAT(1H0,1X,INITIAL PARAMETER VALUES (L.V.),/,
1 1H0,1HMASS,2X,115.5,/,10X SPEC VOL,3X,115.5,/,
2 10X LIQ VOL,3X,115.5,/,10X MASS LOSS,2X,115.5,/,
3 10X LIQ LEVEL,115.5)
7270 FORMAT(1H0,5X,4HTEMP,10X,2HML,10X,2HML,10X,2HML,9X,3HML,9X,
1 3HML,9X,3HML,9X,4HTEMP,9X,2HPT,9X,2HPT)
7280 FORMAT(1H0,5X,
1 15HTOTAL MASS LOSS,/,
2 15X,4HTEMP,11X,4HML,4X,5HVAPOR,
3 12X,1HML)
9000 FORMAT(27H0DATA INPUT ERROR, PINIT = ,115.5,10HLESS THAN DPV =
1 115.5)
9010 FORMAT(26H0DATA INPUT ERROR, ILL = ,115.5,1X,11HGREATER THAN ,
1 5HIT = ,115.5,/,10X,35HRECUOTON CONTINUING WITH ILL-IF =
2 ,115.5)
WRITE(MU,728) ICI
7285 FORMAT(1H0,1,15H ITERATIONS = ,15)
4000 STOP
END

```

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FIN 4-6-43F

SUBROUTINE AIRIN 76/74 OPT=1

```

1      SUBROUTINE AIRIN(ETIME)
2      CALLD SUBROUTINES: ENERGY2,FLOW1,FLOW2,MOMENG2
3      COMMON /GEN/ P(4),T(4),Z(4),ZLH(4),MA(4),MV(4),ML(4),PINF,
4      VU,G,J,DOOR,AOL,AOS,RG1,RG,TINF,TIM(4),CMVI,
5      VI,ZI,IP,DPV,SVD,HUB,CPI,SVL(4),SVV(4),ULO,
6      WVO,WAQ,PAIM,AT,PVI(4),LAMHDA(4),CPL(4),CPVBAR(4),
7      TC,SVA,PINF,ZI,CM1,PA(4)
8      COMMON/ING/ DP,INL,INV,TMA,DML,BMV,DMA,FLOSUB,ENRSUB,AP,INGFLG,
9      IET,XLP,DELTIME,ZIT(4),VOLU,NTEST,NCHANGE
10     COMMON/WAT/ ZLEQT,PTEQI
11     INTEGER HTD,SVD,FLOSUB,ENRSUB,MA2
12     REAL ML,MV,MA,J,LAMBDA
13     NU=12
14     WRITE(NU,95)
15     FORMAT(1H,95)
16     PI=ACOS(-1.0)
17     MA2=MA(2)
18     PT2=PT(2)
19     T2=T(2)
20     COMPUTE VOLUME OF AIR BUBBLE
21     VOLU=PI*5.6*4./3.*((AP/PT)**.5)**3
22
23     KOUNT=1
24     CONTINUE
25     INGFLG=1
26
27     GO TO(120,130),ENRSUB
28     120 CALL ENERGY2(KOUNT)
29     GO TO 140
30     130 CALL MOMENG2(KOUNT)
31     140 CONTINUE
32
33     PT(2)=PT(4)
34     MA(2)=MA(4)
35     T(2)=T(4)
36     CONTINUE
37
38     OLDMLO=ULO
39     OLDMVO=WVO
40     GO TO(420,440),FLOSUB
41     420 CALL FLOW1
42     GO TO 500
43     440 CALL FLOW2
44     500 CONTINUE
45     TANK PRESSURE TOO LOW---LET IN ANOTHER BUBBLE
46     IF(INGFLG.EQ.2) KOUNT=KOUNT+1
47     IF(INGFLG.EQ.2) GO TO 90
48     IF(ULO.GT.0.0360 TO 520
49     DT=DMV/WVO
50     DML=ULO*DT
51     DMA=WAQ*DT
52     GO TO 530
53     520 DT=DML/WLO
54     DMV=WVO*DT
55     DMA=WAQ*DT

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SUBROUTINE AIRRM 76/74 OPT=1

```

510 CONTINUE
TIM(4)=TIM(3)-TIM(2)*DT
GO TO(540,550),ENRSUB
540 CALL ENERGY(TIME)
GO TO 560
550 CALL MOMENTUM(TIME)
560 CONTINUE
IF(ABS(OLDWLO-WLO)-LE-(OLDWLO)-AND-
1 ANS(OLDWVO-WVO)-LE-(OLDWVO)) GO TO 600
C
C CONVERGENCE NOT ACHIEVED, SCROLL
ML(3)=ML(4)
MV(3)=MV(4)
MA(3)=MA(4)
ZL(3)=ZL(4)
SVL(3)=SVL(4)
SVM(3)=SVM(4)
ZLM(3)=ZLM(4)
PT(3)=PT(4)
TIM(3)=TIM(4)
PA(3)=PA(4)
T(3)=T(4)
PV(3)=PV(4)
WRITE(MY,7290) ML(3),MV(3),MA(3),ZL(3),PT(3),T(3),PV(3)
7290 FORMAT(ON NC40,7E12.5,4H ***
GO TO 400
600 CONTINUE
C
C CONVERGENCE ACHIEVED, NEW PT(4) FOUND
TEST PT(4) FOR CONVERGENCE
AT=(PT(4)+(.0001*6/SVL(4))*(ZL(4)-ZP))/(PT2+(-.0001*6/SVL(2))
1 (ZL(2)-ZP))
IF(AT-GE-.95.AND-AT-LE-1.05) GO TO 700
C
C DRL NOT CORRECT FOR PT(4).
C MODIFY DRL AND TRY AGAIN
C
TN=0.0
DRL=DRL*(TIM+1)/(TIM+1.0)
GO TO 400
/100 CONTINUE
C
C CHECK FOR LIQUID LEVEL EQUAL TO TOP OF PUNCTURE
1E((ZL(4)-LE-ZP)/GO TO 720
C
C SCROLL AND ADD ALL RISE TIMES TO TIME STEP
DO 710 I=2,4
ML(I-1)=ML(I)
MV(I-1)=MV(I)
MA(I-1)=MA(I)
PT(I-1)=PT(I)
T(I-1)=T(I)
ZL(I-1)=ZL(I)
ZLM(I-1)=ZLM(I)
PA(I-1)=PA(I)
TIM(I-1)=TIM(I)

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FTM 4.0431F

SUBROUTINE AIRIN 76/74 OPT=1

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115      PV(1)=PV(1)
          SVL(1)=SVL(1)
          SVU(1)=SVU(1)
          CPL(1)=CPL(1)
          CPVBAR(1)=CPVBAR(1)
120      CONTINUE
          PT(2)=PT(1)+PT(4)
          TC(2)=TC(1)+TC(4)
          ML(2)=ML(1)+ML(4)
          MA(2)=MA(1)+MA(4)
          MV(2)=MV(1)+MV(4)
          PV(2)=PV(1)+PV(4)
          TIM(2)=TIM(1)+TIM(4)
          ZL(2)=ZL(1)+ZL(4)
          ZLN(2)=ZLN(1)+ZLN(4)
          PA(2)=PA(1)+PA(4)
          SVL(2)=SVL(1)+SVL(4)
          SVU(2)=SVU(1)+SVU(4)
          CPL(2)=CPL(1)+CPL(4)
          CPVBAR(2)=CPVBAR(1)+CPVBAR(4)
          TML=TML+TML
          TMA=TMA+TMA
          TML=TML+TML
          TMA=TMA+TMA
          WRITE(MU,7200) TIM(2),ML(2),MV(2),MA(2),TML,TMA,TC(2),PT(2),
1          ZL(2)
140      7200 FORMAT(1H,10E12.5)
          GO TO 90
120      CONTINUE
          PT(2)=PT(1)
          MA(2)=MA(1)
          TC(2)=TC(1)
          RETURN
          END
165

```

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DTM 6.6433F

SUBROUTINE AIRPASS 76/74 OPT=1

```

1      SUBROUTINE AIRPASS
      C
      C      CALCULATES MASS OF THE AIR FOR ITERATION IN INGEST
      C      CALLED FUNCTIONS: CPF
      C
5      COMMON /GEW/ PT(4), T(4), ZL(4), ZLW(4), MA(4), MV(4), ML(4), PINE,
      C      VM, G, J, COORF, AOL, ADG, RGT, PG, TATM, TIM(4), MOLE,
      C      VOL(4), EV, ZP, DELPVAL, TVALVE, INEATIR, CPA, VL(4), VU(4),
      C      WLO, WVO, WAO, PAIR, AT, PV(4), LAMBDA(4), CPL(4), CPVBAR(4),
      C      TC, SVA, PINEG, ZT, CRI, PA(4)
10     REAL MA, MV, ML, MOLE, J, LAMHDA
      C
      IL(4)=(VL(4)*(1.-/CPF(2, T(4))))/A1
      VOLG=VOL I- IL(4)*AT
      PA(4)=(28.96*(PT(4)-PV(4))-VOLG)/(T(4)+273.2)*RG
      RETURN
      END
15

```

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FIN 4.6433F

SUBROUTINE CHOKTST 76/74 OPT=1

```

1      SUBROUTINE CHOKTST(KTYP,K5)
2      CALLED FUNCTIONS: CPF,CPLNF,GAMH
3      COMMON /COMMON/ T(4),ZLN(4),MAC(4),MV(4),ML(4),PINF,
4      VM(4),COORF,AOL,RGI,RG,TIMF,TIR(4),CHMT,
5      VT,ZU,ZP,DV,SVD,HTD,CPA,SVL(4),SVV(4),WLO,
6      WVO,MAO,PAIM,AT,PV(4),LAMBDA(4),CPL(4),CPVBAR(4),
7      TC,SVA,PINF,ZT,CMT,PA(4)
8      COMMON/ING/ OP,TML,TMV,TMA,DML,DNV,DMA,FLOSUB,ENRSUB,AP,INGFLG,
9      ICT,XLP,DELTIME,ZMT(4),VOLD,MTEST,NCHANGE
10     COMMON/MAT/ /LEQ1,PTEQ1
11     COMMON/CP/ A,B,C,DA,DB,ZA,ZB,ZC,CA,CO,VA,VB,VC,
12     HA,HD,HG,HD,HE,HF,TA
13     INTEGER HTD,SVD,FLOSUB,ENRSUB
14     REAL ML,RV,MA,J,LAMBDA,LAM01,LAM02,LAMBAR
15     AVERAGE(X)=(X(2)+X(4))/2.0
16     IF(K5.GT.1) RETURN
17     DO NOT CHECK CHOKTST
18     IF(CPL(2).LE.-1.05+PINF) GO TO 675
19
20     ESTABLISH KTYP5 THAT REPRESENT LIQUID AND VAPOR FLOWS
21     ILIQ=IGAS=ILAG=0
22     IK=0
23     IF(KTYP.EQ.1-OR.KTYP.EQ.2-OR.KTYP.EQ.3) ILIQ=1
24     IF(KTYP.EQ.3-OR.KTYP.EQ.9-OR.KTYP.EQ.10) IGAS=1
25     IF(KTYP.EQ.4-OR.KTYP.EQ.5-OR.KTYP.EQ.6-OR.KTYP.EQ.7) ILAG=1
26     GO TO 500 IF ONLY LIQUID ESCAPES
27     GO TO 500 IF ONLY LIQUID ESCAPES
28     IF(ILIQ.EQ.1-AND-IGAS.EQ.0) GO TO 500
29     GO TO 600 IF ONLY VAPOR ESCAPES
30     GO TO 600 IF ONLY VAPOR ESCAPES
31     IF(ILIQ.EQ.0-AND-IGAS.EQ.1) GO TO 600
32
33     LIQUID AND GAS ESCAPES
34     TEST IF MOMENTER AND FLOW? ARE BEING USED
35     IF(FLOSUB.EQ.2) GO TO 600
36
37     ENERGY AND FLOW ARE BEING USED
38
39     IK=1
40     GO TO 505
41     CASE 2-A
42     500 IF(FLOSUB.EQ.2) GO TO 675
43     CASE 1-A
44     ICTF=0
45     505
46     101=CPF(6,PINF)
47     102=(T(2)+T(1))*0.5
48     CPL101=CPF(4,(T(2)+T(1))*0.5)
49     CPL102=CPF(4,(T(2)+T(1))*0.5)
50     SVV(4)=CPF(3,T(2))+RG*(T(2)+273.2)/(CHMT+PV(2))
51     SVL(4)=1./CPF(2,T(2))
52     LAM01=CPLNF(101)
53     LAM02=CPLNF(102)
54     X01=CPL101*(101+273.2)*((ALOG((T(2)+273.2)/(T(1)+273.2)))/LAM01
55     X02=CPL102*(102+273.2)*((ALOG((T(2)+273.2)/(T(2)+273.2)))/LAM02

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60      P01=CPFI(1,T01)
        P02=CPFI(1,T02)
        V001=CPFI(1,T01)*RG*(T01+273.2)/(CNUT*P01)
        V002=CPFI(1,T02)*RG*(T02+273.2)/(CNUT*P02)
        VL01=1./CPFI(2,T01)
        VL02=1./CPFI(2,T02)
        LAMBAR=(LAMN01/LAMN02)/2.
        DERIV1=((CPLT02*(T(2)-T02)-T02*(L+MU2))-CPLT01*(T(2)-T01)-
        1      T01*(LAMN01))/(T02-T01)
        DERIV2=(-(12*(X02+V002-12*(X01+V001+VL02*(1.-12*(X02)-
        1      VL01*(1.-12*(X01))/(T02-T01)
        DER=DERIV1/DERIV2
        CPLAVE=(CPLT01+CPLT02)*.5
        CONST=0*(T(2)+273.2)*ALOG(10./((COT(2))*.42 - 1.-5
        DELT=(T(2)+273.2)*(.12*(SVU(2)+DER/LAMBAR +
        1      ((12*(SVU(2)/LAMBAR)+2*DER+.42*(4*(SVU(2)+DER)
        2      ((1.-12*(SVU(2)+DER*CONST/LAMBAR)/(CPLAVE*(T(2))))*.5/
        3      ((1.-12*(SVU(2)+DER*CONST/LAMBAR)*2.))
        T01=T(2)-DELT
        T02=T01-2.*((T(2)-T01)/5.
        ICTT=ICIT+1
        IF(ICTT-FQ.1) GO TO 550
        PO=CPFI(1,T01)
        IF(PO-LE-PINF-AND-IR-EQ-1) GO TO 600
        IF(PO-LE-PINF) GO TO 675
        IF(PO-GT-PINF) GO TO 300
        350      CONTINUE
        IF(PO-GT-PINF) PINF=PO
        IF(IR-EQ-1) GO TO 600
        GO TO 674
        300      CONTINUE
        GO TO 350
        C      CASE 1-B (USIS FLOW) AND 2-B (USES FLOW2)
        C      ONLY VAPOR CAN ESCAPE
        C
        600      RVT=RG1/CNUT
        GAMMA=GAMN(HA)
        PO=PT(2)*.2/(GAMMA+.1))*(GAMMA/(GAMMA-1.))
        TO=((T(2)+273.2)*(PO/PT(2)))*((GAMMA-1.)/GAMMA)-273.2
        CPVAVE=(CPFI(5,T(2))+(CPFI(5,T01))*.5
        GAMMA=CPVAVE/(CPVAVE-RVT)
        PO=PT(2)*.2/(GAMMA+.1))*(GAMMA/(GAMMA-1.))
        IF(PO-LE-PINF) GO TO 675
        PINFG=PO
        674      K5=1
        RETURN
        675      K5=2
        RETURN
        END

```



FUNCTION CPF 7/274 UPI-1 FIN 4.64356 80/08/15. 13.45.07 PAGE 1  
 1 FUNCTION CPE(FLAG,1)  
 COMMON /CP/ PA,PH,PT,DA,DB,ZA,ZB,IC,CA,CB,VA,VB,VC,  
 1 MA,HU,MC,MO,ME,MT,TA  
 5 C CHEMICAL PROPERTIES FILE FUNCTION SUBPROGRAM  
 C NOTE: ALL TEMPERATURES ARE IN DEG C  
 C GO TO (100,200,300,400,500,600),IFLAG  
 10 C VAPOR PRESSURE VS TEMPERATURE, IFLAG=1  
 C CPE=10.0\*(PA-PH/(T+PT))  
 RETURN  
 C LIQUID DENSITY, IFLAG=2  
 15 C CPE=DA-DB\*T  
 RETURN  
 C VAPOR COMPRESSIBILITY, IFLAG=3  
 20 C CPE=ZA-ZB\*T-7C\*T\*\*2  
 RETURN  
 C SPECIFIC HEAT OF CARGO,IFLAG=4  
 25 C CPE=CA+CB\*T  
 RETURN  
 C SPECIFIC HEAT OF VAPOR, IFLAG=5  
 30 C CPE=VA-VH\*T-VC\*T\*\*2  
 RETURN  
 C TEMPERATURE DETERMINED FROM VAPOR PRESSURE, IFLAG=6  
 400 CPE=-1.0\*(PT + PD/CALOG10(1)-PA))  
 RETURN  
 C END

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FTM 6-604338

76174 OPI=1

**UNITED STATES DEPARTMENT OF JUSTICE**

```

1      FUNCTION CPLMF(I)
2      CALLED FUNCTIONS: CPF
3      COMMON /CPF/ PA,PU,PT,PA,OB,TA,IB,IV,CA,CO,VA,VB,VC,
4      NA,MU,MC,MD,ME,MT,TA
5
6      C
7      C CALCULATES THE LATENT MEAT OF THE CARGO
8      C1=(NA-MU+I-MC+0.2)/(I+273.2)/(I+TA)+0.2
9      C2=MD*CPF(I,3)/(I+273.2)
10     C3=(I+TA)+0.2 * (ME-MF+I)
11     CPLMF=C1 - C2/C3
12     RETURN
13     END

```

SUBROUTINE ENERGY 76/74 OPT=1 PAGE 1  
 1 SUBROUTINE ENERGY(TIME)  
 C  
 C DETERMINES PRESSURE AND TEMPERATURE OF THE TANK CARGO  
 C AS LIQUID OR VAPOR IS VENTED  
 C  
 C  
 C CALLED SUBROUTINES: CORRECT, FLOW, TERNIN, ZLCALC  
 C CALLED FUNCTIONS: CPE, CPLE, CPM, CPMF  
 C REAL LAMBDA, MA, ML, MV, MACPAV, MLCPLAV, MVCPVBA, J,  
 C MOLI  
 1 COMMON /GEN/ PT(4), T(4), ZL(4), ZLH(4), MA(4), MV(4), ML(4), PINF,  
 1 VU(4), ZCD(4), AOL(4), AOG(4), RG(4), TATM(4), TIM(4), MOLE,  
 2 VOLT(4), TP, DELTVAL, IVALVE, IHEATR, CPA, VL(4), VV(4),  
 3 WLO, WVO, WAO, PATM, AT, PV(4), LAMBDA(4), CPL(4), CPVBAR(4),  
 4 TC, SVA, PINF, ZI, CRT, PA(4)  
 1 COMMON /ING/ DP, TML, TRV, TMA, DRL, DMV, DMA, FLOSUB, ENDSUB, AP, INGTLE,  
 1 TCT, XTP, BELTIC, ZOT(4), VOLB, NTEST, NCHANGE  
 1 COMMON /MVAR/ ROMVO  
 1 AVERAGE(4) = (T(4) + X(2)) / 2.0  
 1 LAMBDA(1) = LAMBDA(2) = CPLE(1(2))  
 1 GUESS CONDITIONS OF NEXT TIME STEP  
 1 PT(4) = P(4) - P(2)  
 1 T(4) = T(3) - T(1)  
 1 LAMBDA(4) = LAMBDA(3) = LAMBDA(2)  
 1 VU(4) = CPLE(3, 1(2)) \* (ING/MOLE) \* ((1(2) + 273.2) / CPLE(3, 1(2))  
 1 VV(4) = VV(3) = CPLE(3, 1(3)) \* (ING/MOLE) \* ((1(3) + 273.2) / CPLE(3, 1(3))  
 1 VL(4) = VL(3) = 1. / CPLE(2, 1(3))  
 1 BELTIC = ((VU(4) + VV(4)) \* 5 - WVO \* AVERAGE(VL) \* WLO) \* (TIM(3) - TIM(2)) -  
 1 ((VU(4) + VV(4)) \* 5 - AVERAGE(VL))  
 2 ML(4) = ML(2) - WLO \* (TIM(3) - TIM(2)) - BELTIC  
 1 MV(4) = MV(2) - BELTIC - WVO \* (TIM(3) - TIM(2))  
 1 CPL(4) = CPLE(3) = CPLE(4, 1(3))  
 1 CPVBAR(4) = CPVBAR(3) = CPLE(4, 1(3))  
 1 IF (WLO.GT.0.5) GO TO 60  
 1 DRL = DRL / WVO  
 1 DMA = DMA \* DT  
 1 GO TO 65  
 60 DRL = DRL / WLO  
 1 DMV = DMV \* DT  
 1 DMA = DMA \* DT  
 1 CONTINUE  
 65  
 1 IS VALVE UNSTUCK?  
 1 IF (VALVE.EQ.1) GO TO 500  
 1 RELIEF VALVE STUCK  
 1 IS TANK LIQUID LEVEL AT LEAST PARTIALLY BELOW PUNCTURE  
 1 IF ((ZL(4) - ZP(4)) \* LE - 0.0 - WMO \* P(4) \* LE - PINF)  
 1 CALL TERNIN(ION MATH )  
 1 IF (ZL(4) \* LE - TP) GO TO 400  
 1  
 1 ADIABATIC PROCESS: PUNCTURE BELOW TANK LIQUID LEVEL



PAGE 2

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FTN 4.6433H

76/74 OPT=1

SUBROUTINE ENERGY

```

105  WA-WAO=WVO=0.0
MACPAV=((MA(2)+CPA)+(MA(4)+CPA))/2.0
MLCPVAV=((ML(2)+CPL(2))+((ML(4)+CPL(4)))/2.0
MVCVBA=((MV(2)+CPVBA(2))+((MV(4)+CPVBA(4)))/2.0
OLDMLD=MLD
NCHANGE=1
C
C SECTION 1.1 PUNCTURE BELOW TANK LIQUID LEVEL
C CHECK FOR ISOTHERMAL OPTION
C
IF(HEATR-EG-1) T(4)=T(2)
IF(HEATR-EG-1) GO TO 120
VV(4)=CPL(4)+VTW(4)+((RG/MOLE)*T(4)+273.2)/CPL(1)+T(4)
DELTIME=((((VV(4)+VTW(4))-5*WVO)/AVERAGE(VL)+MLD)+((T(4)-T(2))-
1  AVERAGL(MV)+((VV(4)-VTW(4))-5*WVO)/AVERAGE(ML)+((VL(4)-VL(2)))/
2  ((VV(4)+VTW(4))-5-AVERAGE(VL)))
IF(ZL(2)-EG-0.0) GO TO 106
DELTIME=DELTIME-1.9/(AT+100.+(RG*(AVERAGE(T)+273.2)/MOLE)+.5/
1  AVERAGE(VL))+.25*(DELTIME/DI)+.1.25*DI
106  VV(4)=2.+(AVERAGE(VL)+DELTIME+AVERAGE(VL)+MLD+DI+2.)*AVERAGE(MV)
1  +VV(2)-AVERAGE(ML)+((VL(4)-VL(2)))/(DELTIME-WVO*DI+
2  2.*AVERAGE(MV))-VV(2)
PV(4)=CPL(4)+T(4)+RG*(T(4)+273.2)/(MOLE+VV(4))
PA(4)=(MA(4)+RG/28.96*(T(4)+273.2))/(VOL1-ML(4)+VL(4))
PT(4)=PV(4)+PA(4)
T(4)=(VOL1*(PT(4)-PT(2))+10000.)/J+AVERAGE(LAMBDA)*((ML(4)-ML(2))+
1  AVERAGE(LAMBDA)+MLD+DI)+MACPA*(T(4)-
2  AVERAGE(T))+DI)/(MLCPVAV+MVCVBA+MACPAV)+T(2)
C
C CALCULATE CORRECTED VALUES
C
C
VV4=VV(4)
CALL CORREC
VV(4)=VV4
120  CALL FLOW1
IF(INGELG-EG-2) RETURN
IF(MLD-GT-0.0) GO TO 130
DI=DMV/WVO
DML=MLD+DI
DMA=WAO+DI
GO TO 135
DI=DML/MLD
DMV=WVO+DI
DMA=WAO+DI
135  CONTINUE
ML(4)=ML(2)-MLD+DI-DELTIME
MV(4)=MV(2)+DELTIME
ZL(4)=ML(4)+VL(4)/AT
IF(ZL(4)-LE-IP) GO TO 400
CALL ZINCALC
C
C CHECK FOR CONVERGENCE
C
IF(AUSTWID-OLDMLD).LT(-.001*OLDMLD) GO TO 1000
ZL(4)=ZL(4)
ZLH(4)=ZLH(4)
PA(4)=PA(4)

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SUBROUTINE ENERGY 7/6/74 OPT-1
115 T(3)=T(4)
    GO TO 105
120 CONTINUE
    IF(CHANGE-EQ-1) CALL SCROLL
    IF(CHANGE-IQ-1) CHANGE=NCHANGE+1
    IF(CHANGE-EQ-2) RETURN
125 SECTION 1-2 TANK LIQUID LEVEL IS AT LEAST PARTIALLY BELOW PUNCTURE
    AND PT-61.PAIN
    C
    C
    C
    IF(PT(6)-LE-PAIN) GO TO 450
    MA-MAD-0.0
    MLCPLAV=((ML(2)+CPL(2))*(ML(4)+CPL(4)))/2.0
    MACPAV=((MA(2)+CPA)*(MA(4)+CPA))/2.0
    MUCPUBA=((MV(2)+CPUBAR(2))*(MV(4)+CPUBAR(4)))/2.0
    OLDUVO=UVO
130 C
    C
    C
    CHECK FOR ISOTHERMAL OPTION
    C
    IF(INEATIR-EQ-1) T(4)=T(2)
    IF(INEATIR-EQ-2) GO TO 420
    VV(4)=CPE(1,T(4))*(RG/MOLE)*T(4)+273.23/CPE(1,T(4))
    DELTIME=((((VV(4)+VVTWO)-VUVO)/AVERAGE(VL)+UVO)/T(4)-T(2))/
    1 AVERAGE(MV)+((VV(4)+VVTWO)-AVERAGE(VL)+VL(4)-VL(2))
    2 ((VV(4)+VVTWO)-VUVO)/AVERAGE(VL)
    IF(DELTIME-LT-0.0) GO TO 1000
    IF(DELTIME-GE-0.0) GO TO 406
    DELTIME=DELTIME-1.9/(AT-100.)*ERG*(AVERAGE(T)+273.23)/MOLE)+.5/
    1 AVERAGE(VL)+.5*(DELTIME/81)+.1.25*81
    406 MV(4)=MV(2)+UVO*81*DELTIME
    ML(4)=ML(2)+MLO*81*DELTIME
    IF(ML(4)-LT-0.0) GO TO 1000
    VV(4)=2.*(AVERAGE(VL)+DELTIME/AVERAGE(VL)+VL(4)-VL(2))/DELTIME-UVO*81+
    1 *VV(2)-AVERAGE(MV)+VV(2)
    2 *2-AVERAGE(MV)+VV(2)
    PV(4)=CPE(1,T(4))*RG*(T(4)+273.23)/(MOLE*VV(4))
    PA(4)=(MA(4)+RG/28.96*(T(4)+273.23)/VOLUME*ML(4)+VL(4))
    PT(4)=PV(4)+PA(4)
    T(4)=((VOLUME*(PT(4)-PT(2))+10000.)/)/AVERAGE(LAMUDA)+((ML(4)-ML(2)))/
    1 AVERAGE(LAMUDA)+UVO*81*MACPA+(TAIN-
    2 AVERAGE(T)+81)/(MLCPLAV+MUCPUBA+MACPAV)+T(2)
135 C
    C
    C
    CALCULATE CORRECTED VALUES
    C
    VV4=VV(4)
    CALL CORRECT
    VV(4)=VV4
    IF(TRANS(PT(4)-PTMG)-LT-.005+PTMG) MOVUVO=1
    IF(CARS(PT(4)-PTMG)-LT-.005+PTMG) GO TO 450
    420 CALL FLOW
    IF(INGTLE-EQ-2) RETURN
    GO TO 442
    430 CONTINUE
    442 IF(ULO-GE-0.0) GO TO 415
    DT=DMV/UVO
    ORI=ULO*DT

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FTM 4.04536

SUBROUTINE ENERGY 76/74 OPT=1

```

DMA=WAQ*DT
GO TO 440
435 DT=DM/L*WLO
DMV=WVO*DT
DMA=WAQ*DT
440 CONTINUE
MV(4)=MV(2)-WVO*DT+DELTIME
ML(4)=ML(2)-MLQ*DT+DELTIME
IF (ML(4)-LT-U.O) GO TO 1000
ZL(4)=ML(4)*VL(4)/AT
CALL ZENCALC
C
C CHECK FOR CONVERGENCE
C
185 IF (ABS(WVO-OLDWVO)-LE-(-.001)*OLDWVO) GO TO 1000
ZL(4)=ZL(4)
T(4)=T(4)
PA(4)=PA(4)
ZLH(4)=ZLH(4)
GO TO 405
450 CONTINUE
C
C SECTION 1-3 TANK LIQUID LEVEL IS AT LEAST PARTIALLY BELOW PUNCTURE AND
C PT.EQ.PATM
C
IF (PT(4)-LT.PATM) PT(4)=PATM
WAQ=WVO-U.O
MLCPAV=((ML(2)*CPL(2))+(ML(4)*CPL(4)))/2.0
MACPAV=((MA(2)*CPA)+(MA(4)*CPA))/2.0
MVCPUA=((MV(2)*CPUBAR(2))+(MV(4)*CPUBAR(4)))/2.0
OLDWLO=WLO
C
C CHECK FOR ISOTHERMAL OPTION
C
IF (THEATR-EQ.-1) T(4)=T(2)
IF (THEATR-EQ.-1) GO TO 470
VV(4)=CPF(3,T(4))*(RG/MOLE)*(T(4)+273.2)/(CPF(1,T(4)))
DELTIME=((((VV(4)+VVTMO)*.5+WVO+AVVERAGE(VL))*WLO)+(T(4)-T(2))*
1 AVERAGE(MV)*CVV(4)-VVTMO)-AVVERAGE(ML)*(VL(4)-VL(2)))/
2 ((VV(4)+VVTMO)*.5-AVERAGE(VL))
IF (ZL(2)-EQ.0) GO TO 456
DELTIME=DELTIME-1./((AT+100.)*(RG+(AVERAGE(T)+273.2)/MOLE)*.5/
1 AVERAGE(VL))*.25*(DELTIME/DT)*.1-.25*DT
VV(4)=2.*((AVERAGE(VL)*DELTIME+AVVERAGE(VL))*WLO+DT*2.*AVERAGE(MV)
1 *VV(2)-AVERAGE(ML)*CVL(4)-VL(2))/(DELTIME-WVO*DT)
2 *AVERAGE(MV))-VV(2)
PV(4)=CPF(3,T(4))*RG*(T(4)+273.2)/(MOLE*VV(4))
PT(4)=PATM
T(4)=((VOLT*(PT(4)-PT(2))+1000.)/J+AVVERAGE(LAMBDA))*(ML(4)-ML(2))+
1 AVERAGE(LAMUDA)*WLO+DT*MACPA*(TATM-
2 AVERAGE(T))+DT/(MLCPAV+MVCPUA+MACPAV)*T(2)
C
C CALCULATE CORRECTED VALUES
C
225 WV4=VV(4)
CALL CORRECT
VV(4)=WV4

```

```

230      470 CALL FLOWI
      IF (F6LG-68.23) RETURN
      IF (CLO-GT-0.0) GO TO 485
      DT=DRV/VVO
      DNL=VLO*DT
      DMA=VAG*DT
      GO TO 490
485      DT=DNL/VLO
      DNV=VVO*DT
      DMA=VAG*DT
490      CONTINUE
      ML(4)=ML(2)-VLO*DT-DELTIME
      MV(4)=MV(2)+DELTIME
      MAC(4)=(CPATM-PV(4))*((VOLT-ML(4))*C1/CPH(2,T(4)))
      T(4)=(T(4)+273.2)/28.96
      MA=(MA(4)-MA(2))/DT
      ZL(4)=ML(4)+VL(4)/AT
      CALL ZLUCALC
      T(4)=T(4)
      C
      C CHECK FOR CONVERGENCE
      C
      IF (ABS(CLO-OLDULO)-LE-(-.001*OLDULO)) GO TO 1000
      ZLH(5)=ZLH(4)
      GO TO 455
500      CONTINUE
      C *****RELIEF VALVE UNSTUCK*****
      C
      C SECTION 11-1 TANK LIQUID LEVEL IS ABOVE PUNCTURE AND
      C PT -GT- PATM-DELPVAL
      C
      IF (ZL(4)-GT-ZP-AND-PT(4)-GT-(PATM-DELPVAL)) GO TO 105
      C
      C SECTION 11-2 TANK LIQUID LEVEL IS ABOVE PUNCTURE AND
      C PT -LE- PATM-DELPVAL
      C
      IF (ZL(4)-GT-ZP-AND-PT(4)-LE-(PATM-DELPVAL)) GO TO 515
      GO TO 600
515      IF (PT(4)-LT-(PATM-DELPVAL)) PT(4)=PATM-DELPVAL
      MACPAV=(IMAC(2)+CPA)*MA(4)+CPA)/2.0
      MLCPIAV=(CPL(2)+CPL(2))*ML(4)+CPL(4))/2.0
      MVCPVUA=(CV(2)+CPVUA(2))*MV(4)+CPVUA(4))/2.0
      OLDULO=VLO
      C
      C CHECK FOR ISOTHERMAL OPTION
      C
      IF ((HEATIN-EU-1)*T(4)=T(4))
      IF ((HEATIN-EU-1)*GO TO 520
      VV(4)=CPI(3,T(4))*((RG/MOLE)*C(4)+273.2)/CPH(3,T(4))
      DELTIME=((CV(4)+VVTWO)*.5+VVO)/AVERAGE(VL)+VLO*(TIM(3)-TIM(2))-
      1 AVERAGE(MV)*((V(4)-VVTWO)*.5-AVERAGE(VI))
      2 ((V(4)+VVTWO)*.5-AVERAGE(VI))
      IF (ZL(2)-EQ-0.0) GO TO 516
      DELTIME=DELTIME-1.9/(CAT+100)*((MO-(AVERAGE(1)+273.2)/MOLE)*.5/
      1 AVERAGE(VI))*.25*(DELTIME/DT)+.1*.25*DT
      516      VV(4)=2.*(AVERAGE(VI))*0/(TIM(AVERAGE(VI))*ML*0+DT+.5)*AVERAGE(MV)

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FTM 4-04535F

76/74 DPE-1

SUBROUTINE ENERGY

```

1 0VV(2)=AVERAGE(ML)*(VL(4)-VL(2))/(DELTIME-WVO*DT)
2 2=AVERAGE(ML)*(V(2)-V(1))/(DELTIME-WVO*DT)
PV(4)=CPPE(2,T(4))*RG*(T(4)+273.2)/(HULE*VV(4))
PT(4)=PT(3)+PATM-DELPVAL
T(4)=(VOLT*PT(4)-PT(2))/10000.3/J+AVERAGE(LAMBDA)*(ML(4)-ML(2))*
1 AVERAGE(LAMBDA)*WLO*DT+WA*CPA*(T(4)-
2 AVERAGE(T))+DT)/(HUCPLA+WUCPUBA+MACPAV)*T(2)
C
C CALCULATE CONNECTED VALUES
C
VVA=V(4)
CALL CONNECT
VV(4)=VVA
520 CALL FLGWT
IF(ENGLG-10.2) RETURN
IF(WLO-GT-0.0) GO TO 530
D1=DMV/WVO
DML=WLO*DT
DMA=WAO*DT
GO TO 535
530 DT=DML/WLO
DMV=WVO*DT
DMA=WAO*DT
535 CONTINUE
ML(4)=ML(2)+WLO*DT-DELTIME
MV(4)=MV(2)+DELTIME
MA(4)=(PATM-DELPVAL-PV(4))*(VOLT-ML(4))*(1/CPPE(2,T(4)))/
1 (RG*(T(4)+273.2)/28.96)
WA=(MA(4)-MA(2))/DT
ZL(4)=ML(4)*VL(4)/AT
CALL ZLNCALC
C
C CHECK FOR CONVERGENCE
C
IF(ABS(WLO-OLDWLO)-LE-(.001*OLDWLO)) GO TO 1000
T(3)=T(4)
ZL(3)=ZL(4)
GO TO 515
520
C
C SECTION 11.5 TANK LIQUID LEVEL IS PARTIALLY BELOW PUNCTURE AND
C PT -GT- PATH
C
WA=WAO=0.0
IF(ZL(4)-ZPADP)-LE-0.0 AND WVO-LE-0.0 CALL TERMIN(10H MAJM )
IF(PT(4)-GT-PATH) GO TO 405
600
C
C SECTION 11.4 TANK LIQUID LEVEL IS PARTIALLY BELOW PUNCTURE AND
C PT -PATM-DELPVAL
C
IF(PT(4)-LE-(PATH-DELPVAL)) PT(4)=PATH
IF(ETIME-NE-1)WAO=WVO=0.0
IF(ETIME-NE-1) GO TO 455
PT(2)=PATM-DELPVAL
PT(4)=PATM
PV(4)=CPPE(2,T(4))
MA(4)=(PATM-PV(4))*(VOLT-ML(4))*(1/CPPE(2,T(4)))/
1 (RG*(T(4)+273.2)/28.96)
WA=(MA(4)-MA(2))/DT

```

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FIN 4.0433F

SUBROUTINE ENERGY 76/74 OPT=1

```

345      WAO=WVO=0.0
          WRITE(12,710) PIC(3),PIC(4),PAYM,ZI(3),ZL(4),ZP
          710 FORMAT(12,25HERROR IN CHECK STATEMENTS,/,
          1  2H PIC(3)= ,F15.5,BH PIC(4)= ,F15.5,ZH PAYM= ,F15.5,/,
          2  2H ZL(3)= ,F15.5,BH ZL(4)= ,F15.5,ZH ZP= ,F15.5)
          C CONVERGENCE ACHIEVED
          1000 CONTINUE
          RETURN
          END
350

```

PAGE 1

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FTW 6.61433F

SUBROUTINE ENERGY2 76/74 OPT=1

```

1      SUBROUTINE ENERGY2(KOUNT)
2      C
3      C
4      C
5      C
6      C
7      C
8      C
9      C
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FTW 4.464336

SUBROUTINE ENERGY 76/74 OPT-1

60 C  
C CONVERGENCE ACHIEVED  
C  
1000 CONTINUE  
RETURN  
END



SUBROUTINE FLOW	76/74	OP1=1	FIN 4.6433F	80/08/15. 13.45.07	PAGE 11
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SUBROUTINE FLOW1
C
C   CALLED FUNCTIONS: CPMF,CPLHF
COMMON /GEN/ PT(4),T(4),ZL(4),ZLH(4),MA(4),MV(4),ML(4),PINF,
1   VM,G,J,CD,AOL,AOG,RCI,RC,TAMB,TIME(4),MOLE,VOL1,
2   ZM,JP,DPM,SVB,NIB,CPA,SVL(4),SVV(4),WLO,WVO,MAO,
3   PAIM,AI,PP(4),LAMBDA(4),CPL(4),CPVHAT(4),TC,SVL,
4   PINFG,TI,CML,PA(4)
COMMON/ING/ DP,TML,TMV,THA,DRL,DNV,DMA,FLOSHU,ENRSUH,AP,INGFLG,
1   ICT,XLP,DELTH,ZNT(4),VOLB,MTEST,NCHANGE
REAL MA,MV,ML,MOLE,LAMDA,J
AVERAGE(1)=(X(4)+X(2))/2.0
TSAMU=CPFI6,PINF)
IF(TSAMU-CT.AVERAGE(1)) GO TO 120
VLAMH=1./CPFI2,TSAMU)
VWAMB= CPMF(3,TSAMU)*(RG/MOLE)*(TSAMU*273.2)/PINF
IF(AOL-EQ.0.0) WLO=0.0
IF(AOG-EQ.0.0) GO TO 121
C ***LIQUID
1   SASL=(CPFI(4,T(4))+CPFI(4,T(2)))*.5*ALOG((AVERAGE(1)+273.2)/
2   (TSAMU*273.2))
VOUTL=(VLAMB+SASL*(TSAMU*273.2)+*(VWAMB-VLAMH)/CPLHF(TSAMU)*.12
DUM1=(CPL(2)+CPL(4))*.5*(AVERAGE(1)-TSAMU)-
1   ((TSAMU*273.2)+SASL)*(AVERAGE(PT)-(CPFI(3,T(4))+CPFI(1,T(2)))*
2   -.5))/
2   *10000.*(1./CPFI2,AVERAGE(1))+(AVERAGE(ZL)-AVERAGE(ZLH))*G/J
C
IF(Z2-J*SDUM1-LT.0.0) INGFLG=2
IF(INGFLG-EQ.2) RETURN
WLO=(CO*AOI/VOUTL)*I2-J*SDUM1)*.5
GO TO 121
120 DUM1=(10000.*(AVERAGE(PT)-PINF)*(1./CPFI2,AVERAGE(1)))+G*
1   (AVERAGE(ZL)-AVERAGE(ZLH))*2.0
IF(DUM1-LT.0.0) INGFLG=2
IF(INGFLG-EQ.2) RETURN
WLO=(CO*AOI*SDUM1)*.5)/(1./CPFI2,AVERAGE(1))
121 IF(AOG-EQ.0.0) WVO=WAO=0.0
IF(AOG-LQ.0.0) RETURN
C ***VAPOR
C
C   FIRST CASE
IF(TMA(4).NE.0.0) GO TO 500
TSAMU=CPFI(6,PINF)
VLAMH=1./CPFI2,TSAMU)
VWAMU=CPFI(3,TSAMU)*(RG/MOLE)*(TSAMU*273.2)/PINF
SVSA= CPMF(3,AVERAGE(1))*G3/MOLE*ALOG(AVERAGE(PT)/PINF)-
1   (CPFI(5,T(4))+CPFI(5,T(2)))*.5*ALOG((AVERAGE(1)+273.2)/
2   (TSAMU*273.2))
VOUTV=VWAMB-(SVSA*(TSAMU*273.2)+*(VWAMB-VLAMH)/CPLHF(TSAMU)*.12
DUM2=(CPFI(5,T(4))+CPFI(5,T(2)))*.5*(AVERAGE(1)-TSAMU)
1   *(TSAMU*273.2)+SVSA)*2.0+J
IF(DUM2-LT.0.0) INGFLG=2
IF(INGFLG-EQ.2) RETURN
WVO=(DUM2+.5*CO*AOI)/VOUTV
WAO=0.0
RETURN
C

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```

C      SECOND CASE      MA,ME,0.0
C
500  RV=RG/MOLE
      RA=RG/28.96
      RA=( (RV*RV(4) + RA*MA(4))/(MA(4) + MV(4)) +
      + (RV*RV(2) + RA*MA(2))/(MA(2) + MV(2)))/2.0
      BUMS=( (MV(4)+CP(4),T(4)) + MV(2)+CP(4),T(2)))/2.0 +
      + (MA(4)+CPA(4),2)+CPA(2))/2.0
      CPM= BUMS / ((MA(4)+MV(4)+MA(2)+MV(2))/2.0)
C
      MV1=RG1/MOLF
      RA1=RG1/28.96
      RM1= ((MV1*MV(4)+RA1*MA(4))/(MA(4)+MV(4)) +
      + (MV1*MV(2)+RA1*MA(2))/(MA(2)+MV(2)))/2.0
      GANN= CPM/(CPH-RM1)
      DUM4=((((PINGF/AVERAGE(PF)))+(2./GANN)-(PINGF/AVERAGE(P1)))**
      + (GANN*1.5/GANN))**2./6/GANN/IGANN-1.3)
      IF DUM4<.LT.-0.0) IN6IG=2
      IF DUM4<.EQ.-2) RETURN
      WVO= (100.-AVERAGE(MV)+CP*AOG+AVERAGE(P1)*DUM4**-.53/
      + (RV(4)+MA(4)+MV(2)+MA(2))/2. + (RM-(AVERAGE(B)+275.2))**-.53)
      WAO= AVERAGE(MA)/AVERAGE(MV)*WVO
      RETURN
      END

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FTW 4.0433F

76/74 OPT=1

SUBROUTINE FLOW2

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1      SUBROUTINE FLOW2
2      C
3      C      CALLED FUNCTIONS: CPF
4
5      COMMON /GEN/ PT(4), T(4), ZL(4), ZLH(4), MA(4), MV(4), ML(4), PINF,
6      VW, G, J, CO, AOL, AOG, RGI, RG, TAMB, TIME(4), MOLE, VOLT,
7      ZW, TP, BPV, SVQ, HTD, CPA, SVL(4), SVV(4), WLO, WVO, WAO,
8      PATM, AT, PV(4), LAMBDA(4), CPL(4), CPUBAR(4), TC, SVA,
9      PINTG, ZI, CRI, PA(4)
10     COMMON /ING/ DP, TML, TMV, TMA, DML, DMV, DMA, FLOSUB, ENRSUB, AP, INGFLG,
11     ICT, XLP, DELTIME, ZUT(4), VOLB, NTEST, NCHANGE
12
13     REAL MA, MV, ML, MOLE, LAMBDA, J
14     CALCULATE WLO, WVO, WAO
15     LIQUID
16     AVERAGE(X) = (X(4) + X(2))/2.0
17     MV = T2
18     AT = (10000. * (AVERAGE(PT) - PINF) * (1. / CPF(2, AVERAGE(T))) * G +
19     (AVERAGE(ZL) - AVERAGE(ZLH))) * 2.0
20
21     IF (AT.LT.0.0) INGFLG = 2
22     IF (INGFLG.EQ.2) RETURN
23     WLO = (CO * AOL * AT ** .5) / (1. / CPF(2, AVERAGE(T)))
24     IF (AOG.EQ.0.0) WAO = WVO = 0.0
25     IF (AOG.EQ.0.0) RETURN
26
27     GAS
28     RV = RG / MOLE
29     RA = RG / ZB * 96
30     RM = ((RV * MV(4) + RA * MA(4)) / (MA(4) + MV(4))) +
31     (RV * MV(2) + RA * MA(2)) / (MA(2) + MV(2)) / 2.0
32     GUM3 = (MV(4) * CPF(5, T(4)) + MV(2) * CPF(5, T(2))) / 2.0 +
33     (MA(4) * CPA * MA(2) * CPA) / 2.0
34     CPM = GUM3 / ((MA(4) * MV(4) + MA(2) * MV(2)) / 2.0)
35
36     RV1 = RG1 / MOLE
37     RA1 = RG1 / ZB * 96
38     RM1 = ((RV1 * MV(4) + RA1 * MA(4)) / (MA(4) + MV(4))) +
39     (RV1 * MV(2) + RA1 * MA(2)) / (MA(2) + MV(2)) / 2.0
40     GPM = CPM / (CPM - RM1)
41
42     A2 = (100. * AVERAGE(MV) * CO * AOG * AVERAGE(PT)) / ((MV(4) * MA(4) + MV(2) *
43     MA(2)) / 2.0 * (RM * (AVERAGE(T) * 273.2))) ** .5
44     A3 = (PINF * G / AVERAGE(PT)) * (2. / GPM) - (PINF * AVERAGE(PT)) **
45     ((GPM + 1. ) / GPM)
46     IF ((2. * GPM) / (GPM - 1.)) * A3.LT.0.0 INGFLG = 2
47     IF (INGFLG.EQ.2) RETURN
48     WVO = A2 * ((2. * GPM) / (GPM - 1.)) * A3 ** .5
49     WAO = AVERAGE(MA) / AVERAGE(MV) * WVO
50     RETURN
51     END

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FTW 4.4433F

76/74 OPI=1

FUNCTION GAMM

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1  FUNCTION GAMM(RA)
2  CALLER FUNCTIONS: CPF
3  COMMON /GEN/ PTES, T(4), TL(4), ILH(4), MA(4), MV(4), RL(4), PINF,
4  VM-G, J, COOH, AOL, AOG, RGT, RG, IATM, TIM(4), MOLE,
5  VOLT, ZU, ZP, DELPVAL, IVALVE, IHEATIR, CPA, VL(4), VV(4),
6  WLO, WVU, WAO, PATM, AT, PV(4), LAMBDA(4), CPL(4), CPVBAR(4),
7  TC, SVA, PINFG, ZI, CNI
8  CALCULATE GAMMA N
9  REAL MA, MV, RL, J, MOLE, LAMBDA
10 RV1=RG1/MOLE
11 NV=RG/MOLE
12 RAT=RG1/28.06
13 CPVDAR(2)=CPF(5,1(2))
14 CPVDAR(4)=CPF(5,1(4))
15 RM1=((RV1+NV(2)*RAT+MA(2))/(MA(2)+MV(2)))
16 RM1=((RV1+NV(4)*RAT+MA(4))/(MA(4)+MV(4)))/2.0
17 CPM=((NV(2)+CPVDAR(2)*MV(4)+CPVDAR(4))/2.0)
18 ((MA(2)+MA(4))*CPA/2.0)/
19 ((MA(2)+MA(4)+NV(2)+NV(4))/2.0)
20 GAMM=CPM/(CPM-RM1)
21 RETURN
22 END

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PAGE 1

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FTM 6.46433F

SUBROUTINE INGEST 76/74 OPT=1

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1      SUBROUTINE INGEST
2      CALLED SUBROUTINES: AIRMASS, IERMIN
3      CALLED FUNCTIONS: CPE
4      COMMON /GEN/ PT(4), I(4), ZL(4), ZLH(4), ML(4), MV(4), PINF,
5      VM, G, J, COORE, AOL, AOG, RGI, RG, TIME(4), CMUT,
6      VI, IV, IP, DPV, SVD, MTD, CPA, SVL(4), SVV(4), MLO,
7      WVO, WAO, PATM, AT, PV(4), LAMODA(4), CPL(4), CPVOR(4),
8      IC, SVA, PINIG, ZT, CME, PA(4)
9      COMMON/ING/ DP, IML, IMV, IMA, DML, DMV, DRA, FLOSUB, ENRSUB, AP, INGLG,
10     1 ICT, XLP, DELIME, ZUT(4), VOLB, NTEST, NCHANGE
11     COMMON/WAT/ZLEQI, PTENI
12     INTEGER HTD, SVD, FLOSUB, ENRSUB
13     REAL ML, MV, RA, J, LAMODA, MLEQI, MAI, MVI
14     NV=12
15     WRITE(12,10)
16     FORMAT(1H ,14H ... T TO INGEST)
17     IF (ZM-GT.(ZP-DP)) GO TO 900
18
19     C WATER BELOW PUNCTURE
20     IF (ZL(2)-GT.ZP) GO TO 150
21
22     C INGESTION IMPOSSIBLE
23     RETURN
24
25     C 150 IF(INGLG.NE.Z.AND.PT(3).GE.PATM) GO TO 100
26     IF(INGLG.EQ.2) GO TO 160
27     IF(ZL(3)-ZP+.5*DP+.1*DE+.4*PT(3)+SVL(3)/6.-GT.-1.001+
28     1 .1*DE+.4*PATM+SVL(3)/6) GO TO 100
29     C SOMETHING HAPPENED... ONLY PARAMETERS 1 AND 2 ARE VALID.
30     ICT=ICT+1
31     WRITE(12,90) ZL(3), PT(3), SVL(3), PATM
32     FORMAT(1H ,8H ZL(3)= ,F10.5, 8H PT(3)= ,F10.5, /,
33     1 9H SVL(3)= ,F10.5, 7H PATM= ,F10.5)
34     INGLG=1
35     RETURN
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C
901 CARGO LIQUID WHICH FLOATS WILL AGAIN COME OUT. RETURN MAIN/MOOUT
    ICF=ICF-1
    IF(SVL(2)-LE-VW) CALL TERMINATIONGEST )
    INGLG=4
    RETURN
C
930 WATER IS ABOVE TOP OF PUNCTURE AND LIQUID INTERSECTS PUNCTURE
    IF(SI-PAINT-NE-4+CA-0.5*(ZP+ZL(2)-DNL+SVL(2)/AT)/VW
    IF(INGLG-EQ-0-AND-PT(3)-GE-TEST) GO TO 100
C
    AT THIS POINT THE WATER MAY BE LIGHTER OR HEAVIER THAN THE LIQUID.
C
    IF WATER IS LIGHTER, THEN THIS IS LIKE KTYP=4 EXCEPT NO EVAPORATION
C
    IS POSSIBLE AND SIMPLE OUEFLOW OCCURS
    ICF=ICF-1
    IF(SVL(2)-LI-VW) GO TO 940
    INGLG=4
    RETURN
940
    MAT=MA(2)
    PVT=PV(2)
    DNL=ZL(2)-ZP+DP)*AT/SVL(2)/10.
    ZLI=ZL(2) * PVI-PT(2)
    TTI=TT(2)
    RMT=R63/CRMT
    CPVBAR(2)=CPRES,TTI)
    CPV=CPVBAR(2)
    GAMMA=CPM/(CPM-RMT)
    IF(RT0-EQ-1) GAMMA=1.0
    TOP=VT-AT+ZLI
    GAMLES=(GAMMA-1.0)/GAMMA
    ZL(4)=ZL(2)-DNL+SVL(2)/AT
    IF(ZL(4)-LE-ZP+DP) ZL(4)=ZP+DP
    IF(ZL(4)-LE-ZP+DP) DNL=ZL(2)-ZL(4)*AT/SVL(2)
    ML(4)=ML(2)-DNL
    MV(4)=MV(2)
    VLAVG=SVL(2)
    ZLAVG=(ZL(2)+ZL(4))*0.5
    ADUT=XLP*(ZLAVG-ZP+DP)
    WLO=COORF*ADUT*(G*(ZLAVG-ZP+DP)*(1-VLAVG/VW))*0.5/VLAVG
    DT=(ZL(2)-ZL(4))/VLAVG/WLO
    TML(4)=TIME(2)+DT
    IF(SVD-EQ-1) PT(4)=PVI
    IF(SVD-EQ-1) GO TO 945
    PT(4)=PT(2)
942
    BOT=VT-AT+LI-0.5*(VW/G*(PVTM-PT(1))+ZU)
    PT(4)=PVI*(1+PVT/BOT)*GAMMA
    T(4)=TTI
    IF(RT0-EQ-1) T(4)=(ZT3-ZT1)*PT(4)/PT(1)*GAMLES-273.2
    IF(AUS(PT(3)-PT(4))/PT(4))-LC--01) GO TO 945
    P(4)=PT(4)
    GO TO 942
945
    ZML(4)=(ZL(4)+P-OP)*0.5
    PA(4)=PAI
    PV(4)=PVI
    PV(4)=PV(2)+T(4)/PT(2)
    IF(SVD-EQ-1) T(4)=TTI
    ICF=ICF+1
    ENL=ENL+DNL
    SVL(4)=SVL(2)
    SVD(4)=SVD(2)

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PAGE 3

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SUBROUTINE INGEST 76/74 OPT=1

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115 C
116 C   PARAMETERS SUB 4 MAY BE PRINTED
117 IF(ZL(4).LE.-2P-OP) GO TO 946
118 GO TO 949
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175 IF(INGFLG-NE-2-AND-ZL(4)-LE-2P) GO TO 2100
    IF(INGFLG-NE-2-AND-PT(4)-GE-PATM) GO TO 100
    GO TO 150
180 2100 IF(INGFLG-ML-2-AND-ZL(4)-LT-70) GO TO 2200
    C LIQUID AND WATER INTERSECT PUNCTURE WITH LIQUID ABOVE WATER
    IF(SVL(2)-GT-VW) GO TO 2150
    IF(INGFLG-ML-2-AND-PT(4)-GT-PATM) GO TO 100
    C ADMIT AIR UNTIL PT(2)=PATM. REINITIATE 3 AND 4 VALUES AND GO TO 300 EVENT
    ICF=ICF-1
    INGTG=5
    PT(4)=PATM
    PV(4)=PV(2)
    ML(4)=ML(2)
    TL(4)=TL(2)
    CALL AIRMASS
    PT(2)=PATM
    MA(2)=MA(4)
    PT(1)=PATM
    PINT=PATM
    RETURN
190 2150 TEST=(SVL(4)/VW-1.3)*5*(ZU-ZP+DPI)+SVL(4)+1.64*(PATM-PT(4))/G
    IF(INGFLG-NE-2-AND-ZL(4)-2M-GT-TEST) GO TO 100
    DO 21501 1-3,4
195 21501 ML(1)=1
    MV(1)=MV(1-1)
    MA(1)=MA(1-1)
    TL(1)=TL(1-1)
    ZL(1)=ZL(1-1)
    ZU(1)=ZU(1-1)
    SVL(1)=SVL(1-1)
    21501 SVU(1)=SVU(1-1)
    PT(1)=PT(4)
    ZL1=ZL(2)
    ZU1(1)=ZU1(4)=0.
    DT=TIME(4)-TIME(2)
    IF TIME(4)-TIME(2) GO TO 2152
    IF(MLO-GT-0.) GO TO 2151
    IF(MVO-EG-C-DOF=1-0
    IF(MVO-EG-0.) GO TO 2152
    DT=DWV/DVO
    GO TO 2152
200 2151 DT=DW1/W10
    2152 CONTINUE
    IF(INGFLG-NE-2-AND-PT(4)-GT-PATM) GO TO 2153
    IF(INGFLG-NE-2-AND-PT(4)-LE-PATM) PT(2)=PT(4)=PATM
    IF(PT(2)-LE-PATM) PT(2)=PT(4)=PATM
205 2153 IF(PT(2)-LE-PATM) GO TO 2158
    C ASSUME CARGO NOT VOLATILE AND VENT VAPOR ONLY THRU (ZP-ZL1)*RLP
    C ON (ZP-ZL(2))*RLP, WHICH EVER IS LARGER. VENT VAPOR OVER TIME STEPS
    C DT UNTIL PT(4)=LE-PATM.
    C FOLLOWING IS VAPOR VENT IF IT OCCURS
    MV1=RV1/CMV1
    MV=RG/CMV1
    RM1=(RV1+MV(2)+MA1+MA(2))/(MV(2)+MA(2))
    RM=(RV+MV(2)+MA+MA(2))/(MV(2)+MA(2))
    CPM=(MV(2)+CPE(5, TL(2))+MA(2)+CPA)/(MV(2)+MA(2))
    GAMM=CPM/(CPM+RM1)

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444909-7 M14

SUBROUTINE INGEST 76174 OPI=1

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2230      AOG=(IP-ZL(2))*XLP
      SUB=(IP-ZL(2))*XLP
      IF (DUM-6I.AOG) AOG=0.0M
2235      AVGRV=(MV(12)*MV(4))*-.5
      AVGPT=(PT(12)*PT(4))*-.5
      AVGT=(T(12)*T(4))*-.5+ZT3.2
      AVGMA=(MA(2)*MA(4))*-.5
      A2=(100.*AVGRV+CDORT*AVGPT)/(AVGMA*AVGRV)*(RR*AVGT)*.5)
      A3=(PAIM/AVGPT)*.5*(2./GAMM)-(PAIM/AVGPT)*.5*(GAMM1./GAMM)
      A3=AMAT(0,0,A3)
      VVO=AD2+SQR(2.*GAMM*A3/(GAMM-1.))
      MAO=MVO+AVGPA/AVGRV
      MA(4)=MA(2)-MAO*DT
      MV(4)=MV(2)-MVO*DT
      VM1=(V1-ZL1*AT)/(MV(1)*MA(2))
      VM2=(V1-ZL1*AT)/(MV(4)*MA(4))
      PT(4)=PT(2)*(VM1/VM2)*.6GAMM
      T(4)=T(2)
2240      T(2)=T(2)+ZT3.2
      IF (HTD-.EQ.1) T(4)=T(2)*(PT(4)/PT(2))*((GAMM-1.)/GAMM)-ZT3.2
      IF (AOS(MV(2)*MV(4))-AVGMA*2.)/AVGRV/2.-GE.-01) GO TO 2154
2245      IF ZMT(2).GE.ZM-ZL(1)*ZL(2))*-.5
      AOUT=(ZL1-ZM)*XLP
      ML0=ADT*CDORT*(G*(ZL1-ZM))*-.5/SVL(2)
      MV=SVL(2)*ML0/VW
      ZM(4)=ZMT(2)*MV+MV*DT/AT
      ZL(4)=ZL(2)-ML0*DT*SVL(2)/AT
      IF ZMT(4).LE.ZM-ZL(4)*ZL(4)=ZL(2)
      ML(4)=ML(2)-ML0*DT
      ML(3)=ML(4) $ ZL(3)=ZL(4) $ MV(3)=MV(4) $ MA(3)=MA(4) $ PT(3)=PT(4)
      T(3)=T(4) $ ZL(3)=ZL(4) $ ZLH(3)=ZLH(4)
      TIME(3)=TIME(4)=TIME(2)*DT $ ZC1=ZC1+1
      PV(3)=PV(4)=CP(1,T(4))
      SVL(3)=SVL(4)=SVL(2) $ SVV(3)=SVV(4)=(VT-ZL1*AT)/MV(4)
      TML=TML+ML0*DT
      IMV=TMV+MVO*DT
      TMA=TMA+MAD*DT
2250      PRINTOUT OF PARAMETERS SUB 4 AND TML,TMV,TMA MAY BE DONE HERE
      IF ZMT(4).GE.ZM-AND-(ZL(4).LE.ZM) GO TO 2161
      SCROLL AND CONTINUE
      DO 2160 I=1,5
2255      PT(1)=ML(1+I) $ MV(1)=MV(1+I) $ MAC(1)=MAC(1+I)
      PT(1)=PT(1+I) $ T(1)=T(1+I) $ ZL(1)=ZL(1+I) $ ZLH(1)=ZLH(1+I)
      TIME(1)=TIME(1+I) $ PV(1)=PV(1+I)
      SVV(1)=SVV(1+I)
2260      GO TO 2153
2265      CALL TERMIM(TUNGEST)
      IF (INGELG.ML-2-AND-ZL(4).LE.ZP-OP) GO TO 2300
      WATER AND LIQUID INTERSECT PUNCTURE WITH WATER ABOVE LIQUID CARGO
      IF (SVL(2).LE.VM) GO TO 2230
      GO TO 2260
2270      TEST=PARAM1-DE-4*G*(ZM-ZL(4))*VM*-.5
      IF (INGELG.ML-2-AND-PT(4).GT.HST) GO TO 100
      ICD=ICD-1
      PT(2)=PTM
      DT=TIME(4)-TIME(1)
2275
2280

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SUBROUTINE INGEST 70/74 ,OPT-1

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IF (TIME(4)-.0E+00) GO TO 2212
IF (TIME(4)-.0E+00) GO TO 2211
IF (TIME(4)-.0E+00) DT=1.0
IF (TIME(4)-.0E+00) GO TO 2212
DT=DW/MVO
GO TO 2212
2211 DT=DMR/MLO
2212 CONTINUE
MVO=MVO-U.0
MLO=MLO-U.0
ZL(4)=ZL(2)
ZL(4)=ZL(4)+ZL(2)*.5
ADUT=ELP*(ZLAVG0-ZP*DP)
MLO=ADUT*GOURT*(U*(ZLAVG0-ZP*DP)+(1.-SVL(2)/VWU)*.5)/SVL(2)
ZL(4)=ZL(2)+MLO*DT+SVL(2)/AT
IF (ZL(4)-ZL(2)-ZP*DP) CALL TERMIN(10) INGEST )
IF (ABS(ZL(4)-2.*ZLAVG0-ZL(2))-GE.0.01*ZL(4)) GO TO 2213
2214 ML(4)=ML(2)+ML*DT $ ML(1)=ML(2) $ ML(2)=ML(4)
ML(3)=ML(4)
MV(3)=MV(4)+MV(2)
MV(1)=MV(2)
MA(3)=MA(4)+MA(2) $ MA(1)=MA(2)
PT(3)=PT(4)+PT(2) $ PT(1)=PT(2)
ZL(3)=ZL(4) $ ZL(1)=ZL(2) $ ZL(2)=ZL(4)
T(3)=T(4)+T(2) $ T(1)=T(2)
ZLHC(3)=ZLHC(4)+ZL(4)*ZP*DP*.5 $ ZLHC(1)=ZLHC(2)+ZLHC(4)
TIME(3)=TIME(4)+TIME(2)*DT $ TIME(1)=TIME(2)+TIME(4)
PV(3)=PV(4)+PV(2) $ PV(1)=PV(2)
SVL(3)=SVL(4)+SVL(2) $ SVL(1)=SVL(2)
SVV(3)=SVV(4)+SVV(2) $ SVV(1)=SVV(2)
IML=IML+MLO*DT
C PRINT PARAMETERS 2 HERE
GO TO 2215
2260 IF (ST-PAINT-UE-4*U*(ZM-ZL(4))/VWU*.5
1E+00) GO TO 2261
ICT=ICT+1
PT(2)=PAINT
ZM(2)=APAXI(0.0,ZM-ZL(2))
ADUT=0.5*(ZM-ZP*DP)*ATP $ ZL(2)=ZM
DO 22601 1-3,4
ML(1)=ML(1-1) $ MV(1)=MV(1-1) $ MA(1)=MA(1-1)
T(1)=T(1-1) $ ZL(1)=ZL(1-1) $ ZLHC(1)=ZLHC(1-1)
SVL(1)=SVL(1-1) $ SVV(1)=SVV(1-1) $ ZM(1)=ZM(1-1)
22601 PT(1)=PT(1-1)
DT=TIME(4)-TIME(2)
IF (TIME(4)-.0E+00) GO TO 2262
IF (TIME(4)-.0E+00) DT=1.0
IF (TIME(4)-.0E+00) DT=1.0
IF (TIME(4)-.0E+00) GO TO 2262
DT=DW/MVO
GO TO 2262
2261 DT=DMR/MLO
2262 AC=ADUT*GOURT*(U*(ZM-ZP*DP)+(1.-VWU/SVL(2))
AD=AC*SORT(U.5*(ZM-ZP*DP))
2263 MW=AU $ ZM1AVG=(ZM(2)+ZM(4))*0.5
IF (ZM1AVG-GT-ZP*DP) MW=AC*SORT((ZM-ZM1AVG)*0.5*(ZM-ZM1AVG)
1-(ZP*DP))/(ZM-ZP*DP)
ZM(4)=VWU*U*DT/AT+ZM(2)

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AD-A095 413

SOUTHWEST RESEARCH INST SAN ANTONIO TEX F/G 13/10  
EXPERIMENTAL VERIFICATION AND REVISION OF THE VENTING RATE MODE--ETC(U)  
NOV 80 F T DODGE, E B BOWLES, J E MANN DOT-C6-73623-A

UNCLASSIFIED SWRI-02-5295

USCG-D-63-80

NL

3 of 3  
ALL  
REMOVED

END  
DATE  
FILMED  
4-2-81  
DTIC

PAGE 7

80/08/15. 13.45.07

PTM 4.4433F

76/74 OPT=1

SUBROUTINE INGEST

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345 IF(ABS(ZMT(4))-2.*ZMTAVG-ZMT(2)).GE.0.01*ZMT(4)) GO TO 2263
    WLO=WM*SVL(2)/VM
    ZL(4)=ZL(2) $ ZL(3)=ZL(2)
    ML(4)=ML(2)-WLO*DT $ ML(3)=ML(4)
    ICT=ICT+1
    TIME(3)=TIME(4)-TIME(2)*DT $ ZMT(3)=ZMT(4)
    TML=TML+WLO*DT
    PRINTOUT OF PARAMETERS SUB 4 AND TML,TMV,IMA MAY BE DONE
    IF(ZMT(4).GE.ZM) GO TO 2266
    C
    C
    DO 2264 I=1,1
    ML(1)=ML(1+1) $ MV(1)=MV(1+1) $ MA(1)=MA(1+1) $ PT(1)=PT(1+1)
    ZM(1)=ZM(1+1)
    T(1)=T(1+1) $ ZL(1)=ZL(1+1) $ ZLH(1)=ZLH(1+1)
    TIME(1)=TIME(1+1)
    PV(1)=PV(1+1) $ SVL(1)=SVL(1+1) $ SVV(1)=SVV(1+1)
    GO TO 2265
    2264 CALL TERMIN(IONINGEST )
    2300 CONTINUE
    C
    WATER INTERSECTS PUNCTURE AND LIQUID CARGO IS BELOW THE BOTTOM
    OF THE PUNCTURE
    TEST=PTM*1.0E-4*G*0.5*(ZU-ZPOB)/VM
    IF(INGL6.NE.2-AND.PT(4).GT.1.002*TEST) GO TO 100
    IF(ZL(2).EQ.0.0) GO TO 2264
    IF(SVL(2).GT.VM) GO TO 2360
    C
    WATER LIGHTER- ONLY GAS MAY FLOW OUT
    AOUT=XLP*(ZP-ZM) $ ICT=ICT-1 $ AOG=AOUT
    RV1=RG1/CHW1 $ RV=RG/CHW1 $ RA1=RG1/28.96 $ RA=RG/28.96
    RMI=(RV1-MV(2)*RA1+MA(2))/MV(2)+MA(2)
    RM=(RV+MV(2)*RA+MA(2))/MV(2)+MA(2)
    CPM=(MV(2)*CPF(5,T(2))+MA(2)*CPA)/(MA(2)+MV(2))
    GARM=CPM/(CPM-RMI)
    PT(2)=PT(2)*(ZT-ZL(2))/(ZT-ZU)+GARM
    C
    GARM TRANSFERRED FROM
    ML(3)=ML(4)-ML(2) $ WLO=0.0
    ZL(3)=ZL(4)-ZL(2) $ SVV(3)=SVV(4)-SVV(2)
    ZLH(3)=ZLH(4)-ZLH(2) $ MV(3)=MV(4)-MV(2)
    SVL(3)=SVL(2)-SVL(4) $ MA(3)=MA(4)-MA(2)
    PT(3)=PT(4)-PT(2) $ T(3)=T(4)-T(2)
    IF(WVO.EQ.0.0) DT=1.0 $ ZMT(3)=ZMT(4)-ZMT(2)-ZM
    IF(WVO.EQ.0.0) GO TO 2310
    DT=DMV/WVO
    2310 CONTINUE
    RMI=(RV1-MV(2)*RA1+MA(2))/MV(2)+MA(2)
    RM=(RV+MV(2)*RA+MA(2))/MV(2)+MA(2)
    CPM=(MV(2)*CPF(5,T(2))+MA(2)*CPA)/(MA(2)+MV(2))
    GARM=CPM/(CPM-RMI)
    AVGNV=(MV(2)+MV(4))*0.5 $ AVGPT=(PT(2)+PT(4))*0.5
    AVGT=(T(2)+T(4))*0.5+273.2 $ AVGMA=(MA(2)+MA(4))*0.5
    A2=(100.*AVGNV*CDORF*AOG*AV*PT)/(CAVGMA*AVGNV)*(RM+AVGT)*0.3)
    A3=(PATH/AVGPT)*0.2/GARM-(PATH/AVGPT)*0.1/GARM
    2312 A3=MAX1(0.0,A3)
    WVO=A2*SQRT(2.*GARM*A3/(GARM-1.))
    WAO=WVO*AVGMA/AVGNV
    MA(4)=MA(2)-WAO*DT
    MV(4)=MV(2)-WVO*DT
    VM)=(VM-ZL(2)*AT)/(MV(2)+MA(2))

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SUBROUTINE INGEST  76774  OPT=1          FIM 4.6433F      80/88/15. 13.45.07      PAGE 8
400  UN2=CVT  -ZL(4)*AT3/(UN1(4)*NA(4))
      PT(4)=PT(2)*UN1(4)/UN2(4)*GAMM
      T(4)=T(2)  8  T(2)=T(2)*273.2
      IF(NTB  .EQ.13) T(4)=T(2)*(PT(4)/PT(2))*((GAMM-1.0)/GAMM)-273.2
      IF(ABS(UN2)*UN1(4)-ANGM*2.)/ANGM/2.0.GE.0.01) GO TO 2311
      ICT=ICT+1
      TNU=TRN*UN2*OT
      TNA=TN*UN2*OT
      C  SCROLL , PRINT , CHECK, THEN ALLOW FOR 2311
      2360 ZL(4)=ZF-OP
      GO TO 2260
      END
410

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SUBROUTINE NUMBER 76/74 OPT=1 PAGE 1

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FTN 6.4-53F

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1      SUBROUTINE MOMENT(TIME)
2
3      C
4      C DETERMINE PRESSURE AND TEMPERATURE OF THE TANK CARGO
5      C AS LIQUID OR VAPOR IS VENTED.
6
7      C
8      C CALLED SUBROUTINES: FLOW2, TERNIN, ZLNCALC
9      C CALLED FUNCTIONS: CPT, GAMM
10     C REAL LAMDA, J, MOLE, NA, NL, NV
11     C COMMON /GEN/ PT(3), T(4), ZL(3), ZL(4), ZL(5), NA(3), NV(4), NL(4), PINT,
12     C VU, C, J, COEFF, AOL, AOG, AGI, AG, TAIN, TIN(3), MOLE,
13     C VOL1, ZU, ZP, DELTVAL, IVALVE, IHEATIN, CPA, VU(3), VU(4),
14     C WLB, WVB, WAB, PATR, AT, PV(4), LAMDA(4), CPL(4), CPVAB(4),
15     C TC, SWA, PINTG, IT, CHI, PA(4)
16     C COMMON/ING/ OP, INL, INW, INA, ONL, ONW, ONA, FLOSUB, ENRSUB, AP, INGFLG,
17     C ICT, HLP, DELTIN, INT(4), VOLB, NTEST, NCHANGE
18     C AVERAGEC=(C(2)+C(4))/2.0
19     C NA=RG/ZB.94
20     C NV=RG/MOLE
21
22     C
23     C IS VALVE UNSTUCK?
24     C
25     C IF (IVALVE-.EQ.)GO TO 310
26
27     C RELIEF VALVE STUCK
28     C
29     C IS TANK LIQUID LEVEL PARTIALLY BELOW PUNCTURE
30     C
31     C IF (ZL(4)-ZP+OP)-LE-.0.AND.-WVB.LE-.0) CALL TERNIN(TAIN)
32     C IF (ZL(4)-LE-.10)GO TO 200
33
34     C ADIABATIC PROCESS
35     C
36     C SECTION 1.1 TANK LIQUID LEVEL IS ABOVE PUNCTURE
37     C COMPUTE GAMM FOR INITIAL ITERATION
38     C NA=WAB-WVB=0.0
39     C NA(3)=NA(3)-NA(2)-NA(1)
40     C NV(4)=NV(3)-NV(2)-NV(1)
41     C GAMM=GAMM(NA)
42
43     C IF (IHEATIN-.EQ.-1) GAMM=1.0
44     C ONL=0.0
45     C VL(2)=1./CPT(2,TC)
46     C VL(4)=1./CPT(2,TC)
47     C CALL FLOW2
48     C IF (INGFLG-.EQ.-2) RETURN
49     C IF (WLB-.GT-.0) GO TO 125
50     C DT=ONW/WVB
51     C ONL=VLO/OT
52     C ONA=WAB/OT
53     C GO TO 130
54     C DT=ONL/ULO
55     C ONV=WVB/OT
56     C ONA=WAB/OT
57     C CONTINUE
58     C NL(3)=NL(2)-NL(1)
59     C PT(3)=PT(2)+((VOL1-NL(2)+VL(2))/(VOL1-NL(4)+VL(4)))*GAMM
60     C T(4)=(T(2)+.73.2)/(PT(2)/PT(4))**((GAMM-1.)/GAMM)-273.2
61     C ZL(4)=NL(4)+VL(4)/AT
62     C CALL ZLNCALC

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60      C
        C      UA=UAB-UVO-O.O
        C      CHECK FOR CONVERGENCE
        C
        IF (ABS(WLO-OLDWLO)-LE.(.001*OLDWLO)) GO TO 1000
        PI(3)=PI(4)
        PL(3)=PL(4)
        Y(3)=Y(4)
        VL(3)=VL(4)
        ZL(3)=ZL(4)
        ZLM(3)=ZLM(4)
        GO TO 105
70      C      200
        C      CONTINUE
        C
        C      SECTION 1-2 TANK LIQUID LEVEL IS AT LEAST PARTIALLY BELOW PUNCTURE
        C      AND PT.GT.PATN GO TO 300
75      C      IF(PT(4)-LE.PATN) GO TO 300
        C      OLDWLO=WLO
        C      CALL FLOWZ
        C      IF(INGFLC-EG-2) RETURN
        C      IF(WLO-ET-O.U) GO TO 225
80      C      DT=BHV/UVO
        C      DML=WLO*DT
        C      DMA=UAB*DT
        C      GO TO 230
85      C      DT=DML/WLO
        C      DMV=WLO*DT
        C      DMA=UAB*DT
        C      230
        C      CONTINUE
        C      MA(4)=MA(2)-MA*DT
        C      MV(4)=MV(2)-MV*DT
        C      GAMMA=GAM*(RA)
        C      CHECK FOR ISOTHERMAL OPTION
        C      IF(TIMEATR-EG-1) GAMMA=F.D
        C      VL(2)=1./CP*F(2,I(4))
        C      VL(4)=1./CP*F(2,I(4))
        C      ML(4)=ML(2)-ML*DT
        C      VM1=(VOLT-ML(4)+VL(4))/(MV(2)+MA(2))
        C      VM2=(VOLT-PL(4)+VL(4))/(PV(4)+PA(4))
        C      PI(4)=PI(2)+VM1/VM2**GAMMA
        C      ZL(4)=ML(4)+VL(4)/AT
        C      CALL ZLMCALC
90      C      CHECK FOR CONVERGENCE
        C
        IF (ABS(WLO-OLDWLO)-LE.(.001*OLDWLO)) GO TO 1000
        RA(3)=RA(4)
        RV(3)=RV(4)
        PL(3)=PL(4)
        PI(3)=PI(4)
        Y(3)=Y(4)
        VL(3)=VL(4)
        ZL(3)=ZL(4)
        ZLM(3)=ZLM(4)
        GO TO 205
100     C      300
        C      CONTINUE
        C
105
110
120

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PAGE 3

FTN 4.4433F 80/08/15. 13.45.07

SUBROUTINE NOMEHER 76/74 OPT=1

115 C SECTION 1.3 TANK LIQUID LEVEL IS AT LEAST PARTIALLY BELOW PUNCTURE

C AND PT = PATN

C CHECK FOR ISOTHERMAL OPTION

IF(PT(4)-LT.PATN) PT(4)=PATN

IF(TEMPER-EG.-1)GAMMA=1.0

305 OLQULO=ULO

CALL FLOW2

IF(INGFLG.EQ.2) RETURN

IF(ULO.GT.0.0) GO TO 307

307 DT=DMV/WVO

308 DNL=ULO\*DT

DNA=VWO\*DT

GO TO 308

307 DT=ORL/ULO

DMV=VWO\*DT

DNA=VWO\*DT

308 CONTINUE

NL(4)=NL(2)-ULO\*DT

PT(4)=PATN

2L(4)=NL(4)+VL(4)/AT

CALL ZLMCALC

VA=VWO-VVO=0.0

C CHECK FOR CONVERGENCE

C

140 IF(ABS(ULO-OLQULO)-LE.(.001\*OLQULO)) GO TO 1000

MA(3)=NA(4)-NA(2)

MV(3)=MV(4)-MV(2)

PT(3)=PT(4)

NL(3)=NL(4)

V(3)=V(4)

2L(3)=2L(4)

ZLM(3)=ZLM(4)

GO TO 305

310 CONTINUE

C

C \*\*\*\*\*RELIEF VALVE UNSTUCK\*\*\*\*\*

C

C SECTION 11.1 PUNCTURE BELOW TANK LIQUID LEVEL AND

PT=GT. PATN=DELPVAL

155 IF(2L(4)-GT.-ZP.AND.-PT(4)-GT.(PATN=DELPVAL)) GO TO 104

350 CONTINUE

C

C SECTION 11.2 PUNCTURE BELOW TANK LIQUID LEVEL OR

PT = PATN=DELPVAL

C

OLQULO=ULO

GAMMA=GAM(IRA)

IF(2L(4)-ZP&GT)-LE.0.0.AND.-VVO.LE.0.0) CALL TERRMIN(ON MAIN )

IF(2L(4)-LT.-ZP) GO TO 400

IF(PT(4)-LT.(PATN=DELPVAL)) PT(4)=PATN=DELPVAL

VWO=VVO=0.0

CALL FLOW2

IF(INGFLG.EQ.2) RETURN

IF(ULO.GT.0.0) GO TO 302

307 DT=DMV/WVO

DNL=ULO\*DT

170



SUBROUTINE MORENER 76/74 OPT=1

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175      DHA=DAO*DT
      GO TO 163
362      DT=DRL/WLO
      DRV=UVO*DT
      DHA=DAO*DT
363      CONTINUE
      PT(4)=PT(3)+PAIN-DELPVAL
      T(4)=T(3)+273.2/(PT(2)+PT(4))*((GAMMA-1.)/(GAMMA)-273.2
      NA(4)=(PT(4)-PV(4))*((VOLT-ML(4)+VL(4))/(RA*(T(4)+273.2))
      ML(4)=ML(2)+ML(4)*DT
      ZL(4)=ML(4)+VL(4)/AT
      CALL ZLNCALC
      C
      C CHECK FOR CONVERGENCE
      C
      IF(ABS(ULO-OLDULO)-LE.(1.001*OLDULO)) GO TO 1000
      PT(3)=PT(4)
      ML(3)=ML(4)
      T(3)=T(4)
      ZL(3)=ZL(4)
      ZLM(3)=ZLM(4)
      GO TO 350
400      CONTINUE
195      C
      C SECTION 11.3 TANK LIQUID LEVEL IS AT LEAST PARTIALLY BELOW PUNCTURE
      C AND PT=PAIN
      C
      IF(PT(4)-GT-PAIN) GO TO 200
200      C
      C SECTION 11.4 PUNCTURE BECOMES PARTIALLY BELOW LIQUID LEVEL AND
      C PT=PAIN-DELPVAL
      C
      PT(4)=PAIN
      PT(2)=PAIN-DELPVAL
      IF(ETIME-NE.1) PT(4)=PT(2)+PAIN
      T(4)=T(3)+273.2/(PT(2)+PT(4))*((GAMMA-1.)/(GAMMA)-273.2
      NA(4)=(PT(4)-PV(4))*((VOLT-ML(4)+VL(4))/(RA*(T(4)+273.2))
      CALL FLOW2
      IF(INGTLE-EO.2) RETURN
      IF(ULO-GT.0.0) GO TO 515
      DT=DRV/MVO
      DML=ULO*DT
      DHA=DAO*DT
      GO TO 520
515      DT=DRL/WLO
      DRV=UVO*DT
      DHA=DAO*DT
520      CONTINUE
      ML(4)=ML(2)+ML(4)*DT
      ZL(4)=ML(4)+VL(4)/AT
      CALL ZLNCALC
      C
      C CONVERGENCE ACHIEVED
      C
      1000      CONTINUE
      RETURN
700      WRITE(12,710) PT(3),PT(4),PAIN,ZL(3),ZL(4),P

```

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FTN 4.4433F

76/74 001-1

SUBROUTINE NOMENR

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210 FORMAT(1X,24M ERROR IN CHECK STATEMENTS,/,
1      ON PIC(3)= ,F15.5,ON PIC(4)= ,F15.5,7M PATH= ,F15.5,/,
2      ON ZL(3)= ,F15.5,ON ZL(4)= ,F15.5,5M ZP= ,F15.5)
      RETURN
      END

```

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PAGE 1

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FIN 4.6433F

SUBROUTINE NONENG2 74/74 OPT=1

```

1      SUBROUTINE NONENG2(KOUNT)
2      CALL SUBROUTINE: ZLCALC
3      REAL LAMDA,NA,RL,RV,J,ROLE
4      COMMON /GEN/ PT(4),T(4),ZL(4),ZLN(4),NA(4),RV(4),RL(4),PINF,
5      VM(4),J,COORD,AOL,AGS,AGT,AG,TAIW,TIN(4),MOL,
6      VOLT,IN-IP,BELPVAL,IWALVE,INEATIR,CPA,VL(4),VV(4),
7      WLO,WVO,WAO,PAT,AT,PV(4),LAMBDA(4),CPL(4),CPVARI(4),
8      TC,SVA,PINF,21,CHL,PA(4)
9      COMMON/ING/ DP,INL,INV,INA,DML,DHV,DNA,FLOSUB,ENRSUP,AP,INGFLG,
10     1      AVERAGE(1)=(T(2)+T(4))/2.0
11     1      ICT,XLP,BELTNE,ZUT(4),VOLM,NTST,NCCHANGE
12     1      NA=RG/28.96
13     1      RV=RG/ROLE
14     1      PT(3)=PT(4)
15     1      VLE=1.0/CPF(2,1C)
16     1      NA=WAO-WVO-WLO=0.0
17     1      PUNCTURE BELOW TANK LIQUID LEVEL
18     1      GAMMA=GAM(NA)
19     1      IF(INEATIR.EQ.-1) GAMMA=1.0
20     1      RL(4)=RL(2)
21     1      RV(4)=RV(2)
22     1      NA(4)=NA(2)+(VOLB/SVA)*KOUNT
23     1      PV(4)=CPF(1,T(4))
24     1      VM1=(VOL1-ML(2)+VL1)/(NA(2)+RV(2))
25     1      VM2=(VOL2-ML(2)+VL2)/(NA(2)+RV(2)+(VOLB/SVA))
26     1      PT(4)=PT(2)+(VM1/VM2)*GAMMA
27     1      T(4)=(T(2)+273.2)/(PT(2)/PT(4))*((GAMMA-1.)/GAMMA)-273.2
28     1      ZL(4)=RL(4)+VLE/AT
29     1      CALL ZLCALC
30     1      CHECK FOR CONVERGENCE
31     1      IF(AUS((PT(4)-PT(3))/PT(4)).LE..001) GO TO 1000
32     1      PT(3)=PT(4)
33     1      NA(3)=NA(4)
34     1      RV(3)=RV(4)
35     1      RL(3)=RL(4)
36     1      T(3)=T(4)
37     1      VL(3)=VL(4)
38     1      ZL(3)=ZL(4)
39     1      ZLN(3)=ZLN(4)
40     1      GO TO 105
41     1      CONVERGENCE ACHIEVED
42     1      1000 CONTINUE
43     1      RETURN
44     1      END

```

PAGE 1

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PTM 4.6433F

SUBROUTINE NOOUT 76/74 OPT=1

```

1      SUBROUTINE NOOUT
2      CALLED SUBROUTINE: TERMIN
3      CALLED FUNCTIONS: GAMM
4      COMMON /GEN/ PT(4),T(4),ZL(4),ZLM(4),ML(4),RV(4),RL(4),PINF,
5      VU,G,J,CORF,AOL,AOS,RSI,RG,TINF,TIM(4),CMVT,
6      VT,ZU,IP,DPV,SVB,MTB,CPA,SVL(4),SVU(4),WLO,
7      WVD,WAO,PATN,AT,PV(4),LAMBDA(4),CPL(4),CPWBAR(4),
8      TC,SWA,PINFC,IT,CHI,PA(4)
9      COMMON/IMG/ OP,TNL,TMV,TMA,BML,BMV,DMA,FLOSUD,EMRSUD,AP,INGELG,
10     ICT,XLP,DELTIME,ZUT(4),VOLB,MTEST,NCHANGE
11     COMMON/MAT/ ZLEB1,PTEQ1
12     INTEGER MTB,SVB,FLOSUD,EMRSUD
13     REAL ML,MV,MA,J,LAMBDA
14     AVERAGE(X)=(N(2)*X(4))/2.0
15     WRITE(12,15)
16     FORMAT(13M GOT TO NOOUT)
17     RA=RG/28.96
18     PI=ACOS(-1.0)
19     WV=0.0
20     ATOT=AP
21     PTINT=PT(2)
22     ZLINT=ZL(2)
23     IF(SVL(4)-GJ*WV) GO TO 500
24     CASE 1. WATER LIGHTER THAN CARGO
25     CHECK IF RELIEF VALVE IS STUCK
26     ZUT(1)=ZUT(2)-ZUT(3)-ZUT(4)-ZL(4)
27     IF(SVB.EQ.-1) GO TO 200
28     ZWIEQ=ZU-(WV/SVL(4))-1.)*(ZLINT-ZLM(4))+1000.0*(PINF-PTINT)*WV/G
29     IF(ZWIEQ.GT.-ZT) GO TO 100
30     ZWII=1.1*(ZWIEQ-ZLM(4))*ZLM(4)
31     SQ2=C.2*G*(ZWIEQ-ZLM(4))*0.5
32     SQT=(C.2*G*(WV/SVL(4))-1.)*(AVERAGE(ZL)-AVERAGE(ZLM))*0.5
33     ATN=ATOT+WV*SQ2/(SQ1+SVL(4)+SQ2+WV)
34     AOUT=ATOT-ATN
35     WLO=CBORF*AOUT+SQ2/SVL(4)
36     OLBNW=WV
37     WV=VLO
38     IF(ZJM(4)-WE.0.0) GO TO 52
39     DT=BML/WLO
40     TIM(4)=TIM(2)+DT
41     ML(4)=ML(2)-WLO*DT
42     ZL(4)=ZLINT-WLO*SVL(4)+DT*AT
43     ZLM(4)=ZP-DPA-.5
44     IF(ZL(4)-LT-ZP) ZLM(4)=.5*(ZL(4)+ZP-DP)
45     IF(ZL(4)-LT-ZP) ATOT=(ZL(4)-ZP+DP)*XLP
46     IF(AOS(WV-OLBNW)-GT.-C.001*OLBNW) GO TO 50
47     SCROLL PARAMETERS
48     ZL(1)=ZL(2)
49     ML(1)=ML(2)
50     ZLM(1)=ZLM(2)
51     TIM(1)=TIM(2)
52     ZL(2)=ZL(3)=ZL(4)
53     ML(2)=ML(3)=ML(4)
54     ZLM(2)=ZLM(3)=ZLM(4)
55     TIM(2)=TIM(3)=TIM(4)
56     TNL=TNL+BML
57     TIM(4)=TIM(2)+DT

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      1  WRITE(12,7200) TIM(2),ML(2),MV(2),NA(2),IML,IMV,IMA,IE(2),PIE(2),
      2  ZL(2)
      3  IF(ZL(4)-GT.(EP-OP)) GO TO 50
      4  CALL TERMIN(100 MOOUT)
      5  GO TO 2500
      6  CONTINUE
      7  100 CONTINUE
      8  ZWIEQ-CT-27
      9  ZL(4)-EP-OP+.5
      10 AIM-AP+.5
      11 IF(OP-0.-AND-ZL(4)-LT-.2P) GO TO 108
      12 IF(ZL(4)-LT-.2P) ZL(4)=.5*(ZL(4)+EP-OP+.5)
      13 IF(ZL(4)-LT-.2P) AIM-(EP-ZL(4)).5*(ZL(4)-EP+OP)+XLP
      14 303-12*(PI-PIINT)+VW-10000*.6*(ZL-27)-6*(1./SVL(4)-
      15 1./VW)*AVERAGE(ZL)-ZL(4))*.5
      16 OLWU-VW
      17 WV-CB00F*AIM*SVL/VW
      18 WL=VW*WV/SVL(4)
      19 IF(TIM(4)-WV-0.0) GO TO 104
      20 DT-DWL/DWL0
      21 TIM(4)-TIM(2)+DT
      22 ML(4)-ML(2)+WL*DT
      23 ZL(4)-ZL(2)-VW*WV*DT/AT
      24 IF(AVS(WV-0.0000)-GT.(.001*OLWU)) GO TO 100
      25 TIM(3)-TIM(2)
      26 ZL(3)-ZL(2)
      27 ML(3)-ML(2)
      28 ZL(4)-ZL(2)
      29 TIM(2)-TIM(3)+TIM(4)
      30 ZL(2)=ZL(3)-ZL(4)
      31 ML(2)=ML(3)-ML(4)
      32 ZL(2)-ZL(3)+ZL(4)
      33 TIM(2)-TIM(3)+DT
      34 IML-IML+DWL
      35 WRITE(12,7200) TIM(2),ML(2),MV(2),NA(2),IML,IMV,IMA,IE(2),PIE(2),
      36 ZL(2)
      37 IF(ZL(4)-GT.-EP) GO TO 100
      38 CALL TERMIN(100 MOOUT)
      39 GO TO 2500
      40 SECTION 1-0
      41 RELIEF VALVE IS STUCK
      42 CALCULATE PI BY FILTRATION ROUTINE
      43 GARNAM-GARN(AA)
      44 ZWIEQ2-(ZLINT+ZT)*.5
      45 ZWIEQ2-ZWIEQ
      46 PI(4)-PIINT*((VW-AT*(ZLINT+ZT))/(VW-AT*(ZWIEQ2)+GARNAM
      47 ZWIEQ2-ZW-(VW/SVL(4)-1)*(ZLINT-ZL(4))+10000*(VW*(PI-PIINT-PIE(4))/G
      48 IF(AVS(ZWIEQ2-ZWIEQ1)-GT.(.001*(ZWIEQ1)) GO TO 202
      49 ZWIEQ=ZWIEQ2
      50 GO TO 25
      51 CONTINUE
      52 CARGO LIGHTEN THAN WATER WL-GT-WV
      53 CHECK IF RELIEF VALVE IS STUCK
      54 ZWIE(1)-ZWIE(2)-ZWIE(3) ZWIE(4)-0.0
      55 IF(SVB-EQ.-1) GO TO 700
      56 DEL(ZL-SVL(4)/VW*(ZL-EP)-10000*.5*SVL(4))/6*(PI-PIINT-PIE(4))

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SUBROUTINE HOOUT 76/74 OPT=1

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115      DEL1=ZLINT-BELZLT
      DEL2=ZLINT-(ZT-ZP)
      DEL3=DELZLT-(ZT-ZP)
      IF(DEL1-LE-0.0-AND-DEL2-LE-0.0)GO TO 520
      IF(DEL1-GE-0.0-AND-DEL3-LE-0.0)GO TO 530
      IF(DEL1-LE-0.0-AND-DEL2-GE-0.0)GO TO 540
      IF(DEL1-GE-0.0-AND-DEL3-GE-0.0)GO TO 540

C
C      ERROR IN CHECK STATEMENTS
C      WRITE(12,503)DEL1,DEL2,DEL3
503      FORMAT(1H,7DEL1 = ,F12.5,8H DEL2 = ,F12.5,8H DEL3 = ,F12.5)
      GO TO 2500

C
C      SECTION 11.A I
C      CONTINUE
C      CARGO LISTED ENTIRELY ABOVE PUNCTURE, NO CARGO ESCAPES
      WRITE(12,521)
521      FORMAT(1H,50H WATER ENTERS TANK AND LIFTS CARGO ABOVE PUNCTURE.,
1 /,21H CARGO DOES NOT VENT.,)
      CALL TERMIN(10HOUT )
      GO TO 2500

C
C      SECTION 11.A II
C      CONTINUE
530      WEIGHT OF CARGO IS SUFFICIENT TO BALANCE WATER PRESSURE
      ZWT(2)=ZWT(3)-ZWT(4)-ZWTINT
      ZL(4)=ZLINT+ZWTINT
      ZLH(4)=ZP-DP*.5
532      IF(ZWT(4)-GT-(ZP-DP))GO TO 535
C
C      HEIGHT OF WATER IN TANK IS LESS THAN BOTTOM OF PUNCTURE
      OLDWU=WU
      WU=CDORF*AP/(2+.WU)*(G*(1-.WU/SVL(4))+(ZP-DP/2.-ZWT(4)))**-.5
      WLO=WU+WU/SVL(4)
      IF(TIM(4)-NE-0.0) GO TO 533
      DT=DML/WLO
      TIM(4)=TIM(2)+DT
533      ML(4)=ML(2)-WLO*DT
      ZWT(4)=ZWT(2)+WU+DT+WU/AT
      WRITE(12,533) T(2),T(4),DELTIME,ML(2),ML(4),WU(2),WU(4),PT(2),
1 PT(4),PV(2),PV(4),ZL(2),ZL(4),ZLH(2),ZLH(4),
2 ZWT(2),ZWT(4),WLO,WU
      IF(AUS(WU-OLDWU)-GT-((.001*OLDWU))GO TO 532
C
C      SCROLL PARAMETERS
      ZWT(1)=ZWT(2)
      ML(1)=ML(2)
      TIM(1)=TIM(2)
      ZWT(2)=ZWT(3)-ZWT(4)
      ML(2)=ML(3)-ML(4)
      TIM(2)=TIM(3)-TIM(4)
      TML=TML+DML
      TIM(4)=TIM(2)+DT
      WRITE(12,7200) TIM(2),ML(2),WU(2),ML(4),TML,IMV,IMR,T(2),PT(2),
1 ZL(2)
      IF(ZWT(4)-LT-.ZP)GO TO 552
      CALL TERMIN(10HOUT )

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      GO TO 2500
535 CONTINUE
      OLDMW=WV
      WV=COORDFAP(Z,ZVWV)* (2.-G*(1.-WV/SVL(4))*(ZP-AVERAGE(ZMT)))**0.5
      IF(TIM(4)-NE.O.O) GO TO 536
      DT=DM/LMO
      TIM(4)=TIM(2)+DT
      ML(4)=ML(2)-WLO*DT
      ZMT(4)=ZMT(2)+WV*AT*WV/DT
      WRITE(12,51) T(2),T(4),DELTIME,ML(2),ML(4),MV(2),MV(4),PT(2),
1      PT(4),PV(2),PV(4),ZL(2),ZL(4),ZLM(2),ZLM(4),
2      ZMT(2),ZMT(4),MLO,WV
51 FORMAT(1H ,BN T(2)= ,E12.5,BN T(4)= ,E12.5,9H DELTIME= ,E12.5,,/
1      ,BN ML(2)= ,E12.5,BN ML(4)= ,E12.5,,/
2      ,BN MV(2)= ,E12.5,BN MV(4)= ,E12.5,,/
3      ,BN PV(2)= ,E12.5,BN PV(4)= ,E12.5,,/
4      ,BN ZL(2)= ,E12.5,BN ZL(4)= ,E12.5,,/
5      ,BN ZLM(2)= ,E12.5,9H ZLM(4)= ,E12.5,,/
6      ,9H WMT(2)= ,E12.5,9H WMT(4)= ,E12.5,6H MLO= ,E12.5,
7      ,6H WV= ,E12.5)
8      IF(AUSCW-OLDMW).GT.(-.001*OLDMW) GO TO 535
      ZMT(3)=ZMT(2)
      ML(3)=ML(2)
      TIM(3)=TIM(2)
      ZMT(2)=ZMT(3)-ZMT(4)
      ML(2)=ML(3)-ML(4)
      TIM(2)=TIM(3)-TIM(4)
      TIM(4)=TIM(2)+DT
      TML=TML+DM/LMO
      WRITE(12,7200) TIM(2),TIM(3),ML(2),ML(4),TML,TMV,TMA,T(2),PT(2),
1      ZL(2)
7200 FORMAT(1H ,10E12.5)
      IF(ZMT(4)-LT.ZP) GO TO 535
      CALL TERMINCOM MODUT
      GO TO 2500
C
C SECTION II-A III
C
540 CONTINUE
      ZMTINT=ZT-ZLINT
542 IF(ZMT(4)-GT.ZP-OP) GO TO 545
C
C HEIGHT OF WATER IN TANK IS LESS THAN BOTTOM OF PUNCTURE
C
      OLDMW=WV
      AT=((10000.-*(PI*H-PI*INT)*WV*G*(ZU-ZP*DP/2.))-G*(WV/SVL(4))*
1      (ZT-ZP*DP/2.))**.5
      WV=COORDFAP*AT/(Z.-WV)
      IF(TIM(4)-NE.O.O) GO TO 543
      DT=DM/LMO
      TIM(4)=TIM(2)+DT
      ML(4)=ML(2)-WLO*DT
      ZMT(4)=ZMT(2)+WV*WV*AT/AT
      IF(AUS(WV-OLDMW)-GT.(-.001*OLDMW)) GO TO 542
      ZMT(3)=ZMT(2)

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**REF ID: A64384**

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**SUBMITTING NOON**

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2350 ML(1)=ML(2)
      TIM(1)=TIM(2)
      ZMT(2)=ZMT(3)=ZMT(4)
      ML(2)=ML(3)=ML(4)
      TIM(2)=TIM(3)=TIM(4)
      TIM(4)=TIM(2)+DT
      TML=TML+DML
      WRITE(12,7200) TIM(2),ML(2),MV(2),MA(2),TML,TMV,TMA,T(2),PT(2),
1      ZL(2)
      IF(ZMT(4)-LT-ZP)GO TO 542
545 CONTINUE
      QLOW=WM
      A2=(2.-((10000.-A(PINF-PTINT)*VM*G*(ZM-ZT+VM/SVL(4)))-
1      G*(1.-VM/SVL(4)))*(ZP-AVERAGE(ZMT))+.27*(2.-ODP)-
2      AVERAGE(ZMT)))**.5
      WM=COORF*AP*42/(2.+VM)
      MLO=VM+WM/SVL(4)
      IF(TIM(4)-ME-O-D) GO TO 575
      DT=DML/MLO
      TIM(4)=TIM(2)+DT
575 ML(4)=ML(2)-MLO*DT
      ZMT(4)=ZMT(2)+VM+WM*DT/AT
      IF(ABS(LOW-QLDUM)-GT.(.001+QLDUM))GO TO 545
      ZMT(1)=ZMT(2)
      ML(1)=ML(2)
      TIM(1)=TIM(2)
      ZMT(2)=ZMT(3)=ZMT(4)
      ML(2)=ML(3)=ML(4)
      TIM(2)=TIM(3)=TIM(4)
      TIM(4)=TIM(2)+DT
      TML=TML+DML
      WRITE(12,7200) TIM(2),ML(2),MV(2),MA(2),TML,TMV,TMA,T(2),PT(2),
1      ZL(2)
      IF(ZMT(4)-LT-ZP)GO TO 545.
      CALL TERMINATION(MOOUT )
      GO TO 2500
C 700 CONTINUE
C
C SECTION II.B
C CARGO LIGHTER THAN WATER VL-GT-VM
C RELIEF VALVE IS STUCK
C CALCULATE PT BY ITERATION ROUTINE
      IF(ZL(2)-LE-ZP-DP)GO TO 800
      GAMMA=GAMM(WA)
      DELZL2=SVL(4)/VM*(ZM-ZP)-10000.-4SVL(4)*(PTINT-PATM)/G
      DELZL1=DELZL2
      PT(4)=PTINT*(VM+AT+ALINT)/(VT+AT+(DELZL1+ZP)))**.5+GAMMA
      DELZL2=SVL(4)/VM*(ZM-ZP)-10000.-4SVL(4)*(PT(4)-PATM)/G
      IF(ABS(DELZL2-DELZL1)-GT.(.001+DELZL1))GO TO 701
      DELZL1=DELZL2
      GO TO 501
800 PT(4)=PTINT
      ZMTINT=ZP-ZLINT
      AT=AT+ZMTINT/MVO
      MV(4)=MVO+DT
      DMV=DT+MVO
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2360
2365
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SUBROUTINE MOOUT 76/74 OPT=1

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      M1=10000.4*SVL(2)*PTINT*PATH)
      M2=9999.5-SVL(2)*(20-IP0999.5)/VW
      WLO=AP/SVL(2)*(B1+B2)*.5
      PT=(2P-ZUIMI)*AT/WLO
      ML(4)=WLO*PT
      DML-DT*WLO
      WRITE(12,900) ZUIMI,PTINT,WLO,ZLINT,ML(4),MV(4)
      900  FORMAT(1M,6I12.4)
      RETURN
      295  2500 END

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SUBROUTINE SCROLL 76774 OPT=1 PAGE 1  
 1 SUBROUTINE SCROLL  
 2 C  
 3 C SCROLLS PARAMETERS TO RESTART PROGRAM IN A  
 4 C DIFFERENT KTOP  
 5  
 10 COMMON /GEN/ PT(4),T(4),ZL(4),ZLN(4),MAC(4),MV(4),ML(4),PINF,  
 1 VM,G,J,COORDF,AOL,AGI,AGI,AGI,TIME(4),CRUT,  
 2 VT,ZU,ZP,OPV,SVB,MTD,CPA,SVL(4),SVV(4),ULO,UVO,  
 3 VAO,PATM,AT,PV(4),LAMBDA(4),CPL(4),CPVBAR(4),TC,SVB,  
 4 PINEG,IT,CNE,PAC(4)  
 10 COMMON /ING/ BP,INL,IMU,TRA,DNL,DNV,DNA,FLOSUB,ENRSUB,AP,INSEFLS,  
 1 ICT,XLP,DELTIME,ZUT(4),VOLU,MIESI,MCNAGE  
 1 REAL ML,MV,MA,J,LAMBDA  
 10 INTEGER MTD,POTISC,PE,SVB,FLOSUB,ENRSUB  
 10 WRITE(12,20)  
 15 FORMAT(12,20)  
 20 DO 100 I=2,4  
 20 ML(I-1)=ML(I)  
 20 MV(I-1)=MV(I)  
 20 MAC(I-1)=MAC(I)  
 20 PT(I-1)=PT(I)  
 20 T(I-1)=T(I)  
 20 ZL(I-1)=ZL(I)  
 20 ZLN(I-1)=ZLN(I)  
 20 PAC(I-1)=PAC(I)  
 20 TIME(I-1)=TIME(I)  
 20 PV(I-1)=PV(I)  
 20 SVL(I-1)=SVL(I)  
 20 SVV(I-1)=SVV(I)  
 20 CPL(I-1)=CPL(I)  
 20 CPVBAR(I-1)=CPVBAR(I)  
 100 CONTINUE  
 35 ML(2)=ML(3)=ML(4)  
 35 MV(2)=MV(3)=MV(4)  
 35 MAC(2)=MAC(3)=MAC(4)  
 35 PT(2)=PT(3)=PT(4)  
 35 T(2)=T(3)=T(4)  
 35 ZL(2)=ZL(3)=ZL(4)  
 35 ZLN(2)=ZLN(3)=ZLN(4)  
 35 PAC(2)=PAC(3)=PAC(4)  
 35 TIME(2)=TIME(3)=TIME(4)  
 35 PV(2)=PV(3)=PV(4)  
 35 SVL(2)=SVL(3)=SVL(4)  
 35 SVV(2)=SVV(3)=SVV(4)  
 35 CPL(2)=CPL(3)=CPL(4)  
 35 CPVBAR(2)=CPVBAR(3)=CPVBAR(4)  
 35 TRL=TRL+DNL  
 35 TNU=TNU+DNV  
 35 TRA=TRA+DNA  
 200 FORMAT(4E15.5)  
 50 WRITE(12,200)ML(1),ML(2),ML(3),ML(4)  
 50 WRITE(12,200)MV(1),MV(2),MV(3),MV(4)  
 50 WRITE(12,200)MAC(1),MAC(2),MAC(3),MAC(4)  
 50 WRITE(12,200)PT(1),PT(2),PT(3),PT(4)  
 50 WRITE(12,200)T(1),T(2),T(3),T(4)  
 50 WRITE(12,200)ZL(1),ZL(2),ZL(3),ZL(4)  
 50 WRITE(12,200)ZLN(1),ZLN(2),ZLN(3),ZLN(4)

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CTM 4.4433F

76/74 OPT-1

SUBROUTINE SCROLL

```

WRITE(12,200)PA(1),PA(2),PA(3),PA(4)
WRITE(12,200)TIME(1),TIME(2),TIME(3),TIME(4)
WRITE(12,200)PV(1),PV(2),PV(3),PV(4)
WRITE(12,200)SVL(1),SVL(2),SVL(3),SVL(4)
WRITE(12,200)SVW(1),SVW(2),SVW(3),SVW(4)
WRITE(12,200)CPL(1),CPL(2),CPL(3),CPL(4)
WRITE(12,200)CPVBAR(1),CPVBAR(2),CPVBAR(3),CPVBAR(4)
WRITE(12,200)TML,TMV,TMA
END

```

60

45

SUBROUTINE TEBRIN 76774 OPT=1 PAGE 1  
 FIN 4.6433F 80/08/13. 13.45.07  
 SUBROUTINE TEBRIN(CODE)  
 COMMON /GEN/ PTC(4),TIC(4),ZLC(4),ZLM(4),MAC(4),MV(4),ML(4),PIBF,  
 VU(6),CDNF,ADL,ADG,AGI,AG,TIME(4),CMT,  
 VI,ZP,OPV,SVD,MTD,CPA,SVL(4),SVU(4),MO,UVB,  
 WAB,PATN,AT,PV(4),LAMBDA(4),CPL(4),CPUBAR(4),TC,SVA,  
 PING,PI,CRI  
 COMMON/ING/ OP,TML,TMV,TMA,PML,PMV,PMA,FLOSUB,ENRSMU,MP,INGELG,  
 ICT,XLP,DECLINE,INT(4),VOLD,NTEST,NCHANGE  
 INTEGER MID,SVD,FLOSUB,ENRSMU  
 REAL ML,MV,MA,J,LAMBDA  
 MU=12  
 WRITE(MU,5000)ENCODE  
 WRITE(MU,5100) (TIME(I),I=1,4)  
 WRITE(MU,5110) (ML(I),I=1,4)  
 WRITE(MU,5120) (MV(I),I=1,4)  
 WRITE(MU,5130) (MA(I),I=1,4)  
 WRITE(MU,5140) (PT(I),I=1,4)  
 WRITE(MU,5150) (PV(I),I=1,4)  
 WRITE(MU,5160) (ZL(I),I=1,4)  
 WRITE(MU,5170) (ZC(I),I=1,4)  
 WRITE(MU,5200) TIME(2),TML,TMV,TMA  
 5000 FORMAT(1H1,4E12.2)NIVENT TERMINATED FROM ,A20,/,I,4E12,  
 1 34VALUES OF PARAMETERS AT RUN COMPLETION // 3  
 5100 FORMAT(1H1,4E15.5)  
 5110 FORMAT(2H1,4E15.5)MASS REMAINING ,4E15.5)  
 5120 FORMAT(2H1,4E15.5)MASS REMAINING ,4E15.5)  
 5130 FORMAT(2H1,4E15.5)MASS IN TANK ,4E15.5)  
 5140 FORMAT(2H1,4E15.5)PRESSURE ,4E15.5)  
 5150 FORMAT(2H1,4E15.5)PRESSURE ,4E15.5)  
 5160 FORMAT(2H1,4E15.5)LEVEL ,4E15.5)  
 5170 FORMAT(2H1,4E15.5)TEMPERATURE ,4E15.5)  
 5200 FORMAT(1H1,TOTAL VENT TIME = ,E15.5,/,23H TOTAL MASS LEFT(L,V,A),  
 1 3E15.5)  
 1 STOP  
 2 END

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[illegible]

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REF ID: A66318

ALVIN KILLBOURN 76/76 021-1

```

60      NL(2)=NL(3)=NL(4)
        ZL(2)=ZL(3)=ZL(4)
        ZLM(2)=ZLM(3)=ZLM(4)
        TIM(2)=TIM(3)=TIM(4)
        ZMT(2)=ZMT(3)=ZMT(4)
        TML=TML+TML
        WRITE(12,7200) TIM(2),ML(2),MV(2),MA(2),TML,TMV,TMA,T(2),PI(2),
     1      ZL(2)
        IF(ZL(4)-GT-7P-9P) GO TO 25
        CALL TERMINATION(VATIM)
        RETURN

C
70      CONTINUE
        C WATER LIGHTER THAN CARGO
        C RELIEF VALVE STUCK
        C

75      PTINT=PTERR1*(PT(2)-PTERR1)/KT
        ZLTM1=ZLERR1*(ZL(2)-ZLERR1)/KT
        GAMMA=1.0
        IF(CTD-EO-1) GAMMA=GAM*(RA)
        PT(4)=PT(3)=PT(2)
        ZMT(1)=ZMT(2)=ZMT(3)=ZMT(4)=ZL(2)
        ZL(4)=ZL(2)
        ATOT=AP
        PT(3)=PT(4)
        SOL(4)=1./CPI(2,1(4))
        SOL(2)=SOL(VU)/SOL(4)-1.)*(AVERAGE(ZL)-AVERAGE(ZLM))**.5
        SRT=((10000.-SOL(4))*(PTINT-PTERR1)*G*(ZLTM1-ZLERR1)).2)**.5
        G=C*(VT-AT+AVERAGE(ZMT))/AT +10000.-GAMMA*SOL(4)*AVERAGE(PT)
        G=C*(VT-AT+AVERAGE(ZMT))/AT +10000.-GAMMA*VU*AVERAGE(PT)
        ATR=(ATOT+SRT*G*VU)/(SOL(4)*GL*SOL(VU)*202*GM)
        AOUT=ATOT-AIN
        ML0=CDBRF*AOUT+S02/SVL(4)
        MU=CDBRF*AIN*S01/VU
        DT=TIM(4)-TIM(2)
        DML=ML0*DT
        ML(4)=ML(2)-ML0*DT
        ZL(4)=ZL(2)-ML0*(TIM(4)-TIM(2))*SVL(4)/AT
        CALL ZLNCALC
        ZMT(4)=ZMT(2)+MU*(TIM(4)-TIM(2))*VM/AT*ZL(4)-ZL(2)
        GAMMA=1.0
        IF(CTD-EO-1) GAMMA=GAM*(RA)
        PT(4)=PTINT*(VT-AIN+ZLTM1)/(VT-AT+ZMT(4))*GAMMA
        T(4)=T(2)+25.25*(PT(2)/PT(4))-G*(GAMMA-1.)/GAMMA*-273.2
        DO 710 I=2,4
        710 T(I)=T(1)
        ML(1)=ML(1)
        MV(1)=MV(1)
        MA(1)=MA(1)
        PT(1)=PT(1)
        T(1)=T(1)
        ZL(1)=ZL(1)
        ZLM(1)=ZLM(1)
        PACE(1)=PA(1)
        TIM(1)=TIM(1)
        PV(1)=PV(1)
        SVL(1)=SVL(1)

```



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FIN 4.0433F

SUBROUTINE WATIN 76/74 OPT=1

```

175      TIM(2)=TIM(3)-TIM(4)
          ZMT(2)=ZMT(3)-ZMT(4)
          FORMAT(1M,10E12.5)
          TIM(4)=TIM(2)00Y
          WRITE(12,7200) TIM(2),ML(2),MV(2),MA(2),VM,FMV,FM,1(2),PT(2),
1              ZL(2)
          IF(ZMT(4).LT.1P) GO TO 510
          CALL TERMINATION WATIN
          RETURN
          END
180

```



PAGE 1

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FTM 6.46433F

SUBROUTINE ZLNCALC 76/74 OPT=1

```

1 SUBROUTINE ZLNCALC
COMMON /GEN/ P(4),T(4),ZL(4),ZLH(4),ML(4),MV(4),ML(4),PINF,
1  VM,G,J,COORF,AOL,AOG,AGI,AG,IAIA,TIM(4),MOLE,
3  VOLT,ZM,ZP,BELPVAL,IUALVE,INEATIR,CFA,VL(4),VV(4),
3  WLO,WVO,MAQ,PATM,AT,PV(4),LAMBDA(4),CPL(4),CPUBAR(4),
4  TC,SVA,PINFC,AT,CPI,PA(4)
COMMON/ING/ OP,IRI,INV,INA,DHL,DHV,DMA,FLOSUM,ENRSUD,AP,INGFLS,
1  ICT,ILP,BELINE,ZUT(4),VOLD,NTEST,MCHANGE
DEAL J,LAMBDA,MA,ML,MOLE,MV
PUNCTURE NOT SUBMERGED BY WATER
IF (ZM-GT-ZP-DP) GO TO 310
CARGO BELOW PUNCTURE
IF (ZL(4)-GT-ZP-DP) GO TO 310
KIVP=8
GO TO 400
CARGO ABOVE PUNCTURE
310 CONTINUE
IF (ZL(4)-LT-ZP) GO TO 320
ZLM(4)=ZP-Q.5*DP
KIVP=1
GO TO 400
CARGO INTERSECTS PUNCTURE
320 CONTINUE
ZLM(4)=Q.5*(ZL(4)+ZP-DP)
KIVP=4
GO TO 400
C
C ***
C PUNCTURE SUBMERGED BY WATER
C 330 CONTINUE
IF (ZM-LT-ZP) GO TO 340
CARGO BELOW PUNCTURE
IF (ZL(4)-GT-ZP-DP) GO TO 340
KIVP=9
GO TO 400
CARGO ABOVE PUNCTURE
340 CONTINUE
IF (ZL(4)-LT-ZP) GO TO 350
ZLM(4)=ZP-DP+Q.5
KIVP=2
GO TO 400
CARGO INTERSECTS PUNCTURE
350 ZLM(4)=(ZL(4)+ZP-DP)*Q.5
KIVP=5
GO TO 400
C
C ***
C PUNCTURE INTERSECTS WATER LEVEL
C CARGO BELOW PUNCTURE
360 CONTINUE
IF (ZL(4)-GT-ZP-DP) GO TO 370
KIVP=10
GO TO 400
CARGO ABOVE PUNCTURE
370 CONTINUE
IF (ZL(4)-LT-ZP) GO TO 380
ZLM(4)=ZP-.5*Q.5
KIVP=3

```

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FIN 6.4433F

SUBROUTINE ZLNCALC 76/76 OPT=1

```

60      GO TO 400
        C      CARGO INTERSECTS PUNCTURE, BUT IS ABOVE WATER
        580 CONTINUE
        IF (ZL(S)-L1-ZW) GO TO 390
        ZLN(S)=0.5*(ZL(S)+ZP-OP)
        K1YP=6
        GO TO 400
65      C      CARGO INTERSECTS PUNCTURE, BUT IS BELOW WATER
        390 ZLN(S)=(ZL(S)+ZP-OP)*.5
        K1YP=7
        C      ***
        400 CONTINUE
        RETURN
        END
70

```

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## APPENDIX D. SENSITIVITY ANALYSIS OF MODEL

The sensitivity of the model predictions to slight changes in the initial conditions was determined by the method described in [2]. That is, one of the inputs supplied by the user, say, the cargo temperature, was changed by some percentage and the change in the output, say total mass released or total discharge time, was computed. A sensitivity coefficient was then defined as

$$S = (\Delta\phi/\phi_n)/(\Delta x/x_n) \quad (D.1)$$

where  $\phi$  is the output and  $x$  is the input. The subscript  $n$  refers to the nominal value of  $x$  or  $\phi$ . A value of  $S = 1$  means that the output changes by the same percentage as the input,  $S = 2$  means that the output percentage change is twice the input change, and so on. Strictly speaking, equation (D.1) defines the "first order sensitivity coefficients" because changes in the output as a result of the interaction of changes in two or more inputs are not considered.

In this analysis, only the sensitivity coefficients for the input selected by the user will be determined. The sensitivity to the various internal parameters of the model, say, the Chemical Properties File correlations, will not be determined since these are presumed to be "accurate." Changes in the numerical scheme will not be investigated either because a predictor-corrector type of integration is employed in which the discharge mass increment is changed automatically within the program to ensure that the changes in tank pressure and liquid level during each time step remain relatively constant; this kind of numerical integration avoids the sensitivity problems of numerical integration that are discussed in [2]. The sensitivity of the results to changes in the various empirical constants of the model was discussed in the main text; since

the "best" values have already been incorporated in the model, the sensitivity of the output to these parameters is not relevant.

The sensitivity coefficients were computed for a tank containing LNG, which is a volatile cargo displaying most of the thermodynamic interactions of interest; adiabatic conditions were assumed. The nominal input conditions were:

$$V_T, \text{ Tank volume} = 2 \times 10^8 \text{ cm}^3$$

$$Z_T, \text{ Tank height} = 700 \text{ cm}$$

$$A_p, \text{ Circular puncture area} = 2688 \text{ cm}^2$$

$$Z_p, \text{ Puncture centerline elevation} = 175 \text{ cm}$$

$$C_D, \text{ Discharge coefficient} = 0.68$$

Relief valve = stuck

$$M_i, \text{ Initial cargo mass} = 8 \times 10^7 \text{ grams}$$

$$T_i, \text{ Initial temperature} = -159.2^\circ\text{C} = 114^\circ\text{K}$$

$$P_{\text{atm}}, \text{ Atmospheric pressure} = 101.3565 \text{ KPa}$$

$$T_{\text{atm}}, \text{ Air temperature} = 21^\circ\text{C} = 294.2^\circ\text{K}$$

All the input parameters were varied by  $\pm 5\%$  except for the temperature, which was varied by  $\pm 1\%$  of its absolute value; the large change of LNG vapor pressure with temperature at pressures near atmospheric required a smaller percentage change for  $T_i$  than for the other inputs. The relief valve was always kept "stuck" to emphasize two-phase flow effects; it ought to be realized, however, that when the relief valve is functioning it controls the vapor space pressure, and the sensitivity of the output to the initial temperature (i.e., the saturation pressure) is much diminished. For all the calculations, the puncture was open to the atmosphere (nonsubmerged puncture). Note that the nominal value of temperature was taken as  $114^\circ\text{K}$ , rather than as  $-159.2^\circ\text{C}$ , so that the sensitivity coefficients will be

positive if the output increases when the temperature increases, and vice versa.

Only the sensitivity coefficients for the liquid outflow duration of the discharge are presented here. After the tank liquid level falls below the puncture, LNG vapor is slowly discharged until all the remaining liquid is boiled away. The sensitivity coefficients for the entire discharge are, consequently, not very revealing. Sensitivity coefficients for the total mass discharged are all zero (e.g., all the initial mass is discharged, regardless of the initial conditions), except for the two coefficients related to the changes in the initial mass, which are both one. All the sensitivity coefficients for the total discharge time are nearly zero (because the boil-off time of the liquid dominates the calculations), except for those coefficients related to changes in the puncture area, discharge coefficient, and elevation of the puncture; these are all nearly equal to one.

The nominal values of the output for the liquid discharge part of the venting are: liquid mass discharged =  $5.69 \times 10^7$  grams; discharge time = 82.4 sec; and average discharge rate of the liquid =  $6.9 \times 10^5$ . The sensitivity coefficients for these outputs are given in Table D.1, for both increases (+) and decreases (-) of the indicated input. A discussion of these coefficients follows.

Tank volume. Increasing the tank volume increases the percentage of the liquid below the puncture, and similarly for decreasing the tank volume. The sensitivity coefficients for discharge mass and time therefore depend on where the puncture is with respect to the liquid surface; this explains the non-zero values of the coefficients. The average flow rate changes little.

Tank height. Changes in tank height have the same kind of effects as changes in tank volume, for the same reason, except that the direction of the change of the output is reversed.

TABLE D.1. SENSITIVITY COEFFICIENTS FOR LIQUID OUTFLOW  
OF LNG AT  $-159.2^{\circ}\text{C}$

Input Parameter		Discharged Mass	Discharge Time	Average Flow Rate
Tank Volume	+	-0.36	-0.52	0.16
	-	0.42	0.66	-0.23
Tank Height	+	0.91	0.73	0.08
	-	-0.86	-0.75	-0.11
Puncture Area	+	0	-0.85	0.89
	-	0	0.93	-0.89
Discharge	+	0	-0.85	0.89
	-	0	0.93	-0.89
Initial Mass	+	1.48	1.72	-0.22
	-	-1.37	-1.48	0.12
Initial Temperature	+	$\approx 0$	-22.59	29.19
	-	$\approx 0$	65.69	-39.64
Atmospheric Pressure	+	0	0.20	-0.19
	-	0	-0.16	0.16
Atmospheric Temperature	+	0	0	0
	-	0	0	0

Puncture area and discharge coefficient. These inputs always occur in the combination  $A_p C_D$ ; their individual effects are identical. Changes in them do not affect the total mass discharged. The average flow rate and discharge time, however, are almost related one-to-one (i.e.,  $S = \pm 1$ ) to the changes in  $Z_p$  or  $C_D$ . If the liquid did not change phase and the tank pressure remained constant, the coefficients would be exactly  $\pm 1$ .

Initial mass. Since the quantity of liquid below the puncture is constant, changes in the total initial liquid mass have a slightly greater than one-to-one effect on both the mass discharged and discharge time.

Initial temperature. Changes in the initial temperature have a negligible effect on the total mass that is discharged, but have a great effect on the discharge time. A one percent change in temperature changes the initial tank pressure by nine percent and the initial pressure differential at the puncture by over fifty percent, so the large sensitivity coefficients are understandable.

Atmospheric pressure. Changing the atmospheric pressure changes the pressure differential from the tank to the outside, so the discharge rates and times are affected. If air ingestion had occurred, the sensitivity coefficients would have been larger. On the other hand, if the vacuum relief valve had been operable, the sensitivity to atmospheric pressure would have been very small.

Atmospheric temperature. Since air was not ingested, the temperature of the atmosphere had no effect.

It can be concluded that only the sensitivity of the results to the initial temperature might create a problem in practice. Even then, the sensitivity would be large only for volatile liquids carried in pressurized tanks or in tanks whose relief valve is stuck, so that super- or sub-atmospheric tank pressures are possible.

